

## Safety Data Sheet

### Section 1. Product and Company Identification

Product Identification: 1500 µg Pd/mL- 1000 µg Mg(NO<sub>3</sub>)<sub>2</sub>/mL in 10% HNO<sub>3</sub> + Tr HCl  
SDS Number: MM-9101  
Recommended Use: For Laboratory Use.  
Company Identification: High-Purity Standards  
P.O. Box 41727  
Charleston, SC 29423  
Telephone: (843) 767-7900  
FAX: (843) 767-7906

In case of emergency call INFOTRAC: 800-535-5053

### Section 2. Hazard Identification

Classification:

**Skin Corrosion/Irritation, Category 1**

**Serious Eye Damage/ Eye Irritation, Category 1**

Labeling:



Symbol:

Signal Word: **Danger.**

Hazard Statement: **Causes severe skin burns and eye damage.**

Precautionary Statement: **Wear protective gloves/clothing and eye/face protection. Wash thoroughly after handling.**

### Section 3. Composition

Component	CAS/EINECS Registry #	Percent Concentration
Palladium	7440-05-3/ 231-115-6	0.15
Magnesium Nitrate Hexahydrate (Mg(NO <sub>3</sub> ) <sub>2</sub> · 6H <sub>2</sub> O)	13446-18-9/233-826-7	0.1
Nitric Acid	7697-37-2/231-714-2	10
Hydrochloric Acid	7647-01-0/231-595-7	<0.01
Water, deionized	7732-18-5/231-791-2	Balance

### Section 4. First Aid Measures

**IF ON SKIN (or hair):** Remove/Take off immediately all contaminated clothing. Rinse skin with water/shower. Wash contaminated clothing before reuse. Call a physician if irritation develops.

**IF IN EYES:** Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.

**IF SWALLOWED:** Rinse mouth. Do NOT induce vomiting. Call a physician. May cause nausea, vomiting, and diarrhea.

**IF INHALED:** Remove to fresh air and keep at rest in a position comfortable for breathing.

**Target Organs:** Eyes, skin.

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#### Section 5. Fire Fighting Measures

Fire & Explosion hazards: While nitric acid is not combustible, it is a strong oxidizing agent that can react with combustible materials. NO<sub>x</sub> compounds can be released in event of fire.

Extinguishing Media: Use any extinguishing media that is suitable for the surrounding area. Use a water spray to dilute nitric acid and to absorb liberated nitrogen oxides.

Specific Methods: Firefighters should wear proper protective equipment and self-contained breathing apparatus with full face piece operated in positive pressure mode

#### Section 6. Accidental Release Measures

Ventilate area of leak or spill. Wear appropriate personal protective equipment as specified in Section 8. Remove source of ignition if hydrogen is a hazard. Cover the spill with sodium bicarbonate or a soda ash-slaked lime mixture (50:50) to neutralize the acid. Place the neutralized material into containers suitable for eventual disposal, reclamation, or destruction. Always dispose of in accordance with local regulations.

#### Section 7. Handling and Storage

Store in a cool, dry, ventilated storage area with acid resistant floors and good drainage. Keep out of direct sunlight and away from heat, water, and incompatible materials. When diluting, the acid should always be added slowly to water and in small amounts. Refer to Section 8 for personal handling instructions.

#### Section 8. Exposure Controls and Personal Protection

Engineering Controls: Provide general and local (e.g., fume hood) ventilation systems to maintain airborne concentrations below the TLV. Local exhaust ventilation is preferred because it can control contaminant emissions at the source, preventing dispersion into the general work area. Ensure the availability of eyewash stations and safety showers.

Respiratory Protection: Provide approved respiratory apparatus for non-routine or emergency use. Use an approved filter and vapor respirator when the vapor or mist concentrations are high. If necessary, refer to the NIOSH document Guide to the Selection and Use of Particulate Respirators Certified under 42 CFR 84 for selection and use of respirators certified by NIOSH.

Personal Protection: Wear proper gloves, safety glasses with side shields, lab coat/apron.

##### Exposure Limits:

Component	ACGIH TLV	OSHA PEL
Palladium	Not Available	Not Available
Magnesium Nitrate Hexahydrate	Not Available	Not Available
Nitric Acid	2 mg/kg	5 mg/m <sup>3</sup>
Hydrochloric Acid	C 5ppm C 7 mg/m <sup>3</sup>	C 5ppm C 7 mg/m <sup>3</sup>

#### Section 9. Physical and Chemical Properties

Appearance: Clear, dark brown liquid

Physical State: Liquid

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Odor: Odorless to a faint pungent odor

Odor threshold: None

pH: <2

Melting point: N/A

Freezing Point: N/A

Boiling Point: Approximately 100°C

Flash point: N/A

Evaporation rate: N/A

Flammability: N/A

Explosion limits: N/A

Vapor Pressure (mm): N/A

Vapor Density (air+1): N/A

Relative density: (H<sub>2</sub>O = 1): Approximately 1.0

Solubility in H<sub>2</sub>O: Complete

Auto ignition temperature: N/A

Decomposition temperature: N/A

Molecular Weight: N/A

#### Section 10. Stability and Reactivity

Stability Indicator: YES

Conditions to Avoid: Metals, chlorine, organic materials, strong alkali, cyanides.

Incompatibles: Organic materials, strong reducing agents.

Hazardous Decomposition Products: NO<sub>x</sub> compounds including nitric oxide (NO), nitrogen dioxide (NO<sub>2</sub>), nitrous oxide (N<sub>2</sub>O) and nitric acid mist or vapor.

Hazardous Polymerization: None expected

#### Section 11. Toxicological Information

May affect skin, mucous membranes and eyes. Swallowing may lead to a negative effect on mouth and throat and to the risk of perforation or the corrosion of esophagus and stomach.

**RTECS# :**

HNO<sub>3</sub>-QU5775000 Pd-RT3480500 Mg(NO<sub>3</sub>)<sub>2</sub>-OM3756000 HCl-MW4025000

**Toxicity Data:**

LD<sub>LO</sub> Oral, Human: (Nitric Acid) 430 mg/kg; LD<sub>50</sub> Oral, Rat: (Mg(NO<sub>3</sub>)<sub>2</sub>) 5440 mg/kg. LD<sub>50</sub>

Oral, Rabbit: (Hydrochloric Acid) 900 mg/kg.

#### Section 12. Ecological Information

Ecotoxicological information: Do not allow material to reach ground water, water bodies, or sewage system.

#### Section 13. Disposal Considerations

General: Follow federal, state and local regulations for waste.

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#### Section 14. Transport Information

D.O.T. Classification: Hazardous by IATA and 49CFR regulations (based on concentration of acid).

D.O.T. Shipping Name: Corrosive liquid, Acidic, Inorganic, n.o.s. (Nitric Acid Solution)

D.O.T. Hazard Class: 8

U.N./N.A. Number: 3264

Packing Group: II

D.O.T. Label: Corrosive (8)

#### Section 15. Regulations (Not meant to be all inclusive-selected regulation listed)

TSCA Status: Components of this solution are listed on the TSCA Inventory.

RCRA Status: No.

SARA: Subject to the reporting requirements of Section 313 of SARA Title III and of 40 CFR 372

Risk Phrases: R23/24/25, R34 Toxic by inhalation, skin contact, or ingestion. Causes burns.

Safety Phrases: S36/37/39 Wear suitable protective clothing, gloves and eye/face protection

WHMIS Information (Canada): E: Corrosive

#### Section 16. Other Information

HPS products are intended for laboratory use only. All products should be handled and used by trained professional personnel only. The responsibility for the safe handling and use of these products rests solely with the buyer and/or user. The MSDS was prepared carefully and represents the best data currently available to us; however, HPS does not certify the data on the MSDS. Certified values for this material are given only on the Certificate of Analysis.

Theodore C. Rains, Ph.D.