# Section 1. Product and Company Identification

Product Identification: ANALCS-R MSDS Number: ANALCS-R

Company Identification: High-Purity Standards

P.O. Box 41727

Charleston, SC 29423 Telephone: (843) 767-7900 FAX: (843) 767-7906

In case of emergency call INFOTRAC: 800-535-5053

### Section 2. Hazard Identification

#### **Classification:**

Skin Corrosion/Irritation, Category 1

Serious Eye Damage/ Eye Irritation, Category 1

Labeling:

Symbol:

Signal Word: Danger.

**Hazard Statement:** Causes severe skin burns and eye damage.

**Precautionary Statement:** Wear protective gloves/clothing and eye/face protection. Wash

thoroughly after handling.

Section 3. Hazard Identification			
Component	CAS/EINECS Registry #	Percent Concentration	
Antimony	7440-36-0/231-146-5	0.006	
Arsenic	7440-38-2/231-148-6	0.001	
Barium Carbonate (BaCO <sub>3</sub> )	513-77-9/208-167-3	0.005 ( D )	
Barium Nitrate (Ba(NO <sub>3</sub> ) <sub>2</sub> )	10022-31-8/233-020-5	0.005 (as Ba)	
Beryllium Acetate (Be <sub>4</sub> O(C <sub>2</sub> H <sub>3</sub> O <sub>2</sub> ) <sub>6</sub> )	19049-40-2/242-785-4	0.005 (as Be)	
Cadmium	7440-43-9/231-152-8	0.01	
Chromium	7440-47-3/231-157-5	0.005	
Cobalt	7440-48-4/231-158-0	0.005	
Copper	7440-50-8/231-159-6	0.005	
Lead	7439-92-1/231-100-4	<0.001	
Manganese Acetate Tetrahydrate (Mn(CH <sub>3</sub> CO <sub>2</sub> ) <sub>2</sub> )*4H <sub>2</sub> O	6156-78-1/211-334-3	0.1 (as Mn)	
Nickel	7440-02-0/231-111-4	0.01	
Selenium	7782-49-2/231-957-4	< 0.001	
Silver	7440-22-4/231-131-3	0.002	
Thallium	7440-28-0/231-138-1	0.001	
Ammonium Metavanadate (NH <sub>4</sub> VO <sub>3</sub> )	7803-55-6/232-261-3	0.005 (as V)	
Zinc	7440-66-6/231-175-3	0.01	
Nitric Acid	7697-37-2/231-714-2	2	
Hydrofluoric Acid	7664-39-3/231-634-8	< 0.001	
Water, deionized	7732-18-5/231-791-2	Balance	

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<sup>\*</sup>Note: Barium is derived from either Barium carbonate or Barium Nitrate. For this reason both sources are listed on the SDS. Refer to the product's Certificate of Analysis to determine which source was used in the production of the lot number received.

### Section 4. First Aid Measures

**IF ON SKIN (or hair):** Remove/Take off immediately all contaminated clothing. Gently wash with plenty of soap and water. Rub calcium gluconate gel immediately to skin. Obtain medical assistance. Wash contaminated clothing before reuse.

**IF IN EYES:** Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing. Immediately call a POISON CENTER doctor/physician.

**IF SWALLOWED:** Rinse mouth. Do NOT induce vomiting.

**IF INHALED:** Remove to fresh air and keep at rest in a position comfortable for breathing.

Target Organs: Eyes, skin, respiratory system, teeth, and skeletal system.

### Section 5. Fire Fighting Measures

Fire & Explosion hazards: While nitric acid is not combustible, it is a strong oxidizing agent that can react with combustible materials; however, it is present in limited quantities in this solution. NO<sub>x</sub> compounds can be released in case of fire. Hydrofluoric acid may ignite or explode on contact with combustible materials.

Extinguishing Media: Use any extinguishing media that is suitable for the surrounding area. Use a water spray to dilute nitric acid and to absorb liberated nitrogen oxides.

Specific Methods: Firefighters should wear proper protective equipment and self-contained breathing apparatus with full face piece operated in positive pressure mode.

#### Section 6. Accidental Release Measures

Ventilate area of leak or spill. Wear appropriate personal protective equipment as specified in Section 8. Cover the spill with sodium bicarbonate or a soda ash-slaked lime mixture (50:50) to neutralize the acid. Place the neutralized material into containers suitable for eventual disposal, reclamation, or destruction. Always dispose of in accordance with local regulations.

# Section 7. Handling and Storage

Store in a cool, dry, ventilated storage area with acid resistant floors and good drainage. Keep out of direct sunlight and away from heat, water, and incompatible materials. When diluting, the acid should always be added slowly to water and in small amounts. Refer to Section 8 for personal handling instructions. Wash exposed skin area thoroughly after handling.

### Section 8. Exposure Controls and Personal Protection

Engineering Controls: Provide exhaust ventilation or other engineering controls to keep any buildup of airborne contaminants below their respective threshold limit value. Ensure the availability of eyewash stations and safety showers.

Personal Protection: Wear proper gloves, safety glasses with side shields, lab coat/apron.

## **Exposure Limits:**

Component	ACGIH TLV	OSHA PEL
Antimony	$0.5 \text{ mg/m}^3$	$0.5 \text{ mg/m}^3$
Arsenic	$0.01 \text{ mg/m}^3$	$10 \mu\text{g/m}^3$

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Barium	$0.5 \text{ mg/m}^3$	$0.5 \text{ mg/m}^3$
Beryllium Acetate	$0.002 \text{ mg/m}^3$	$0.002 \text{ mg/m}^3$
Cadmium	0.002 mg/m <sup>3</sup> (respirable	$0.005 \text{ mg/m}^3$
	particulate)	
Chromium	$0.5 \text{ mg/m}^3$	$1 \text{ mg/m}^3$
Cobalt	$0.02 \text{ mg/m}^3$	$0.1 \text{ mg/m}^3$
Copper	$0.2 \text{ mg/m}^3 \text{ (fumes)}$	$0.1 \text{ mg/m}^3 \text{ (fumes)}$
Lead	$0.05 \text{ mg/m}^3$	$0.05 \text{ mg/m}^3$
Manganese Acetate Tetrahydrate	$0.2 \text{ mg/m}^3$	C 5 mg/m <sup>3</sup>
Nickel	$1.5 \text{ mg/m}^3$	$1 \text{ mg/m}^3$
Selenium	$0.2 \text{ mg/m}^3$	$0.2 \text{ mg/ m}^3$
Silver	$0.1 \text{ mg/m}^3$	Not Available
Thallium	$0.1 \text{ mg/m}^3$	$0.1 \text{ mg/m}^3$
Ammonium Metavanadate	$0.05 \text{ mg/m}^3$	Not Available
Zinc	$5 \text{ mg/m}^3$	$1 \text{ mg/m}^3$
Nitric Acid	2 mg/kg	5 mg/m <sup>3</sup>
Hydrofluoric Acid	C: 3 mg/ml	$2.5 \text{ mg/m}^3$
		STEL: 6 mg/ml
Water, deionized	Not Available	Not Available

# Section 9. Physical and Chemical Properties

Physical State: Liquid Color: Clear, colorless liquid

Odor: Odorless to a faint pungent odor

Odor threshold: None

pH: <2

Melting point: N/A Freezing Point: N/A

Boiling Point: Approximately 100°C

Flash point: N/A
Evaporation rate: N/A
Flammability: N/A
Explosion limits: N/A
Vapor Pressure (mm): N/A
Vapor Density (air+1): N/A

Relative density:  $(H_2O = 1)$ : Approximately 1.0

Solubility in H<sub>2</sub>O: Complete Auto ignition temperature: N/A Decomposition temperature: N/A

Molecular Weight: N/A

### Section 10. Stability and Reactivity

Stability Indicator: YES

Conditions to Avoid: Avoid heat and contact with combustible and other incompatible materials. Incompatibles: Strong reducing agents, metallic powders, strong bases, chlorine, calcium compounds, hydroxides, organic materials, strong alkali, cyanides.

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Hazardous Decomposition Products: HF and NO<sub>x</sub> compounds including nitric oxide (NO), nitrogen dioxide (NO<sub>2</sub>), nitrous oxide (N<sub>2</sub>O) and nitric acid mist or vapor. Hazardous Polymerization: Will not occur.

## Section 11. Toxicological Information

May affect skin, mucous membranes and eyes. Swallowing may lead to a negative effect on mouth and throat and to the risk of perforation or the corrosion of esophagus and stomach. RTECS#

HNO<sub>3</sub>- OU5775000 HF- MW7875000 BaCO<sub>3</sub>- CQ8600000 Sb-CC4025000 As- CG0525000 Ba(NO<sub>3</sub>)<sub>2</sub>- CQ9625000 Be-DS1750000 Cd- EU9800000 Co- GF8750000 Cr- GB4200000 Cu-GL5325000 Pb- OF7525000 Mn- AI5775000 Ni- OR5950000 Tl- XG3425000 Se- VS7700000 Ag- VW3500000 V- YW0875000 Zn- ZG8600000

LD<sub>LO</sub> Oral, Human: (Nitric Acid) 430 mg/kg;

LC<sub>LO</sub> Inhalation, Human: (Hydrofluoric Acid) 50 mg/kg/30 min;

LD<sub>50</sub> Oral, Rat: (Antimony) 7g/kg; LD<sub>50</sub>, Oral, Rat: (Arsenic) 763 mg/kg;

LD<sub>LO</sub> Oral, Human: (Barium Carbonate) 17 mg/kg;

LD<sub>50</sub> Oral, Rat: (Ba(NO<sub>3</sub>)<sub>2</sub>) 355 mg/kg.

TD<sub>LO</sub> Intratracheal, Rat: (Beryllium Acetate) 13 mg/kg;

LD<sub>LO</sub> Oral, Human: (Cadmium) 2330 mg/kg;

LD<sub>50</sub> Unreported Route, Rat: (Chromium) 27.5 mg/kg;

LD<sub>LO</sub> Oral, Rabbit: (Cobalt) 750 mg/kg; TD<sub>LO</sub> Oral, Human: (Copper) 120 μg/kg;

TD<sub>50</sub> Oral, Woman: (Lead) 450 mg/kg/6 year;

LD<sub>50</sub> Oral, Rat: (Manganese) 3730mg/kg;

LD<sub>50</sub>, Intravenous, Mouse: (Nickel) 50 mg/kg;

LD<sub>50</sub>, Oral, Rat: (Selenium) 6700 mg/kg;

TD<sub>LO</sub> Implant, Mouse: (Silver) 11 g/kg; LD<sub>50;</sub>

TD<sub>LO</sub> Oral, Man: (Thallium) 5,714 μg/kg;

LD<sub>50</sub> Oral, Rat: (Ammonium Metavanadate) 58,100 μg/kg;

LD<sub>LO</sub> Oral, Duck: (Zinc) 388 mg/kg.

# Section 12. Ecological Information

Ecotoxicological information: Do not allow material to reach ground water, water bodies, or sewage system. High concentrations of zinc have been shown to be detrimental to aquatic life. Beryllium and its compounds are considered to have high acute and chronic toxicity to aquatic life. Beryllium is more toxic in soft water than in hard water.

### Section 13. Disposal Considerations

Follow federal, state and local regulations for acid waste.

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### Section 14. Transport Information

D.O.T. Classification: Hazardous by IATA and 49CFR regulations (based on concentration of acid).

D.O.T. Shipping Name: Corrosive liquid, Acidic, Inorganic, n.o.s. (Nitric Acid Solution)

D.O.T. Hazard Class: 8 U.N./N.A. Number: 3264

Packing Group: II

D.O.T. Label: Corrosive (8)

## Section 15. Regulations (Not meant to be all inclusive-selected regulation listed)

TSCA Status: Components of this solution are listed on the TSCA Inventory.

RCRA Status: Yes (Hydrofluoric Acid-U134)

SARA: Subject to the reporting requirements of Section 313 of SARA Title III and of 40 CFR

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Risk Phrases: R20/21/22, R24, R25, R34 Harmful by inhalation, skin contact, or if ingested.

Toxic in contact with skin. Toxic if swallowed. Causes burns.

Safety Phrases: S24, S25, S36/37/39 Avoid contact with the skin. Avoid contact with eyes. Wear

suitable protective clothing, gloves and eye/face protection

WHMIS Information (Canada): E: Corrosive

### Section 16. Other Information

HPS products are intended for laboratory use only. All products should be handled and used by trained professional personnel only. The responsibility for the safe handling and use of these products rests solely with the buyer and/or user. The SDS was prepared carefully and represents the best data currently available to us; however, HPS does not certify the data on the SDS. Certified values for this material are given only on the Certificate of Analysis.

Theodore C. Rains, Ph.D.