

High-Purity Standards

Catalogue number: ICP-AM-4

Version No: 1.1

Safety Data Sheet according to OSHA HazCom Standard (2012) requirements

Chemwatch Hazard Alert Code: 3

Issue Date: **06/03/2017** Print Date: **06/03/2017** S.GHS.USA.EN

SECTION 1 IDENTIFICATION

Product Identifier

Product name	ICP Analytical Mixture 4
Synonyms	ICP-AM-4
Proper shipping name	Hydrochloric acid
Other means of identification	ICP-AM-4

Recommended use of the chemical and restrictions on use

Name, address, and telephone number of the chemical manufacturer, importer, or other responsible party

Registered company name	High-Purity Standards
Address	PO Box 41727 SC 29423 United States
Telephone	843-767-7900
Fax	843-767-7906
Website	highpuritystandards.com
Email	Not Available

Emergency phone number

0 , .	
Association / Organisation	INFOTRAC
Emergency telephone numbers	1-800-535-5053
Other emergency telephone numbers	1-352-323-3500

SECTION 2 HAZARD(S) IDENTIFICATION

Classification of the substance or mixture

Classification

Specific target organ toxicity - single exposure Category 3 (respiratory tract irritation), Metal Corrosion Category 1, Skin Corrosion/Irritation Category 1A, Serious Eye Damage Category 1

Label elements

Hazard pictogram(s)





SIGNAL WORD

DANGER

Hazard statement(s)

H335	lay cause respiratory irritation.	
H290	y be corrosive to metals.	
H314	Causes severe skin burns and eye damage.	

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Not Applicable

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Precautionary statement(s) Prevention

P260 Do not breathe dust/fume/gas/mist/vapours/spray.

Precautionary statement(s) Response

P301+P330+P331

IF SWALLOWED: Rinse mouth. Do NOT induce vomiting.

Precautionary statement(s) Storage

P405

Store locked up.

Precautionary statement(s) Disposal

P501

Dispose of contents/container in accordance with local regulations.

SECTION 3 COMPOSITION / INFORMATION ON INGREDIENTS

Substances

See section below for composition of Mixtures

Mixtures

CAS No	%[weight]	Name
7440-36-0	0.01	antimony
7440-70-2	0.01	calcium
7439-95-4	0.01	magnesium
7782-49-2	0.01	selenium
497-19-8	0.01 (as Na)	sodium carbonate
7664-93-9	0.01 (as S)	sulfuric acid
13494-80-9	0.01	tellurium
7440-31-5	0.01	<u>tin</u>
7647-01-0	20	hydrochloric acid
7732-18-5	balance	water

SECTION 4 FIRST-AID MEASURES

Description of first aid measures

Description of first aid me	easures
Eye Contact	If this product comes in contact with the eyes: Immediately hold eyelids apart and flush the eye continuously with running water. Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids. Continue flushing until advised to stop by the Poisons Information Centre or a doctor, or for at least 15 minutes. Transport to hospital or doctor without delay. Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.
Skin Contact	If skin or hair contact occurs: Immediately flush body and clothes with large amounts of water, using safety shower if available. Quickly remove all contaminated clothing, including footwear. Wash skin and hair with running water. Continue flushing with water until advised to stop by the Poisons Information Centre. Transport to hospital, or doctor.
Inhalation	 If furnes or combustion products are inhaled remove from contaminated area. Lay patient down. Keep warm and rested. Prostheses such as false teeth, which may block airway, should be removed, where possible, prior to initiating first aid procedures. Apply artificial respiration if not breathing, preferably with a demand valve resuscitator, bag-valve mask device, or pocket mask as trained. Perform CPR if necessary. Transport to hospital, or doctor, without delay. Inhalation of vapours or aerosols (mists, furnes) may cause lung oedema. Corrosive substances may cause lung damage (e.g. lung oedema, fluid in the lungs). As this reaction may be delayed up to 24 hours after exposure, affected individuals need complete rest (preferably in semi-recumbent posture) and must be kept under medical observation even if no symptoms are (yet) manifested. Before any such manifestation, the administration of a spray containing a dexamethasone derivative or beclomethasone derivative may be considered. This must definitely be left to a doctor or person authorised by him/her. (ICSC13719)
Ingestion	 For advice, contact a Poisons Information Centre or a doctor at once. Urgent hospital treatment is likely to be needed. If swallowed do NOT induce vomiting. If vomiting occurs, lean patient forward or place on left side (head-down position, if possible) to maintain open airway and prevent aspiration. Observe the patient carefully.

▶ Never give liquid to a person showing signs of being sleepy or with reduced awareness; i.e. becoming unconscious. ▶ Give water to rinse out mouth, then provide liquid slowly and as much as casualty can comfortably drink.

▶ Transport to hospital or doctor without delay.

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See Section 11

Indication of any immediate medical attention and special treatment needed

For acute or short term repeated exposures to strong acids:

- ▶ Airway problems may arise from laryngeal edema and inhalation exposure. Treat with 100% oxygen initially.
- Respiratory distress may require cricothyroidotomy if endotracheal intubation is contraindicated by excessive swelling
- Intravenous lines should be established immediately in all cases where there is evidence of circulatory compromise
- Strong acids produce a coagulation necrosis characterised by formation of a coagulum (eschar) as a result of the dessicating action of the acid on proteins in specific tissues. INGESTION:
- ▶ Immediate dilution (milk or water) within 30 minutes post ingestion is recommended.
- DO NOT attempt to neutralise the acid since exothermic reaction may extend the corrosive injury
- ▶ Be careful to avoid further vomit since re-exposure of the mucosa to the acid is harmful. Limit fluids to one or two glasses in an adult.
- ▶ Charcoal has no place in acid management.
- ▶ Some authors suggest the use of lavage within 1 hour of ingestion.

SKIN:

- ▶ Skin lesions require copious saline irrigation. Treat chemical burns as thermal burns with non-adherent gauze and wrapping.
- ▶ Deep second-degree burns may benefit from topical silver sulfadiazine.

EYE:

- ▶ Eye injuries require retraction of the eyelids to ensure thorough irrigation of the conjuctival cul-de-sacs. Irrigation should last at least 20-30 minutes. DO NOT use neutralising agents or any other additives. Several litres of saline are required.
- Cycloplegic drops, (1% cyclopentolate for short-term use or 5% homatropine for longer term use) antibiotic drops, vasoconstrictive agents or artificial tears may be indicated dependent on the severity of the injury.
- ▶ Steroid eye drops should only be administered with the approval of a consulting ophthalmologist).

[Ellenhorn and Barceloux: Medical Toxicology]

SECTION 5 FIRE-FIGHTING MEASURES

Extinguishing media

- ▶ There is no restriction on the type of extinguisher which may be used.
- Use extinguishing media suitable for surrounding area.

Special hazards arising from the substrate or mixture

Fire Incompatibility None known.

Special protective equipment and precautions for fire-fighters

Fire Fighting	
Fire/Explosion Hazard	 Non combustible. Not considered to be a significant fire risk. Acids may react with metals to produce hydrogen, a highly flammable and explosive gas. Heating may cause expansion or decomposition leading to violent rupture of containers. May emit corrosive, poisonous fumes. May emit acrid smoke. Decomposition may produce toxic fumes of: hydrogen chloride

SECTION 6 ACCIDENTAL RELEASE MEASURES

Personal precautions, protective equipment and emergency procedures

See section 8

Environmental precautions

See section 12

Methods and material for containment and cleaning up

Minor Spills	 Drains for storage or use areas should have retention basins for pH adjustments and dilution of spills before discharge or disposal of material. Check regularly for spills and leaks. Clean up all spills immediately. Avoid breathing vapours and contact with skin and eyes. Control personal contact with the substance, by using protective equipment. Contain and absorb spill with sand, earth, inert material or vermiculite. Wipe up. Place in a suitable, labelled container for waste disposal.
Major Spills	#

Personal Protective Equipment advice is contained in Section 8 of the SDS

SECTION 7 HANDLING AND STORAGE

Safe handling

Precautions for safe handling

Avoid al	l persona	al contact	, inclu	ding ir	halation.

- ► Wear protective clothing when risk of exposure occurs.
- Use in a well-ventilated area.
- ▶ WARNING: To avoid violent reaction, ALWAYS add material to water and NEVER water to material
- Avoid smoking, naked lights or ignition sources.
- Avoid contact with incompatible materials.
- ► When handling, **DO NOT** eat, drink or smoke.

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- Keep containers securely sealed when not in use. Avoid physical damage to containers
- Always wash hands with soap and water after handling.
- Work clothes should be laundered separately. Launder contaminated clothing before re-use.
- Use good occupational work practice.
- Observe manufacturer's storage and handling recommendations contained within this SDS.
- Atmosphere should be regularly checked against established exposure standards to ensure safe working conditions are maintained.
- DO NOT allow clothing wet with material to stay in contact with skin

- ► Store in original containers. Keep containers securely sealed.
- Store in a cool, dry, well-ventilated area.
- Store away from incompatible materials and foodstuff containers.
- Protect containers against physical damage and check regularly for leaks.
- ▶ Observe manufacturer's storage and handling recommendations contained within this SDS

Conditions for safe storage, including any incompatibilities

► DO NOT use aluminium or galvanised containers

- ► Check regularly for spills and leaks
- Lined metal can, lined metal pail/ can.
- Plastic pail.
- Polyliner drum.
- ► Packing as recommended by manufacturer.
- Check all containers are clearly labelled and free from leaks.

For low viscosity materials

Suitable container

Other information

Drums and ierricans must be of the non-removable head type.

▶ Where a can is to be used as an inner package, the can must have a screwed enclosure.

For materials with a viscosity of at least 2680 cSt. (23 deg. C) and solids (between 15 C deg. and 40 deg C.):

- Removable head packaging:
- Cans with friction closures and
- low pressure tubes and cartridges

may be used.

Where combination packages are used, and the inner packages are of glass, porcelain or stoneware, there must be sufficient inert cushioning material in contact with inner and outer packages unless the outer packaging is a close fitting moulded plastic box and the substances are not incompatible with the plastic

- ▶ Inorganic acids are generally soluble in water with the release of hydrogen ions. The resulting solutions have pH's of less than 7.0.
- ▶ Inorganic acids neutralise chemical bases (for example: amines and inorganic hydroxides) to form salts neutralisation can generate dangerously large amounts of heat in small spaces
- ▶ The dissolution of inorganic acids in water or the dilution of their concentrated solutions with additional water may generate significant heat.
- ► The addition of water to inorganic acids often generates sufficient heat in the small region of mixing to cause some of the water to boil explosively. The resulting "bumping" can spatter the acid.
- Inorganic acids react with active metals, including such structural metals as aluminum and iron, to release hydrogen, a flammable gas.
- ▶ Inorganic acids can initiate the polymerisation of certain classes of organic compounds.
- Inorganic acids react with cyanide compounds to release gaseous hydrogen cyanide.
- ▶ Inorganic acids generate flammable and/or toxic gases in contact with dithiocarbamates, isocyanates, mercaptans, nitrides, nitrides, sulfides, and strong reducing agents. Additional gas-generating reactions occur with sulfites, nitrites, thiosulfates (to give H2S and SO3), dithionites (SO2), and even carbonates.
- ▶ Acids often catalyse (increase the rate of) chemical reactions.

Storage incompatibility

Hydrogen chloride:

- reacts strongly with strong oxidisers (releasing chlorine gas), acetic anhydride, caesium cyanotridecahydrodecaborate(2-), ethylidene difluoride, hexalithium disilicide, metal acetylide, sodium, silicon dioxide, tetraselenium tetranitride, and many organic materials
- ▶ is incompatible with alkaline materials, acetic anhydride, acetylides, aliphatic amines, alkanolamines, alkylene oxides, aluminium, aluminium-titanium alloys, aromatic amines, amines, amides, 2-aminoethanol, ammonia, ammonium hydroxide, borides, calcium phosphide, carbides, carbonates, cyanides, chlorosulfonic acid, ethylenediamine, ethyleneimine, epichlorohydrin, formaldehyde, isocyanates, metals, metal oxides, metal hydroxides, metal acetylides, metal carbides, oleum, organic anhydrides, potassium permanganate, perchloric acid, phosphides, 3-propiolactone, silicides, sulfites, sulfuric acid, uranium phosphide, vinyl acetate, vinylidene fluoride
- attacks most metals forming flammable hydrogen gas, and some plastics, rubbers and coatings
- reacts with zinc, brass, galvanised iron, aluminium, copper and copper alloys
- Reacts vigorously with alkalis
- ▶ Reacts with mild steel, galvanised steel / zinc producing hydrogen gas which may form an explosive mixture with air.
- Avoid strong acids, acid chlorides, acid anhydrides and chloroformates.

SECTION 8 EXPOSURE CONTROLS / PERSONAL PROTECTION

Control parameters

OCCUPATIONAL EXPOSURE LIMITS (OEL)

INGREDIENT DATA

Source	Ingredient	Material name	TWA	STEL	Peak	Notes
US NIOSH Recommended Exposure Limits (RELs)	antimony	Antimony metal, Antimony powder, Stibium	0.5 mg/m3	Not Available	Not Available	[*Note: The REL also applies to other antimony compounds (as Sb).]
US NIOSH Recommended Exposure Limits (RELs)	selenium	Elemental selenium, Selenium alloy	0.2 mg/m3	Not Available	Not Available	[*Note: The REL also applies to other selenium compounds (as Se) except Selenium hexafluoride.]
US OSHA Permissible Exposure Levels (PELs) - Table Z1	sulfuric acid	Sulfuric acid	1 mg/m3	Not Available	Not Available	TLV® Basis: Pulm func
US NIOSH Recommended Exposure Limits (RELs)	sulfuric acid	Battery acid, Hydrogen sulfate, Oil of vitriol, Sulfuric acid (aqueous)	1 mg/m3	Not Available	Not Available	Not Available

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US ACGIH Threshold Limit Values (TLV)	sulfuric acid	Sulfuric acid	0.2 mg/m3	Not Available	Not Available	Not Available
US NIOSH Recommended Exposure Limits (RELs)	tellurium	Aurum paradoxum, Metallum problematum	0.1 mg/m3	Not Available	Not Available	[*Note: The REL also applies to other tellurium compounds (as Te) except Tellurium hexafluoride and Bismuth telluride.]
US NIOSH Recommended Exposure Limits (RELs)	tin	Metallic tin, Tin flake, Tin metal, Tin powder	2 mg/m3	Not Available	Not Available	[*Note: The REL also applies to other inorganic tin compounds (as Sn) except tin oxides.]
US OSHA Permissible Exposure Levels (PELs) - Table Z1	hydrochloric acid	Hydrogen chloride	Not Available	Not Available	7 mg/m3 / 5 ppm	TLV® Basis: URT irr
US NIOSH Recommended Exposure Limits (RELs)	hydrochloric acid	Anhydrous hydrogen chloride; Aqueous hydrogen chloride (i.e., Hydrochloric acid, Muriatic acid) [Note: Often used in an aqueous solution.]	Not Available	Not Available	7 mg/m3 / 5 ppm	Not Available
US ACGIH Threshold Limit Values (TLV)	hydrochloric acid	Hydrogen chloride	Not Available	Not Available	2 ppm	Not Available

EMERGENCY LIMITS

Ingredient	Material name	TEEL-1	TEEL-2	TEEL-3
antimony	Antimony	1.5 mg/m3	13 mg/m3	80 mg/m3
magnesium	Magnesium	18 mg/m3	200 mg/m3	1,200 mg/m3
selenium	Selenium	0.6 mg/m3	6.6 mg/m3	40 mg/m3
sodium carbonate	Sodium carbonate	7.6 mg/m3	83 mg/m3	500 mg/m3
sulfuric acid	Sulfuric acid	Not Available	Not Available	Not Available
tellurium	Tellurium	1.8 mg/m3	20 mg/m3	110 mg/m3
tin	Tin	6 mg/m3	67 mg/m3	400 mg/m3
hydrochloric acid	Hydrogen chloride; (Hydrochloric acid)	Not Available	Not Available	Not Available
hydrochloric acid	Deuterochloric acid; (Deuterium chloride)	1.8 ppm	22 ppm	100 ppm

Ingredient	Original IDLH	Revised IDLH
antimony	80 mg/m3	50 mg/m3
calcium	Not Available	Not Available
magnesium	Not Available	Not Available
selenium	Unknown mg/m3 / Unknown ppm	1 mg/m3
sodium carbonate	Not Available	Not Available
sulfuric acid	80 mg/m3	15 mg/m3
tellurium	N.E. / N.E.	25 mg/m3
tin	Unknown mg/m3 / 400 mg/m3 / Unknown ppm	25 mg/m3 / 100 mg/m3
hydrochloric acid	100 ppm	50 ppm
water	Not Available	Not Available

Exposure controls

Engineering controls are used to remove a hazard or place a barrier between the worker and the hazard. Well-designed engineering controls can be highly effective in protecting workers and will typically be independent of worker interactions to provide this high level of protection.

The basic types of engineering controls are:

Process controls which involve changing the way a job activity or process is done to reduce the risk.

Enclosure and/or isolation of emission source which keeps a selected hazard "physically" away from the worker and ventilation that strategically "adds" and "removes" air in the work environment. Ventilation can remove or dilute an air contaminant if designed properly. The design of a ventilation system must match the particular process and chemical or contaminant in use.

Employers may need to use multiple types of controls to prevent employee overexposure.

Local exhaust ventilation usually required. If risk of overexposure exists, wear approved respirator. Correct fit is essential to obtain adequate protection. Supplied-air type respirator may be required in special circumstances. Correct fit is essential to ensure adequate protection.

An approved self contained breathing apparatus (SCBA) may be required in some situations.

Provide adequate ventilation in warehouse or closed storage area. Air contaminants generated in the workplace possess varying "escape" velocities which, in turn, determine the "capture velocities" of fresh circulating air required to effectively remove the contaminant.

Appropriate engineering controls

Type of Contaminant:	Air Speed:
solvent, vapours, degreasing etc., evaporating from tank (in still air).	0.25-0.5 m/s (50-100 f/min.)
aerosols, fumes from pouring operations, intermittent container filling, low speed conveyer transfers, welding, spray drift, plating acid fumes, pickling (released at low velocity into zone of active generation)	0.5-1 m/s (100-200 f/min.)
direct spray, spray painting in shallow booths, drum filling, conveyer loading, crusher dusts, gas discharge (active generation into zone of rapid air motion)	1-2.5 m/s (200-500 f/min.)
grinding, abrasive blasting, tumbling, high speed wheel generated dusts (released at high initial velocity into zone of very high rapid air motion).	2.5-10 m/s (500-2000 f/min.)

Within each range the appropriate value depends on:

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Lower end of the range

1: Room air currents minimal or favourable to capture

1: Disturbing room air currents

2: Contaminants of low toxicity or of nuisance value only.

2: Contaminants of high toxicity

3: Intermittent, low production.

3: High production, heavy use

4: Large hood or large air mass in motion

4: Small hood-local control only

Simple theory shows that air velocity falls rapidly with distance away from the opening of a simple extraction pipe. Velocity generally decreases with the square of distance from the extraction point (in simple cases). Therefore the air speed at the extraction point should be adjusted, accordingly, after reference to distance from the contaminating source. The air velocity at the extraction fan, for example, should be a minimum of 1-2 m/s (200-400 f/min) for extraction of solvents generated in a tank 2 meters distant from the extraction point. Other mechanical considerations, producing performance deficits within the extraction apparatus, make it essential that theoretical air velocities are multiplied by factors of 10 or more when extraction systems are installed or used.

Personal protection











Eye and face protection

- Safety glasses with unperforated side shields may be used where continuous eye protection is desirable, as in laboratories; spectacles are not sufficient where complete eye protection is needed such as when handling bulk-quantities, where there is a danger of splashing, or if the material may be under pressure.
- ▶ Chemical goggles.whenever there is a danger of the material coming in contact with the eyes; goggles must be properly fitted.
- Full face shield (20 cm, 8 in minimum) may be required for supplementary but never for primary protection of eyes; these afford face protection.
- Alternatively a gas mask may replace splash goggles and face shields.
 - Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants. A written policy document, describing the wearing of lenses or restrictions on use, should be created for each workplace or task. This should include a review of lens absorption and adsorption for the class of chemicals in use and an account of injury experience. Medical and first-aid personnel should be trained in their removal and suitable equipment should be readily available. In the event of chemical exposure, begin eye irrigation immediately and remove contact lens as soon as practicable. Lens should be removed at the first signs of eye redness or irritation lens should be removed in a clean environment only after workers have washed hands thoroughly. [CDC NIOSH Current Intelligence Bulletin 59], [AS/NZS 1336 or national equivalent]

Skin protection

See Hand protection below

- ▶ Elbow length PVC gloves
- ▶ When handling corrosive liquids, wear trousers or overalls outside of boots, to avoid spills entering boots.

The selection of suitable gloves does not only depend on the material, but also on further marks of quality which vary from manufacturer to manufacturer. Where the chemical is a preparation of several substances, the resistance of the glove material can not be calculated in advance and has therefore to be checked prior to the application.

The exact break through time for substances has to be obtained from the manufacturer of the protective gloves and has to be observed when making a final choice.

Personal hygiene is a key element of effective hand care. Gloves must only be worn on clean hands. After using gloves, hands should be washed and dried thoroughly. Application of a non-perfumed moisturizer is recommended.

Suitability and durability of glove type is dependent on usage. Important factors in the selection of gloves include:

- frequency and duration of contact
- · chemical resistance of glove material,
- glove thickness and
- dexterity

Select gloves tested to a relevant standard (e.g. Europe EN 374, US F739, AS/NZS 2161.1 or national equivalent).

- · When prolonged or frequently repeated contact may occur, a glove with a protection class of 5 or higher (breakthrough time greater than 240 minutes according to EN 374, AS/NZS 2161.10.1 or national equivalent) is recommended.
- When only brief contact is expected, a glove with a protection class of 3 or higher (breakthrough time greater than 60 minutes according to EN 374. AS/NZS 2161.10.1 or national equivalent) is recommended.
- Some glove polymer types are less affected by movement and this should be taken into account when considering gloves for long-term use.
- Contaminated gloves should be replaced.

For general applications, gloves with a thickness typically greater than 0.35 mm, are recommended.

It should be emphasised that glove thickness is not necessarily a good predictor of glove resistance to a specific chemical, as the permeation efficiency of the glove will be dependent on the exact composition of the glove material. Therefore, glove selection should also be based on consideration of the task requirements and knowledge of breakthrough times.

Glove thickness may also vary depending on the glove manufacturer, the glove type and the glove model. Therefore, the manufacturers' technical data should always be taken into account to ensure selection of the most appropriate glove for the task.

Note: Depending on the activity being conducted, gloves of varying thickness may be required for specific tasks. For example:

- Thinner gloves (down to 0.1 mm or less) may be required where a high degree of manual dexterity is needed. However, these gloves are only likely to give short duration protection and would normally be just for single use applications, then disposed of.
- Thicker gloves (up to 3 mm or more) may be required where there is a mechanical (as well as a chemical) risk i.e. where there is abrasion or puncture potential

Gloves must only be worn on clean hands. After using gloves, hands should be washed and dried thoroughly. Application of a non-perfumed moisturiser is recommended.

Body protection

Hands/feet protection

See Other protection below

Other protection

- Overalls.PVC Apron.
- PVC protective suit may be required if exposure severe.
- ► Eyewash unit.
- Ensure there is ready access to a safety shower.

Thermal hazards

Not Available

Respiratory protection

Type B-P Filter of sufficient capacity. (AS/NZS 1716 & 1715, EN 143:2000 & 149:2001, ANSI Z88 or national equivalent)

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SECTION 9 PHYSICAL AND CHEMICAL PROPERTIES

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Appearance	Not Available		
Physical state	Liquid	Relative density (Water = 1)	Not Available
Odour	Not Available	Partition coefficient n-octanol / water	Not Available
Odour threshold	Not Available	Auto-ignition temperature (°C)	Not Available
pH (as supplied)	<2	Decomposition temperature	Not Available
Melting point / freezing point (°C)	Not Available	Viscosity (cSt)	Not Available
Initial boiling point and boiling range (°C)	Not Available	Molecular weight (g/mol)	Not Available
Flash point (°C)	Not Available	Taste	Not Available
Evaporation rate	Not Available	Explosive properties	Not Available
Flammability	Not Available	Oxidising properties	Not Available
Upper Explosive Limit (%)	Not Available	Surface Tension (dyn/cm or mN/m)	Not Available
Lower Explosive Limit (%)	Not Available	Volatile Component (%vol)	Not Available
Vapour pressure (kPa)	Not Available	Gas group	Not Available
Solubility in water (g/L)	Miscible	pH as a solution (1%)	Not Available
Vapour density (Air = 1)	Not Available	VOC g/L	Not Available

SECTION 10 STABILITY AND REACTIVITY

Reactivity	See section 7
Chemical stability	► Contact with alkaline material liberates heat
Possibility of hazardous reactions	See section 7
Conditions to avoid	See section 7
Incompatible materials	See section 7
Hazardous decomposition products	See section 5

SECTION 11 TOXICOLOGICAL INFORMATION

Eye

Chronic

Information on toxicological effects

The material can cause respiratory irritation in some persons. The body's response to such irritation can cause further lung damage.
Corrective exists can express irritation of the respiratory treat, with courding and muceus membrane demans. There may be dizziness, headache

nausea and weakness.

The material has NOT been classified by EC Directives or other classification systems as "harmful by inhalation". This is because of the lack of corroborating

animal or human evidence.

Hydrogen chloride (HCI) vapour or fumes present a hazard from a single acute exposure. Exposures of 1300 to 2000 ppm have been lethal to humans in a few

minutes.
Inhalation of HCI may cause choking, coughing, burning sensation and may cause ulceration of the nose, throat and larynx. Fluid on the lungs followed by

generalised lung damage may follow.

Breathing of HCl vapour may aggravate asthma and inflammatory or fibrotic pulmonary disease.

High concentrations cause necrosis of the tracheal and bronchial epithelium, pulmonary oedema, atelectasis and emphysema and damage to the pulmonary blood vessels and liver.

Ingestion of acidic corrosives may produce burns around and in the mouth, the throat and oesophagus. Immediate pain and difficulties in swallowing and speaking may also be evident.

The material has NOT been classified by EC Directives or other classification systems as "harmful by ingestion". This is because of the lack of corroborating

animal or human evidence.

Skin contact with acidic corrosives may result in pain and burns; these may be deep with distinct edges and may heal slowly with the formation of scar tissue. Skin contact is not thought to have harmful health effects (as classified under EC Directives); the material may still produce health damage following entry

Skin Contact through wounds, lesions or abrasions.

Open cuts, abraded or irritated skin should not be exposed to this material

Entry into the blood-stream, through, for example, cuts, abrasions or lesions, may produce systemic injury with harmful effects. Examine the skin prior to the use of the material and ensure that any external damage is suitably protected.

If applied to the eyes, this material causes severe eye damage.

completely.

Repeated or prolonged exposure to acids may result in the erosion of teeth, swelling and/or ulceration of mouth lining. Irritation of airways to lung, with cough,

Direct eye contact with acid corrosives may produce pain, tears, sensitivity to light and burns. Mild burns of the epithelia generally recover rapidly and

Repeated or prolonged exposure to acids may result in the erosion of teeth, swelling and/or diceration of mouth lining. Irritation of allways to lung, with cough, and inflammation of lung tissue often occurs.

Long-term exposure to respiratory irritants may result in airways disease, involving difficulty breathing and related whole-body problems. Substance accumulation, in the human body, may occur and may cause some concern following repeated or long-term occupational exposure.

There has been some concern that this material can cause cancer or mutations but there is not enough data to make an assessment. Chronic minor exposure to hydrogen chloride (HCl) vapour or fume may cause discolouration or erosion of the teeth, bleeding of the nose and gums; and ulceration of the mucous membranes of the nose. Workers exposed to hydrochloric acid suffered from stomach inflammation and a number of cases of chronic bronchitis (airway inflammation) have also been reported. Repeated or prolonged exposure to dilute solutions of hydrogen chloride may cause skin inflammation.

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	TOXICITY	IDI	RITATION			
ICP Analytical Mixture 4	Not Available		t Available			
	TOTAL	110	· · · · · · · · · · · · · · · · · · ·			
	TOXICITY			IRRITATION		
	Dermal (rabbit) LD50: >8300 mg/kg ^[1]			Not Available		
antimony	Oral (rat) LD50: 100 mg/kg ^[2]			TTOT / Wallabio		
	Oral (rat) LD50: 100 mg/kg ^{c-2}					
				In the second se		
	TOXICITY	IRRITATION				
calcium	Dermal (rabbit) LD50: >2500 mg/kg ^[1]	Not Available				
	Oral (rat) LD50: >2000 mg/kg ^[1]					
magnesium	TOXICITY			IRRITATION		
	Oral (rat) LD50: >2000 mg/kg ^[1]			Not Available		
selenium	TOXICITY			IRRITATION		
	Oral (rat) LD50: 6700 mg/kgd ^[2]			Not Available		
	TOXICITY		IRRITATION			
	dermal (rat) LD50: >2000 mg/kg*E ^[2]	oderate				
sodium carbonate	Oral (rat) LD50: 2800 mg/kg*d ^[2]	s mild				
		ERE				
			Skin (rabbit): 500 mg/24	h mild		
			TATION			
	TOXICITY					
sulfuric acid	Oral (rat) LD50: 2140 mg/kgE ^[2]		rabbit): 1.38 mg SEVER			
	Eye (rabbit): 5 mg/30sec SEVERE					
tellurium	TOXICITY	RRITATION Not Available				
	Oral (rat) LD50: 83 mg/kgd ^[2]			Not Available		
	TOXICITY	IRRITATION				
tin	dermal (rat) LD50: >2000 mg/kg ^[1]	Not Available				
	Oral (rat) LD50: >2000 mg/kg ^[1]					
			1			
	TOXICITY		IRRITATION	30s - mild		
hydrochloric acid	Inhalation (rat) LC50: 781 ppm/1hr ^[2]		Eye (rabbit): 5mg	t): 5mg/30s - mild		
	Oral (rat) LD50: 900 mg/kg ^[2]					
			RITATION			
water	TOXICITY					
	Not Available	No	t Available			
Lamarete	1 Value obtained from Furence ECLIA Positional Substant	noge - Acute toxicity 2 * 1/a	lue obtained from many	facturarie SDS Inlans athornian according		
Legend:	Value obtained from Europe ECHA Registered Substan extracted from RTECS - Register of Toxic Effect of chemic		iue obiairieu irom mänul	acturer's סטט. Onless otherwise specified		
CALCIUM	The solid may react violently on contact with wet skin tissue or tissue death, severe eye damage (corneal burns or opar calcium) will cause shortness of breath, nausea, headache	cification), and probable b	lindness. Inhalation of de	ust or fumes (especially from a fire involving		
SODIUM CARBONATE	For sodium carbonate: Sodium carbonate has little potential for skin irritation, but The material may cause skin irritation after prolonged or rescaling and thickening of the skin.		y produce on contact skir	n redness, swelling, the production of vesicl		
SULFURIC ACID	WARNING: For inhalation exposure ONLY: This substant	ce has been classified by	the IARC as Group 1: C	ARCINOGENIC TO HUMANS		

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TELLURIUM	Changes to respiratory system, kidney, ureter, bladder and sp system.	Changes to respiratory system, kidney, ureter, bladder and specific developmental abnormalities involving central nervous, craniofacial, musculoskeletal system.						
HYDROCHLORIC ACID	For acid mists, aerosols, vapours Test results suggest that eukaryotic cells are susceptible to g The material may be irritating to the eye, with prolonged contr		about 6.5.					
CALCIUM & TIN & HYDROCHLORIC ACID & WATER	No significant acute toxicological data identified in literature	search.						
CALCIUM & SODIUM CARBONATE & SULFURIC ACID & HYDROCHLORIC ACID	Asthma-like symptoms may continue for months or even years after exposure to the material ends.							
SELENIUM & HYDROCHLORIC ACID	The substance is classified by IARC as Group 3: NOT classifiable as to its carcinogenicity to humans.							
Acute Toxicity	0	Carcinogenicity	0					
Skin Irritation/Corrosion	✓	Reproductivity	0					
Serious Eye Damage/Irritation	✓	STOT - Single Exposure	~					
Respiratory or Skin sensitisation	0	STOT - Repeated Exposure	0					
Mutagenicity	0	Aspiration Hazard	0					

Legend:

- X − Data available but does not fill the criteria for classification
 ✓ − Data available to make classification
- - Data Not Available to make classification

SECTION 12 ECOLOGICAL INFORMATION

Toxicity

ICP Analytical Mixture 4	ENDPOINT TEST DURATION (HR)		SPECIES		VALU	VALUE		SOURCE			
or Analytical Mixture 4	Not Applicable		Not Applicable		Not Appli	cable	Not A	pplicabl	plicable Not Applicable		plicable
	ENDPOINT	TES	T DURATION (HR)	SPEC	IES			\	/ALUE		SOURCE
	LC50	96		Fish				0.93mg/L		2	
	EC50	48		Crustacea			1	1mg/L		2	
antimony	EC50	72		Algae	Algae or other aquatic plants			>	>2.4mg/L		2
	EC50	96		Crusta	acea			C).5mg/L		2
	NOEC	720		Fish				>	-0.0075mg/L		2
	ENDPOINT		TEST DURATION (HR)			SPECIES		VALU	E	so	URCE
calcium	EC50	24			Crustacea			6934mg/L		5	
	NOEC		48			Crustacea		33.3m	g/L	2	
											0011000
magnesium	ENDPOINT		ST DURATION (HR)		SPECIES				VALUE 541mg/l		SOURCE
	LC50	96		Fish	Algae or other aquatic plants				541mg/L >20mg/L		2
	EC50				Algae or other aquatic plants				-		2
	EC50	72		-					>20mg/L		2
	NOEC	72		Alga	le or otner a	aquatic plants			>25.5mg/L		2
	ENDPOINT	TES	T DURATION (HR)	SPECI	ES			VA	ALUE		SOURCE
	LC50	96		Fish				>0).0262mg/L		2
	EC50	48		Crustacea			>0).1603mg/L		2	
selenium	EC50	72		Algae or other aquatic plants		>0	>0.00173mg/L		2		
	BCF	504		Crustacea			0.711mg/L		4		
	EC50	96			Algae or other aquatic plants			0.355mg/L		2	
	NOEC	72		Algae	Algae or other aquatic plants		0.0	0.000547mg/L		2	
	ENDPOINT	TES	T DURATION (HR)	SPEC	CIES			1	VALUE		SOURCE
	LC50	96		Fish				3	300mg/L		4
sodium carbonate	EC50	48		Crusta	acea				=176mg/L		1
	EC50	96		Algae	or other ac	quatic plants		2	242mg/L		4

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tellurium LC50 EC51 EC50 NOE EC51 EC50 EC50	DPOINT T 50 9 50 4 50 2 EC 7 DPOINT T 50 9 50 4 50 7 50 3 EC 7	TEST DURATION (HR) 16 18 140 1200 TEST DURATION (HR) 16 18 12 184 12 TEST DURATION (HR)	Fish SPEC Fish Crusta Algae Crusta Algae	acea or other aquatic plants CIES acea or other aquatic plants acea or other aquatic plants acea		VALUE =8mg/L =42.5mg/L 2.5000mg/L 0.13mg/L VALUE 18.413mg/L 5.79mg/L >11.7mg/L 4.440mg/L 3.34mg/L VALUE	SOURCE 1 1 4 2 SOURCE 3 2 2 3 2 SOURCE 3 SOURCE
tellurium LC50 EC51 EC50 NOE EC51 EC50 EC50	50 9 50 4 50 2 EC 7 DPOINT T 50 9 50 4 50 7 50 3 EC 7 DPOINT T 50 9 9 9 9 9 9 9 9 9 9 9 9 9 9 9 9 9 9 9	166 188 140 1200 1EST DURATION (HR) 166 18 18 12 1884 12 1EST DURATION (HR)	Fish Crusta Algae Fish SPEC Fish Crusta Algae Crusta Algae	acea or other aquatic plants CIES acea or other aquatic plants acea or other aquatic plants		=8mg/L =42.5mg/L 2.5000mg/L 0.13mg/L VALUE 18.413mg/L 5.79mg/L >11.7mg/L 4.440mg/L 3.34mg/L	1 1 4 2 2 SOURCE 3 2 2 3 2 2
tellurium LC50 EC51 EC50 NOE EC51 EC50 EC50	50 9 50 4 50 2 EC 7 DPOINT T 50 9 50 4 50 7 50 3 EC 7 DPOINT T 50 9 9 9 9 9 9 9 9 9 9 9 9 9 9 9 9 9 9 9	166 188 140 1200 1EST DURATION (HR) 166 18 18 12 1884 12 1EST DURATION (HR)	Fish Crusta Algae Fish SPEC Fish Crusta Algae Crusta Algae	acea or other aquatic plants CIES acea or other aquatic plants acea or other aquatic plants		=8mg/L =42.5mg/L 2.5000mg/L 0.13mg/L VALUE 18.413mg/L 5.79mg/L >11.7mg/L 4.440mg/L 3.34mg/L	1 1 4 2 2 SOURCE 3 2 2 3 2 2
### Sulfuric acid EC50 EC50 NOE	50 4 50 2 50 7 50 9 50 9 50 4 50 7 50 3 EC 7	18 140 140 140 140 140 140 140 140 140 140	Crusta Algae Fish SPEC Fish Crusta Algae Crusta Algae SPECI	cites acea or other aquatic plants acea or other aquatic plants acea		=42.5mg/L 2.5000mg/L 0.13mg/L VALUE 18.413mg/L 5.79mg/L >11.7mg/L 4.440mg/L 3.34mg/L	1 4 2 SOURCE 3 2 2 3 2 2
tellurium	DPOINT T 50 9 50 3 EC 7. DPOINT T 50 9 50 4 50 7 50 3	2200 TEST DURATION (HR) 66 18 12 184 12 TEST DURATION (HR)	Algae Fish SPEC Fish Crusta Algae Crusta Algae	cites acea or other aquatic plants acea or other aquatic plants acea		2.5000mg/L 0.13mg/L VALUE 18.413mg/L 5.79mg/L >11.7mg/L 4.440mg/L 3.34mg/L	SOURCE 3 2 2 3 2
tellurium tellurium tellurium tellurium tellurium tellurium tellurium tellurium tellurium	DPOINT T 50 9 50 4 50 7 50 3 EC 7 DPOINT T 50 9 9 9 9 9 9 9 9 9 9 9	TEST DURATION (HR) 16 18 17 18 18 17 18 18 18 18 18 18 18 18 18 18 18 18 18	Fish SPEC Fish Crusta Algae Crusta Algae	acea or other aquatic plants acea or other aquatic plants		0.13mg/L VALUE 18.413mg/L 5.79mg/L >11.7mg/L 4.440mg/L 3.34mg/L	SOURCE 3 2 2 3 2 2
tellurium	DPOINT T 50 9 50 4 50 7 50 3 EC 7 DPOINT T 50 96	TEST DURATION (HR) 16 18 17 18 18 18 18 18 18 18 18 18 18 18 18 18	SPECE	acea or other aquatic plants acea or other aquatic plants		VALUE 18.413mg/L 5.79mg/L >11.7mg/L 4.440mg/L 3.34mg/L	SOURCE 3 2 2 2 3 3 2 2
tellurium	50 9 50 4 50 7 50 3 EC 7 DPOINT T 50 96	66 88 '22 884 '22 'EST DURATION (HR)	Fish Crusta Algae Crusta Algae SPECI	acea or other aquatic plants acea or other aquatic plants		18.413mg/L 5.79mg/L >11.7mg/L 4.440mg/L 3.34mg/L	3 2 2 2 3 2
tellurium	50 9 50 4 50 7 50 3 EC 7 DPOINT T 50 96	66 88 '22 884 '22 'EST DURATION (HR)	Fish Crusta Algae Crusta Algae SPECI	acea or other aquatic plants acea or other aquatic plants		18.413mg/L 5.79mg/L >11.7mg/L 4.440mg/L 3.34mg/L	3 2 2 2 3 2
tellurium EC50 EC50 NOE tin	50 4 50 7 50 3 50 7 DPOINT T	18 2 2 4 4 4 7 2 2 1 EST DURATION (HR)	Crusta Algae Crusta Algae	or other aquatic plants acea or other aquatic plants		5.79mg/L >11.7mg/L 4.440mg/L 3.34mg/L	2 2 3 2 2
tellurium EC56 EC56 NOE tin	50 7. 50 3 EC 7. DPOINT T 50 96	284 22 EST DURATION (HR)	Algae Crusta Algae	or other aquatic plants acea or other aquatic plants		>11.7mg/L 4.440mg/L 3.34mg/L	2 3 2
EC50 EC50 NOE	50 3 EC 7 DPOINT T 50 96	EST DURATION (HR)	Crusta Algae SPECI	acea or other aquatic plants		4.440mg/L 3.34mg/L	3 2
tin NOE	DPOINT T 50 96	EST DURATION (HR)	Algae	or other aquatic plants		3.34mg/L	2
tin EC50	DPOINT T	EST DURATION (HR)	SPECI				
tin LC50 EC50	50 96	, ,		ES		VALUE	SOURCE
tin LC50 EC50	50 96	, ,		ES		VALUE	SOURCE
tin EC50		6					
tin EC50	1		Fish			>0.0124mg/L	2
EC50	50 48	8	Crusta	cea		0.00018mg/L	5
	50 72	2	Algae	or other aquatic plants		>0.0192mg/L	2
EC50	50 72	2	Algae	or other aquatic plants		>0.0192mg/L	2
NOE	EC 16	68	Crustacea			<0.005mg/L	2
END	DPOINT T	EST DURATION (HR)	SPECIE	ES		VALUE	SOURCE
LC50	50 96	6	Fish			70.057mg/L	3
hydrochloric acid EC50	50 96	6	Algae o	or other aquatic plants		344.947mg/L	3
EC50	50 9.	.33	Fish			0.014000mg/L	4
NOE	EC 0.	.08	Fish	Fish		10mg/L	4
water	ENDPOINT TEST DURATION (HR)		SPECIES VALUE				SOURCE
Not A	Applicable	Not Applicable		Not Applicable	Not Applic	oplicable Not Applicable	

Ecotoxicity:

The tolerance of water organisms towards pH margin and variation is diverse. Recommended pH values for test species listed in OECD guidelines are between 6.0 and almost 9. Acute testing with fish showed 96h-LC50 at about pH 3.5

For Chloride: Although inorganic chloride ions are not normally considered toxic they can exist in effluents at acutely toxic levels. Incidental exposure to inorganic chloride may occur in occupational settings where chemicals management policies are improperly applied. The toxicity of chloride salts depends on the counter-ion (cation) present; that of chloride itself is unknown. Chloride toxicity has not been observed in humans except in the special case of impaired sodium chloride metabolism, e.g. in congestive heart failure. Healthy individuals can tolerate the intake of large quantities of chloride provided that there is an intake of fresh water following ingestion. Although excessive intake of drinking-water containing sodium chloride at concentrations above 2.5 g/L has been reported to produce hypertension, this effect is believed to be related to the sodium ion concentration. Chloride concentrations in excess of about 250 mg/L can give rise to detectable taste in water. Consumers can, however, become accustomed to concentrations in excess of 250 mg/L. No health-based guideline value is proposed for chloride in drinking-water. Chloride is almost completely absorbed in normal individuals. In metal pipes, chloride reacts with metal ions to form soluble salts thus increasing levels of metals in drinking-water. Chloride enhances galvanic corrosion in lead pipes and can also increase the rate of pitting corrosion of metal pipes.

Aquatic Fate: Inorganic chlorine eventually finds its way into aquatic systems and becomes bio-available. Chloride increases the electrical conductivity of water and thus increases its corrosivity. Ecotoxicity: When excessive inorganic chloride ions are introduced to aquatic environments, the resulting salinity can exceed the tolerances of most freshwater organisms.

Prevent, by any means available, spillage from entering drains or water courses.

DO NOT discharge into sewer or waterways

Persistence and degradability

Ingredient	Persistence: Water/Soil	Persistence: Air	
sodium carbonate	LOW	LOW	
tellurium	HIGH	HIGH	
hydrochloric acid	LOW	LOW	
water	LOW	LOW	

Bioaccumulative potential

Ingredient	Bioaccumulation
sodium carbonate	LOW (LogKOW = -0.4605)
tellurium	LOW (LogKOW = 2.229)

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hydrochloric acid	LOW (LogKOW = 0.5392)
water	LOW (LogKOW = -1.38)

Mobility in soil

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Ingredient	Mobility
sodium carbonate	HIGH (KOC = 1)
tellurium	LOW (KOC = 23.74)
hydrochloric acid	LOW (KOC = 14.3)
water	LOW (KOC = 14.3)

SECTION 13 DISPOSAL CONSIDERATIONS

Waste treatment methods

Legislation addressing waste disposal requirements may differ by country, state and/ or territory. Each user must refer to laws operating in their area. In some areas, certain wastes must be tracked.

A Hierarchy of Controls seems to be common - the user should investigate:

- ▶ Reduction
- ▶ Reuse
- ▶ Recycling
- ▶ Disposal (if all else fails)

Product / Packaging disposal

This material may be recycled if unused, or if it has not been contaminated so as to make it unsuitable for its intended use. If it has been contaminated, it may be possible to reclaim the product by filtration, distillation or some other means. Shelf life considerations should also be applied in making decisions of this type. Note that properties of a material may change in use, and recycling or reuse may not always be appropriate.

- ▶ DO NOT allow wash water from cleaning or process equipment to enter drains
- ▶ It may be necessary to collect all wash water for treatment before disposal.
- ▶ In all cases disposal to sewer may be subject to local laws and regulations and these should be considered first.
- Where in doubt contact the responsible authority.
- Recycle wherever possible.
- Consult manufacturer for recycling options or consult local or regional waste management authority for disposal if no suitable treatment or disposal facility can be identified.
- ► Treat and neutralise at an approved treatment plant. Treatment should involve: Neutralisation with soda-ash or soda-lime followed by: burial in a land-fill specifically licensed to accept chemical and / or pharmaceutical wastes or Incineration in a licensed apparatus (after admixture with suitable combustible material)
- ▶ Decontaminate empty containers with 5% aqueous sodium hydroxide or soda ash, followed by water. Observe all label safeguards until containers are cleaned and destroyed.

SECTION 14 TRANSPORT INFORMATION

Labels Required



Marine Pollutant

NC

Land transport (DOT)

UN number	1789
UN proper shipping name	Hydrochloric acid
Transport hazard class(es)	Class 8 Subrisk Not Applicable
Packing group	П
Environmental hazard	Not Applicable
Special precautions for user	Hazard Label 8 Special provisions 386, A3, A6, B3, B15, B133, IB2, N41, T8, TP2

Air transport (ICAO-IATA / DGR)

UN number	1789		
UN proper shipping name	Hydrochloric acid		
Transport hazard class(es)	ICAO/IATA Class ICAO / IATA Subrisk ERG Code	8 Not Applicable 8L	
Packing group	II		

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Environmental hazard Not Applicable Special provisions A3A803 Cargo Only Packing Instructions 855 Cargo Only Maximum Qty / Pack 30 L Special precautions for user Passenger and Cargo Packing Instructions 851 Passenger and Cargo Maximum Qty / Pack 1 L Passenger and Cargo Limited Quantity Packing Instructions Y840 Passenger and Cargo Limited Maximum Qty / Pack 0.5 L

Sea transport (IMDG-Code / GGVSee)

UN number	1789	
UN proper shipping name	HYDROCHLORIC ACID	
Transport hazard class(es)	IMDG Class 8 IMDG Subrisk Not Applicable	
Packing group	II .	
Environmental hazard	Not Applicable	
Special precautions for user	EMS Number F-A, S-B Special provisions Not Applicable Limited Quantities 1 L	

Transport in bulk according to Annex II of MARPOL and the IBC code

ANTIMONY(7440-36-0) IS FOUND ON THE FOLLOWING REGULATORY LISTS

Source	Product name	Pollution Category	Ship Type
IMO MARPOL (Annex II) - List of Noxious Liquid Substances Carried in Bulk	Hydrochloric acid	Z	3

SECTION 15 REGULATORY INFORMATION

Safety, health and environmental regulations / legislation specific for the substance or mixture

US - Alaska Limits for Air Contaminants	US - Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air	
US - California Permissible Exposure Limits for Chemical Contaminants	Contaminants	
US - Hawaii Air Contaminant Limits	US - Washington Permissible exposure limits of air contaminants	
US - Idaho - Limits for Air Contaminants	US - Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants	
US - Massachusetts - Right To Know Listed Chemicals	US ACGIH Threshold Limit Values (TLV)	
US - Michigan Exposure Limits for Air Contaminants	US Clean Air Act - Hazardous Air Pollutants	
US - Minnesota Permissible Exposure Limits (PELs)	US CWA (Clean Water Act) - Priority Pollutants	
US - Oregon Permissible Exposure Limits (Z-1)	US CWA (Clean Water Act) - Toxic Pollutants	
US - Pennsylvania - Hazardous Substance List	US EPCRA Section 313 Chemical List	
US - Rhode Island Hazardous Substance List	US NIOSH Recommended Exposure Limits (RELs)	
US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants	US OSHA Permissible Exposure Levels (PELs) - Table Z1	
US - Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants	US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory	
CALCIUM(7440-70-2) IS FOUND ON THE FOLLOWING REGULATORY LISTS		
US - Massachusetts - Right To Know Listed Chemicals	US - Rhode Island Hazardous Substance List	
US - Pennsylvania - Hazardous Substance List	US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory	
MAGNESIUM(7439-95-4) IS FOUND ON THE FOLLOWING REGULATORY LISTS		
International Agency for Research on Cancer (IARC) - Agents Classified by the IARC	US - Pennsylvania - Hazardous Substance List	
Monographs	US - Rhode Island Hazardous Substance List	
US - California OEHHA/ARB - Chronic Reference Exposure Levels and Target Organs	US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants	
(CRELs)	US - Washington Permissible exposure limits of air contaminants	
US - California Permissible Exposure Limits for Chemical Contaminants	US - Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants	
US - Hawaii Air Contaminant Limits	US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory	
US - Massachusetts - Right To Know Listed Chemicals		
US - Michigan Exposure Limits for Air Contaminants		
US - Oregon Permissible Exposure Limits (Z-1)		

\parallel SELENIUM(7782-49-2) IS FOUND ON THE FOLLOWING REGULATORY LISTS

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International Agency for Research on Cancer (IARC) - Agents Classified by the IARC Monographs

- US Alaska Limits for Air Contaminants
- US California OEHHA/ARB Acute Reference Exposure Levels and Target Organs (RELs)
- US California OEHHA/ARB Chronic Reference Exposure Levels and Target Organs (CRELs)
- US Hawaii Air Contaminant Limits
- US Idaho Limits for Air Contaminants
- US Massachusetts Right To Know Listed Chemicals
- US Minnesota Permissible Exposure Limits (PELs)
- US Pennsylvania Hazardous Substance List
- US Rhode Island Hazardous Substance List
- US Tennessee Occupational Exposure Limits Limits For Air Contaminants
- US Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants
- US Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air Contaminants

- US Washington Permissible exposure limits of air contaminants
- US Washington Toxic air pollutants and their ASIL, SQER and de minimis emission values
- US ACGIH Threshold Limit Values (TLV)
- US ATSDR Minimal Risk Levels for Hazardous Substances (MRLs)
- US Clean Air Act Hazardous Air Pollutants
- US CWA (Clean Water Act) Priority Pollutants
- US CWA (Clean Water Act) Toxic Pollutants
- US EPA Carcinogens Listing
- US EPCRA Section 313 Chemical List
- US NIOSH Recommended Exposure Limits (RELs)
- US OSHA Permissible Exposure Levels (PELs) Table Z1
- US Toxic Substances Control Act (TSCA) Chemical Substance Inventory

SODIUM CARBONATE(497-19-8) IS FOUND ON THE FOLLOWING REGULATORY LISTS

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

SULFURIC ACID(7664-93-9) IS FOUND ON THE FOLLOWING REGULATORY LISTS

International Agency for Research on Cancer (IARC) - Agents Classified by the IARC Monographs

International Air Transport Association (IATA) Dangerous Goods Regulations - Prohibited List Passenger and Cargo Aircraft

- US Alaska Limits for Air Contaminants
- US California OEHHA/ARB Acute Reference Exposure Levels and Target Organs (RELs)
- US California OEHHA/ARB Chronic Reference Exposure Levels and Target Organs
- US California Permissible Exposure Limits for Chemical Contaminants
- US Hawaii Air Contaminant Limits
- US Idaho Limits for Air Contaminants
- US Massachusetts Right To Know Listed Chemicals
- US Michigan Exposure Limits for Air Contaminants
- US Minnesota Permissible Exposure Limits (PELs)
- US New Jersey Right to Know Special Health Hazard Substance List (SHHSL):
- Carcinogens
- US Oregon Permissible Exposure Limits (Z-1)
- US Rhode Island Hazardous Substance List
- US Tennessee Occupational Exposure Limits Limits For Air Contaminants

- US Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants
- US Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air
- US Washington Permissible exposure limits of air contaminants
- US Washington Toxic air pollutants and their ASIL, SQER and de minimis emission values
- US Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants
- US ACGIH Threshold Limit Values (TLV)
- US ACGIH Threshold Limit Values (TLV) Carcinogens
- US CWA (Clean Water Act) List of Hazardous Substances
- US Drug Enforcement Administration (DEA) List I and II Regulated Chemicals
- US EPCRA Section 313 Chemical List
- US National Toxicology Program (NTP) 14th Report Part A Known to be Human Carcinogens
- US NIOSH Recommended Exposure Limits (RELs)
- US OSHA Permissible Exposure Levels (PELs) Table Z1
- US SARA Section 302 Extremely Hazardous Substances
- US Toxic Substances Control Act (TSCA) Chemical Substance Inventory

TELLURIUM(13494-80-9) IS FOUND ON THE FOLLOWING REGULATORY LISTS

- US Alaska Limits for Air Contaminants
- US California Permissible Exposure Limits for Chemical Contaminants
- US Hawaii Air Contaminant Limits
- US Idaho Limits for Air Contaminants
- US Massachusetts Right To Know Listed Chemicals
- US Michigan Exposure Limits for Air Contaminants
- US Minnesota Permissible Exposure Limits (PELs) US - Oregon Permissible Exposure Limits (Z-1)
- US Pennsylvania Hazardous Substance List
- US Rhode Island Hazardous Substance List

- US Tennessee Occupational Exposure Limits Limits For Air Contaminants
- US Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants
- US Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air
- US Washington Permissible exposure limits of air contaminants
- US Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants
- US ACGIH Threshold Limit Values (TLV)
- US NIOSH Recommended Exposure Limits (RELs)
- US OSHA Permissible Exposure Levels (PELs) Table Z1
- US Toxic Substances Control Act (TSCA) Chemical Substance Inventory

TIN(7440-31-5) IS FOUND ON THE FOLLOWING REGULATORY LISTS

- US Alaska Limits for Air Contaminants
- US California Permissible Exposure Limits for Chemical Contaminants
- US Hawaii Air Contaminant Limits
- US Idaho Limits for Air Contaminants
- US Massachusetts Right To Know Listed Chemicals
- US Minnesota Permissible Exposure Limits (PELs) US - Pennsylvania - Hazardous Substance List
- US Rhode Island Hazardous Substance List

- US Tennessee Occupational Exposure Limits Limits For Air Contaminants
- US Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants
- US Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air Contaminants
- US Washington Permissible exposure limits of air contaminants
- US NIOSH Recommended Exposure Limits (RELs)
- US OSHA Permissible Exposure Levels (PELs) Table Z1
- US Toxic Substances Control Act (TSCA) Chemical Substance Inventory

HYDROCHLORIC ACID(7647-01-0) IS FOUND ON THE FOLLOWING REGULATORY LISTS

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ICP Analytical Mixture 4

Issue Date: 06/03/2017 Print Date: 06/03/2017

International Agency for Research on Cancer (IARC) - Agents Classified by the IARC US - Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air Monographs Contaminants US - Alaska Limits for Air Contaminants US - Washington Permissible exposure limits of air contaminants US - California OEHHA/ARB - Acute Reference Exposure Levels and Target Organs (RELs) US - Washington Toxic air pollutants and their ASIL, SQER and de minimis emission values US - California OEHHA/ARB - Chronic Reference Exposure Levels and Target Organs US - Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants US ACGIH Threshold Limit Values (TLV) US - California Permissible Exposure Limits for Chemical Contaminants US ACGIH Threshold Limit Values (TLV) - Carcinogens US - Hawaii Air Contaminant Limits US Clean Air Act - Hazardous Air Pollutants US - Idaho - Limits for Air Contaminants US CWA (Clean Water Act) - List of Hazardous Substances US - Massachusetts - Right To Know Listed Chemicals US Drug Enforcement Administration (DEA) List I and II Regulated Chemicals US - Michigan Exposure Limits for Air Contaminants US EPCRA Section 313 Chemical List US - Minnesota Permissible Exposure Limits (PELs) US NIOSH Recommended Exposure Limits (RELs) US - Oregon Permissible Exposure Limits (Z-1) US OSHA Permissible Exposure Levels (PELs) - Table Z1 US - Pennsylvania - Hazardous Substance List US SARA Section 302 Extremely Hazardous Substances US - Rhode Island Hazardous Substance List US Spacecraft Maximum Allowable Concentrations (SMACs) for Airborne Contaminants US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory US - Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants WATER(7732-18-5) IS FOUND ON THE FOLLOWING REGULATORY LISTS

US - Pennsylvania - Hazardous Substance List

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

Federal Regulations

Superfund Amendments and Reauthorization Act of 1986 (SARA)

SECTION 311/312 HAZARD CATEGORIES

Immediate (acute) health hazard	Yes
Delayed (chronic) health hazard	No
Fire hazard	No
Pressure hazard	No
Reactivity hazard	No

US. EPA CERCLA HAZARDOUS SUBSTANCES AND REPORTABLE QUANTITIES (40 CFR 302.4)

Name	Reportable Quantity in Pounds (lb)	Reportable Quantity in kg
Antimony	5000	2270
Selenium	100	45.4
Sulfuric acid	1000	454
Hydrochloric acid	5000	2270

State Regulations

US. CALIFORNIA PROPOSITION 65

None Reported

National Inventory	Status
Australia - AICS	Y
Canada - DSL	Y
Canada - NDSL	N (calcium; magnesium; hydrochloric acid; water; antimony; selenium; sulfuric acid; tin; tellurium; sodium carbonate)
China - IECSC	Y
Europe - EINEC / ELINCS / NLP	Y
Japan - ENCS	N (calcium; magnesium; water; antimony; selenium; tin; tellurium; sodium carbonate)
Korea - KECI	Y
New Zealand - NZIoC	Y
Philippines - PICCS	Y
USA - TSCA	Y
Legend:	Y = All ingredients are on the inventory N = Not determined or one or more ingredients are not on the inventory and are not exempt from listing(see specific ingredients in brackets)

SECTION 16 OTHER INFORMATION

Other information

Ingredients with multiple cas numbers

Name	CAS No
calcium	7440-70-2, 8047-59-4
sodium carbonate	497-19-8, 7542-12-3, 1314087-39-2, 1332-57-6

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references

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The SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

Definitions and abbreviations

PC — TWA: Permissible Concentration-Time Weighted Average PC — STEL: Permissible Concentration-Short Term Exposure Limit

IARC: International Agency for Research on Cancer

ACGIH: American Conference of Governmental Industrial Hygienists

STEL: Short Term Exposure Limit

TEEL: Temporary Emergency Exposure Limit。

IDLH: Immediately Dangerous to Life or Health Concentrations

OSF: Odour Safety Factor

NOAEL :No Observed Adverse Effect Level LOAEL: Lowest Observed Adverse Effect Level

TLV: Threshold Limit Value LOD: Limit Of Detection OTV: Odour Threshold Value BCF: BioConcentration Factors BEI: Biological Exposure Index

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TEL (+61 3) 9572 4700.