Section 1. Product and Company Identification

Product Identification: ICP-200.7-10
MSDS Number: ICP-200.7-10
Recommended Use: For Laboratory Use.
Company Identification: High-Purity Standards
P.O. Box 41727
Charleston, SC 29423
Telephone: (843) 767-7900
FAX: (843) 767-7906
In case of emergency call INFOTRAC: 800-535-5053

Section 2. Hazard Identification

Classification:
Skin Corrosion/Irritation, Category 1
Serious Eye Damage/ Eye Irritation, Category 1
Labeling:

Symbol:
Signal Word: Danger.
Hazard Statement: Causes severe skin burns and eye damage.
Precautionary Statement: Wear protective gloves/clothing and eye/face protection. Wash thoroughly after handling.

Section 3. Composition

Component	CAS/EINECS Registry #	Percent Concentration
Aluminum	7429-90-5/231-072-3	0.02
Arsenic	7440-38-2/231-148-6	0.02
Barium Nitrate ($Ba(NO_3)_2$)	10022-31-8/233-020-5	0.02 (as Ba)
Barium Carbonate ($BaCO_3$)	513-77-9/208-167-3	
Beryllium Acetate ($Be_4O(C_2H_3O_2)_6$)	19049-40-2/242-785-4	0.02 (as Be)
Boric Acid (H_3BO_3)	10043-35-3/233-139-2	0.02 (as B)
Cadmium	7440-43-9/231-152-8	0.02
Calcium Carbonate ($CaCO_3$)	471-34-1/207-439-9	0.02 (as Ca)
Cerium Oxide (CeO_2)	1306-38-3/215-150-4	0.02 (as Ce)
Chromium	7440-47-3/231-157-5	0.02
Cobalt	7440-48-4/231-158-0	0.02
Copper	7440-50-8/231-159-6	0.02
Iron	7439-89-6/231-096-4	0.02
Lead	7439-92-1/231-100-4	0.02

Lithium Carbonate (Li_2CO_3)	554-13-2/209-062-5	0.02 (as Li)
Magnesium	7439-95-4/231-104-6	0.02
Manganese	7439-96-5/231-105-1	0.02
Mercury	7439-97-6/231-106-7	0.02
Nickel	7440-02-0/231-111-4	0.02
Ammonium Dihydrogen Phosphate ($NH_4H_2PO_4$)	7722-76-1/231-764-5	0.1 (as P)
Potassium Nitrate (KNO_3)	7757-79-1/231-818-8	0.1 (as K)
Selenium	7782-49-2/231-957-4	0.02
Thallium	7440-28-0/231-138-1	0.02
Ammonium Metavanadate (NH_4VO_3)	7803-55-6/232-261-3	0.02 (as V)
Silver	7440-22-4/231-131-3	0.0025
Zinc	7440-66-6/231-175-3	0.02
Nitric Acid (HNO_3)	7697-37-2/231-714-2	5
Water, deionized	7732-18-5/231-791-2	Balance

**Note: Barium is derived from either Barium carbonate or Barium Nitrate. For this reason both sources are listed on the SDS. Refer to the product's Certificate of Analysis to determine which source was used in the production of the lot number received.*

Section 4. First Aid Measures

IF ON SKIN (or hair): Remove/Take off immediately all contaminated clothing. Rinse skin with water/shower. Wash contaminated clothing before reuse. Call a physician if irritation develops.

IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.

IF SWALLOWED: Rinse mouth. Do NOT induce vomiting. Call a physician. May cause nausea, vomiting, and diarrhea.

IF INHALED: Remove to fresh air and keep at rest in a position comfortable for breathing.

Target Organs: Eyes, skin.

Section 5. Fire Fighting Measures

Fire & Explosion hazards: While nitric acid is not combustible, it is a strong oxidizing agent that can react with combustible materials; however, it is present in limited quantities in this solution. NO_x compounds can be released in case of fire.

Extinguishing Media: Use any extinguishing media that is suitable for the surrounding area. Use a water spray to dilute nitric acid and to absorb liberated nitrogen oxides.

Specific Methods: Firefighters should wear proper protective equipment and self-contained breathing apparatus with full face piece operated in positive pressure mode.

Section 6. Accidental Release Measures

Ventilate area of leak or spill. Wear appropriate personal protective equipment as specified in Section 8. Do not allow to enter drainage systems or water ways. Dike area and dilute spill with water and neutralize with soda ash, limestone, etc. Place the neutralized material into containers suitable for eventual disposal, reclamation, or destruction. Always dispose of in accordance with local regulations.

Section 7. Handling and Storage

Store in a cool, dry, ventilated storage area with acid resistant floors and good drainage. Keep out of direct sunlight and away from heat, water, and incompatible materials. When diluting, the acid should always be added slowly to water and in small amounts. Refer to Section 8 for personal handling instructions.

Section 8. Exposure Controls and Personal Protection

Engineering Controls: Provide exhaust ventilation or other engineering controls to keep any buildup of airborne contaminants below their respective threshold limit value. Ensure the availability of eyewash stations and safety showers.

Respiratory Protection: Provide approved respiratory apparatus for non-routine or emergency use. Use an approved vapor respirator when the vapor or mist concentrations are high. If necessary, refer to the NIOSH document Guide to the Selection and Use of Particulate Respirators Certified under 42 CFR 84 for selection and use of respirators certified by NIOSH.

Personal Protection: Wear proper gloves, safety glasses with side shields, lab coat/apron.

Exposure Limits:

Component	ACGIH TLV	OSHA PEL
Aluminum	10 mg/m^3	15 mg/m^3
Arsenic	0.01 mg/m^3	10 $\mu g/ m^3$
Barium	0.5 mg/m^3	0.5 mg/m^3
Beryllium Acetate	0.002 mg/m^3	0.002 mg/m^3
Boric Acid	Not Available	Not Available
Cadmium	0.002 mg/m^3 (respirable particulate)	0.005 mg/m^3
Calcium Carbonate	0.5 mg/m^3	0.5 mg/m^3
Cerium Oxide	Not Available	Not Available
Chromium	0.5 mg/m^3	1 mg/m^3
Cobalt	0.02 mg/m^3	0.1 mg/m^3
Copper	0.2 mg/m^3 (fumes)	0.1 mg/m^3 (fumes)
Iron	10 mg/m^3	5 mg/m^3
Lead	0.05 mg/m^3	0.05 mg/m^3
Lithium Carbonate	Not Available	Not Available
Magnesium	Not Available	Not Available
Manganese	0.2 mg/m^3	C 5 mg/m^3
Mercury	0.05 mg/m^3	0.025 mg/m^3
Nickel	1.5 mg/m^3	1 mg/m^3
Ammonium Dihydrogen Phosphate	Not Available	Not Available
Potassium Nitrate	Not Available	Not Available
Selenium	0.2 mg/m^3	0.2 mg/ m^3

Thallium	0.1 mg/m^3	0.1 mg/m^3
Ammonium Metavanadate	0.05 mg/m^3	Not Available
Silver	0.1 mg/m^3	Not Available
Zinc	5 mg/m^3	1 mg/m^3
Nitric Acid	2 mg/kg	5 mg/m^3
Water, deionized	Not Available	Not Available

Section 9. Physical and Chemical Properties

Physical State: Liquid
Color: Clear, light brown to grey liquid
Odor: Odorless to a faint pungent odor
Odor threshold: None
pH: <2
Melting point: N/A
Freezing Point: N/A
Boiling Point: Approximately 100°C
Flash point: N/A
Evaporation rate: N/A
Flammability: N/A
Explosion limits: N/A
Vapor Pressure (mm): N/A
Vapor Density (air+1): N/A
Relative density: (H_2O = 1): Approximately 1.0
Solubility in H_2O: Complete
Auto ignition temperature: N/A
Decomposition temperature: N/A
Molecular Weight: N/A

Section 10. Stability and Reactivity

Stability Indicator: Decomposes slowly to release oxygen.
Conditions to Avoid: Metals, chlorine, organic materials, strong alkali, cyanides.
Incompatibles: Strong reducing agents.
Hazardous Decomposition Products: NO_x compounds including nitric oxide (NO), nitrogen dioxide (NO_2), nitrous oxide (N_2O) and nitric acid mist or vapor.
Hazardous Polymerization: Will not occur.

Section 11. Toxicological Information

May affect skin, mucous membranes and eyes. Swallowing may lead to a negative effect on mouth and throat and to the risk of perforation or the corrosion of esophagus and stomach.

RTECS#

HNO_3- QU5775000	Al- BD0330000
As- CG0525000	$Be_4O(C_2H_3O_2)_6$ – DS2900000
H_3BO_3- ED4560000	Cd- EU9800000
$CaCO_3$- EV9580000	CeO_2- FK6310000
Co- GF8750000	Cr- GB4200000
Cu- GL5325000	Fe- NO4565500
Pb- OF7525000	Li_2CO_3- OJ5800000

Mg- FW6475100
Hg- OV4550000
KNO_3- TT3700000
Tl- XG3425000
Zn- ZG8600000
$BaCO_3$; CQ8600000
Mn- OO9275000
Ni- QR5950000
Se- VS7700000
NH_4VO_3- YW0875000
Ag- VW3500000
$Ba(NO_3)_2$- CQ9625000

LD_{LO} Oral, Human: (Nitric Acid) 430 mg/kg
LD_{50} Oral, Rat: (Aluminum) >5000 mg/kg
LD_{50} Oral, Rat: (Arsenic) 763 mg/kg
LD_{50} Oral, Rat: (Barium Nitrate) 355 mg/kg
LD_{LO} Oral, Human: (Barium Carbonate) 17 mg/kg.
TD_{LO} Intratracheal, Rat: (Beryllium Acetate) 13 mg/kg
LD_{50} Oral, Rat: (Boric Acid) 2660 mg/kg
LD_{50} Oral, Rat: (Calcium Carbonate) 6450 mg/kg
LD_{LO} Oral, Human: (Cadmium) 2330 mg/kg
LD_{50} Oral, Rat: (Cerium Oxide) >5 g/kg
LD_{50} Unreported Route, Rat: (Chromium) 27.5 mg/kg
LD_{LO} Oral, Rabbit: (Cobalt 750 mg/kg
TD_{LO} Oral, Human: (Copper) 120 μg/kg
LD_{50} Oral, Rat: (Iron) 30 g/kg
LD_{50} Oral, Rat: (Lithium Carbonate) 525 mg/kg
LD_{50} Oral, Rat: (Potassium Nitrate) 3750 mg/kg
LD_{50} Oral, Rat: (Selenium) 6700 mg/kg
LD_{50} Oral, Rat: (Manganese) 9 g/kg
LD_{50}, Intravenous; Mouse: (Nickel) 50 mg/kg
TD_{LO} Oral, Man: (Thallium) 5,714 μg/kg
LD_{50} Oral, Mouse: (Ammonium Metavanadate) 25 mg/kg

Section 12. Ecological Information

Ecotoxicological information: Do not allow material to reach ground water, water bodies, or sewage system. Beryllium and its compounds are considered to have high acute and chronic toxicity to aquatic life. Beryllium is more toxic in soft water than in hard water.

Section 13. Disposal Considerations

General: Follow Federal, state and local regulations for waste.

Section 14. Transport Information

D.O.T. Classification: Hazardous by IATA and 49CFR regulations (based on concentration of acid).
D.O.T. Shipping Name: Corrosive liquid, Acidic, Inorganic, n.o.s. (Nitric Acid Solution)
D.O.T. Hazard Class: 8
U.N./N.A. Number: 3264
Packing Group: II
D.O.T. Label: Corrosive (8)

Section 15. Regulations (Not meant to be all inclusive-selected regulation listed)

TSCA Status: Components of this solution are listed on the TSCA Inventory.
RCRA: Status: 7439-97-6 (Mercury), 7803-55-6 (Ammonium Metavanadate)
SARA: Subject to the reporting requirements of Section 313 of SARA Title III and of 40 CFR 372
Risk Phrases: R20/21/22 Harmful by inhalation, skin contact, or if swallowed.
Safety Phrases: S36/37/39 Wear suitable protective clothing, gloves and eye/face protection
WHMIS Information (Canada): E: Corrosive

Section 16. Other Information

HPS products are intended for laboratory use only. All products should be handled and used by trained professional personnel only. The responsibility for the safe handling and use of these products rests solely with the buyer and/or user. The SDS was prepared carefully and represents the best data currently available to us; however, HPS does not certify the data on the SDS. Certified values for this material are given only on the Certificate of Analysis.

Theodore C. Rains, Ph.D.