

Safety Data Sheet

Section 1. Product and Company Identification

Product Identification: CRM-BL
 MSDS Number: CRM-BL
 Recommended Use: For Laboratory Use.
 Company Identification: High-Purity Standards
 P.O. Box 41727
 Charleston, SC 29423
 Telephone: (843) 767-7900
 FAX: (843) 767-7906

In case of emergency call INFOTRAC: 800-535-5053

Section 2. Hazard Identification

Classification:

Skin Corrosion/Irritation, Category 1

Serious Eye Damage/ Eye Irritation, Category 1

Labeling:



Symbol:

Signal Word: Danger.

Hazard Statement: Causes severe skin burns and eye damage.

Precautionary Statement: Wear protective gloves/clothing and eye/face protection. Wash thoroughly after handling.

Section 3. Composition

Component	CAS/EINECS Registry #	Percent Concentration
Aluminum	7429-90-5/231-072-3	<0.001
Arsenic	7440-38-2/231-148-6	<0.001
Barium Carbonate (BaCO ₃)	513-77-9/208-167-3	<0.001 (as Ba)
Barium Nitrate (Ba(NO ₃) ₂)	10022-31-8/233-020-5	
Ammonium Bromide (NH ₄ Br)	12124-97-9/235-183-8	<0.001 (as Br)
Calcium Carbonate (CaCO ₃)	471-34-1/207-439-9	<0.001 (as Ca)
Cadmium	7440-43-9/231-152-8	<0.001
Ammonium Chloride (NH ₄ Cl)	12125-02-9/235-186-4	0.003 (as Cl)
Chromium	7440-47-3/231-157-5	<0.001
Cobalt	7440-48-4/231-158-0	<0.001
Copper	7440-50-8/231-159-6	<0.001
Iron	7439-89-6/231-096-4	<0.001
Lead	7439-92-1/231-100-4	<0.001

Safety Data Sheet No. CRM-BL	Date: April 11, 2014	
CRM-BL	Revision: 001	Page 2 of 6

Magnesium	7439-95-4/231-104-6	<0.001
Manganese Acetate Tetrahydrate (Mn(CH ₃ CO ₂) ₂)*4H ₂ O	6156-78-1/211-334-3	<0.001 (as Mn)
Nickel	7440-02-0/231-111-4	<0.001
Ammonium Dihydrogen Phosphate (NH ₄ H ₂ PO ₄)	7722-76-1/231-764-5	0.011 (as P)
Potassium Nitrate (KNO ₃)	7757-79-1/231-818-8	0.01 (as K)
Rubidium Carbonate (Rb ₂ CO ₃)	584-09-8/209-530-9	<0.001 (as Rb)
Selenium	7782-49-2/231-957-4	<0.001
Ammonium Hexafluorosilicate ((NH ₄) ₂ SiF ₆)	16919-19-0/240-968-3	<0.001 (as Si)
Sodium Carbonate (Na ₂ CO ₃)	497-19-8/207-838-8	0.003 (as Na)
Sulfuric Acid (H ₂ SO ₄)	7664-93-9/321-639-5	0.008 (as S)
Ammonium Metavanadate (NH ₄ VO ₃)	7803-55-6/232-261-3	<0.001 (as V)
Zinc	7440-66-6/231-175-3	<0.001
Nitric Acid	7697-37-2/ 231-714-2	4
Water, deionized	7732-18-5/ 231-791-2	Balance

**Note: Barium is derived from either Barium carbonate or Barium Nitrate. For this reason both sources are listed on the SDS. Refer to the product's Certificate of Analysis to determine which source was used in the production of the lot number received.*

Section 4. First Aid Measures

IF ON SKIN (or hair): Remove/Take off immediately all contaminated clothing. Gently wash with plenty of soap and water. Rub calcium gluconate gel immediately to skin. Obtain medical assistance. Wash contaminated clothing before reuse.

IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing. Immediately call a POISON CENTER doctor/physician.

IF SWALLOWED: Rinse mouth. Do NOT induce vomiting.

IF INHALED: Remove to fresh air and keep at rest in a position comfortable for breathing.

Target Organs: Eyes, skin, respiratory system, teeth, and skeletal system.

Section 5. Fire Fighting Measures

Fire & Explosion hazards: While nitric acid is not combustible, it is a strong oxidizing agent that can react with combustible materials; however, it is present in limited quantities in this solution. NO_x compounds can be released in case of fire.

Extinguishing Media: Use any extinguishing media that is suitable for the surrounding area. Use a water spray to dilute nitric acid and to absorb liberated nitrogen oxides.

Safety Data Sheet No. CRM-BL	Date: April 11, 2014	
CRM-BL	Revision: 001	Page 3 of 6

Specific Methods: Firefighters should wear proper protective equipment and self-contained breathing apparatus with full face piece operated in positive pressure mode.

Section 6. Accidental Release Measures

Ventilate area of leak or spill. Wear appropriate personal protective equipment as specified in Section 8. Do not allow to enter drainage systems or water ways. Dike area and dilute spill with water and neutralize with soda ash, limestone, etc. Place the neutralized material into containers suitable for eventual disposal, reclamation, or destruction. Always dispose of in accordance with local regulations.

Section 7. Handling and Storage

Store in a cool, dry, ventilated storage area with acid resistant floors and good drainage. Keep out of direct sunlight and away from heat, water, and incompatible materials. When diluting, the acid should always be added slowly to water and in small amounts. Refer to Section 8 for personal handling instructions.

Section 8. Exposure Controls and Personal Protection

Engineering Controls: Provide exhaust ventilation or other engineering controls to keep any buildup of airborne contaminants below their respective threshold limit value. Ensure the availability of eyewash stations and safety showers.

Personal Protection: Wear proper gloves, safety glasses with side shields, lab coat/apron.

Exposure Limits:

Component	ACGIH TLV	OSHA PEL
Aluminum	10 mg/m ³	15 mg/m ³
Arsenic	0.01 mg/m ³	10 µg/ m ³
Barium	0.5 mg/m ³	0.5 mg/m ³
Ammonium Bromide	Not Available	Not Available
Calcium Carbonate	0.5 mg/m ³	0.5 mg/m ³
Cadmium	0.002 mg/m ³ (respirable particulate)	0.005 mg/m ³
Ammonium Chloride	10 mg/m ³	Not Available
Chromium	0.5 mg/m ³	1 mg/m ³
Cobalt	0.02 mg/m ³	0.1 mg/m ³
Copper	0.2 mg/m ³ (fumes)	0.1 mg/m ³ (fumes)
Iron	10 mg/m ³	5 mg/m ³
Lead	0.05 mg/m ³	0.05 mg/m ³
Magnesium	Not Available	Not Available
Manganese Acetate Tetrahydrate	0.2 mg/m ³	C 5 mg/m ³
Nickel	1.5 mg/m ³	1 mg/m ³
Ammonium Dihydrogen Phosphate	Not Available	Not Available
Potassium Nitrate	Not Available	Not Available
Rubidium Carbonate	Not Available	Not Available
Selenium	0.2 mg/m ³	0.2 mg/ m ³
Ammonium Hexafluorosilicate	Not Available	Not Available
Sodium Carbonate	Not Available	Not Available

Safety Data Sheet No. CRM-BL	Date: April 11, 2014	
CRM-BL	Revision: 001	Page 4 of 6

Sulfuric Acid	5 mg/m ³	1 mg/m ³
Ammonium Metavanadate	0.05 mg/m ³	Not Available
Zinc	5 mg/m ³	1 mg/m ³
Nitric Acid	2 mg/kg	5 mg/m ³

Section 9. Physical and Chemical Properties

Physical State: Liquid
 Color: Clear, colorless liquid
 Odor: Odorless to a faint pungent odor
 Odor threshold: None
 pH: <2
 Melting point: N/A
 Freezing Point: N/A
 Boiling Point: Approximately 100°C
 Flash point: N/A
 Evaporation rate: N/A
 Flammability: N/A
 Explosion limits: N/A
 Vapor Pressure (mm): N/A
 Vapor Density (air+1): N/A
 Relative density: (H₂O = 1): Approximately 1.0
 Solubility in H₂O: Complete
 Auto ignition temperature: N/A
 Decomposition temperature: N/A
 Molecular Weight: N/A

Section 10. Stability and Reactivity

Stability Indicator: Decomposes slowly to release oxygen.
 Conditions to Avoid: Metals, chlorine, organic materials, strong alkali, cyanides.
 Incompatibles: Strong reducing agents.
 Hazardous Decomposition Products: NO_x compounds including nitric oxide (NO), nitrogen dioxide (NO₂), nitrous oxide (N₂O) and nitric acid mist or vapor.
 Hazardous Polymerization: Will not occur.

Section 11. Toxicological Information

May affect skin, mucous membranes and eyes. Swallowing may lead to a negative effect on mouth and throat and to the risk of perforation or the corrosion of esophagus and stomach.

RTECS#

HNO ₃ - QU5775000	Al - BD0330000	As - CG0525000
BaCO ₃ - CQ8600000	NH ₄ Br - BO9155000	Cd - EU9800000
CaCO ₃ - FF9335000	NH ₄ Cl - BP4550000	Co - GF8750000
Cr - GB4200000	Cu - GL5325000	Fe - NO4565500
Pb - OF7525000	Mg - FW6475100	Mn - AI5775000
Ni - QR5950000	KNO ₃ - TT3700000	Rb ₂ CO ₃ - FG0650000
Se - VS7700000	Na ₂ CO ₃ - VZ4050000	(NH ₄) ₂ SiF ₆ - VV7800000
H ₂ SO ₄ - WS5600000	Zn - ZG8600000	NH ₄ VO ₃ - YW0875000
Ba(NO ₃) ₂ - CQ9625000		

Safety Data Sheet No. CRM-BL	Date: April 11, 2014	
CRM-BL	Revision: 001	Page 5 of 6

LD_{LO} Oral, Human: (Nitric Acid) 430 mg/kg
 LD₅₀ Oral, Rat: (Aluminum) >5000 mg/kg
 LD₅₀, Oral, Rat: (Arsenic) 763 mg/kg
 LD_{LO} Oral, Human: (Barium Carbonate) 17 mg/kg.
LD₅₀ Oral, Rat: (Ba(NO₃)₂) 355 mg/kg.
 LD₅₀ Oral, Rat: (Ammonium Bromide) 2700 mg/kg
 LD₅₀ Oral, Rat: (Calcium Carbonate) 6450 mg/kg
 LD₅₀ Oral, Rat: (Ammonium Chloride) 1.65 g/kg
 LD_{LO} Oral, Rabbit: (Cobalt) 750 mg/kg
 LD₅₀ Unreported Route, Rat: (Chromium) 27.5 mg/kg
 TD_{LO} Oral, Human: (Copper) 120 µg/kg
 LD₅₀ Oral, Rat: (Iron) 30 g/kg
 LD₅₀ Oral, Rat: (Potassium Nitrate) 3750 mg/kg
 LD₅₀ Oral, Rat: (Manganese Acetate) 3730mg/kg
 LD₅₀, Oral, Mouse: (Sodium Carbonate) 6600 mg/kg
 LD₅₀, Oral, Rat: (Rubidium Carbonate) 2625 mg/kg
 TD₅₀ Oral, Woman: (Lead) 450 mg/kg/6 years
 LD₅₀ Oral, Rat: (Sulfuric Acid) 2140 mg/kg
 LD₅₀ Oral, Mouse: (Ammonium Hexafluorosilicate) 70 mg/kg
 LC₅₀ Inhalation, Rat: (Sulfuric Acid) 510 mg/m³/2H. No toxic effect noted
 LD₅₀, Oral, Rat: (Selenium) 6700 mg/kg
 LD₅₀ Oral, Rat: (Ammonium Metavanadate) 58.1 mg/kg
 LD_{LO} Oral, Duck: (Zinc) 388 mg/kg

Section 12. Ecological Information

Ecotoxicological information: Do not allow material to reach ground water, water bodies, or sewage system.

Section 13. Disposal Considerations

General: Follow Federal, state and local regulations for waste.

Section 14. Transport Information

D.O.T. Classification: Hazardous by IATA and 49CFR regulations (based on concentration of acid).
 D.O.T. Shipping Name: Corrosive liquid, Acidic, Inorganic, n.o.s. (Nitric Acid Solution)
 D.O.T. Hazard Class: 8
 U.N./N.A. Number: 3264
 Packing Group: II
 D.O.T. Label: Corrosive (8)

Section 15. Regulations (Not meant to be all inclusive-selected regulation listed)

TSCA Status: Components of this solution are listed on the TSCA Inventory.
 RCRA Status: Ammonium metavanadate (7803-55-6)
 SARA: Subject to the reporting requirements of Section 313 or SARA Title III and of 40 CFR 372
 Risk Phrases: R20/21/22, Harmful by inhalation, skin contact, or if swallowed.
 Safety Phrases: S36/37/39 Wear suitable protective clothing, gloves and eye/face protection.

Safety Data Sheet No. CRM-BL	Date: April 11, 2014	
CRM-BL	Revision: 001	Page 6 of 6

WHMIS Information (Canada): E: Corrosive

Section 16. Other Information

HPS products are intended for laboratory use only. All products should be handled and used by trained professional personnel only. The responsibility for the safe handling and use of these products rests solely with the buyer and/or user. The SDS was prepared carefully and represents the best data currently available to us; however, HPS does not certify the data on the SDS. Certified values for this material are given only on the Certificate of Analysis.

Theodore C. Rains, Ph.D.