

# **ICP Analytical Mixture 15**

### **High-Purity Standards**

Catalogue number: ICP-AM-15-1M

Version No: 1.1 Safety Data Sheet according to OSHA HazCom Standard (2012) requirements

#### Chemwatch Hazard Alert Code: 3 Issue Date: 03/10/2017 Print Date: 03/10/2017

S.GHS.USA.EN

### **SECTION 1 IDENTIFICATION**

### **Product Identifier**

Product name	ICP Analytical Mixture 15
Synonyms	ICP-AM-15-1M
Proper shipping name	Corrosive liquid, acidic, inorganic, n.o.s. (contains nitric acid)
Other means of identification	ICP-AM-15-1M

#### Recommended use of the chemical and restrictions on use

INTEGRITY CHECK: Product contains BOTH an acid and a base as ingredients. Relevant identified uses

#### Name, address, and telephone number of the chemical manufacturer, importer, or other responsible party

Registered company name	High-Purity Standards
Address	PO Box 41727 SC 29423 United States
Telephone	843-767-7900
Fax	843-767-7906
Website	highpuritystandards.com
Email	Not Available

#### Emergency phone number

Association / Organisation	INFOTRAC
Emergency telephone numbers	1-800-535-5053
Other emergency telephone numbers	1-352-323-3500

#### SECTION 2 HAZARD(S) IDENTIFICATION

Classification	sification Metal Corrosion Category 1, Skin Corrosion/Irritation Category 1A, Serious Eye Damage Category 1	
abel elements		
GHS label elements		
SIGNAL WORD	DANGER	
lazard statement(s)		
H290	May be corrosive to metals.	
H314	Causes severe skin burns and eye damage.	

#### Hazard(s) not otherwise specified

Not Applicable

Chemwatch: 9-357917

Version No: 1.1

Catalogue number: ICP-AM-15-1M

#### Page 2 of 11

ICP Analytical Mixture 15

P260	Do not breathe dust/fume/gas/mist/vapours/spray.
Precautionary statement(s	Response
P301+P330+P331	IF SWALLOWED: Rinse mouth. Do NOT induce vomiting.
Precautionary statement(s) P405	-
Precautionary statement(s) P405 Precautionary statement(s)	Store locked up.

#### Substances

See section below for composition of Mixtures

#### Mixtures

CAS No	%[weight]	Name
7440-70-2	0.1	calcium
7439-95-4	0.1	magnesium
7440-09-7	0.1	potassium
7440-23-5	0.1	sodium
7697-37-2	2	nitric acid
7732-18-5	balance	water

## SECTION 4 FIRST-AID MEASURES

#### Description of first aid measures

Eye Contact	If this product comes in contact with the eyes: <ul> <li>Immediately hold eyelids apart and flush the eye continuously with running water.</li> <li>Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids.</li> <li>Continue flushing until advised to stop by the Poisons Information Centre or a doctor, or for at least 15 minutes.</li> <li>Transport to hospital or doctor without delay.</li> <li>Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.</li> </ul>
Skin Contact	<ul> <li>If skin or hair contact occurs:</li> <li>Immediately flush body and clothes with large amounts of water, using safety shower if available.</li> <li>Quickly remove all contaminated clothing, including footwear.</li> <li>Wash skin and hair with running water. Continue flushing with water until advised to stop by the Poisons Information Centre.</li> <li>Transport to hospital, or doctor.</li> </ul>
Inhalation	<ul> <li>If fumes or combustion products are inhaled remove from contaminated area.</li> <li>Lay patient down. Keep warm and rested.</li> <li>Prostheses such as false teeth, which may block airway, should be removed, where possible, prior to initiating first aid procedures.</li> <li>Apply artificial respiration if not breathing, preferably with a demand valve resuscitator, bag-valve mask device, or pocket mask as trained. Perform CPR if necessary.</li> <li>Transport to hospital, or doctor, without delay.</li> <li>Inhalation of vapours or aerosols (mists, fumes) may cause lung oedema.</li> <li>Corrosive substances may cause lung damage (e.g. lung oedema, fluid in the lungs).</li> <li>As this reaction may be delayed up to 24 hours after exposure, affected individuals need complete rest (preferably in semi-recumbent posture) and must be kept under medical observation, the administration of a spray containing a dexamethasone derivative or beclomethasone derivative may be considered.</li> <li>This must definitely be left to a doctor or person authorised by him/her. (ICSC13719)</li> </ul>
Ingestion	<ul> <li>For advice, contact a Poisons Information Centre or a doctor at once.</li> <li>Urgent hospital treatment is likely to be needed.</li> <li>If swallowed do NOT induce vomiting.</li> <li>If vomiting occurs, lean patient forward or place on left side (head-down position, if possible) to maintain open airway and prevent aspiration.</li> <li>Observe the patient carefully.</li> <li>Never give liquid to a person showing signs of being sleepy or with reduced awareness; i.e. becoming unconscious.</li> <li>Give water to rinse out mouth, then provide liquid slowly and as much as casuality can comfortably drink.</li> <li>Transport to hospital or doctor without delay.</li> </ul>

#### Most important symptoms and effects, both acute and delayed

See Section 11

#### Indication of any immediate medical attention and special treatment needed

For acute or short term repeated exposures to strong acids:

- + Airway problems may arise from laryngeal edema and inhalation exposure. Treat with 100% oxygen initially.
- ▶ Respiratory distress may require cricothyroidotomy if endotracheal intubation is contraindicated by excessive swelling
- ► Intravenous lines should be established immediately in all cases where there is evidence of circulatory compromise.
- Strong acids produce a coagulation necrosis characterised by formation of a coagulum (eschar) as a result of the dessicating action of the acid on proteins in specific tissues. INGESTION:
- Immediate dilution (milk or water) within 30 minutes post ingestion is recommended.

Chemwatch: 9-357917	Page <b>3</b> of <b>11</b>	Issue Date: 03/10/2017
Catalogue number: ICP-AM-15-1M	ICP Analytical Mixture 15	Print Date: 03/10/2017
Version No: 1.1	for Analytical mixture to	

• DO NOT attempt to neutralise the acid since exothermic reaction may extend the corrosive injury.

> Be careful to avoid further vomit since re-exposure of the mucosa to the acid is harmful. Limit fluids to one or two glasses in an adult.

- Charcoal has no place in acid management.
- Some authors suggest the use of lavage within 1 hour of ingestion.

### SKIN:

> Skin lesions require copious saline irrigation. Treat chemical burns as thermal burns with non-adherent gauze and wrapping.

Deep second-degree burns may benefit from topical silver sulfadiazine.

EYE:

Eye injuries require retraction of the eyelids to ensure thorough irrigation of the conjuctival cul-de-sacs. Irrigation should last at least 20-30 minutes. DO NOT use neutralising agents or any other additives. Several litres of saline are required.

- Cycloplegic drops, (1% cyclopentolate for short-term use or 5% homatropine for longer term use) antibiotic drops, vasoconstrictive agents or artificial tears may be indicated dependent on the severity of the injury.
- Steroid eye drops should only be administered with the approval of a consulting ophthalmologist).

[Ellenhorn and Barceloux: Medical Toxicology]

### **SECTION 5 FIRE-FIGHTING MEASURES**

#### Extinguishing media

- There is no restriction on the type of extinguisher which may be used.
- Use extinguishing media suitable for surrounding area.

## Special hazards arising from the substrate or mixture

Fire Incompatibility None known.

#### Special protective equipment and precautions for fire-fighters

Fire Fighting	
Fire/Explosion Hazard	<ul> <li>Non combustible.</li> <li>Not considered to be a significant fire risk.</li> <li>Acids may react with metals to produce hydrogen, a highly flammable and explosive gas.</li> <li>Heating may cause expansion or decomposition leading to violent rupture of containers.</li> <li>May emit corrosive, poisonous fumes. May emit acrid smoke.</li> </ul>

#### SECTION 6 ACCIDENTAL RELEASE MEASURES

## Personal precautions, protective equipment and emergency procedures

See section 8

#### **Environmental precautions**

See section 12

#### Methods and material for containment and cleaning up

Minor Spills	<ul> <li>Drains for storage or use areas should have retention basins for pH adjustments and dilution of spills before discharge or disposal of material.</li> <li>Check regularly for spills and leaks.</li> <li>Clean up all spills immediately.</li> <li>Avoid breathing vapours and contact with skin and eyes.</li> <li>Control personal contact with the substance, by using protective equipment.</li> <li>Contain and absorb spill with sand, earth, inert material or vermiculite.</li> <li>Wipe up.</li> <li>Place in a suitable, labelled container for waste disposal.</li> </ul>
Major Spills	#

Personal Protective Equipment advice is contained in Section 8 of the SDS.

## SECTION 7 HANDLING AND STORAGE

#### Precautions for safe handling

Safe handling	<ul> <li>Avoid all personal contact, including inhalation.</li> <li>Wear protective clothing when risk of exposure occurs.</li> <li>Use in a well-ventilated area.</li> <li>WARNING: To avoid violent reaction, ALWAYS add material to water and NEVER water to material.</li> <li>Avoid smoking, naked lights or ignition sources.</li> <li>Avoid contact with incompatible materials.</li> <li>When handling, DO NOT eat, drink or smoke.</li> <li>Keep containers securely sealed when not in use.</li> <li>Avoid physical damage to contairers.</li> <li>Always wash hands with soap and water after handling.</li> <li>Work clothes should be laundered separately. Launder contaminated clothing before re-use.</li> <li>Use good occupational work practice.</li> <li>Observe manufacturer's storage and handling recommendations contained within this SDS.</li> <li>Atmosphere should be regularly checked against established exposure standards to ensure safe working conditions are maintained.</li> </ul>
Other information	<ul> <li>Store in original containers.</li> <li>Keep containers securely sealed.</li> <li>Store in a cool, dry, well-ventilated area.</li> <li>Store away from incompatible materials and foodstuff containers.</li> <li>Protect containers against physical damage and check regularly for leaks.</li> </ul>

Version No: 1.1

## Page 4 of 11

## **ICP** Analytical Mixture 15

Observe manufacturer's storage and handling recommendations contained within this SDS.

## Conditions for safe storage, including any incompatibilities

Conditions for safe storage, including any incompatibilities				
Suitable container	<ul> <li>DO NOT use aluminium or galvanised containers</li> <li>Check regularly for spills and leaks</li> <li>Lined metal can, lined metal pail/ can.</li> <li>Plastic pail.</li> <li>Polyliner drum.</li> <li>Packing as recommended by manufacturer.</li> <li>Check all containers are clearly labelled and free from leaks.</li> <li>For low viscosity materials</li> <li>Drums and jerricans must be of the non-removable head type.</li> <li>Where a can is to be used as an inner package, the can must have a screwed enclosure.</li> <li>For materials with a viscosity of at least 2680 cSt. (23 deg. C) and solids (between 15 C deg. and 40 deg C.):</li> <li>Removable head packaging;</li> <li>Cans with friction closures and</li> <li>low pressure tubes and cartridges</li> <li>may be used.</li> <li>-</li> </ul>			
Storage incompatibility	<ul> <li>Inorganic acids are generally soluble in water with the release of hydrogen ions. The resulting solutions have pH's of less than 7.0.</li> <li>Inorganic acids neutralise chemical bases (for example: amines and inorganic hydroxides) to form salts - neutralisation can generate dangerously large amounts of heat in small spaces.</li> <li>The dissolution of inorganic acids in water or the dilution of their concentrated solutions with additional water may generate significant heat.</li> <li>The addition of water to inorganic acids often generates sufficient heat in the small region of mixing to cause some of the water to boil explosively. The resulting "bumping" can spatter the acid.</li> <li>Inorganic acids react with active metals, including such structural metals as aluminum and iron, to release hydrogen, a flammable gas.</li> <li>Inorganic acids react with oryanide compounds to release gaseous hydrogen cyanide.</li> <li>Inorganic acids generate flammable and/or toxic gases in contact with dithiccarbamates, isocyanates, mercaptans, nitrides, nitriles, sulfides, and strong reducing agents. Additional gas-generating reactions occur with sulfites, nitrites, thiosulfates (to give H2S and SO3), dithionites (SO2), and even carbonates.</li> <li>Acids often catalyse (increase the rate of) chemical reactions.</li> <li>Avoid strong acids, acid chlorides, acid anhydrides and chloroformates.</li> </ul>			

## SECTION 8 EXPOSURE CONTROLS / PERSONAL PROTECTION

## **Control parameters**

## OCCUPATIONAL EXPOSURE LIMITS (OEL)

INGREDIENT DATA

Source	Ingredient	Material name	TWA	STEL	Peak	Notes
US OSHA Permissible Exposure Levels (PELs) - Table Z1	magnesium	Manganese compounds / Manganese fume	Not Available	Not Available	5 mg/m3	(as Mn)
US OSHA Permissible Exposure Levels (PELs) - Table Z3	magnesium	Inert or Nuisance Dust	5 mg/m3 / 15 mg/m3 / 15 mppcf / 50 mppcf	Not Available	Not Available	Respirable fraction;All inert or nuisance dusts, whether mineral, inorganic, or organic, not listed specifically by substance name are covered by this limit, which is the same as the Particulates Not Otherwise Regulated (PNOR) limit in Table Z-1. / Total dust;All inert or nuisance dusts, whether mineral, inorganic, or organic, not listed specifically by substance name are covered by this limit, which is the same as the Particulates Not Otherwise Regulated (PNOR) limit in Table Z-1.
US NIOSH Recommended Exposure Limits (RELs)	magnesium	Synonyms vary depending upon the specific aluminum compound.	5 mg/m3 / 2 mg/m3	Not Available	Not Available	Not Available
US OSHA Permissible Exposure Levels (PELs) - Table Z1	nitric acid	Nitric acid	5 mg/m3 / 2 ppm	Not Available	Not Available	Not Available
US ACGIH Threshold Limit Values (TLV)	nitric acid	Nitric acid	2 ppm	4 ppm	Not Available	TLV® Basis: URT & eye irr; dental erosion
US NIOSH Recommended Exposure Limits (RELs)	nitric acid	Aqua fortis, Engravers acid, Hydrogen nitrate, Red furning nitric acid (RFNA), White furning nitric acid (WFNA)	5 mg/m3 / 2 ppm	10 mg/m3 / 4 ppm	Not Available	Not Available

## EMERGENCY LIMITS

Ingredient	Material name	TEEL-1	TEEL-2	TEEL-3
magnesium	Magnesium	18 mg/m3	200 mg/m3	1,200 mg/m3
potassium	Potassium	2.3 mg/m3	25 mg/m3	150 mg/m3
sodium	Sodium	13 mg/m3	140 mg/m3	870 mg/m3
nitric acid	Nitric acid	Not Available	Not Available	Not Available
Ingredient	Original IDLH		Revised IDLH	

Chemwatch: 9-357917

## **ICP** Analytical Mixture 15

calcium	Not Available	Not Available
magnesium	N.E. / N.E.	500 mg/m3
potassium	Not Available	Not Available
sodium	Not Available	Not Available
nitric acid	100 ppm	25 ppm
water	Not Available	Not Available

## Exposure controls

Appropriate engineering controls	Engineering controls are used to remove a hazard or place a barrier between the worker and the haz effective in protecting workers and will typically be independent of worker interactions to provide this hi The basic types of engineering controls are: Process controls which involve changing the way a job activity or process is done to reduce the risk. Enclosure and/or isolation of emission source which keeps a selected hazard "physically" away from th "removes" air in the work environment. Ventilation can remove or dilute an air contaminant if designed the particular process and chemical or contaminant in use. Employers may need to use multiple types of controls to prevent employee overexposure. Local exhaust ventilation usually required. If risk of overexposure exists, wear approved respirator. Co Supplied-air type respirator may be required in special circumstances. Correct fit is essential to ensu An approved self contained breathing apparatus (SCBA) may be required in some situations. Provide adequate ventilation in warehouse or closed storage area. Air contaminants generated in the turn, determine the "capture velocities" of fresh circulating air required to effectively remove the contain Type of Contaminant: solvent, vapours, degreasing etc., evaporating from tank (in still air). aerosols, fumes from pouring operations, intermittent container filling, low speed conveyer transfers acid fumes, pickling (released at low velocity into zone of active generation) direct spray, spray painting in shallow booths, drum filling, conveyer loading, crusher dusts, gas dis- zone of rapid air motion)	igh level of protection. he worker and ventilation that stra properly. The design of a ventilation prrect fit is essential to obtain adequive re adequate protection. workplace possess varying "escan minant.	tegically "adds" and on system must match quate protection.
	grinding, abrasive blasting, tumbling, high speed wheel generated dusts (released at high initial velo air motion).	ocity into zone of very high rapid	2.5-10 m/s (500-2000 f/min.)
	Within each range the appropriate value depends on:		
	Lower end of the range	Upper end of the range	
		1: Disturbing room air currents	
	2: Contaminants of low toxicity or of nuisance value only.	2: Contaminants of high toxicity	
	3: Intermittent, low production.	3: High production, heavy use	
	4: Large hood or large air mass in motion 4: Small hood-local control only		
	of distance from the extraction point (in simple cases). Therefore the air speed at the extraction point s distance from the contaminating source. The air velocity at the extraction fan, for example, should be a		
Personal protection	solvents generated in a tank 2 meters distant from the extraction point. Other mechanical consideratio apparatus, make it essential that theoretical air velocities are multiplied by factors of 10 or more when	ns, producing performance deficit	ts within the extraction
Personal protection	solvents generated in a tank 2 meters distant from the extraction point. Other mechanical consideratio	ns, producing performance deficit extraction systems are installed o esirable, as in laboratories; specta a danger of splashing, or if the mat oggles must be properly fitted. rotection of eyes; these afford face nts. A written policy document, de review of lens absorption and ad rained in their removal and suitabl contact lens as soon as practicable	ts within the extraction r used. Acles are not sufficient terial may be under e protection. escribing the wearing of sorption for the class of le equipment should be e. Lens should be remove
	<ul> <li>solvents generated in a tank 2 meters distant from the extraction point. Other mechanical consideratio apparatus, make it essential that theoretical air velocities are multiplied by factors of 10 or more when</li> <li>With the extraction point. Other mechanical consideration of the extraction point. Other mechanical consideration apparatus, make it essential that theoretical air velocities are multiplied by factors of 10 or more when</li> <li>Safety glasses with unperforated side shields may be used where continuous eye protection is de where complete eye protection is needed such as when handling bulk-quantities, where there is a pressure.</li> <li>Chemical goggles.whenever there is a danger of the material coming in contact with the eyes; got</li> <li>Full face shield (20 cm, 8 in minimum) may be required for supplementary but never for primary p</li> <li>Alternatively a gas mask may replace splash goggles and face shields.</li> <li>Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irrital lenses or restrictions on use, should be created for each workplace or task. This should include a a chemicals in use and an account of injury experience. Medical and first-aid personnel should be the readily available. In the event of chemical exposure, begin eye irrigation immediately and remove or at the first signs of eye redness or irritation - lens should be removed in a clean environment only at the first signs of eye redness or irritation - lens should be removed in a clean environment only at the first signs of eye redness or irritation - lens should be removed in a clean environment only at the first signs of eye redness or irritation - lens should be removed in a clean environment only at the first signs of eye redness or irritation - lens should be removed in a clean environment only at the first signs of eye redness or irritation - lens should be removed in a clean environment only at the first signs of eye redness or irritation - lens should be removed in a clean</li></ul>	ns, producing performance deficit extraction systems are installed o esirable, as in laboratories; specta a danger of splashing, or if the mat oggles must be properly fitted. rotection of eyes; these afford face nts. A written policy document, de review of lens absorption and ad rained in their removal and suitabl contact lens as soon as practicable	ts within the extraction r used. Acles are not sufficient terial may be under e protection. escribing the wearing of sorption for the class of le equipment should be e. Lens should be remove
Eye and face protection	<ul> <li>solvents generated in a tank 2 meters distant from the extraction point. Other mechanical consideratio apparatus, make it essential that theoretical air velocities are multiplied by factors of 10 or more when</li> <li>Safety glasses with unperforated side shields may be used where continuous eye protection is de where complete eye protection is needed such as when handling bulk-quantities, where there is a pressure.</li> <li>Chemical goggles.whenever there is a danger of the material coming in contact with the eyes; got</li> <li>Full face shield (20 cm, 8 in minimum) may be required for supplementary but never for primary p</li> <li>Alternatively a gas mask may replace splash goggles and face shields.</li> <li>Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irrital lenses or restrictions on use, should be created for each workplace or task. This should include a chemicals in use and an account of injury experience. Medical and first-aid personnel should be treadily available. In the event of chemical exposure, begin eye irrigation immediately and remove or at the first signs of eye redness or irritation - lens should be removed in a clean environment only a Current Intelligence Bulletin 59], [AS/NZS 1336 or national equivalent]</li> </ul>	ns, producing performance deficit extraction systems are installed o esirable, as in laboratories; specta a danger of splashing, or if the mat oggles must be properly fitted. rotection of eyes; these afford face ints. A written policy document, de review of lens absorption and ad rained in their removal and suitabl contact lens as soon as practicable after workers have washed hands	ts within the extraction r used. Acles are not sufficient terial may be under e protection. escribing the wearing of sorption for the class of le equipment should be e. Lens should be remove
Eye and face protection	<ul> <li>solvents generated in a tank 2 meters distant from the extraction point. Other mechanical consideratio apparatus, make it essential that theoretical air velocities are multiplied by factors of 10 or more when</li> <li>Image: The second se</li></ul>	ns, producing performance deficit extraction systems are installed o esirable, as in laboratories; specta a danger of splashing, or if the mat oggles must be properly fitted. rotection of eyes; these afford face ints. A written policy document, de review of lens absorption and ad rained in their removal and suitabl contact lens as soon as practicable after workers have washed hands	ts within the extraction r used. Acles are not sufficient terial may be under e protection. escribing the wearing of sorption for the class of le equipment should be e. Lens should be remove
Eye and face protection Skin protection Hands/feet protection	<ul> <li>solvents generated in a tank 2 meters distant from the extraction point. Other mechanical consideratio apparatus, make it essential that theoretical air velocities are multiplied by factors of 10 or more when</li> <li>Safety glasses with unperforated side shields may be used where continuous eye protection is de where complete eye protection is needed such as when handling bulk-quantities, where there is a pressure.</li> <li>Chemical goggles.whenever there is a danger of the material coming in contact with the eyes; gr</li> <li>Full face shield (20 cm, 8 in minimum) may be required for supplementary but never for primary p</li> <li>Atternatively a gas mask may replace splash goggles and face shields.</li> <li>Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irrita lenses or restrictions on use, should be created for each workplace or task. This should include a chemicals in use and an account of injury experience. Medical and first-aid personnel should be treadily available. In the event of chemical exposure, begin eye irrigation immediately and remove or at the first signs of eye redness or irritation - lens should be removed in a clean environment only a Current Intelligence Bulletin 59]. [AS/NZS 1336 or national equivalent]</li> <li>See Hand protection below</li> <li>Elbow length PVC gloves</li> <li>When handling corrosive liquids, wear trousers or overalls outside of boots, to avoid spills entering the spills entering in the spille entering in the spille entering in the spille entering in the spille of the spille of the spille entering in the spille of the spille of the spille entering in the spill</li></ul>	ns, producing performance deficit extraction systems are installed o esirable, as in laboratories; specta a danger of splashing, or if the mat oggles must be properly fitted. rotection of eyes; these afford face ints. A written policy document, de review of lens absorption and ad rained in their removal and suitabl contact lens as soon as practicable after workers have washed hands	ts within the extraction r used. Acles are not sufficient terial may be under e protection. escribing the wearing of sorption for the class of le equipment should be e. Lens should be remove

## **ICP** Analytical Mixture 15

Type A Filter of sufficient capacity. (AS/NZS 1716 & 1715, EN 143:2000 & 149:2001, ANSI Z88 or national equivalent)

## SECTION 9 PHYSICAL AND CHEMICAL PROPERTIES

#### Information on basic physical and chemical properties

Appearance	Colourless		
Physical state	Liquid	Relative density (Water = 1)	Not Available
Odour	Not Available	Partition coefficient n-octanol / water	Not Available
Odour threshold	Not Available	Auto-ignition temperature (°C)	Not Available
pH (as supplied)	<2	Decomposition temperature	Not Available
Melting point / freezing point (°C)	Not Available	Viscosity (cSt)	Not Available
Initial boiling point and boiling range (°C)	Not Available	Molecular weight (g/mol)	Not Available
Flash point (°C)	Not Available	Taste	Not Available
Evaporation rate	Not Available	Explosive properties	Not Available
Flammability	Not Available	Oxidising properties	Not Available
Upper Explosive Limit (%)	Not Available	Surface Tension (dyn/cm or mN/m)	Not Available
Lower Explosive Limit (%)	Not Available	Volatile Component (%vol)	Not Available
Vapour pressure (kPa)	Not Available	Gas group	Not Available
Solubility in water (g/L)	Miscible	pH as a solution (1%)	Not Available
Vapour density (Air = 1)	Not Available	VOC g/L	Not Available

#### SECTION 10 STABILITY AND REACTIVITY

Reactivity	See section 7
Chemical stability	Contact with alkaline material liberates heat
Possibility of hazardous reactions	See section 7
Conditions to avoid	See section 7
Incompatible materials	See section 7
Hazardous decomposition products	See section 5

## SECTION 11 TOXICOLOGICAL INFORMATION

### Information on toxicological effects

Inhaled	The material can cause respiratory irritation in some persons. The body's resp Corrosive acids can cause irritation of the respiratory tract, with coughing, chok nausea and weakness. The material has <b>NOT</b> been classified by EC Directives or other classification s animal or human evidence.	ing and mucous membrane damage. There may be dizziness, headache,
Ingestion	Ingestion of acidic corrosives may produce burns around and in the mouth, the speaking may also be evident. The material has <b>NOT</b> been classified by EC Directives or other classification stanimal or human evidence.	
Skin Contact	Skin contact with acidic corrosives may result in pain and burns; these may be of Skin contact is not thought to have harmful health effects (as classified under E through wounds, lesions or abrasions. Open cuts, abraded or irritated skin should not be exposed to this material Entry into the blood-stream, through, for example, cuts, abrasions or lesions, ma of the material and ensure that any external damage is suitably protected.	
Eye	If applied to the eyes, this material causes severe eye damage. Direct eye contact with acid corrosives may produce pain, tears, sensitivity to lin completely.	ght and burns. Mild burns of the epithelia generally recover rapidly and
Chronic	Repeated or prolonged exposure to acids may result in the erosion of teeth, sw and inflammation of lung tissue often occurs. Long-term exposure to respiratory irritants may result in disease of the airways Substance accumulation, in the human body, may occur and may cause some c	involving difficult breathing and related systemic problems.
ICP Analytical Mixture 15	TOXICITY	IRRITATION
ICF Analytical Mixture 15	Not Available	Not Available
		1

Chemwatch: 9-357917 Catalogue number: ICP-AM-15-1M Version No: 1.1

#### **ICP Analytical Mixture 15**

ΤΟΧΙΟΙΤΥ IRRITATION Dermal (rabbit) LD50: >2500 mg/kg<sup>[1]</sup> Not Available calcium Oral (rat) LD50: >2000 mg/kg<sup>[1]</sup> ΤΟΧΙΟΙΤΥ IRRITATION magnesium Oral (rat) LD50: >2000 mg/kg<sup>[1]</sup> Not Available TOXICITY IRRITATION potassium Not Available Not Available TOXICITY IRRITATION sodium Not Available Not Available TOXICITY IRRITATION Inhalation (rat) LC50: 0.13 mg/L/4hr<sup>[2]</sup> nitric acid Not Available Inhalation (rat) LC50: 2500 ppm/1h \*t<sup>[2]</sup> ΤΟΧΙΟΙΤΥ IRRITATION wate Oral (rat) LD50: >90000 mg/kg<sup>[2]</sup> Not Available 1. Value obtained from Europe ECHA Registered Substances - Acute toxicity 2.\* Value obtained from manufacturer's SDS. Unless otherwise specified data Legend: extracted from RTECS - Register of Toxic Effect of chemical Substances The solid may react violently on contact with wet skin tissue, i.e. eyes, mouth, causing chemical and thermal burns. The acute effects include burns, ulceration, CALCIUM or tissue death, severe eye damage (corneal burns or opacification), and probable blindness. Inhalation of dust or fumes (especially from a fire involving calcium) will cause shortness of breath, nausea, headache, nose and respiratory tract irritation and in extreme, pneumonitis for acid mists, aerosols, vapours Data from assays for genotoxic activity in vitro suggest that eukaryotic cells are susceptible to genetic damage when the pH falls to about 6.5. The material may produce severe irritation to the eye causing pronounced inflammation. NITRIC ACID The material may produce respiratory tract irritation, and result in damage to the lung including reduced lung function. The material may cause severe skin irritation after prolonged or repeated exposure and may produce on contact skin redness, swelling, the production of vesicles, scaling and thickening of the skin. Oral (?) LD50: 50-500 mg/kg \* [Various Manufacturers] CALCIUM & POTASSIUM & No significant acute toxicological data identified in literature search. SODIUM & WATER CALCIUM & POTASSIUM & SODIUM & NITRIC ACID Asthma-like symptoms may continue for months or even years after exposure to the material ceases.  $\bigcirc$  $\bigcirc$ Acute Toxicity Carcinogenicity Skin Irritation/Corrosion Ý Reproductivity 0 Serious Eve ~ STOT - Single Exposure  $\bigcirc$ Damage/Irritation Respiratory or Skin  $\bigcirc$  $\bigcirc$ STOT - Repeated Exposure sensitisation  $\bigcirc$  $\bigcirc$ Mutagenicity Aspiration Hazard X – Data available but does not fill the criteria for classification Legend: - Data available to make classification

🚫 – Data Not Available to make classification

#### **SECTION 12 ECOLOGICAL INFORMATION**

#### Toxicity

,					
Ingredient	Endpoint	Test Duration (hr)	Species	Value	Source
calcium	EC50	24	Crustacea	6934mg/L	5
calcium	NOEC	48	Crustacea	33.3mg/L	2
magnesium	LC50	96	Fish	541mg/L	2
magnesium	EC50	72	Algae or other aquatic plants	>20mg/L	2
magnesium	EC50	72	Algae or other aquatic plants	>20mg/L	2
magnesium	NOEC	72	Algae or other aquatic plants	>25.5mg/L	2
potassium	EC50	24	Crustacea	400mg/L	5
sodium	EC50	48	Crustacea	1640mg/L	4

Version No: 1.1

Catalogue number: ICP-AM-15-1M

## **ICP** Analytical Mixture 15

sodium	EC50	504	Crustacea	1020mg/L	4
nitric acid	NOEC	16	Crustacea	107mg/L	4
Legend:	(QSAR) - Aquatic Toxicit	, , ,	stered Substances - Ecotoxicological Information x database - Aquatic Toxicity Data 5. ECETOC / ration Data 8. Vendor Data	, ,	

Ecotoxicity:

The tolerance of water organisms towards pH margin and variation is diverse. Recommended pH values for test species listed in OECD guidelines are between 6.0 and almost 9. Acute testing with fish showed 96h-LC50 at about pH 3.5

Prevent, by any means available, spillage from entering drains or water courses.

DO NOT discharge into sewer or waterways

### Persistence and degradability

water LQW LQW	Ingredient	Persistence: Water/Soil	Persistence: Air
	water	LOW	LOW

#### **Bioaccumulative potential**

Ingredient	Bioaccumulation
water	LOW (LogKOW = -1.38)

#### Mobility in soil

Ingredient	Mobility
water	LOW (KOC = 14.3)

## SECTION 13 DISPOSAL CONSIDERATIONS

#### Waste treatment methods

	<ul> <li>Recycle wherever possible.</li> <li>Consult manufacturer for recycling options or consult local or regional waste management authority for disposal if no suitable treatment or disposal facility can be identified.</li> </ul>
Product / Packaging disposal	Treat and neutralise at an approved treatment plant. Treatment should involve: Neutralisation with soda-ash or soda-lime followed by: burial in a land-fill specifically licensed to accept chemical and / or pharmaceutical wastes or Incineration in a licensed apparatus (after admixture with suitable combustible material).
	<ul> <li>Decontaminate empty containers with 5% aqueous sodium hydroxide or soda ash, followed by water. Observe all label safeguards until containers are cleaned and destroyed.</li> </ul>

#### **SECTION 14 TRANSPORT INFORMATION**

#### Labels Required



#### Land transport (DOT)

• • •	
UN number	3264
UN proper shipping name	Corrosive liquid, acidic, inorganic, n.o.s. (contains nitric acid)
Transport hazard class(es)	Class8SubriskNot Applicable
Packing group	II
Environmental hazard	Not Applicable
Special precautions for user	Hazard Label8Special provisions386, B2, IB2, T11, TP2, TP27

## Air transport (ICAO-IATA / DGR)

	=		
UN number	3264		
UN proper shipping name	Corrosive liquid, acidio	c, inorganic, n.o.s	. * (contains nitric acid)
		1	
	ICAO/IATA Class	8	
Transport hazard class(es)	ICAO / IATA Subrisk	Not Applicable	
	ERG Code	8L	

## **ICP** Analytical Mixture 15

## Sea transport (IMDG-Code / GGVSee)

UN number	3264
UN proper shipping name	CORROSIVE LIQUID, ACIDIC, INORGANIC, N.O.S. (contains nitric acid)
Transport hazard class(es)	IMDG Class8IMDG SubriskNot Applicable
Packing group	II Contraction of the second
Environmental hazard	Not Applicable
Special precautions for user	EMS NumberF-A, S-BSpecial provisions274Limited Quantities1 L

## Transport in bulk according to Annex II of MARPOL and the IBC code

Source	Product name	Pollution Category	Ship Type
IMO MARPOL (Annex II) - List of Noxious Liquid Substances Carried in Bulk	Nitric acid (70% and over) Nitric acid (less than 70%)	Y; Y	2 2

## SECTION 15 REGULATORY INFORMATION

## Safety, health and environmental regulations / legislation specific for the substance or mixture

## CALCIUM(7440-70-2) IS FOUND ON THE FOLLOWING REGULATORY LISTS

US - Massachusetts - Right To Know Listed Chemicals	US - Rhode Island Hazardous Substance List
US - Pennsylvania - Hazardous Substance List	US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory
MAGNESIUM(7439-95-4) IS FOUND ON THE FOLLOWING REGULATORY LISTS	
US - Alaska Limits for Air Contaminants	US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants
US - California OEHHA/ARB - Chronic Reference Exposure Levels and Target Organs	US - Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants
(CRELs)	US - Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air
US - California Permissible Exposure Limits for Chemical Contaminants	Contaminants
US - Hawaii Air Contaminant Limits	US - Washington Permissible exposure limits of air contaminants
US - Idaho - Limits for Air Contaminants	US - Washington Toxic air pollutants and their ASIL, SQER and de minimis emission values
US - Massachusetts - Right To Know Listed Chemicals	US - Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants
US - Michigan Exposure Limits for Air Contaminants	US Clean Air Act - Hazardous Air Pollutants
US - Minnesota Permissible Exposure Limits (PELs)	US NIOSH Recommended Exposure Limits (RELs)
US - Oregon Permissible Exposure Limits (Z-1)	US OSHA Permissible Exposure Levels (PELs) - Table Z1
US - Pennsylvania - Hazardous Substance List	US OSHA Permissible Exposure Levels (PELs) - Table Z3
US - Rhode Island Hazardous Substance List	US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory
POTASSIUM(7440-09-7) IS FOUND ON THE FOLLOWING REGULATORY LISTS	
International Air Transport Association (IATA) Dangerous Goods Regulations - Prohibited List	US - Rhode Island Hazardous Substance List
Passenger and Cargo Aircraft	US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory
US - Massachusetts - Right To Know Listed Chemicals	
US - Pennsylvania - Hazardous Substance List	
SODIUM(7440-23-5) IS FOUND ON THE FOLLOWING REGULATORY LISTS	
International Air Transport Association (IATA) Dangerous Goods Regulations - Prohibited List	US - Rhode Island Hazardous Substance List
Passenger and Cargo Aircraft	US CWA (Clean Water Act) - List of Hazardous Substances
US - Massachusetts - Right To Know Listed Chemicals	US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

NITRIC ACID(7697-37-2) IS FOUND ON THE FOLLOWING REGULATORY LISTS

US - Pennsylvania - Hazardous Substance List

Continued...

hemwatch: 9-357917	Page 10 of 11 Issue Date: 03/10/20
atalogue number: ICP-AM-15-1M	ICP Analytical Mixture 15 Print Date: 03/10/20
ersion No: 1.1	
International Air Transport Association (IATA) Dangerous Goods Regulations - Prohi Passenger and Cargo Aircraft	ted List US - Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air Contaminants
US - Alaska Limits for Air Contaminants	US - Washington Permissible exposure limits of air contaminants
US - California OEHHA/ARB - Acute Reference Exposure Levels and Target Organs	RELs) US - Washington Toxic air pollutants and their ASIL, SQER and de minimis emission values
US - California OEHHA/ARB - Chronic Reference Exposure Levels and Target Orga (CRELs)	s US - Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants US ACGIH Threshold Limit Values (TLV)
US - California Permissible Exposure Limits for Chemical Contaminants	US Clean Air Act - Hazardous Air Pollutants
US - California Proposition 65 - Carcinogens	US CWA (Clean Water Act) - List of Hazardous Substances
US - Hawaii Air Contaminant Limits	US CWA (Clean Water Act) - Priority Pollutants
US - Idaho - Limits for Air Contaminants	US CWA (Clean Water Act) - Toxic Pollutants
US - Massachusetts - Right To Know Listed Chemicals	US EPCRA Section 313 Chemical List
US - Michigan Exposure Limits for Air Contaminants	US National Toxicology Program (NTP) 14th Report Part A Known to be Human Carcinogens
US - Minnesota Permissible Exposure Limits (PELs)	US NIOSH Recommended Exposure Limits (RELs)
US - Oregon Permissible Exposure Limits (Z-1)	US OSHA Permissible Exposure Levels (PELs) - Table Z1
US - Pennsylvania - Hazardous Substance List	US Priority List for the Development of Proposition 65 Safe Harbor Levels - No Significant Risl
US - Rhode Island Hazardous Substance List	Levels (NSRLs) for Carcinogens and Maximum Allowable Dose Levels (MADLs) for
US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants	Chemicals Causing Reproductive Toxicity
US - Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Cont	
	US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

WATER(7732-18-5) IS FOUND ON THE FOLLOWING REGULATORY LISTS US - Pennsylvania - Hazardous Substance List

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

#### Federal Regulations

### Superfund Amendments and Reauthorization Act of 1986 (SARA)

#### SECTION 311/312 HAZARD CATEGORIES

Immediate (acute) health hazard	Yes
Delayed (chronic) health hazard	No
Fire hazard	No
Pressure hazard	No
Reactivity hazard	No

#### US. EPA CERCLA HAZARDOUS SUBSTANCES AND REPORTABLE QUANTITIES (40 CFR 302.4)

Name	Reportable Quantity in Pounds (lb)	Reportable Quantity in kg	
Sodium	10	4.54	
Nitric acid	1000	454	

#### State Regulations

#### US. CALIFORNIA PROPOSITION 65

WARNING: This product contains a chemical known to the State of California to cause cancer and birth defects or other reproductive harm

## US - CALIFORNIA PREPOSITION 65 - CARCINOGENS & REPRODUCTIVE TOXICITY (CRT): LISTED SUBSTANCE

#### Nickel compounds Listed

National Inventory	Status
Australia - AICS	Υ
Canada - DSL	Υ
Canada - NDSL	N (sodium; calcium; potassium; magnesium; water; nitric acid)
China - IECSC	Υ
Europe - EINEC / ELINCS / NLP	Y
Japan - ENCS	N (sodium; calcium; potassium; magnesium; water)
Korea - KECI	Υ
New Zealand - NZIoC	Υ
Philippines - PICCS	Y
USA - TSCA	Υ
Legend:	Y = All ingredients are on the inventory N = Not determined or one or more ingredients are not on the inventory and are not exempt from listing(see specific ingredients in brackets)

## **SECTION 16 OTHER INFORMATION**

#### Other information

#### Ingredients with multiple cas numbers

calcium 7440-70-2, 8047-59-4	Name	CAS No
	calcium	7440-70-2, 8047-59-4

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references.

Chemwatch: 9-357917	Page <b>11</b> of <b>11</b>	Issue Date: 03/10/2017
Catalogue number: ICP-AM-15-1M	ICP Analytical Mixture 15	Print Date: 03/10/2017
Version No: 1.1	let sinalytical mixture to	

The SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

#### Definitions and abbreviations

PC – TWA: Permissible Concentration-Time Weighted Average PC – STEL: Permissible Concentration-Short Term Exposure Limit IARC: International Agency for Research on Cancer ACGIH: American Conference of Governmental Industrial Hygienists STEL: Short Term Exposure Limit TEEL: Temporary Emergency Exposure Limit, IDLH: Immediately Dangerous to Life or Health Concentrations OSF: Odour Safety Factor NOAEL: No Observed Adverse Effect Level LOAEL: Lowest Observed Adverse Effect Level TLV: Threshold Limit Value LOD: Limit Of Detection OTV: Odour Threshold Value BCF: BioConcentration Factors

BEI: Biological Exposure Index This document is copyright.

Apart from any fair dealing for the purposes of private study, research, review or criticism, as permitted under the Copyright Act, no part may be reproduced by any process without written permission from CHEMWATCH. TEL (+61 3) 9572 4700.

end of SDS