Section 1. Product and Company Identification

Product Identification: AAS-SSWS-R MSDS Number: AAS-SSWS-R Recommended Use: For Laboratory Use

High-Purity Standards Company Identification:

P.O. Box 41727

Charleston, SC 29423 Telephone: (843) 767-7900

FAX: (843) 767-7906

In case of emergency call INFOTRAC: 800-535-5053

Section 2. Hazard Identification

Classification:

Skin Corrosion/Irritation, Category 1

Serious Eye Damage/ Eye Irritation, Category 1

Labeling:



Symbol:

Signal Word: Danger.

Hazard Statement: Causes severe skin burns and eye damage.

Precautionary Statement: Wear protective gloves/clothing and eye/face protection. Wash thoroughly

after handling.

Section 3. Composition				
Component	CAS/EINECS Registry #	Percent Concentration		
Antimony	7440-36-0/231-146-5	0.01		
Arsenic	7440-38-2/231-148-6	0.004		
Cadmium	7440-43-9/231-152-8	<0.001		
Lead	7439-92-1/231-100-4	0.002		
Selenium	7782-49-2/231-957-4	0.001		
Thallium	7440-28-0/231-138-1	0.005		
Nitric Acid (HNO ₃)	7697-37-2/231-714-2	2		
Hydrofluoric Acid	7664-39-3/231-634-8	0.05		
Water, deionized	7732-18-5/231-791-2	Balance		

Section 4. First Aid Measures

IF ON SKIN (or hair): Remove/Take off immediately all contaminated clothing. Gently wash with plenty of soap and water. Rub calcium gluconate gel immediately to skin. Obtain medical assistance. Wash contaminated clothing before reuse.

IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing. Immediately call a POISON CENTER doctor/physician.

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IF SWALLOWED: Rinse mouth. Do NOT induce vomiting.

IF INHALED: Remove to fresh air and keep at rest in a position comfortable for breathing.

Target Organs: Eyes, skin, respiratory system, teeth, and skeletal system.

Section 5. Fire Fighting Measures

Fire & Explosion hazards: While nitric acid is not combustible, it is a strong oxidizing agent that can react with combustible materials; however, it is present in limited quantities in this solution. NO_x compounds can be released in case of fire.

Extinguishing Media: Use any extinguishing media that is suitable for the surrounding area. Use a water spray to dilute nitric acid and to absorb liberated nitrogen oxides.

Specific Methods: Firefighters should wear proper protective equipment and self-contained breathing apparatus with full face piece operated in positive pressure mode.

Section 6. Accidental Release Measures

Ventilate area of leak or spill. Wear appropriate personal protective equipment as specified in Section 8. Do not allow to enter drainage systems or water ways. Dike area and dilute spill with water and neutralize with soda ash, limestone, etc. Place the neutralized material into containers suitable for eventual disposal, reclamation, or destruction. Always dispose of in accordance with local regulations.

Section 7. Handling and Storage

Store in a cool, dry, ventilated storage area with acid resistant floors and good drainage. Keep out of direct sunlight and away from heat, water, and incompatible materials. When diluting, the acid should always be added slowly to water and in small amounts. Refer to Section 8 for personal handling instructions.

Section 8. Exposure Controls and Personal Protection

Engineering Controls: Provide exhaust ventilation or other engineering controls to keep any buildup of airborne contaminants below their respective threshold limit value. Ensure the availability of eyewash stations and safety showers.

Personal Protection: Wear proper gloves, safety glasses with side shields, lab coat/apron.

Exposure Limits:

Component	ACGIH TLV OSHA PEI		
Antimony	0.5 mg/m^3	0.5 mg/m^3	
Arsenic	0.01 mg/m^3	$10 \mu\mathrm{g/m}^3$	
Cadmium	0.002 mg/m^3	0.005 mg/m^3	
	(respirable		
	particulate)		
Lead	0.05 mg/m^3	0.05 mg/m^3	
Selenium	0.2 mg/m^3	0.2 mg/m^3	
Thallium	0.1 mg/m^3	0.1 mg/m^3	
Nitric Acid	2 mg/kg	5 mg/m^3	
Hydrofluoric Acid	C: 3 mg/ml	2.5 mg/m^3	
	_	STEL: 6 mg/ml	
Water, deionized	Not Available	Not Available	

Section 9. Physical and Chemical Properties

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Physical State: Liquid Color: Colorless to light gray

Odor: Odorless to a faint pungent odor

Odor threshold: None

pH: <2

Melting point: N/A Freezing Point: N/A

Boiling Point: Approximately 100°C

Flash point: N/A Evaporation rate: N/A Flammability: N/A Explosion limits: N/A Vapor Pressure (mm): N/A Vapor Density (air+1): N/A

Relative density: $(H_2O = 1)$: Approximately 1.0

Solubility in H₂O: Complete Auto ignition temperature: N/A Decomposition temperature: N/A

Molecular Weight: N/A

Section 10. Stability and Reactivity

Stability Indicator: Decomposes slowly to release oxygen.

Conditions to Avoid: Metals, chlorine, organic materials, strong alkali, cyanides.

Incompatibles: Strong reducing agents.

Hazardous Decomposition Products: NO_x compounds including nitric oxide (NO), nitrogen dioxide

(NO₂), nitrous oxide (N₂O) and nitric acid mist or vapor.

Hazardous Polymerization: Will not occur.

Section 11. Toxicological Information

May affect skin, mucous membranes and eyes. Swallowing may lead to a negative effect on mouth and throat and to the risk of perforation or the corrosion of esophagus and stomach.

Arsenic, Cadmium, and Lead are known or investigated as possible carcinogenic substances

RTECS#:

HNO₃- QU5775000 HF- MW7875000 Sb- CC4025000 As- CG0525000 Cd- EU9800000 Pb- OF7525000 Se- VS7700000 Tl- XG3425000

Toxicity Data:

LD_{LO} Oral, Human: (Nitric Acid) 430 mg/kg

LC_{LO} Inhalation, Human: (Hydrofluoric Acid) 50 mg/kg/30 min

LD₅₀ Oral, Rat: (Antimony) 7g/kg LD₅₀, Oral, Rat: (Arsenic) 763 mg/kg

LD_{LO} Oral, Human: (Cadmium) 2330 mg/kg TD₅₀ Oral, Woman: (Lead) 450 mg/kg/6 year LD₅₀, Oral, Rat: (Selenium) 6700 mg/kg TD_{LO} Oral, Man: (Thallium) 5,714 μg/kg.

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Section 12. Ecological Information

Ecotoxicological information: Do not allow material to reach ground water, water bodies, or sewage system.

Section 13. Disposal Considerations

General: Follow Federal, state and local regulations for waste.

Section 14. Transport Information

D.O.T. Classification: Hazardous by IATA and 49CFR regulations (based on concentration of acid).

D.O.T. Shipping Name: Corrosive liquid, Acidic, Inorganic, n.o.s. (Nitric Acid Solution)

D.O.T. Hazard Class: 8 U.N./N.A. Number: 3264 Packing Group: II

D.O.T. Label: Corrosive (8)

Section 15. Regulations (Not meant to be all inclusive-selected regulation listed)

TSCA Status: Components of this solution are listed on the TSCA Inventory.

RCRA Status: Yes (Hydrofluoric Acid-U134)

SARA: Subject to the reporting requirements of Section 313 or SARA Title III and of 40 CFR 372

Risk Phrases: R20/21/22. Harmful by inhalation, skin contact, or if swallowed.

Safety Phrases: S36/37/39. Wear suitable protective clothing, gloves and eye/face protection.

WHMIS Information (Canada): E: Corrosive

Section 16. Other Information

HPS products are intended for laboratory use only. All products should be handled and used by trained professional personnel only. The responsibility for the safe handling and use of these products rests solely with the buyer and/or user. The SDS was prepared carefully and represents the best data currently available to us; however, HPS does not certify the data on the SDS. Certified values for this material are given only on the Certificate of Analysis.

Theodore C. Rains, Ph.D.