

High-Purity Standards

Catalogue number: 10M56-3

Version No: 1.1 Safety Data Sheet according to OSHA HazCom Standard (2012) requirements

SECTION 1 IDENTIFICATION

Product Identifier

Product name	10M56-3 Tellurium (10,000µg/mL in 5% HNO3 + 2% HF)
Synonyms	10,000µg/mL Tellurium in 5% HNO3 + 2% HF
Proper shipping name	Corrosive liquid, acidic, inorganic, n.o.s (contains nitric acid and hydrofluoric acid)
Other means of identification	10M56-3
Recommended use of the chemical and restrictions on use	

Relevant identified uses Use according to manufacturer's directions.

Name, address, and telephone number of the chemical manufacturer, importer, or other responsible party

Registered company name	High-Purity Standards
Address	PO Box 41727 SC 29423 United States
Telephone	843-767-7900
Fax	843-767-7906
Website	highpuritystandards.com
Email	Not Available

Emergency phone number

Association / Organisation	Not Available
Emergency telephone numbers	1-800-535-5053
Other emergency telephone numbers	1-352-323-3500

SECTION 2 HAZARD(S) IDENTIFICATION

Classification of the substance or mixture

Classification Serious Eye Damage Category 1, Skin Corrosion/Irritation Category 1B, Acute Toxicity (Oral) Category 3, Acute Toxicity (Dermal) Category Label elements Empty and the serious explore the serie the									
GHS label elements Image: Constraint of the system SIGNAL WORD DANGER Hazard statement(s) Example of the system H318 Causes serious eye damage. H314 Causes serious eye damage. H314 Causes severe skin burns and eye damage. H301 Toxic if swallowed.			Toxici	ity (Dei	rmal)	Catego	ry 3, Me	ətal	
SIGNAL WORD DANGER Hazard statement(s) Causes serious eye damage. H318 Causes serious eye damage. H314 Causes severe skin burns and eye damage. H301 Toxic if swallowed.									
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H314 Causes severe skin burns and eye damage. H301 Toxic if swallowed.									
H301 Toxic if swallowed.									
H311 Toxic in contact with skin.									
H290 May be corrosive to metals.									
H335 May cause respiratory irritation.									

Chemwatch Hazard Alert Code: 3

Issue Date: 08/06/2016 Print Date: 08/06/2016

S.GHS.USA.EN

Hazard(s) not otherwise specified

Not Applicable

Precautionary statement(s) Prevention

recautionary statement(s) Prevention				
P260	Do not breathe dust/fume/gas/mist/vapours/spray.			
Precautionary statement(s)	Precautionary statement(s) Response			
P301+P310	IF SWALLOWED: Immediately call a POISON CENTER or doctor/physician.			
Precautionary statement(s) P405	Storage Store locked up.			
Precautionary statement(s) Disposal				
P501	Dispose of contents/container in accordance with local regulations.			
SECTION 3 COMPOSITION / INFORMATION ON INGREDIENTS				

Substances

See section below for composition of Mixtures

Mixtures

CAS No	%[weight]	Name
13494-80-9	1	tellurium
7697-37-2	5	nitric acid
7664-39-3	2	hydrofluoric acid
7732-18-5	balance	water

SECTION 4 FIRST-AID MEASURES

Description of first aid measures

Eye Contact	If this product comes in contact with the eyes: Immediately hold eyelids apart and flush the eye continuously with running water. Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids. Continue flushing until advised to stop by the Poisons Information Centre or a doctor, or for at least 15 minutes. Transport to hospital or doctor without delay. Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.
Skin Contact	 If there is evidence of severe skin irritation or skin burns: Avoid further contact. Immediately remove contaminated clothing, including footwear. Flush skin under running water for 15 minutes. Avoiding contamination of the hands, massage calcium gluconate gel into affected areas, pay particular attention to creases in skin. Contact the Poisons Information Centre. Continue gel application for at least 15 minutes after burning sensation ceases. If pain recurs, repeat application of calcium gluconate gel or apply every 20 minutes. If no gel is available, continue washing for at least 15 minutes, using soap if available. If patient is conscious, give six calcium gluconate or calcium carbonate tablets in water by mouth. Transport to hospital, or doctor, urgently.
Inhalation	 If fumes or combustion products are inhaled remove from contaminated area. Lay patient down. Keep warm and rested. Prostheses such as false teeth, which may block airway, should be removed, where possible, prior to initiating first aid procedures. Apply artificial respiration if not breathing, preferably with a demand valve resuscitator, bag-valve mask device, or pocket mask as trained. Perform CPR if necessary. Transport to hospital, or doctor, without delay. For massive exposures: If dusts, vapours, aerosols, fumes or combustion products are inhaled, remove from contaminated area. Lay patient down. Keep warm and rested. Prostheses such as false teeth, which may block airway, should be removed, where possible, prior to initiating first aid procedures. Apply artificial respiration if not breathing, preferably with a demand valve resuscitator, bag-valve mask device, or pocket mask as trained. Perform CPR if necessary. If outst, vapours, aerosols, fumes or combustion products are inhaled, remove from contaminated area. Lay patient down. Keep warm and rested. Prostheses such as false teeth, which may block airway, should be removed, where possible, prior to initiating first aid procedures. Apply artificial respiration if not breathing, preferably with a demand valve resuscitator, bag-valve mask device, or pocket mask as trained. Perform CPR if necessary. If victim is conscious, give six calcium gluconate or calcium carbonate tablets in water by mouth. Transport to hospital, or doctor, urgently.
Ingestion	 For advice, contact a Poisons Information Centre or a doctor at once. Urgent hospital treatment is likely to be needed. If swallowed do NOT induce vomiting. If vomiting occurs, lean patient forward or place on left side (head-down position, if possible) to maintain open airway and prevent aspiration. Observe the patient carefully. Never give liquid to a person showing signs of being sleepy or with reduced awareness; i.e. becoming unconscious. Give water to rinse out mouth, then provide liquid slowly and as much as casualty can comfortably drink. Transport to hospital or doctor without delay.

Most important symptoms and effects, both acute and delayed

Indication of any immediate medical attention and special treatment needed

Following acute or short term repeated exposure to hydrofluoric acid:

- Subcutaneous injections of Calcium Gluconate may be necessary around the burnt area. Continued application of Calcium Gluconate Gel or subcutaneous Calcium Gluconate should then continue for 3-4 days at a frequency of 4-6 times per day. If a "burning" sensation recurs, apply more frequently.
- Systemic effects of extensive hydrofluoric acid burns include renal damage, hypocalcaemia and consequent cardiac arrhythmias. Monitor haematological, respiratory, renal, cardiac and electrolyte status at least daily. Tests should include FBE, blood gases, chest X-ray, creatinine and electrolytes, urine output, Ca ions, Mg ions and phosphate ions. Continuous ECG monitoring may be required.
- Where serum calcium is low, or clinical, or ECG signs of hypocalcaemia develop, infusions of calcium gluconate, or if less serious, oral Sandocal, should be given. Hydrocortisone 500 mg in a four to six hourly infusion may help.
- Antibiotics should not be given as a routine, but only when indicated.
- Eye contact pain may be excruciating and 2-3 drops of 0.05% pentocaine hydrochloride may be instilled, followed by further irrigation

BIOLOGICAL EXPOSURE INDEX - BEI

These represent the determinants observed in specimens collected from a healthy worker exposed at the Exposure Standard (ES or TLV):

Determinant	Index	Sampling Time	Comments
1. Methaemoglobin in blood	1.5% of haemoglobin	During or end of shift	B, NS, SQ

B: Background levels occur in specimens collected from subjects NOT exposed.

NS: Non-specific determinant; Also seen after exposure to other materials

SQ: Semi-quantitative determinant - Interpretation may be ambiguous; should be used as a screening test or confirmatory test.

For acute or short term repeated exposures to fluorides:

Fluoride absorption from gastro-intestinal tract may be retarded by calcium salts, milk or antacids.

- Fluoride particulates or fume may be absorbed through the respiratory tract with 20-30% deposited at alveolar level.
- Peak serum levels are reached 30 mins. post-exposure; 50% appears in the urine within 24 hours.
- For acute poisoning (endotracheal intubation if inadequate tidal volume), monitor breathing and evaluate/monitor blood pressure and pulse frequently since shock may supervene with little warning. Monitor ECG immediately; watch for arrhythmias and evidence of Q-T prolongation or T-wave changes. Maintain monitor. Treat shock vigorously with isotonic saline (in 5% glucose) to restore blood volume and enhance renal excretion.
- Where evidence of hypocalcaemic or normocalcaemic tetany exists, calcium gluconate (10 ml of a 10% solution) is injected to avoid tachycardia.

BIOLOGICAL EXPOSURE INDEX - BEI

These represent the determinants observed in specimens collected from a healthy worker exposed at the Exposure Standard (ES or TLV):

Determinant	Index	Sampling Time	Comments
Fluorides in urine	3 mg/gm creatinine	Prior to shift	B, NS
	10mg/gm creatinine	End of shift	B, NS

B: Background levels occur in specimens collected from subjects NOT exposed

NS: Non-specific determinant; also observed after exposure to other exposures.

BAL appears to be contraindicated for tellurium poisonings as when given before tellurium, in test animals, it appears to intensify the effect. Ascorbate has had only equivocal effects in acutely poisoned animals.

[GOSSELIN & HODGE: Clinical Toxicology of Commercial Products]

SECTION 5 FIRE-FIGHTING MEASURES

Extinguishing media

There is no restriction on the type of extinguisher which may be used.

Use extinguishing media suitable for surrounding area.

Special hazards arising from the substrate or mixture

Fire Incompatibility	None known.
Special protective equipme	ent and precautions for fire-fighters
Fire Fighting	 Alert Fire Brigade and tell them location and nature of hazard.
Fire/Explosion Hazard	 Non combustible. May emit poisonous fumes.

SECTION 6 ACCIDENTAL RELEASE MEASURES

Personal precautions, protective equipment and emergency procedures

See section 8

Environmental precautions

See section 12

Methods and material for containment and cleaning up

Minor Spills	 Clean up all spills immediately.
Major Spills #	#

Personal Protective Equipment advice is contained in Section 8 of the SDS.

SECTION 7 HANDLING AND STORAGE

Safe handling	Avoid all personal contact, including inhalation.
Other information	► Store in original containers.
Conditions for safe storag	e, including any incompatibilities
Suitable container	 DO NOT use aluminium or galvanised containers Lined metal can, lined metal pail/ can. For low viscosity materials Drums and jerricans must be of the non-removable head type. All inner and sole packagings for substances that have been assigned to Packaging Groups I or II on the basis of inhalation toxicity criteria, must be hermetically sealed. Material is corrosive to most metals, glass and other siliceous materials.
Storage incompatibility	 Salts of inorganic fluoride: react with water forming acidic solutions. Reacts with mild steel, galvanised steel / zinc producing hydrogen gas which may form an explosive mixture with air. Hydrogen fluoride: reacts violently with strong oxidisers, acetic anhydride, alkalis, 2-aminoethanol, arsenic trioxide (with generation of heat), bismuthic acid, calcium oxide, chlorosulfonic acid, cyanogen fluoride, ethylenediamine, ethylenemine, fluorine (fluorine gas reacts vigorously with a 50% hydrofluoric acid solution and mat burst into flame), nitrogen trifluoride, N-phenylazopiperidine, oleum, oxygen difluoride, phosphorus pentoxide, potassium permanganate, potassium tetrafluorosilicate(2-), beta-propiolactone, propylene oxide, sodium, sodium tetrafluorosilicate, (2-), beta-propiolactone, propylene oxide, salkanolamines, alkylene oxides, aromatic amines, amides, ammonia, ammonium hydroxide, epichlorohydrin, isocyanates, metal acetylides, metal silicides, methanesulfonic acid, nitrogen compounds, organic anhydrides, oxides, silicon compounds, vinylidene fluoride attacks glass and siliceous materials, concrete, ceramics, metals (flammable hydrogen gas may be produced), metal alloys, some plastics, rubber coatings leather, and most other materials with the exception of lead, platinum, polyethylene, wax. Avoid strong bases.

SECTION 8 EXPOSURE CONTROLS / PERSONAL PROTECTION

Control parameters

OCCUPATIONAL EXPOSURE LIMITS (OEL)

INGREDIENT DATA

INGREDIENT DATA						
Source	Ingredient	Material name	TWA	STEL	Peak	Notes
US OSHA Permissible Exposure Levels (PELs) - Table Z1	tellurium	Tellurium and compounds	0.1 mg/m3	Not Available	Not Available	(as Te)
US ACGIH Threshold Limit Values (TLV)	tellurium	Tellurium and compounds (NOS), as Te, excluding hydrogen telluride	0.1 mg/m3	Not Available	Not Available	TLV® Basis: Halitosis
US NIOSH Recommended Exposure Limits (RELs)	tellurium	Aurum paradoxum, Metallum problematum	0.1 mg/m3	Not Available	Not Available	[*Note: The REL also applies to other tellurium compounds (as Te) except Tellurium hexafluoride and Bismuth telluride.]
US OSHA Permissible Exposure Levels (PELs) - Table Z1	nitric acid	Nitric acid	5 mg/m3 / 2 ppm	Not Available	Not Available	Not Available
US ACGIH Threshold Limit Values (TLV)	nitric acid	Nitric acid	2 ppm	4 ppm	Not Available	TLV® Basis: URT & eye irr; dental erosion
US NIOSH Recommended Exposure Limits (RELs)	nitric acid	Aqua fortis, Engravers acid, Hydrogen nitrate, Red fuming nitric acid (RFNA), White fuming nitric acid (WFNA)	5 mg/m3 / 2 ppm	10 mg/m3 / 4 ppm	Not Available	Not Available
US OSHA Permissible Exposure Levels (PELs) - Table Z1	hydrofluoric acid	Hydrogen fluoride	Not Available	Not Available	Not Available	See Table Z-2;(as F)
US OSHA Permissible Exposure Levels (PELs) - Table Z2	hydrofluoric acid	Hydrogen fluoride	3 ppm	Not Available	Not Available	(Z37.28–1969)
US ACGIH Threshold Limit Values (TLV)	hydrofluoric acid	Hydrogen fluoride, as F	0.5 ppm	Not Available	2 ppm	TLV® Basis: URT, LRT, skin, & eye irr; fluorosis; BEI
US NIOSH Recommended Exposure Limits (RELs)	hydrofluoric acid	Anhydrous hydrogen fluoride; Aqueous hydrogen fluoride (i.e., Hydrofluoric acid); HF-A	2.5 mg/m3 / 3 ppm	Not Available	5 mg/m3 / 6 ppm	[15-minute]

EMERGENCY LIMITS

Ingredient	Material name		-1	TEEL-2	TEEL-3
tellurium	Tellurium 1		ı/m3	20 mg/m3	20 mg/m3
nitric acid	Nitric acid Not		vailable	Not Available	Not Available
hydrofluoric acid	Hydrogen fluoride; (Hydrofluoric acid)		vailable	Not Available	Not Available
Ingredient	Original IDLH		Revised IDLH		
tellurium	N.E. / N.E.		25 mg/m3		
nitric acid	100 ppm		25 ppm		

hydroflu	uoric acid	30 ppm	30 [Unch] ppm
water		Not Available	Not Available

Exposure controls

Appropriate engineering controls	Engineering controls are used to remove a hazard or place a barrier between the worker and the hazard.
Personal protection	
Eye and face protection	▶ Chemical goggles.
Skin protection	See Hand protection below
Hands/feet protection	 Wear chemical protective gloves, e.g. PVC. When handling corrosive liquids, wear trousers or overalls outside of boots, to avoid spills entering boots.
Body protection	See Other protection below
Other protection	► Overalls.
Thermal hazards	Not Available

Respiratory protection

Type A Filter of sufficient capacity.

SECTION 9 PHYSICAL AND CHEMICAL PROPERTIES

Information on basic physical and chemical properties

Appearance	colorless		
Physical state	Liquid	Relative density (Water = 1)	Not Available
Odour	Not Available	Partition coefficient n-octanol / water	Not Available
Odour threshold	Not Available	Auto-ignition temperature (°C)	Not Available
pH (as supplied)	<2	Decomposition temperature	Not Available
Melting point / freezing point (°C)	Not Available	Viscosity (cSt)	Not Available
Initial boiling point and boiling range (°C)	Not Available	Molecular weight (g/mol)	Not Available
Flash point (°C)	Not Available	Taste	Not Available
Evaporation rate	Not Available	Explosive properties	Not Available
Flammability	Not Available	Oxidising properties	Not Available
Upper Explosive Limit (%)	Not Available	Surface Tension (dyn/cm or mN/m)	Not Available
Lower Explosive Limit (%)	Not Available	Volatile Component (%vol)	Not Available
Vapour pressure (kPa)	Not Available	Gas group	Not Available
Solubility in water (g/L)	Miscible	pH as a solution (1%)	Not Available
Vapour density (Air = 1)	Not Available	VOC g/L	Not Available

SECTION 10 STABILITY AND REACTIVITY

Reactivity	See section 7
Chemical stability	 Unstable in the presence of incompatible materials.
Possibility of hazardous reactions	See section 7
Conditions to avoid	See section 7
Incompatible materials	See section 7
Hazardous decomposition products	See section 5

SECTION 11 TOXICOLOGICAL INFORMATION

Information on toxicological effects

Inhaled Inhalation of vapours or aerosols (mists, fumes), generated by the material during the course of normal handling, may produce toxic effects. The material can cause respiratory irritation in some persons. Corrosive acids can cause irritation of the respiratory tract, with coughing, choking and mucous membrane damage. Acute effects of fluoride inhalation include irritation of nose and throat, coughing and chest discomfort. Acute inhalation exposures to hydrogen fluoride (hydrofluoric acid) vapours produce severe eye, nose, and throat irritation; delayed fever, cyanosis, and 1

10M56-3 Tellurium (10,000µg/mL in 5% HNO3 + 2% HF)

	pulmonary edema; and may cause death.		
Ingestion	Toxic effects may result from the accidental ingestion of the material; animal experiments indicate that ingestion of less than 40 gram may be fatal or may produce serious damage to the health of the individual. The material can produce chemical burns within the oral cavity and gastrointestinal tract following ingestion. Ingestion of acidic corrosives may produce burns around and in the mouth, the throat and oesophagus. Fluoride causes severe loss of calcium in the blood, with symptoms appearing several hours later including painful and rigid muscle contractions of the limbs.		
Skin Contact	Skin contact with the material may produce toxic effects; systemic effects may result following absorption. The material can produce chemical burns following direct contact with the skin. Skin contact with acidic corrosives may result in pain and burns; these may be deep with distinct edges and may heal slowly with the formation of scar tissue. Contact of the skin with liquid hydrofluoric acid (hydrogen fluoride) may cause severe burns, erythema, and swelling, vesiculation, and serious crusting. Fluorides are easily absorbed through the skin and cause death of soft tissue and erode bone. Open cuts, abraded or irritated skin should not be exposed to this material Entry into the blood-stream, through, for example, cuts, abrasions or lesions, may produce systemic injury with harmful effects.		
Eye	The material can produce chemical burns to the eye following direct contact. If applied to the eyes, this material causes severe eye damage. Direct eye contact with acid corrosives may produce pain, tears, sensitivity to light and burns. Experiments in which a 20-percent aqueous solution of hydrofluoric acid (hydrogen fluoride) was instilled into the eyes of rabbits caused immediate damage in the form of total corneal opacification and conjunctival ischemia; within an hour, corneal stroma edema occurred, followed by necrosis of anterior ocular structures.		
Chronic	Long-term exposure to respiratory irritants may result in disease of the airways involving difficult breathing and related systemic problems. Substance accumulation, in the human body, may occur and may cause some concern following repeated or long-term occupational exposure. Repeated or prolonged exposure to acids may result in the erosion of teeth, swelling and/or ulceration of mouth lining. Extended exposure to inorganic fluorides causes fluorosis, which includes signs of joint pain and stiffness, tooth discolouration, nausea and vomiting, loss of appetite, diarrhoea or constipation, weight loss, anaemia, weakness and general unwellness. Hydrogen fluoride easily penetrates the skin and causes destruction and corrosion of the bone and underlying tissue.		
10M56-3 Tellurium	ΤΟΧΙΟΙΤΥ	IRRITATION	
(10,000µg/mL in 5% HNO3 + 2% HF)	Not Available	Not Available	
	ΤΟΧΙΟΙΤΥ	IRRITATION	
tellurium	Inhalation (rat) LC50: >2.42 mg/L/4hr ^[2]	Nil reported	
	Oral (rat) LD50: 83 mg/kg ^[2]		
	ΤΟΧΙΟΙΤΥ	IRRITATION	
nitric acid	Inhalation (rat) LC50: 0.13 mg/L/4hr ^[2]	* DuPont	
	Inhalation (rat) LC50: 2500 ppm/1h *t ^[2]	Nil reported	
	ΤΟΧΙΟΙΤΥ	IRRITATION	
hydrofluoric acid	Inhalation (rat) LC50: 1.1 mg/L/60M ^[2]	Eye (human): 50 mg - SEVERE	
	Inhalation (rat) LC50: 1276 ppm/1hr ^[2]		
	ΤΟΧΙΟΙΤΥ	IRRITATION	
water	Oral (rat) LD50: >90000 mg/kg ^[2]	Not Available	
Legend:	 Value obtained from Europe ECHA Registered Substances - A extracted from RTECS - Register of Toxic Effect of chemical Sub- 	Acute toxicity 2.* Value obtained from manufacturer's SDS. Unless otherwise specified data Instances	
TELLURIUM	Changes to respiratory system, kidney, ureter, bladder and specifi system.	ic developmental abnormalities involving central nervous, craniofacial, musculoskeletal	

TELLURIUM	system.
NITRIC ACID	for acid mists, aerosols, vapours Data from assays for genotoxic activity in vitro suggest that eukaryotic cells are susceptible to genetic damage when the pH falls to about 6.5. Cells from the respiratory tract have not been examined in this respect. Mucous secretion may protect the cells of the airways from direct exposure to inhaled acidic mists, just as mucous plays an important role in protecting the gastric epithelium from its auto-secreted hydrochloric acid. In considering whether pH itself induces genotoxic events in vivo in the respiratory system, comparison should be made with the human stomach, in which gastric juice may be at pH 1-2 under fasting or nocturnal conditions, and with the human urinary bladder, in which the pH of urine can range from <5 to > 7 and normally averages 6.2. Furthermore, exposures to low pH in vivo differ from exposures <i>in vitro</i> in that, <i>in vivo</i> , only a portion of the cell surface is subjected to the adverse conditions, so that perturbation of intracellular homeostasis may be maintained more readily than in vitro. The material may cause severe skin irritation after prolonged or repeated exposure and may produce on contact skin redness, swelling, the production of vesicles, scaling and thickening of the skin. Repeated exposures may produce severe ulceration. Oral (?) LD50: 50-500 mg/kg * [Various Manufacturers]
HYDROFLUORIC ACID	(liver and kidney damage) [Manufacturer] for hydrogen fluoride (as vapour)
10M56-3 Tellurium (10,000µg/mL in 5% HNO3 + 2% HF) & NITRIC ACID & HYDROFLUORIC ACID	Asthma-like symptoms may continue for months or even years after exposure to the material ceases. This may be due to a non-allergenic condition known as reactive airways dysfunction syndrome (RADS) which can occur following exposure to high levels of highly irritating compound. Key criteria for the diagnosis of RADS include the absence of preceding respiratory disease, in a non-atopic individual, with abrupt onset of persistent asthma-like symptoms within minutes to hours of a documented exposure to the irritant. A reversible airflow pattern, on spirometry, with the presence of moderate to severe bronchial hyperreactivity on methacholine challenge testing and the lack of minimal lymphocytic inflammation, without eosinophilia, have also been included in the criteria for diagnosis of RADS. RADS (or asthma) following an irritating inhalation is an infrequent disorder with rates related to the concentration of and duration of exposure to the irritating substance. Industrial bronchitis, on the other hand, is a disorder that occurs as result of exposure due to high concentrations of irritating substance (often particulate in nature) and is completely reversible after exposure ceases. The disorder is characterised by dyspnea, cough and mucus production.
NITRIC ACID & HYDROFLUORIC ACID	The material may produce severe irritation to the eye causing pronounced inflammation. Repeated or prolonged exposure to irritants may produce conjunctivitis.

NITRIC ACID & HYDROFLUORIC ACID	The material may produce respiratory tract irritation, and result in damage to the lung including reduced lung function.		
HYDROFLUORIC ACID & WATER	No significant acute toxicological data identified in literature search.		
Acute Toxicity	v	Carcinogenicity	\otimes
Skin Irritation/Corrosion	✓	Reproductivity	\otimes
Serious Eye Damage/Irritation	*	STOT - Single Exposure	*
Respiratory or Skin sensitisation	0	STOT - Repeated Exposure	0
Mutagenicity	0	Aspiration Hazard	0
		Legend: 🗙	- Data available but does not fill the criteria for classification

Data required to make classification available

S - Data Not Available to make classification

SECTION 12 ECOLOGICAL INFORMATION

Toxicity

Ingredient	Endpoint	Test Duration (hr)	Species	Value	Source
tellurium	EC50	384	Crustacea	4.440mg/L	3
tellurium	LC50	96	Fish	18.413mg/L	3
tellurium	EC50	48	Crustacea	5.79mg/L	2
tellurium	EC50	72	Algae or other aquatic plants	>11.7mg/L	2
tellurium	NOEC	72	Algae or other aquatic plants	3.34mg/L	2
nitric acid	NOEC	2160	Fish	97.8mg/L	2
nitric acid	EC50	48	Crustacea	490mg/L	2
nitric acid	EC50	96	Crustacea	39mg/L	2
hydrofluoric acid	LC50	96	Fish	51mg/L	2
hydrofluoric acid	EC50	48	Crustacea	97mg/L	2
hydrofluoric acid	EC50	96	Crustacea	10.5mg/L	2
hydrofluoric acid	NOEC	504	Crustacea	3.7mg/L	2
hydrofluoric acid	EC50	96	Algae or other aquatic plants	43mg/L	2
water	EC50	384	Crustacea	199.179mg/L	3
water	EC50	96	Algae or other aquatic plants	8768.874mg/L	3
water	LC50	96	Fish	897.520mg/L	3

Bioconcentration Data 7. METI (Japan) - Bioconcentration Data 8. Vendor Data

For tellurium and its compounds:

Species of Fusarium sp.

Ecotoxicity:

The tolerance of water organisms towards pH margin and variation is diverse.

For Fluorides: Small amounts of fluoride have beneficial effects however, excessive intake over long periods may cause dental and/or skeletal fluorosis.

DO NOT discharge into sewer or waterways.

Persistence and degradability

Ingredient	Persistence: Water/Soil	Persistence: Air
tellurium	HIGH	HIGH
water	LOW	LOW

Bioaccumulative potential

Ingredient	Bioaccumulation
tellurium	LOW (LogKOW = 2.229)
water	LOW (LogKOW = -1.38)

Mobility in soil

Ingredient	Mobility
tellurium	LOW (KOC = 23.74)
water	LOW (KOC = 14.3)

SECTION 13 DISPOSAL CONSIDERATIONS

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10M56-3 Tellurium (10,000µg/mL in 5% HNO3 + 2% HF)

Product / Packaging disposal

Containers may still present a chemical hazard/ danger when empty.
Recycle wherever possible.

SECTION 14 TRANSPORT INFORMATION

Labels Required



Marine Pollutant

Land transport (DOT)

• • • •	
UN number	3264
UN proper shipping name	Corrosive liquid, acidic, inorganic, n.o.s (contains nitric acid and hydrofluoric acid)
Transport hazard class(es)	Class 8 Subrisk Not Applicable
Packing group	II Contraction of the second sec
Environmental hazard	Not Applicable
Special precautions for user	Hazard Label 8 Special provisions B2, IB2, T11, TP2, TP27

Air transport (ICAO-IATA / DGR)

UN number	3264	
UN proper shipping name	Corrosive liquid, acidic, inorganic, n.o.s. * (contains nitric acid a	nd hydrofluoric acid)
Transport hazard class(es)	ICAO/IATA Class 8 ICAO / IATA Subrisk Not Applicable ERG Code 8L	
Packing group	Ш	
Environmental hazard	Not Applicable	
Special precautions for user	Special provisions Cargo Only Packing Instructions Cargo Only Maximum Qty / Pack Passenger and Cargo Packing Instructions Passenger and Cargo Maximum Qty / Pack Passenger and Cargo Limited Quantity Packing Instructions Passenger and Cargo Limited Maximum Qty / Pack	A3A803 855 30 L 851 1 L Y840 0.5 L

Sea transport (IMDG-Code / GGVSee)

UN number	3264
UN proper shipping name	CORROSIVE LIQUID, ACIDIC, INORGANIC, N.O.S. (contains nitric acid and hydrofluoric acid)
Transport hazard class(es)	IMDG Class 8 IMDG Subrisk Not Applicable
Packing group	II
Environmental hazard	Not Applicable
Special precautions for user	EMS NumberF-A, S-BSpecial provisions274Limited Quantities1 L

Transport in bulk according to Annex II of MARPOL and the IBC code Not Applicable

SECTION 15 REGULATORY INFORMATION

Safety, health and environmental regulations / legislation specific for the substance or mixture

TELLURIUM(13494-80-9) IS FOUND ON THE FOLLOWING REGULATORY LISTS

- US Alaska Limits for Air Contaminants US - Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants US - California Permissible Exposure Limits for Chemical Contaminants US - Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air Contaminants US - Hawaii Air Contaminant Limits US - Washington Permissible exposure limits of air contaminants US - Idaho - Limits for Air Contaminants US - Michigan Exposure Limits for Air Contaminants US - Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants US - Minnesota Permissible Exposure Limits (PELs) US ACGIH Threshold Limit Values (TLV) US NIOSH Recommended Exposure Limits (RELs) US - Oregon Permissible Exposure Limits (Z-1) US OSHA Permissible Exposure Levels (PELs) - Table Z1 US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory NITRIC ACID(7697-37-2) IS FOUND ON THE FOLLOWING REGULATORY LISTS International Air Transport Association (IATA) Dangerous Goods Regulations - Prohibited List US - Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air Passenger and Cargo Aircraft Contaminants US - Alaska Limits for Air Contaminants US - Washington Permissible exposure limits of air contaminants US - California OEHHA/ARB - Acute Reference Exposure Levels and Target Organs (RELs) US - Washington Toxic air pollutants and their ASIL. SQER and de minimis emission values US - California Permissible Exposure Limits for Chemical Contaminants US - Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants US - Hawaii Air Contaminant Limits US ACGIH Threshold Limit Values (TLV) US EPCRA Section 313 Chemical List US - Idaho - Limits for Air Contaminants US - Michigan Exposure Limits for Air Contaminants US NIOSH Recommended Exposure Limits (RELs) US OSHA Permissible Exposure Levels (PELs) - Table Z1 US - Minnesota Permissible Exposure Limits (PELs) US - Oregon Permissible Exposure Limits (Z-1) US SARA Section 302 Extremely Hazardous Substances US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory US - Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants HYDROFLUORIC ACID(7664-39-3) IS FOUND ON THE FOLLOWING REGULATORY LISTS International Agency for Research on Cancer (IARC) - Agents Classified by the IARC US - Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air Monographs Contaminants US - Alaska Limits for Air Contaminants US - Washington Permissible exposure limits of air contaminants US - California OEHHA/ARB - Acute Reference Exposure Levels and Target Organs (RELs) US - Washington Toxic air pollutants and their ASIL. SQER and de minimis emission values US - California OEHHA/ARB - Chronic Reference Exposure Levels and Target Organs US - Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants (CRELs) US - Wyoming Toxic and Hazardous Substances Table Z-2 Acceptable ceiling concentration, US - California Permissible Exposure Limits for Chemical Contaminants Acceptable maximum peak above the acceptable ceiling concentration for an 8-hr shift US - Hawaii Air Contaminant Limits US ACGIH Threshold Limit Values (TLV) US - Idaho - Acceptable Maximum Peak Concentrations US ACGIH Threshold Limit Values (TLV) - Carcinogens US ATSDR Minimal Risk Levels for Hazardous Substances (MRLs) US - Idaho - Limits for Air Contaminants US - Michigan Exposure Limits for Air Contaminants US EPCRA Section 313 Chemical List US - Minnesota Permissible Exposure Limits (PELs) US NIOSH Recommended Exposure Limits (RELs) US - Oregon Permissible Exposure Limits (Z-1) US OSHA Permissible Exposure Levels (PELs) - Table Z1 US - Oregon Permissible Exposure Limits (Z-2) US OSHA Permissible Exposure Levels (PELs) - Table Z2 US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants US SARA Section 302 Extremely Hazardous Substances
- US Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants

WATER(7732-18-5) IS FOUND ON THE FOLLOWING REGULATORY LISTS

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

Federal Regulations

Superfund Amendments and Reauthorization Act of 1986 (SARA)

SECTION 311/312 HAZARD CATEGORIES

Immediate (acute) health hazard	YES
Delayed (chronic) health hazard	NO
Fire hazard	NO
Pressure hazard	NO
Reactivity hazard	NO

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

US. EPA CERCLA HAZARDOUS SUBSTANCES AND REPORTABLE QUANTITIES (40 CFR 302.4) Name Reportable Quantity in Pounds (lb) Reportable Quantity in kg Nitric acid 1000 454 Hydrofluoric acid 100 45.4

State Regulations

US. CALIFORNIA PROPOSITION 65

None Reported

National Inventory	Status
Australia - AICS	Y
Canada - DSL	Y
Canada - NDSL	N (water; tellurium; hydrofluoric acid; nitric acid)
China - IECSC	Υ
Europe - EINEC / ELINCS / NLP	Y
Japan - ENCS	N (water; tellurium)
Korea - KECI	Υ

New Zealand - NZIoC	Y
Philippines - PICCS	Y
USA - TSCA	Y
Legend:	Y = All ingredients are on the inventory N = Not determined or one or more ingredients are not on the inventory and are not exempt from listing(see specific ingredients in brackets)

SECTION 16 OTHER INFORMATION

Other information

Ingredients with multiple cas numbers

Name	CAS No
hydrofluoric acid	7664-39-3, 790596-14-4

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chernwatch Classification committee using available literature references.

The SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment.

Definitions and abbreviations

PC – TWA: Permissible Concentration-Time Weighted Average PC – STEL: Permissible Concentration-Short Term Exposure Limit IARC: International Agency for Research on Cancer ACGIH: American Conference of Governmental Industrial Hygienists STEL: Short Term Exposure Limit TEEL: Temporary Emergency Exposure Limit. IDLH: Immediately Dangerous to Life or Health Concentrations OSF: Odour Safety Factor NOAEL: No Observed Adverse Effect Level LOAEL: Lowest Observed Adverse Effect Level TLV: Threshold Limit Value LOD: Limit Of Detection OTV: Odour Threshold Value BCF: BioConcentration Factors BEI: Biological Exposure Index

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