SECTION 1 IDENTIFICATION

Product Identifier

- **Product name**: 10005-1 (Beryllium 1000µg/mL in 2% HNO3)
- **Synonyms**: 1000µg/mL Beryllium in 2% HNO3
- **Proper shipping name**: Corrosive liquid, acidic, inorganic, n.o.s. (contains nitric acid)
- **Other means of identification**: 10005-1

Recommended use of the chemical and restrictions on use

- **Relevant identified uses**: Use according to manufacturer's directions.

Name, address, and telephone number of the chemical manufacturer, importer, or other responsible party

- **Registered company name**: High-Purity Standards
- **Address**: PO Box 41727 SC 29423 United States
- **Telephone**: 843-767-7900
- **Fax**: 843-767-7906
- **Website**: highpuritystandards.com
- **Email**: Not Available

Emergency phone number

- **Association / Organisation**: INFOTRAC
- **Emergency telephone numbers**: 1-800-535-5053
- **Other emergency telephone numbers**: 1-352-323-3500

SECTION 2 HAZARD(S) IDENTIFICATION

Classification of the substance or mixture

- **Classification**: Skin Sensitizer Category 1, Carcinogenicity Category 1B, Metal Corrosion Category 1, Skin Corrosion/Irritation Category 1A, Serious Eye Damage Category 1

Label elements

- **GHS label elements**: Image of hazard symbols

SIGNAL WORD

- **DANGER**

Hazard statement(s)

- **H317**: May cause an allergic skin reaction.
- **H350**: May cause cancer.
- **H290**: May be corrosive to metals.
- **H314**: Causes severe skin burns and eye damage.
Hazard(s) not otherwise specified
Not Applicable

Precautionary statement(s) Prevention
P201 Obtain special instructions before use.

Precautionary statement(s) Response
P301+P330+P331 If SWALLOWED: Rinse mouth. Do NOT induce vomiting.

Precautionary statement(s) Storage
P405 Store locked up.

Precautionary statement(s) Disposal
P501 Dispose of contents/container in accordance with local regulations.

SECTION 3 COMPOSITION / INFORMATION ON INGREDIENTS

Substances
See section below for composition of Mixtures

Mixtures

<table>
<thead>
<tr>
<th>CAS No</th>
<th>% [weight]</th>
<th>Name</th>
</tr>
</thead>
<tbody>
<tr>
<td>7697-37-2</td>
<td>2</td>
<td>nitric acid</td>
</tr>
<tr>
<td>543-81-7</td>
<td>0.1 (as Be)</td>
<td>beryllium acetate</td>
</tr>
<tr>
<td>7732-18-5</td>
<td>balance</td>
<td>water</td>
</tr>
</tbody>
</table>

SECTION 4 FIRST-AID MEASURES

Description of first aid measures

Eye Contact
If this product comes in contact with the eyes:
- Immediately hold eyelids apart and flush the eye continuously with running water.
- Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids.
- Continue flushing until advised to stop by the Poisons Information Centre or a doctor, or for at least 15 minutes.
- Transport to hospital or doctor without delay.
- Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.

Skin Contact
If skin or hair contact occurs:
- Immediately flush body and clothes with large amounts of water, using safety shower if available.
- Quickly remove all contaminated clothing, including footwear.
- Wash skin and hair with running water. Continue flushing with water until advised to stop by the Poisons Information Centre.
- Transport to hospital, or doctor.

Inhalation
- If fumes or combustion products are inhaled remove from contaminated area.
- Lay patient down. Keep warm and rested.
- Prostheses such as false teeth, which may block airway, should be removed, where possible, prior to initiating first aid procedures.
- Apply artificial respiration if not breathing, preferably with a demand valve resuscitator, bag-valve mask device, or pocket mask as trained. Perform CPR if necessary.
- Transport to hospital, or doctor without delay.
- Inhalation of vapours or aerosols (mists, fumes) may cause lung edema.
- Corrosive substances may cause lung damage (e.g. lung oedema, fluid in the lungs).
- As this reaction may be delayed up to 24 hours after exposure, affected individuals need complete rest (preferably in semi-recumbent posture) and must be kept under medical observation even if no symptoms are (yet) manifested.
- Before any such manifestation, the administration of a spray containing a dexamethasone derivative or beclomethasone derivative may be considered. This must definitely be left to a doctor or person authorised by him/her. (ICSG13719)

Ingestion
- For advice, contact a Poisons Information Centre or a doctor at once.
- Urgent hospital treatment is likely to be needed.
- If swallowed do NOT induce vomiting.
- If vomiting occurs, lean patient forward or place on left side (head-down position, if possible) to maintain open airway and prevent aspiration.
- Observe the patient carefully.
- Never give liquid to a person showing signs of being sleepy or with reduced awareness; i.e. becoming unconscious.
- Give water to rinse out mouth, then provide liquid slowly and as much as casualty can comfortably drink.
- Transport to hospital or doctor without delay.

Most important symptoms and effects, both acute and delayed
See Section 11

Indication of any immediate medical attention and special treatment needed
For acute or short term repeated exposures to strong acids:
- Airway problems may arise from laryngeal oedema and inhalation exposure. Treat with 100% oxygen initially.
- Respiratory distress may require intubation if endotracheal intubation is contraindicated by excessive swelling.
- Intravenous lines should be established immediately in all cases where there is evidence of circulatory compromise.
- Strong acids produce a coagulation necrosis characterised by formation of a coagulum (eschar) as a result of the dessicating action of the acid on proteins in specific tissues.

INGESTION:
Immediate dilution (milk or water) within 30 minutes post ingestion is recommended.

**DO NOT** attempt to neutralise the acid since exothermic reaction may extend the corrosive injury.

> Be careful to avoid further vomit since re-exposure of the mucosa to the acid is harmful. Limit fluids to one or two glasses in an adult.
> Charcoal has no place in acid management.

> Some authors suggest the use of lavage within 1 hour of ingestion.

**SKIN:**

> Skin lesions require copious saline irrigation. Treat chemical burns as thermal burns with non-adherent gauze and wrapping.
> Deep second-degree burns may benefit from topical silver sulfadiazine.

**EYE:**

> Eye injuries require retraction of the eyelids to ensure thorough irrigation of the conjunctival cul-de-sacs. Irrigation should last at least 20-30 minutes. **DO NOT** use neutralising agents or any other additives. Several litres of saline are required.
> Cycloplegic drops, (1% cyclopentolate for short-term use or 5% homatropine for longer term use) antibiotic drops, vasoconstrictive agents or artificial tears may be indicated dependent on the severity of the injury.
> Steroid eye drops should only be administered with the approval of a consulting ophthalmologist.

[Ellenhorn and Barceloux: Medical Toxicology]

### SECTION 5 FIRE-FIGHTING MEASURES

**Extinguishing media**

- Water spray or fog.
- Foam.
- Dry chemical powder.
- BCF (where regulations permit).
- Carbon dioxide.

**Special hazards arising from the substrate or mixture**

<table>
<thead>
<tr>
<th>Fire Incompatibility</th>
<th>None known.</th>
</tr>
</thead>
</table>

**Special protective equipment and precautions for fire-fighters**

<table>
<thead>
<tr>
<th>Fire Fighting</th>
<th>Alert Fire Brigade and tell them location and nature of hazard.</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>Wear full body protective clothing with breathing apparatus.</td>
</tr>
<tr>
<td></td>
<td>Prevent, by any means available, spillage from entering drains or water course.</td>
</tr>
<tr>
<td></td>
<td>Use fire fighting procedures suitable for surrounding area.</td>
</tr>
<tr>
<td></td>
<td>Do not approach containers suspected to be hot.</td>
</tr>
<tr>
<td></td>
<td>Cool fire exposed containers with water spray from a protected location.</td>
</tr>
<tr>
<td></td>
<td>If safe to do so, remove containers from path of fire.</td>
</tr>
<tr>
<td></td>
<td>Equipment should be thoroughly decontaminated after use.</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Fire/Explosion Hazard</th>
<th>Non combustible.</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>Not considered to be a significant fire risk.</td>
</tr>
<tr>
<td></td>
<td>Acids may react with metals to produce hydrogen, a highly flammable and explosive gas.</td>
</tr>
<tr>
<td></td>
<td>Heating may cause expansion or decomposition leading to violent rupture of containers.</td>
</tr>
<tr>
<td></td>
<td>May emit corrosive, poisonous fumes. May emit acid smoke.</td>
</tr>
</tbody>
</table>

### SECTION 6 ACCIDENTAL RELEASE MEASURES

#### Personal precautions, protective equipment and emergency procedures

See section 8

#### Environmental precautions

See section 12

#### Methods and material for containment and cleaning up

<table>
<thead>
<tr>
<th>Minor Spills</th>
<th>Drains for storage or use areas should have retention basins for pH adjustments and dilution of spills before discharge or disposal of material.</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>Check regularly for spills and leaks.</td>
</tr>
<tr>
<td></td>
<td>Clean up all spills immediately.</td>
</tr>
<tr>
<td></td>
<td>Avoid breathing vapours and contact with skin and eyes.</td>
</tr>
<tr>
<td></td>
<td>Control personal contact with the substance, by using protective equipment.</td>
</tr>
<tr>
<td></td>
<td>Contain and absorb spill with sand, earth, inert material or vermiculite.</td>
</tr>
<tr>
<td></td>
<td>Wipe up.</td>
</tr>
<tr>
<td></td>
<td>Place in a suitable, labelled container for waste disposal.</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Major Spills</th>
<th>Clear area of personnel and move upward.</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>Alert Fire Brigade and tell them location and nature of hazard.</td>
</tr>
<tr>
<td></td>
<td>Wear full body protective clothing with breathing apparatus.</td>
</tr>
<tr>
<td></td>
<td>Prevent, by any means available, spillage from entering drains or water course.</td>
</tr>
<tr>
<td></td>
<td>Consider evacuation (or protect in place).</td>
</tr>
<tr>
<td></td>
<td>Stop leak if safe to do so.</td>
</tr>
<tr>
<td></td>
<td>Contain spill with sand, earth or vermiculite.</td>
</tr>
<tr>
<td></td>
<td>Collect recoverable product into labelled containers for recycling.</td>
</tr>
<tr>
<td></td>
<td>Neutralise/decontaminate residue (see Section 13 for specific agent).</td>
</tr>
<tr>
<td></td>
<td>Collect solid residues and seal in labelled drums for disposal.</td>
</tr>
<tr>
<td></td>
<td>Wash area and prevent runoff into drains.</td>
</tr>
<tr>
<td></td>
<td>After clean up operations, decontaminate and launder all protective clothing and equipment before storing and re-using.</td>
</tr>
<tr>
<td></td>
<td>If contamination of drains or waterways occurs, advise emergency services.</td>
</tr>
</tbody>
</table>

Personal Protective Equipment advice is contained in Section 8 of the SDS.

### SECTION 7 HANDLING AND STORAGE

Continued...
### Precautions for safe handling

- Avoid all personal contact, including inhalation.
- Wear protective clothing when risk of exposure occurs.
- Use in a well-ventilated area.
- **WARNING:** To avoid violent reaction, **ALWAYS** add material to water and **NEVER** water to material.
- Avoid smoking, naked lights or ignition sources.
- Avoid contact with incompatible materials.
- When handling, **DO NOT** eat, drink or smoke.
- Keep containers securely sealed when not in use.
- Avoid physical damage to containers.
- Always wash hands with soap and water after handling.
- Work clothes should be laundered separately. Launder contaminated clothing before re-use.
- Use good occupational work practice.
- Observe manufacturer's storage and handling recommendations contained within this SDS.
- Atmosphere should be regularly checked against established exposure standards to ensure safe working conditions are maintained.

#### Safe handling

- **Store in original containers.**
- Keep containers securely sealed.
- Store in a cool, dry, well-ventilated area.
- Store away from incompatible materials and foodstuff containers.
- Protect containers against physical damage and check regularly for leaks.
- Observe manufacturer's storage and handling recommendations contained within this SDS.

### Conditions for safe storage, including any incompatibilities

- **DO NOT** use aluminium or galvanised containers
- Check regularly for spills and leaks
- Lined metal can, lined metal pull/ can.
- Plastic pail.
- Polyliner drum.
- Packing as recommended by manufacturer.
- Check all containers are clearly labelled and free from leaks.

For low viscosity materials:
- Drums and jenicans must be of the non-removable head type.
- Where a can is to be used as an inner package, the can must have a screwed enclosure.
- For materials with a viscosity of at least 2680 cSt. (23 deg. C) and solids (between 15 C deg. and 40 deg C.):
  - Removable head packaging;
  - Cans with friction closures and low pressure tubes and cartridges may be used.

Where combination packages are used, and the inner packages are of glass, porcelain or stoneware, there must be sufficient inert cushioning material in contact with inner and outer packages unless the outer packaging is a close fitting moulded plastic box and the substances are not incompatible with the plastic.

### Storage incompatibility

- Inorganic acids are generally soluble in water with the release of hydrogen ions. The resulting solutions have pHs of less than 7.0.
- Inorganic acids neutralise chemical bases (for example: amines and inorganic hydroxides) to form salts - neutralisation can generate dangerously large amounts of heat in small spaces.
- The dissolution of inorganic acids in water or the dilution of their concentrated solutions with additional water may generate significant heat.
- The addition of water to inorganic acids often generates sufficient heat in the small region of mixing to cause some of the water to boil explosively. The resulting “bumping” can spatter the acid.
- Inorganic acids react with active metals, including such structural metals as aluminum and iron, to release hydrogen, a flammable gas.
- Inorganic acids can initiate the polymerisation of certain classes of organic compounds.
- Inorganic acids react with cyanide compounds to release gaseous hydrogen cyanide.
- Inorganic acids generate flammable and/or toxic gases in contact with dithiocarbamates, isocyanates, mercaptans, nitrides, nitriles, sulfides, and strong reducing agents. Additional gas-generating reactions occur with sulfites, nitrites, thiosulfates (to give H2S and SO3), dithionites (SO2), and even carbonates.
- Acids often catalyse (increase the rate of) chemical reactions.

### SECTION 8 EXPOSURE CONTROLS / PERSONAL PROTECTION

#### Control parameters

**OCCIDENTAL EXPOSURE LIMITS (OEL)**

<table>
<thead>
<tr>
<th>Source</th>
<th>Ingredient</th>
<th>Material name</th>
<th>TWA</th>
<th>STEL</th>
<th>Peak</th>
<th>Notes</th>
</tr>
</thead>
<tbody>
<tr>
<td>US OSHA Permissible Exposure Levels (PELs) - Table Z1</td>
<td>nitric acid</td>
<td>Nitric acid</td>
<td>5 mg/m³ / 2 ppm</td>
<td>Not Available</td>
<td>Not Available</td>
<td>Not Available</td>
</tr>
<tr>
<td>US ACGIH Threshold Limit Values (TLV)</td>
<td>nitric acid</td>
<td>Nitric acid</td>
<td>2 ppm</td>
<td>4 ppm</td>
<td>Not Available</td>
<td>TLV® Basis: URT &amp; eye irrit; dental erosion</td>
</tr>
<tr>
<td>US NIOSH Recommended Exposure Limits (RELs)</td>
<td>nitric acid</td>
<td>Aqua fortis, Engravers acid, Hydrogen nitrate, Red fuming nitric acid (RFNA), White fuming nitric acid (WFNA)</td>
<td>5 mg/m³ / 2 ppm</td>
<td>10 mg/m³ / 4 ppm</td>
<td>Not Available</td>
<td>Not Available</td>
</tr>
<tr>
<td>US OSHA Permissible Exposure Levels (PELs) - Table Z1</td>
<td>beryllium acetate</td>
<td>Silicates - Mica / Silicates - Soapstone / Silicates- Soapstone / Silicates - Talc / Silicates - Tremolite,</td>
<td>0.1 mg/m³</td>
<td>Not Available</td>
<td>Not Available</td>
<td>See Table Z-3;less than 1% crystalline silica(respirable dust) / See Table Z-3;less than 1% crystalline silica, total dust / See Table Z-3;less than 1% crystalline silica,respirable dust / less than 1% crystalline silica;see 29 CFR 1910.1001;See Table Z-3(containing asbestos); use asbestos limit; (STEL (Excursion limit)as averaged over a sampling</td>
</tr>
</tbody>
</table>

**Notes:**

- Continuation...
### EMERGENCY LIMITS

<table>
<thead>
<tr>
<th>Ingredient</th>
<th>Material name</th>
<th>TEEL-1</th>
<th>TEEL-2</th>
<th>TEEL-3</th>
</tr>
</thead>
<tbody>
<tr>
<td>Nitric acid</td>
<td>Nitric acid</td>
<td>Not Available</td>
<td>Not Available</td>
<td>Not Available</td>
</tr>
<tr>
<td>Nitric acid</td>
<td>100 ppm</td>
<td>25 ppm</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Beryllium acetate</td>
<td>10 mg/m³</td>
<td>4 mg/m³</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Water</td>
<td>Not Available</td>
<td></td>
<td></td>
<td></td>
</tr>
</tbody>
</table>

### Exposure controls

Engineering controls are used to remove a hazard or place a barrier between the worker and the hazard. Well-designed engineering controls can be highly effective in protecting workers and will typically be independent of worker interactions to provide this high level of protection.

The basic types of engineering controls are:

- Enclosure and/or isolation of emission source which keeps a selected hazard "physically" away from the worker and ventilation that strategically "adds" and "removes" air in the work environment. Ventilation can remove or dilute an air contaminant if designed properly. The design of a ventilation system must match the particular process and chemical or contaminant in use.
- Process controls which involve changing the way a job activity or process is done to reduce the risk.
- Employers may need to use multiple types of controls to prevent employee overexposure.

Local exhaust ventilation usually required. If risk of overexposure exists, wear approved respirator. Correct fit is essential to obtain adequate protection. Supplied-air type respirator may be required in special circumstances. Correct fit is essential to ensure adequate protection.

An approved self contained breathing apparatus (SCBA) may be required in some situations. Provide adequate ventilation in warehouse or closed storage area. Air contaminants generated in the workplace possess varying "escape" velocities which, in turn, determine the "capture velocities" of fresh circulating air required to effectively remove the contaminant.

### Appropriate engineering controls

<table>
<thead>
<tr>
<th>Type of Contaminant</th>
<th>Air Speed:</th>
</tr>
</thead>
<tbody>
<tr>
<td>solvent, vapours, degreasing etc... evaporating from tank (in still air).</td>
<td>0.25-0.5 m/s (50-100 f/min.)</td>
</tr>
<tr>
<td>aerosols, fumes from pouring operations, intermittent container filling, low speed conveyor transfers, welding, spray drift, plating acid fumes, pickling (released at low velocity into zone of active generation)</td>
<td>0.5-1 m/s (100-200 f/min.)</td>
</tr>
<tr>
<td>direct spray, spray painting in shallow booths, drum filling, conveyor loading, crusher dusts, gas discharge (active generation into zone of rapid air motion)</td>
<td>1-2.5 m/s (200-500 f/min.)</td>
</tr>
<tr>
<td>grinding, abrasive blasting, tumbling, high speed wheel generated dusts (released at high initial velocity into zone of very high rapid air motion).</td>
<td>2.5-10 m/s (500-2000 f/min.)</td>
</tr>
</tbody>
</table>

Within each range the appropriate value depends on:

<table>
<thead>
<tr>
<th>Lower end of the range</th>
<th>Upper end of the range</th>
</tr>
</thead>
<tbody>
<tr>
<td>1: Room air currents minimal or favourable to capture</td>
<td>1: Disturbing room air currents</td>
</tr>
<tr>
<td>2: Contaminants of low toxicity or of nuisance value only.</td>
<td>2: Contaminants of high toxicity</td>
</tr>
<tr>
<td>3: Intermittent, low production.</td>
<td>3: High production, heavy use</td>
</tr>
<tr>
<td>4: Large hood or large air mass in motion.</td>
<td>4: Small hood-local control only</td>
</tr>
</tbody>
</table>

Simple theory shows that air velocity falls rapidly with distance away from the opening of a simple extraction pipe. Velocity generally decreases with the square of distance from the extraction point (in simple cases). Therefore the air speed at the extraction point should be adjusted, accordingly, after reference to distance from the contaminating source. The air velocity at the extraction fan, for example, should be a minimum of 1-2 m/s (200-400 f/min) for extraction of solvents generated in a tank 2 meters distant from the extraction point. Other mechanical considerations, producing performance deficits within the extraction apparatus, make it essential that theoretical air velocities are multiplied by factors of 10 or more when extraction systems are installed or used.
Eye and face protection

Safety glasses with unperforated side shields may be used where continuous eye protection is desirable, as in laboratories; spectacles are not sufficient where complete eye protection is needed such as when handling bulk-quantities, where there is a danger of splashing, or if the material may be under pressure. Chemical goggles whenever there is a danger of the material coming in contact with the eyes; goggles must be properly fitted. Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants. A written policy document, describing the wearing of lenses or restrictions on use, should be created for each workplace or task. This should include a review of lens absorption and adsorption for the class of chemicals in use and an account of injury experience. Medical and first-aid personnel should be trained in their removal and suitable equipment should be readily available. In the event of chemical exposure, begin eye irrigation immediately and remove contact lens as soon as practicable. Lens should be removed at the first signs of eye redness or irritation - lens should be removed in a clean environment only after workers have washed hands thoroughly. [CDC NIOSH Current Intelligence Bulletin 59], [AS/NZS 1336 or national equivalent]

Skin protection

See Hand protection below

Hands/feet protection

Elbow length PVC gloves

When handling corrosive liquids, wear trousers or overalls outside of boots, to avoid spills entering boots.

Body protection

See Other protection below

Other protection

Overalls.
PVC Apron.
PVC protective suit may be required if exposure severe.
Eyewash unit.
Ensure there is ready access to a safety shower.

Thermal hazards

Not Available

Respiratory protection

Type A Filter of sufficient capacity. (AS/NZS 1716 & 1715, EN 143:2000 & 149:2001, ANSI Z88 or national equivalent)

SECTION 9 PHYSICAL AND CHEMICAL PROPERTIES

Information on basic physical and chemical properties

<table>
<thead>
<tr>
<th>Appearance</th>
<th>colorless</th>
</tr>
</thead>
<tbody>
<tr>
<td>Physical state</td>
<td>Liquid</td>
</tr>
<tr>
<td>Odour</td>
<td>Not Available</td>
</tr>
<tr>
<td>Odour threshold</td>
<td>Not Available</td>
</tr>
<tr>
<td>pH (as supplied)</td>
<td>&lt;2</td>
</tr>
<tr>
<td>Melting point / freezing point (°C)</td>
<td>Not Available</td>
</tr>
<tr>
<td>Initial boiling point and boiling range (°C)</td>
<td>Not Available</td>
</tr>
<tr>
<td>Flash point (°C)</td>
<td>Not Available</td>
</tr>
<tr>
<td>Evaporation rate</td>
<td>Not Available</td>
</tr>
<tr>
<td>Flammability</td>
<td>Not Available</td>
</tr>
<tr>
<td>Upper Explosive Limit (%)</td>
<td>Not Available</td>
</tr>
<tr>
<td>Lower Explosive Limit (%)</td>
<td>Not Available</td>
</tr>
<tr>
<td>Vapour pressure (kPa)</td>
<td>Not Available</td>
</tr>
<tr>
<td>Solubility in water (g/L)</td>
<td>Miscible</td>
</tr>
<tr>
<td>Vapour density (Air = 1)</td>
<td>Not Available</td>
</tr>
<tr>
<td>Relative density (Water = 1)</td>
<td>Not Available</td>
</tr>
<tr>
<td>Partition coefficient n-octanol / water</td>
<td>Not Available</td>
</tr>
<tr>
<td>Auto-ignition temperature (°C)</td>
<td>Not Available</td>
</tr>
<tr>
<td>Decomposition temperature</td>
<td>Not Available</td>
</tr>
<tr>
<td>Viscosity (cSt)</td>
<td>Not Available</td>
</tr>
<tr>
<td>Molecular weight (g/mol)</td>
<td>Not Available</td>
</tr>
<tr>
<td>Taste</td>
<td>Not Available</td>
</tr>
<tr>
<td>Explosive properties</td>
<td>Not Available</td>
</tr>
<tr>
<td>Oxidising properties</td>
<td>Not Available</td>
</tr>
<tr>
<td>Surface Tension (dyn/cm or mN/m)</td>
<td>Not Available</td>
</tr>
<tr>
<td>Volatile Component (%vol)</td>
<td>Not Available</td>
</tr>
<tr>
<td>Gas group</td>
<td>Not Available</td>
</tr>
<tr>
<td>pH as a solution (1%)</td>
<td>Not Available</td>
</tr>
<tr>
<td>VOC g/L</td>
<td>Not Available</td>
</tr>
</tbody>
</table>

SECTION 10 STABILITY AND REACTIVITY

Reactivity

See section 7

Chemical stability

Contact with alkaline material liberates heat.
Unstable in the presence of incompatible materials.
Product is considered stable.
Hazardous polymerisation will not occur.

Possibility of hazardous reactions

See section 7

Conditions to avoid

See section 7

Incompatible materials

See section 7

Hazardous decomposition products

See section 5

SECTION 11 TOXICOLOGICAL INFORMATION

Information on toxicological effects

Inhaled

Inhalation of vapours or aerosols (mists, fumes), generated by the material during the course of normal handling, may be harmful.
### Acute Toxicity

**Toxicity**

<table>
<thead>
<tr>
<th>Substance</th>
<th>Ingestion</th>
<th>Skin Contact</th>
<th>Eye</th>
</tr>
</thead>
<tbody>
<tr>
<td>Beryllium acetate</td>
<td>Not Available</td>
<td>Not Available</td>
<td>Not Available</td>
</tr>
<tr>
<td>Nitric acid</td>
<td>Inhalation (rat) LC50: 0.13 mg/L/4hr[^2]</td>
<td>Not Available</td>
<td>Not Available</td>
</tr>
<tr>
<td>Water</td>
<td>Oral (rat) LD50: &gt;90000 mg/kg[^2]</td>
<td>Not Available</td>
<td>Not Available</td>
</tr>
</tbody>
</table>

**Legend:**

1. Value obtained from Europe ECHA Registered Substances - Acute toxicity 2. Value obtained from manufacturer's SDS. Unless otherwise specified data extracted from RTECS - Register of Toxic Effect of chemical Substances

### Chronic Toxicity

- The material can cause respiratory irritation in some persons. The body's response to such irritation can cause further lung damage.
- Skin contact with acidic corrosives may result in pain and burns; these may be deep with distinct edges and may heal slowly with the formation of scar tissue.
- Repeated or prolonged exposure to acids may result in the erosion of teeth, swelling and/or ulceration of mouth lining.
- Skin contact with acidic corrosives may produce pain, tears, sensitivity to light and burns. Mild burns of the epithelium generally recover rapidly and completely.

### Asthma-like symptoms

- Asthma-like symptoms may continue for months or even years after exposure to the material ceases.
- Studies show that inhaling this substance for over a long period (e.g. in an occupational setting) may increase the risk of cancer.
- Substance accumulation, in the human body, may occur and may cause some concern following repeated or long-term occupational exposure.

### Chemical properties

- *Nitric acid*:
  - Inhalation (rat) LC50: 0.13 mg/L/4hr[^2]
  - Not Available

- *Beryllium acetate*:
  - Not Available
  - Not Available

- *Water*:
  - Oral (rat) LD50: >90000 mg/kg[^2]
  - Not Available

### NITRIC ACID

- For acid mists, aerosols, vapours
  - Data from assays for genotoxic activity in vitro suggest that eukaryotic cells are susceptible to genetic damage when the pH falls to about 6.5.
  - The material may produce severe irritation to the eye causing pronounced inflammation.
  - The material may produce respiratory tract irritation, and result in damage to the lung including reduced lung function.
  - The material may cause severe skin irritation after prolonged or repeated exposure and may produce on contact skin redness, swelling, the production of vesicles, scaling and thickening of the skin.
  - Oral (rat) LD50: 50-500 mg/kg * [Various Manufacturers]

### BERYLLIUM ACETATE

- The following information refers to contact allergens as a group and may not be specific to this product.
  - **WARNING:** This substance has been classified by the IARC as Group 1: CARCINOGENIC TO HUMANS.

### WATER

- No significant acute toxicological data identified in literature search.

### NITRIC ACID & BERYLLIUM ACETATE

- Asthma-like symptoms may continue for months or even years after exposure to the material ceases.

### Toxicity

- **Acute Toxicity**
  - ![Carcinogenicity](image)
  - ![Reproductivity](image)
  - ![STOT - Single Exposure](image)
  - ![STOT - Repeated Exposure](image)
  - ![Aspiration Hazard](image)

**Legend:**

- ![Data available but does not fill the criteria for classification](image)
- ![Data available to make classification](image)
- ![Data Not Available to make classification](image)
### Ecotoxicity:
The tolerance of water organisms towards pH margin and variation is diverse. Recommended pH values for test species listed in OECD guidelines are between 6.0 and almost 9. Acute testing with fish showed 96h-LC50 at about pH 3.5.

Prevent, by any means available, spillage from entering drains or water courses.

**DO NOT** discharge into sewer or waterways.

### Persistence and degradability

<table>
<thead>
<tr>
<th>Ingredient</th>
<th>Endpoint</th>
<th>Test Duration (hr)</th>
<th>Species</th>
<th>Value</th>
<th>Source</th>
</tr>
</thead>
<tbody>
<tr>
<td>nitric acid</td>
<td>NOEC</td>
<td>16</td>
<td>Crustacea</td>
<td>107mg/L</td>
<td>4</td>
</tr>
</tbody>
</table>

**Legend:**
- Extracted from 1. IUCLID Toxicity Data
- 2. Europe ECHA Registered Substances - Ecotoxicological Information - Aquatic Toxicity 3.
- EPIWIN Suite V3.12 (QSAR) - Aquatic Toxicity Data (Estimated)
- 4. US EPA, Ecotox database - Aquatic Toxicity Data 5.
- ECETOC Aquatic Hazard Assessment Data 6.
- NITE (Japan) - Bioconcentration Data 7.
- METI (Japan) - Bioconcentration Data 8. Vendor Data

### Bioaccumulative potential

<table>
<thead>
<tr>
<th>Ingredient</th>
<th>Bioaccumulation</th>
</tr>
</thead>
<tbody>
<tr>
<td>water</td>
<td>LOW (LogKOW = -1.38)</td>
</tr>
</tbody>
</table>

### Mobility in soil

<table>
<thead>
<tr>
<th>Ingredient</th>
<th>Mobility</th>
</tr>
</thead>
<tbody>
<tr>
<td>water</td>
<td>LOW (KOC = 14.3)</td>
</tr>
</tbody>
</table>

### SECTION 13 DISPOSAL CONSIDERATIONS

**Waste treatment methods**

- Recycle wherever possible.
- Consult manufacturer for recycling options or consult local or regional waste management authority for disposal if no suitable treatment or disposal facility can be identified.
- Treat and neutralise at an approved treatment plant. Treatment should involve: Neutralisation with soda ash or soda-lime followed by: burial in a land-fill specifically licensed to accept chemical and / or pharmaceutical wastes or Incineration in a licensed apparatus (after admixture with suitable combustible material).
- Decontaminate empty containers with 5% aqueous sodium hydroxide or soda ash, followed by water. Observe all label safeguards until containers are cleaned and destroyed.

### SECTION 14 TRANSPORT INFORMATION

**Labels Required**

- Marine Pollutant: NO

#### Land transport (DOT)

<table>
<thead>
<tr>
<th>UN number</th>
<th>3264</th>
</tr>
</thead>
<tbody>
<tr>
<td>UN proper shipping name</td>
<td>Corrosive liquid, acidic, inorganic, n.o.s. (contains nitric acid)</td>
</tr>
<tr>
<td>Transport hazard class(es)</td>
<td>Class 8</td>
</tr>
<tr>
<td>Subrisk</td>
<td>Not Applicable</td>
</tr>
<tr>
<td>Packing group</td>
<td>II</td>
</tr>
<tr>
<td>Environmental hazard</td>
<td>Not Applicable</td>
</tr>
</tbody>
</table>

**Special precautions for user**

- Hazard Label: 8
- Special provisions: 386, B2, IB2, T11, TP2, TP27

#### Air transport (ICAO-IATA / DGR)

<table>
<thead>
<tr>
<th>UN number</th>
<th>3264</th>
</tr>
</thead>
<tbody>
<tr>
<td>UN proper shipping name</td>
<td>Corrosive liquid, acidic, inorganic, n.o.s. * (contains nitric acid)</td>
</tr>
<tr>
<td>Transport hazard class(es)</td>
<td>ICAO/IATA Class 8</td>
</tr>
<tr>
<td>ICAO / IATA Subrisk</td>
<td>Not Applicable</td>
</tr>
<tr>
<td>ERG Code</td>
<td>8L</td>
</tr>
</tbody>
</table>

*Continued...*
Sea transport (IMDG-Code / GGVSee)

<table>
<thead>
<tr>
<th>Source Product name</th>
<th>Pollution Category</th>
<th>Ship Type</th>
</tr>
</thead>
<tbody>
<tr>
<td>Nitric acid (70% and over)/Nitric acid (less than 70%)</td>
<td>Y; Y</td>
<td>2/2</td>
</tr>
</tbody>
</table>

Transport in bulk according to Annex II of MARPOL and the IBC code

<table>
<thead>
<tr>
<th>Source</th>
<th>Product name</th>
<th>Pollution Category</th>
<th>Ship Type</th>
</tr>
</thead>
<tbody>
<tr>
<td>Nitric acid (70% and over)/Nitric acid (less than 70%)</td>
<td>Y; Y</td>
<td>2/2</td>
<td></td>
</tr>
</tbody>
</table>

SECTION 15 REGULATORY INFORMATION

Safety, health and environmental regulations / legislation specific for the substance or mixture

NITRIC ACID(7697-37-2) IS FOUND ON THE FOLLOWING REGULATORY LISTS

- International Air Transport Association (IATA) Dangerous Goods Regulations - Prohibited List
- Passenger and Cargo Aircraft
- US - Alaska Limits for Air Contaminants
- US - California OEHHA/ARB - Acute Reference Exposure Levels and Target Organs (RELS)
- US - California Permissible Exposure Limits for Chemical Contaminants
- US - Hawaii Air Contaminant Limits
- US - Idaho - Limits for Air Contaminants
- US - Massachusetts - Right To Know Listed Chemicals
- US - Michigan Exposure Limits for Air Contaminants
- US - Minnesota Permissible Exposure Limits (PELs)
- US - Oregon Permissible Exposure Limits (Z-1)
- US - Pennsylvania - Hazardous Substance List
- US - Rhode Island Hazardous Substance List
- US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants

BERYLLIUM ACETATE(543-81-7) IS FOUND ON THE FOLLOWING REGULATORY LISTS

- International Agency for Research on Cancer (IARC) - Agents Classified by the IARC Monographs
- US - Alaska Limits for Air Contaminants
- US - California OEHHA/ARB - Chronic Reference Exposure Levels and Target Organs (CRELS)
- US - California Permissible Exposure Limits for Chemical Contaminants
- US - Hawaii Air Contaminant Limits
- US - Idaho - Acceptable Maximum Peak Concentrations
- US - Idaho - Limits for Air Contaminants
- US - Michigan Exposure Limits for Air Contaminants
- US - Minnesota Permissible Exposure Limits (PELs)
- US - Oregon Permissible Exposure Limits (Z-1)
- US - Oregon Permissible Exposure Limits (Z-2)
- US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants
- US - Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants

WATER(7732-18-5) IS FOUND ON THE FOLLOWING REGULATORY LISTS

- US - Pennsylvania - Hazardous Substance List

Federal Regulations

Superfund Amendments and Reauthorization Act of 1986 (SARA)

SECTION 311/312 HAZARD CATEGORIES

<table>
<thead>
<tr>
<th>Hazard Category</th>
<th>Yes</th>
<th>No</th>
</tr>
</thead>
<tbody>
<tr>
<td>Immediate (acute) health hazard</td>
<td>Yes</td>
<td></td>
</tr>
<tr>
<td>Delayed (chronic) health hazard</td>
<td>Yes</td>
<td></td>
</tr>
<tr>
<td>Fire hazard</td>
<td>No</td>
<td></td>
</tr>
<tr>
<td>Pressure hazard</td>
<td>No</td>
<td></td>
</tr>
</tbody>
</table>
Reactivity hazard: No

### US. EPA CERCLA HAZARDOUS SUBSTANCES AND REPORTABLE QUANTITIES (40 CFR 302.4)

<table>
<thead>
<tr>
<th>Name</th>
<th>Reportable Quantity in Pounds (lb)</th>
<th>Reportable Quantity in kg</th>
</tr>
</thead>
<tbody>
<tr>
<td>Nitric acid</td>
<td>1000</td>
<td>454</td>
</tr>
</tbody>
</table>

### State Regulations

#### US. CALIFORNIA PROPOSITION 65

None Reported

<table>
<thead>
<tr>
<th>National Inventory</th>
<th>Status</th>
</tr>
</thead>
<tbody>
<tr>
<td>Australia - AICS</td>
<td>N (beryllium acetate)</td>
</tr>
<tr>
<td>Canada - DSL</td>
<td>N (beryllium acetate)</td>
</tr>
<tr>
<td>Canada - NDSL</td>
<td>N (water; beryllium acetate; nitric acid)</td>
</tr>
<tr>
<td>China - IECSC</td>
<td>N (beryllium acetate)</td>
</tr>
<tr>
<td>Europe - EINEC / ELINCS / NLP</td>
<td>Y</td>
</tr>
<tr>
<td>Japan - ENCS</td>
<td>N (water; beryllium acetate)</td>
</tr>
<tr>
<td>Korea - KECI</td>
<td>N (beryllium acetate)</td>
</tr>
<tr>
<td>New Zealand - NZIoC</td>
<td>N (beryllium acetate)</td>
</tr>
<tr>
<td>Philippines - PICCS</td>
<td>N (beryllium acetate)</td>
</tr>
<tr>
<td>USA - TSCA</td>
<td>N (beryllium acetate)</td>
</tr>
</tbody>
</table>

**Legend:**  
Y = All ingredients are on the inventory  
N = Not determined or one or more ingredients are not on the inventory and are not exempt from listing (see specific ingredients in brackets)

### SECTION 16 OTHER INFORMATION

#### Other information

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references.

The SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

#### Definitions and abbreviations

- **PC** – TWA: Permissible Concentration-Time Weighted Average  
- **PC** – STEL: Permissible Concentration-Short Term Exposure Limit  
- IARC: International Agency for Research on Cancer  
- ACGIH: American Conference of Governmental Industrial Hygienists  
- STEL: Short Term Exposure Limit  
- TEEL: Temporary Emergency Exposure Limit  
- IDLH: Immediately Dangerous to Life or Health Concentrations  
- OSF: Odour Safety Factor  
- NOAEL: No Observed Adverse Effect Level  
- LOAEL: Lowest Observed Adverse Effect Level  
- TLV: Threshold Limit Value  
- LOD: Limit Of Detection  
- OTV: Odour Threshold Value  
- BCF: BioConcentration Factors  
- BEI: Biological Exposure Index

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