

Estuarian Sediment Solution

High-Purity Standards

Catalogue number: CRM-ES

Version No: 1.1

Safety Data Sheet according to OSHA HazCom Standard (2012) requirements

Chemwatch Hazard Alert Code: 3

Issue Date: **06/06/2017** Print Date: **06/06/2017** S.GHS.USA.EN

SECTION 1 IDENTIFICATION

Product Identifier

Product name	Estuarian Sediment Solution
Synonyms	CRM-ES
Proper shipping name	Corrosive liquid, acidic, inorganic, n.o.s. (contains nitric acid)
Other means of identification	CRM-ES

Recommended use of the chemical and restrictions on use

Name, address, and telephone number of the chemical manufacturer, importer, or other responsible party

Registered company name	High-Purity Standards
Address	PO Box 41727 SC 29423 United States
Telephone	843-767-7900
Fax	843-767-7906
Website	highpuritystandards.com
Email	Not Available

Emergency phone number

0 , .	
Association / Organisation	INFOTRAC
Emergency telephone numbers	1-800-535-5053
Other emergency telephone numbers	1-352-323-3500

SECTION 2 HAZARD(S) IDENTIFICATION

Classification of the substance or mixture

Classification

Metal Corrosion Category 1, Skin Corrosion/Irritation Category 1A

Label elements

Hazard pictogram(s)



SIGNAL WORD

DANGER

Hazard statement(s)

• •	
H290	May be corrosive to metals.
H314	Causes severe skin burns and eye damage.

Hazard(s) not otherwise specified

Not Applicable

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P260

Do not breathe dust/fume/gas/mist/vapours/spray.

Precautionary statement(s) Response

P301+P330+P331

IF SWALLOWED: Rinse mouth. Do NOT induce vomiting.

Precautionary statement(s) Storage

P405

Store locked up.

Precautionary statement(s) Disposal

P501

Dispose of contents/container in accordance with local regulations.

SECTION 3 COMPOSITION / INFORMATION ON INGREDIENTS

Substances

See section below for composition of Mixtures

Mixtures

CAS No	%[weight]	Name
7429-90-5	0.07	aluminium
7440-36-0	0.0000004	antimony
7440-38-2	0.00001	arsenic
7440-43-9	0.0000004	<u>cadmium</u>
7440-70-2	0.008	calcium
7440-47-3	0.00008	chromium
7440-48-4	0.00001	cobalt
7440-50-8	0.00002	copper
7439-89-6	0.035	<u>iron</u>
7439-92-1	0.00003	<u>lead</u>
7439-95-4	0.01	<u>magnesium</u>
638-38-0	0.0004 (as Mn)	manganese(II) acetate
7440-02-0	0.00003	nickel
7440-09-7	0.015	potassium
7782-49-2	0.000005	selenium
7440-23-5	0.02	sodium
7803-55-6	0.0001 (as V)	ammonium metavanadate
7440-66-6	0.00015	zinc
7697-37-2	4	nitric acid
7732-18-5	balance	water
7722-76-1	0.0005 (as P)	ammonium phosphate, monobasic
16919-19-0	0.3 (as Si)	ammonium fluorosilicate
1314-20-1	0.00001 (as Th)	thorium oxide
543-81-7	0.000002 (as Be)	beryllium acetate

SECTION 4 FIRST-AID MEASURES

Description of first aid measures

Description of first aid me	asures
Eye Contact	If this product comes in contact with the eyes: Immediately hold eyelids apart and flush the eye continuously with running water. Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids. Continue flushing until advised to stop by the Poisons Information Centre or a doctor, or for at least 15 minutes. Transport to hospital or doctor without delay. Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.
Skin Contact	If skin or hair contact occurs: Immediately flush body and clothes with large amounts of water, using safety shower if available. Quickly remove all contaminated clothing, including footwear. Wash skin and hair with running water. Continue flushing with water until advised to stop by the Poisons Information Centre. Transport to hospital, or doctor.
Inhalation	 If fumes or combustion products are inhaled remove from contaminated area. Lay patient down. Keep warm and rested. Prostheses such as false teeth, which may block airway, should be removed, where possible, prior to initiating first aid procedures. Apply artificial respiration if not breathing, preferably with a demand valve resuscitator, bag-valve mask device, or pocket mask as trained. Perform CPR if necessary. Transport to hospital, or doctor, without delay.

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Inhalation of vapours or aerosols (mists, fumes) may cause lung oedema. Corrosive substances may cause lung damage (e.g. lung oedema, fluid in the lungs) As this reaction may be delayed up to 24 hours after exposure, affected individuals need complete rest (preferably in semi-recumbent posture) and must be kept under medical observation even if no symptoms are (vet) manifested. ▶ Before any such manifestation, the administration of a spray containing a dexamethasone derivative or beclomethasone derivative may be considered. This must definitely be left to a doctor or person authorised by him/her. (ICSC13719) ▶ For advice, contact a Poisons Information Centre or a doctor at once Urgent hospital treatment is likely to be needed. If swallowed do **NOT** induce vomiting If vomiting occurs, lean patient forward or place on left side (head-down position, if possible) to maintain open airway and prevent aspiration. Ingestion Observe the patient carefully. Never give liquid to a person showing signs of being sleepy or with reduced awareness; i.e. becoming unconscious. ▶ Give water to rinse out mouth, then provide liquid slowly and as much as casualty can comfortably drink.

Most important symptoms and effects, both acute and delayed

See Section 11

Indication of any immediate medical attention and special treatment needed

For acute or short term repeated exposures to strong acids:

- Airway problems may arise from laryngeal edema and inhalation exposure. Treat with 100% oxygen initially.
- Respiratory distress may require cricothyroidotomy if endotracheal intubation is contraindicated by excessive swelling

Transport to hospital or doctor without delay.

- Intravenous lines should be established immediately in all cases where there is evidence of circulatory compromise
- Figure 3. Strong acids produce a coagulation necrosis characterised by formation of a coagulum (eschar) as a result of the dessicating action of the acid on proteins in specific tissues. INGESTION:
- ▶ Immediate dilution (milk or water) within 30 minutes post ingestion is recommended.
- ▶ DO NOT attempt to neutralise the acid since exot
- ▶ Be careful to avoid further vomit since re-exposure of the mucosa to the acid is harmful. Limit fluids to one or two glasses in an adult.
- ▶ Charcoal has no place in acid management.
- ▶ Some authors suggest the use of lavage within 1 hour of ingestion.

SKIN:

- > Skin lesions require copious saline irrigation. Treat chemical burns as thermal burns with non-adherent gauze and wrapping.
- ▶ Deep second-degree burns may benefit from topical silver sulfadiazine.

FYF:

- Fey e injuries require retraction of the eyelids to ensure thorough irrigation of the conjuctival cul-de-sacs. Irrigation should last at least 20-30 minutes. DO NOT use neutralising agents or any other additives. Several litres of saline are required
- Cycloplegic drops, (1% cyclopentolate for short-term use or 5% homatropine for longer term use) antibiotic drops, vasoconstrictive agents or artificial tears may be indicated dependent on the severity of the injury
- ▶ Steroid eye drops should only be administered with the approval of a consulting ophthalmologist).

[Ellenhorn and Barceloux: Medical Toxicology]

Both dermal and oral toxicity of manganese salts is low because of limited solubility of manganese. No known permanent pulmonary sequelae develop after acute manganese exposure. Treatment is supportive

[Ellenhorn and Barceloux: Medical Toxicology]

In clinical trials with miners exposed to manganese-containing dusts, L-dopa relieved extrapyramidal symptoms of both hypo kinetic and dystonic patients. For short periods of time symptoms could also be controlled with scopolamine and amphetamine. BAL and calcium EDTA prove ineffective.

[Gosselin et al: Clinical Toxicology of Commercial Products.]

SECTION 5 FIRE-FIGHTING MEASURES

Extinguishing media

- There is no restriction on the type of extinguisher which may be used.
- Use extinguishing media suitable for surrounding area.

Special hazards arising from the substrate or mixture

Fire Incompatibility

None known

Special protective equipment and precautions for fire-fighters

Fire Fighting

- ▶ Non combustible.
- Not considered to be a significant fire risk.
- Acids may react with metals to produce hydrogen, a highly flammable and explosive gas.

Fire/Explosion Hazard

Heating may cause expansion or decomposition leading to violent rupture of containers.

May emit corrosive, poisonous fumes. May emit acrid smoke.

When aluminium oxide dust is dispersed in air, firefighters should wear protection against inhalation of dust particles, which can also contain hazardous substances from the fire absorbed on the alumina particles.

SECTION 6 ACCIDENTAL RELEASE MEASURES

Personal precautions, protective equipment and emergency procedures

See section 8

Environmental precautions

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See section 12

Methods and material for containment and cleaning up

▶ Drains for storage or use areas should have retention basins for pH adjustments and dilution of spills before discharge or disposal of material.

- Check regularly for spills and leaks.
- Clean up all spills immediately.
- Avoid breathing vapours and contact with skin and eyes. Minor Spills
 - Control personal contact with the substance, by using protective equipment.
 - Contain and absorb spill with sand, earth, inert material or vermiculite
 - ▶ Wipe up.
 - Place in a suitable, labelled container for waste disposal.

Major Spills

Personal Protective Equipment advice is contained in Section 8 of the SDS.

SECTION 7 HANDLING AND STORAGE

Precautions for safe handling

- Avoid all personal contact, including inhalation.
- Wear protective clothing when risk of exposure occurs.
- Use in a well-ventilated area.
- WARNING: To avoid violent reaction, ALWAYS add material to water and NEVER water to material
- Avoid smoking, naked lights or ignition sources.
- Avoid contact with incompatible materials.
- Safe handling
- ▶ When handling, **DO NOT** eat, drink or smoke
- Keep containers securely sealed when not in use.
- Avoid physical damage to containers.
- Always wash hands with soap and water after handling.
- Work clothes should be laundered separately. Launder contaminated clothing before re-use.
- Use good occupational work practice
- ▶ Observe manufacturer's storage and handling recommendations contained within this SDS.
- Atmosphere should be regularly checked against established exposure standards to ensure safe working conditions are maintained.

Other information

Suitable container

- Store in original containers.
- Keep containers securely sealed.
- Store in a cool, dry, well-ventilated area.
- Store away from incompatible materials and foodstuff containers.
- Protect containers against physical damage and check regularly for leaks.
- ▶ Observe manufacturer's storage and handling recommendations contained within this SDS

Conditions for safe storage, including any incompatibilities

- ▶ DO NOT use aluminium or galvanised containers
- Check regularly for spills and leaks
- ▶ Lined metal can, lined metal pail/ can.
- Plastic pail.
- Polyliner drum.
- Packing as recommended by manufacturer.
- Check all containers are clearly labelled and free from leaks.

For low viscosity materials

- ▶ Drums and jerricans must be of the non-removable head type.
- Where a can is to be used as an inner package, the can must have a screwed enclosure.

For materials with a viscosity of at least 2680 cSt. (23 deg. C) and solids (between 15 C deg. and 40 deg C.):

- Removable head packaging;
- Cans with friction closures and
- ▶ low pressure tubes and cartridges

may be used.

Where combination packages are used, and the inner packages are of glass, porcelain or stoneware, there must be sufficient inert cushioning material in contact with inner and outer packages unless the outer packaging is a close fitting moulded plastic box and the substances are not incompatible with the plastic.

For aluminas (aluminium oxide):

Incompatible with hot chlorinated rubber.

In the presence of chlorine trifluoride may react violently and ignite. -May initiate explosive polymerisation of olefin oxides including ethylene oxide.

-Produces exothermic reaction above 200 C with halocarbons and an exothermic reaction at ambient temperatures with halocarbons in the presence of other metals.

-Produces exothermic reaction with oxygen difluoride

-May form explosive mixture with oxygen difluoride

-Forms explosive mixtures with sodium nitrate

-Reacts vigorously with vinyl acetate

Storage incompatibility

Aluminium oxide is an amphoteric substance, meaning it can react with both acids and bases, such as hydrofluoric acid and sodium hydroxide, acting as an acid with a base and a base with an acid, neutralising the other and producing a salt.

- ▶ Inorganic acids are generally soluble in water with the release of hydrogen ions. The resulting solutions have pH's of less than 7.0.
- Inorganic acids neutralise chemical bases (for example: amines and inorganic hydroxides) to form salts neutralisation can generate dangerously large amounts of heat in small spaces
- ► The dissolution of inorganic acids in water or the dilution of their concentrated solutions with additional water may generate significant heat.
- The addition of water to inorganic acids often generates sufficient heat in the small region of mixing to cause some of the water to boil explosively. The resulting "bumping" can spatter the acid.
- Inorganic acids react with active metals, including such structural metals as aluminum and iron, to release hydrogen, a flammable gas.
- Inorganic acids can initiate the polymerisation of certain classes of organic compounds.
- ▶ Inorganic acids react with cyanide compounds to release gaseous hydrogen cyanide.
- Inorganic acids generate flammable and/or toxic gases in contact with dithiocarbamates, isocvanates, mercaptans, nitrides, nitrides, sulfides, and strong reducing agents. Additional gas-generating reactions occur with sulfites, nitrites, thiosulfates (to give H2S and SO3), dithionites (SO2), and even

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- carbonates.
- ▶ Acids often catalyse (increase the rate of) chemical reactions.
- ► WARNING: Avoid or control reaction with peroxides. All *transition metal* peroxides should be considered as potentially explosive. For example transition metal complexes of alkyl hydroperoxides may decompose explosively.

STEL

Peak

Notes

- ► The pi-complexes formed between chromium(0), vanadium(0) and other transition metals (haloarene-metal complexes) and mono-or poly-fluorobenzene show extreme sensitivity to heat and are explosive.
- ► Avoid reaction with borohydrides or cyanoborohydrides

Material name

Avoid strong acids, acid chlorides, acid anhydrides and chloroformates.

SECTION 8 EXPOSURE CONTROLS / PERSONAL PROTECTION

Ingredient

Control parameters

OCCUPATIONAL EXPOSURE LIMITS (OEL)

INGREDIENT DATA

Source

	g. ca.c	material name		0		
US OSHA Permissible Exposure Levels (PELs) - Table Z1	aluminium	Aluminum, metal	15 mg/m3	Not Available	Not Available	Total dust; (as Al)
US OSHA Permissible Exposure Levels (PELs) - Table Z1	aluminium	Aluminum, metal- Respirable fraction	5 mg/m3	Not Available	Not Available	(as Al)
US NIOSH Recommended Exposure Limits (RELs)	aluminium	Aluminium, Aluminum metal, Aluminum powder, Elemental aluminum	10 (total), 5 (resp) mg/m3	Not Available	Not Available	Not Available
US NIOSH Recommended Exposure Limits (RELs)	antimony	Antimony metal, Antimony powder, Stibium	0.5 mg/m3	Not Available	Not Available	[*Note: The REL also applies to other antimony compounds (as Sb).]
US NIOSH Recommended Exposure Limits (RELs)	arsenic	Arsenic metal: Arsenia	Not Available	Not Available	0.002 mg/m3	Ca See Appendix A
US OSHA Permissible Exposure Levels (PELs) - Table Z1	cadmium	Cadmium	0.005 mg/m3	Not Available	Not Available	see 1910.1027;(as Cd)
US NIOSH Recommended Exposure Limits (RELs)	cadmium	Cadmium metal: Cadmium	0.01 mg/m3	Not Available	Not Available	Ca See Appendix A [*Note: The REL applies to all Cadmium compounds (as Cd).]
US ACGIH Threshold Limit Values (TLV)	cadmium	Cadmium	Not Available	Not Available	Not Available	TLV® Basis: Kidney dam; BEI
US NIOSH Recommended Exposure Limits (RELs)	chromium	Chrome, Chromium	0.5 mg/m3	Not Available	Not Available	Not Available
US OSHA Permissible Exposure Levels (PELs) - Table Z1	cobalt	Cobalt metal, dust, and fume	0.1 mg/m3	Not Available	Not Available	(as Co)
US NIOSH Recommended Exposure Limits (RELs)	cobalt	Cobalt metal dust, Cobalt metal fume	0.05 mg/m3	Not Available	Not Available	TLV® Basis: Pneumonitis
US ACGIH Threshold Limit Values (TLV)	cobalt	Hard metals containing Cobalt and Tungsten carbide, as Co	0.005 mg/m3	Not Available	Not Available	Not Available
US NIOSH Recommended Exposure Limits (RELs)	copper	Copper metal dusts, Copper metal fumes	1 mg/m3	Not Available	Not Available	[*Note: The REL also applies to other copper compounds (as Cu) except Copper fume.]
US ACGIH Threshold Limit Values (TLV)	copper	Copper - Fume, as Cu	0.2 mg/m3	Not Available	Not Available	TLV® Basis: Irr; GI; metal fume fever; BEI
US ACGIH Threshold Limit Values (TLV)	copper	Copper - Dusts and mists, as Cu	1 mg/m3	Not Available	Not Available	TLV® Basis: Irr; GI; metal fume fever; BEI
US NIOSH Recommended Exposure Limits (RELs)	lead	Lead metal, Plumbum	0.050 mg/m3	Not Available	Not Available	See Appendix C [*Note: The REL also applies to other lead compounds (as Pb) see Appendix C.]
US NIOSH Recommended Exposure Limits (RELs)	nickel	Nickel metal: Elemental nickel, Nickel catalyst	0.015 mg/m3	Not Available	Not Available	Ca See Appendix A [*Note: The REL does not apply to Nickel carbonyl.]
US ACGIH Threshold Limit Values (TLV)	nickel	Nickel and inorganic compounds including Nickel subsulfide, as Ni - Elemental	1.5 mg/m3	Not Available	Not Available	TLV® Basis: Dermatitis; pneumoconiosis
US NIOSH Recommended Exposure Limits (RELs)	selenium	Elemental selenium, Selenium alloy	0.2 mg/m3	Not Available	Not Available	[*Note: The REL also applies to other selenium compounds (as Se) except Selenium hexafluoride.]
US OSHA Permissible Exposure Levels (PELs) - Table Z1	nitric acid	Nitric acid	5 mg/m3 / 2 ppm	10 mg/m3 / 4 ppm	Not Available	TLV® Basis: URT & eye irr; dental erosion
US NIOSH Recommended Exposure Limits (RELs)	nitric acid	Aqua fortis, Engravers acid, Hydrogen nitrate, Red fuming nitric acid (RFNA), White fuming nitric acid (WFNA)	5 mg/m3 / 2 ppm	4 ppm	Not Available	Not Available
US ACGIH Threshold Limit Values (TLV)	nitric acid	Nitric acid	2 ppm	Not Available	Not Available	Not Available

EMERGENCY LIMITS

Ingredient	Material name	TEEL-1	TEEL-2	TEEL-3
antimony	Antimony	1.5 mg/m3	13 mg/m3	80 mg/m3

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cadmium	Cadmium	Not Available	Not Available	Not Available
chromium	Chromium	1.5 mg/m3	17 mg/m3	99 mg/m3
cobalt	Cobalt	0.18 mg/m3	2 mg/m3	20 mg/m3
copper	Copper	3 mg/m3	33 mg/m3	200 mg/m3
iron	Iron	3.2 mg/m3	35 mg/m3	150 mg/m3
lead	Lead	0.15 mg/m3	120 mg/m3	700 mg/m3
magnesium	Magnesium	18 mg/m3	200 mg/m3	1,200 mg/m3
manganese(II) acetate	Acetic acid, manganese(II) salt (2:1)	9.4 mg/m3	16 mg/m3	96 mg/m3
nickel	Nickel	4.5 mg/m3	50 mg/m3	99 mg/m3
potassium	Potassium	2.3 mg/m3	25 mg/m3	150 mg/m3
selenium	Selenium	0.6 mg/m3	6.6 mg/m3	40 mg/m3
sodium	Sodium	13 mg/m3	140 mg/m3	870 mg/m3
ammonium metavanadate	Ammonium vanadate; (Ammonium vanadium oxide; Ammonium metavanadate)	0.01 mg/m3	0.11 mg/m3	80 mg/m3
zinc	Zinc	6 mg/m3	21 mg/m3	120 mg/m3
nitric acid	Nitric acid	Not Available	Not Available	Not Available
ammonium phosphate, monobasic	Ammonium dihydrogen phosphate; (Monoammonium phosphate)	17 mg/m3	190 mg/m3	1,100 mg/m3
ammonium fluorosilicate	Ammonium hexafluorosilicate; (Ammonium silicofluoride)	12 mg/m3	130 mg/m3	780 mg/m3
thorium oxide	Thorium oxide; (Thorium dioxide)	30 mg/m3	330 mg/m3	2,000 mg/m3

Ingredient	Original IDLH	Revised IDLH
aluminium	Not Available	Not Available
antimony	80 mg/m3	50 mg/m3
arsenic	100 mg/m3	5 mg/m3
cadmium	50 mg/m3 / 9 mg/m3	9 mg/m3 / 9 [Unch] mg/m3
calcium	Not Available	Not Available
chromium	N.E. / N.E.	250 mg/m3
cobalt	20 mg/m3	20 [Unch] mg/m3
copper	N.E. / N.E.	100 mg/m3
iron	Not Available	Not Available
lead	700 mg/m3	100 mg/m3
magnesium	Not Available	Not Available
manganese(II) acetate	N.E. / N.E.	500 mg/m3
nickel	N.E. / N.E.	10 mg/m3
potassium	Not Available	Not Available
selenium	Unknown mg/m3 / Unknown ppm	1 mg/m3
sodium	Not Available	Not Available
ammonium metavanadate	Not Available	Not Available
zinc	Not Available	Not Available
nitric acid	100 ppm	25 ppm
water	Not Available	Not Available
ammonium phosphate, monobasic	Not Available	Not Available
ammonium fluorosilicate	Not Available	Not Available
thorium oxide	Not Available	Not Available
beryllium acetate	10 mg/m3	4 mg/m3

Exposure controls

Engineering controls are used to remove a hazard or place a barrier between the worker and the hazard. Well-designed engineering controls can be highly effective in protecting workers and will typically be independent of worker interactions to provide this high level of protection.

The basic types of engineering controls are:

Process controls which involve changing the way a job activity or process is done to reduce the risk.

Enclosure and/or isolation of emission source which keeps a selected hazard "physically" away from the worker and ventilation that strategically "adds" and "removes" air in the work environment. Ventilation can remove or dilute an air contaminant if designed properly. The design of a ventilation system must match the particular process and chemical or contaminant in use.

Employers may need to use multiple types of controls to prevent employee overexposure.

Appropriate engineering controls

Local exhaust ventilation usually required. If risk of overexposure exists, wear approved respirator. Correct fit is essential to obtain adequate protection. Supplied-air type respirator may be required in special circumstances. Correct fit is essential to ensure adequate protection.

An approved self contained breathing apparatus (SCBA) may be required in some situations.

Provide adequate ventilation in warehouse or closed storage area. Air contaminants generated in the workplace possess varying "escape" velocities which, in turn, determine the "capture velocities" of fresh circulating air required to effectively remove the contaminant.

Type of Contaminant:	Air Speed:
Type or Contaminant:	Air Speed:

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0.25-0.5 m/s (50-100 solvent, vapours, degreasing etc., evaporating from tank (in still air). f/min.) aerosols, fumes from pouring operations, intermittent container filling, low speed conveyer transfers, welding, spray drift, plating 0.5-1 m/s (100-200 acid fumes, pickling (released at low velocity into zone of active generation) f/min.) direct spray, spray painting in shallow booths, drum filling, conveyer loading, crusher dusts, gas discharge (active generation into 1-2.5 m/s (200-500 zone of rapid air motion) f/min.) grinding, abrasive blasting, tumbling, high speed wheel generated dusts (released at high initial velocity into zone of very high rapid 2.5-10 m/s (500-2000 air motion). f/min.)

Within each range the appropriate value depends on:

Lower end of the range	Upper end of the range
1: Room air currents minimal or favourable to capture	1: Disturbing room air currents
2: Contaminants of low toxicity or of nuisance value only.	2: Contaminants of high toxicity
3: Intermittent, low production.	3: High production, heavy use
4: Large hood or large air mass in motion	4: Small hood-local control only

Simple theory shows that air velocity falls rapidly with distance away from the opening of a simple extraction pipe. Velocity generally decreases with the square of distance from the extraction point (in simple cases). Therefore the air speed at the extraction point should be adjusted, accordingly, after reference to distance from the contaminating source. The air velocity at the extraction fan, for example, should be a minimum of 1-2 m/s (200-400 f/min) for extraction of solvents generated in a tank 2 meters distant from the extraction point. Other mechanical considerations, producing performance deficits within the extraction apparatus, make it essential that theoretical air velocities are multiplied by factors of 10 or more when extraction systems are installed or used.

Personal protection







Alternatively a gas mask may replace splash goggles and face shields.





- Safety glasses with unperforated side shields may be used where continuous eye protection is desirable, as in laboratories; spectacles are not sufficient where complete eye protection is needed such as when handling bulk-quantities, where there is a danger of splashing, or if the material may be under
- Chemical goggles.whenever there is a danger of the material coming in contact with the eyes; goggles must be properly fitted.
- Full face shield (20 cm, 8 in minimum) may be required for supplementary but never for primary protection of eyes; these afford face protection.

Eye and face protection

Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants. A written policy document, describing the wearing of lenses or restrictions on use, should be created for each workplace or task. This should include a review of lens absorption and adsorption for the class of chemicals in use and an account of injury experience. Medical and first-aid personnel should be trained in their removal and suitable equipment should be readily available. In the event of chemical exposure, begin eye irrigation immediately and remove contact lens as soon as practicable. Lens should be removed at the first signs of eye redness or irritation - lens should be removed in a clean environment only after workers have washed hands thoroughly. [CDC NIOSH Current Intelligence Bulletin 59], [AS/NZS 1336 or national equivalent]

Skin protection See Hand protection below

Hands/feet protection

- ▶ Elbow length PVC gloves
- ▶ When handling corrosive liquids, wear trousers or overalls outside of boots, to avoid spills entering boots.
- See Other protection below

Body protection

Overalls

Other protection

- ▶ PVC Apron.
- ▶ PVC protective suit may be required if exposure severe
- Eyewash unit.
- ▶ Ensure there is ready access to a safety shower

Thermal hazards

Not Available

Respiratory protection

Type A Filter of sufficient capacity. (AS/NZS 1716 & 1715, EN 143:2000 & 149:2001, ANSI Z88 or national equivalent)

SECTION 9 PHYSICAL AND CHEMICAL PROPERTIES

Information on basic physical and chemical properties

Appearance	Not Available		
Physical state	Liquid	Relative density (Water = 1)	Not Available
Odour	Not Available	Partition coefficient n-octanol / water	Not Available
Odour threshold	Not Available	Auto-ignition temperature (°C)	Not Available
pH (as supplied)	Not Available	Decomposition temperature	Not Available
Melting point / freezing point (°C)	Not Available	Viscosity (cSt)	Not Available
Initial boiling point and boiling range (°C)	Not Available	Molecular weight (g/mol)	Not Available
Flash point (°C)	Not Available	Taste	Not Available
Evaporation rate	Not Available	Explosive properties	Not Available
Flammability	Not Available	Oxidising properties	Not Available

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Upper Explosive Limit (%)	Not Available	Surface Tension (dyn/cm or mN/m)	Not Available
Lower Explosive Limit (%)	Not Available	Volatile Component (%vol)	Not Available
Vapour pressure (kPa)	Not Available	Gas group	Not Available
Solubility in water (g/L)	Miscible	pH as a solution (1%)	Not Available
Vapour density (Air = 1)	Not Available	VOC g/L	Not Available

SECTION 10 STABILITY AND REACTIVITY

Reactivity	See section 7
Chemical stability	► Contact with alkaline material liberates heat
Possibility of hazardous reactions	See section 7
Conditions to avoid	See section 7
Incompatible materials	See section 7
Hazardous decomposition products	See section 5

SECTION 11 TOXICOLOGICAL INFORMATION

Information	on	toxicolog	ical effects

nformation on toxicologic	cal effects					
Inhaled	The material can cause respiratory irritation in some persons. The body's respicorrosive acids can cause irritation of the respiratory tract, with coughing, chokinausea and weakness. The material has NOT been classified by EC Directives or other classification sanimal or human evidence.	ng and mucous membrane damage. There may be dizziness, headache,				
Ingestion	Ingestion of acidic corrosives may produce burns around and in the mouth, the throat and oesophagus. Immediate pain and difficulties in swallowing and speaking may also be evident. The material has NOT been classified by EC Directives or other classification systems as "harmful by ingestion". This is because of the lack of corroboratin animal or human evidence. Poisonings rarely occur after oral administration of manganese salts because they are poorly absorbed from the gut.					
Skin Contact	Skin contact with acidic corrosives may result in pain and burns; these may be of Skin contact is not thought to have harmful health effects (as classified under Esthrough wounds, lesions or abrasions. Though considered non-harmful, slight irritation may result from contact because itching and skin reaction and inflammation. Open cuts, abraded or irritated skin should not be exposed to this material. Entry into the blood-stream, through, for example, cuts, abrasions or lesions, may of the material and ensure that any external damage is suitably protected.	C Directives); the material may still produce health damage following entry to of the abrasive nature of the aluminium oxide particles. Thus it may cause				
Еуе	If applied to the eyes, this material causes severe eye damage. Direct eye contact with acid corrosives may produce pain, tears, sensitivity to liq completely.	ght and burns. Mild burns of the epithelia generally recover rapidly and				
Chronic	Repeated or prolonged exposure to acids may result in the erosion of teeth, sw and inflammation of lung tissue often occurs. Long-term exposure to respiratory irritants may result in airways disease, involv Substance accumulation, in the human body, may occur and may cause some of Animal testing shows long term exposure to aluminium oxides may cause lung the greater the tendencies of causing harm. Manganese is an essential trace element. Chronic exposure to low levels of mas slurred speech, disordered muscle tone, fatigue, anorexia, loss of strength and	ing difficulty breathing and related whole-body problems. oncern following repeated or long-term occupational exposure. disease and cancer, depending on the size of the particle. The smaller the size, anganese can include a mask-like facial expression, spastic gait, tremors,				
Estuarian Sediment Solution	TOXICITY Not Available	IRRITATION Not Available				
	TOXICITY	IRRITATION				

Solution	Not Available	Not Available			
aluminium	TOXICITY Oral (rat) LD50: >2000 mg/kg ^[1]	RRITATION Not Available			
antimony	TOXICITY Dermal (rabbit) LD50: >8300 mg/kg ^[1] Oral (rat) LD50: 100 mg/kg ^[2]		IRRITATION Not Available		
arsenic	TOXICITY Oral (rat) LD50: 763 mg/kg ^[2]		Available		
cadmium	TOXICITY Oral (rat) LD50: >63<259 mg/kg> ^[1]	IRRITATION Not Available			

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TOXICITY IRRITATION Dermal (rabbit) LD50: >2500 $mg/kg^{[1]}$ Not Available calcium Oral (rat) LD50: >2000 mg/kg^[1] TOXICITY IRRITATION chromium Not Available Not Available TOXICITY IRRITATION dermal (rat) LD50: >2000 mg/kg^[1] Not Available cobalt Oral (rat) LD50: 6170 mg/kgd^[2] TOXICITY IRRITATION dermal (rat) LD50: >2000 mg/kg^[1] Not Available Inhalation (rat) LC50: 0.733 mg/l/4hr^[1] copper Inhalation (rat) LC50: 1.03 mg/l/4hr^[1] Inhalation (rat) LC50: 1.67 mg/l/4hr^[1] Oral (rat) LD50: 300-500 $mg/kg^{[1]}$ TOXICITY IRRITATION Oral (rat) LD50: 98600 mg/kg]^[2] Not Available TOXICITY IRRITATION dermal (rat) LD50: >2000 mg/kg^[1] Not Available Inhalation (rat) LC50: >5.05 $mg/l/4hr^{[1]}$ Oral (rat) LD50: >2000 mg/kg^[1] TOXICITY IRRITATION magnesium Oral (rat) LD50: >2000 mg/kg^[1] Not Available TOXICITY IRRITATION manganese(II) acetate Oral (rat) LD50: 2940 mg/kga^[2] Not Available TOXICITY IRRITATION Oral (rat) LD50: 5000 mg/kg^[2] Not Available TOXICITY IRRITATION potassium Not Available Not Available TOXICITY IRRITATION selenium Oral (rat) LD50: 6700 mg/kgd^[2] Not Available TOXICITY IRRITATION sodium Not Available Not Available TOXICITY IRRITATION dermal (rat) LD50: 2102 $mg/kg^{[2]}$ Not Available ammonium metavanadate Oral (rat) LD50: 160 mg/kgd^[2] TOXICITY IRRITATION zinc Dermal (rabbit) LD50: 1130 mg/kg^[2] Not Available

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	Oral (rat) LD50: >2000 mg/kg ^[1]						
	TOXICITY		IRRITATION				
nitric acid	Inhalation (rat) LC50: 625 ppm/1h*t ^[2]		Not Available				
	Innalation (rat) LCou: 625 ppm/ In to		Not Available				
	TOXICITY	IRRITATION					
water	Not Available	Not Available					
		1111111111111111111111111111111111111					
	TOXICITY		IRRITATION				
ammonium phosphate,	dermal (rat) LD50: >5000 mg/kg ^[1]		Not Available				
monobasic	Oral (rat) LD50: >2000 mg/kg ^[1]						
	TOXICITY		IRRITATION				
ammonium fluorosilicate	Oral (mouse) LD50: 70 mg/kg ^[2]		Not Available				
	Grant (models) = 2001 / Grang rig						
	тохісіту	IRRITATION					
thorium oxide	Not Available	Not Available					
	TOXICITY	IRRITATION					
beryllium acetate	Not Available	Not Available					
		<u>'</u>	·				
Legend:	Value obtained from Europe ECHA Registered Substances - Acute toxicit extracted from RTECS - Register of Toxic Effect of chemical Substances	y 2.* Value obtained from manufa	cturer's SDS. Unless otherwise specified data				
ARSENIC	Arsenic compounds are classified by the European Union as toxic by inhalat Tumorigenic - Carcinogenic by RTECS criteria.	ion and ingestion and toxic to aqu	atic life and long lasting in the environment.				
CALCIUM	The solid may react violently on contact with wet skin tissue, i.e. eyes, mouth, causing chemical and thermal burns. The acute effects include burns, ulceration, or tissue death, severe eye damage (corneal burns or opacification), and probable blindness. Inhalation of dust or furnes (especially from a fire involving calcium) will cause shortness of breath, nausea, headache, nose and respiratory tract irritation and in extreme, pneumonitis						
CHROMIUM	On skin and inhalation exposure, chromium and its compounds (except hexa Gastrointestinal tumours, lymphoma, musculoskeletal tumours and tumours		as particulates.				
COBALT	Allergic reactions involving the respiratory tract are usually due to interactions between IgE antibodies and allergens and occur rapidly. Attention should be paid to atopic diathesis, characterised by increased susceptibility to nasal inflammation, asthma and eczema. Exogenous allergic alveolitis is induced essentially by allergen specific immune-complexes of the IgG type; cell-mediated reactions (T lymphocytes) may be involved.						
COPPER	for copper and its compounds (typically copper chloride): Acute toxicity: There are no reliable acute oral toxicity results available. WARNING: Inhalation of high concentrations of copper fume may cause "milike respiratory tract irritation with fever.	for copper and its compounds (typically copper chloride): Acute toxicity: There are no reliable acute oral toxicity results available. WARNING: Inhalation of high concentrations of copper fume may cause "metal fume fever", an acute industrial disease of short duration. tiredness, influenza					
LEAD	WARNING: Lead is a cumulative poison and has the potential to cause abor	tion and intellectual impairment to	unborn children of pregnant workers.				
MANGANESE(II) ACETATE	Laboratory tests have shown mutagenic effects: Positive B. rec.	· · · · · · · · · · · · · · · · · · ·					
NICKEL	Tenth Annual Report on Carcinogens: Substance anticipated to be Carcinog [National Toxicology Program: U.S. Dep. Oral (rat) TDLo: 500 mg/kg/5D-I Inhalation (rat) TCLo: 0.1 mg/m3/24H/17V						
ZINC	The material may cause skin irritation after prolonged or repeated exposure and may produce on contact skin redness, swelling, the production of vesicles, scaling and thickening of the skin.						
NITRIC ACID	For acid mists, aerosols, vapours Test results suggest that eukaryotic cells are susceptible to genetic damage The material may produce severe irritation to the eye causing pronounced in The material may produce respiratory tract irritation, and result in damage to The material may cause severe skin irritation after prolonged or repeated exp vesicles, scaling and thickening of the skin. Oral (?) LD50: 50-500 mg/kg * [Various Manufacturers]	lammation. the lung including reduced lung					
THORIUM OXIDE	Thorium and its compounds are mainly alpha particle emitters although beta radiation is also encountered The radiological danger is considerably more serious than the chemical dan they are deposited (mainly in bones, lungs, lymphatic glands etc.) leading to (liver turnours) Substance has been investigated as a turnorigen; Turnorige lymphoma recorded.	ger in view of the long time that al long-term alpha-irradiation of the	tissues.				
ALUMINIUM & CALCIUM & CHROMIUM & POTASSIUM & SODIUM & WATER & AMMONIUM PHOSPHATE, MONOBASIC	No significant acute toxicological data identified in literature search.						

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ARSENIC & BERYLLIUM ACETATE WARNING: This substance has been classified by the IARC as Group 1: CARCINOGENIC TO HUMANS. **CALCIUM & POTASSIUM & SODIUM & AMMONIUM** METAVANADATE & NITRIC Asthma-like symptoms may continue for months or even years after exposure to the material ends. ACID & AMMONIUM PHOSPHATE, MONOBASIC & BERYLLIUM ACETATE The substance is classified by IARC as Group 3: **CHROMIUM & SELENIUM** NOT classifiable as to its carcinogenicity to humans. **CHROMIUM & THORIUM** Tenth Annual Report on Carcinogens: Substance known to be Carcinogenic OXIDE [National Toxicology Program: U.S. Dep. **COBALT & NICKEL &** The following information refers to contact allergens as a group and may not be specific to this product. BERYLLIUM ACETATE **COBALT & NICKEL** WARNING: This substance has been classified by the IARC as Group 2B: Possibly Carcinogenic to Humans. **Acute Toxicity** 0 Carcinogenicity 0 Skin Irritation/Corrosion Reproductivity 0 Serious Eye 0 STOT - Single Exposure 0 Damage/Irritation Respiratory or Skin 0 0 STOT - Repeated Exposure sensitisation

> **Aspiration Hazard** Legend:

X − Data available but does not fill the criteria for classification
 ✓ − Data available to make classification

- Data available to make classification

0

Data Not Available to make classification

SECTION 12 ECOLOGICAL INFORMATION

Mutagenicity

0

Toxicity

Estuarian Sediment	ENDPOINT	TEST DURATION (HR) S	PECIES	VALUE		SOURCE
Solution	Not Applicable	Not Applicable	<u> </u>	Not Applicable Not		licable	Not Applicable
	ENDROINE	TEGT BURATION (UB)	OPEOIEO			VALUE	2011000
	ENDPOINT	TEST DURATION (HR)	SPECIES			VALUE	SOURCE
	LC50	96	Fish			0.078-0.108mg/L	2
	EC50	48	Crustacea			0.7364mg/L	2
aluminium	EC50	96		ner aquatic plants		0.0054mg/L	2
	BCF	360		ner aquatic plants		9mg/L	4
	EC50	120	Fish			0.000051mg/L	5
	NOEC	72	Algae or ot	ner aquatic plants		>=0.004mg/L	2
	ENDPOINT	TEST DURATION (HR)	SPECIES	3		VALUE	SOURCE
	LC50	96	Fish	Fish		0.93mg/L	2
antimony	EC50	48 Crustacea		1mg/L	2		
·	EC50	72	Algae or	other aquatic plants	S	>2.4mg/L	2
	EC50	96	Crustace	a		0.5mg/L	2
	NOEC	720	720 Fish		>0.0075mg/L	2	
	ENDPOINT	TEST DURATION (HR)	SPECII	S		VALUE	SOURCE
arsenic	LC50	96	Fish	Fish			4
arsenic	EC50	336	Algae o	Algae or other aquatic plants			4
	NOEC	336	Algae o	r other aquatic plan	ts	<0.75mg/L	4
	ENDPOINT	TEST DURATION (HR)	SPECIES			VALUE	SOURCE
	LC50	96	Fish			0.001mg/L	4
	EC50	48	Crustacea			0.0033mg/L	5
cadmium	EC50	72	Algae or ot	her aquatic plants		0.018mg/L	2
	BCF	960	Fish			500mg/L	4
	EC50	336	Crustacea			0.00065mg/L	5
	NOEC	168	Fish			0.00001821mg/L	4

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	ENDPOINT		TEST DUBATION (US)			SPECIES	1/41	.UE	SOURCE
1-1			TEST DURATION (HR)						
calcium	EC50		24			Crustacea		4mg/L	5
	NOEC		48			Crustacea	33.3	Bmg/L	2
	ENDPOINT	TEST	DURATION (HR)		SPECIES			VALUE	SOURCE
			DOKATION (TIK)		Fish				
	LC50	96						13.9mg/L	4
	EC50	48			Crustacea			0.0225mg/L	5
chromium	EC50	72			Algae or other aquatic plants			0.104mg/L	4
	BCF	1440			Algae or other a	aquatic plants		0.0495mg/L	4
	EC50	48			Crustacea			0.0245mg/L	5
	NOEC	672			Fish			0.00019mg/L	4
	ENDPOINT	TEST	TOURATION (HR)		SPECIES			VALUE	SOURCE
	ENDPOINT TEST DURATION (HR) SPECIES LC50 96 Fish								
								1.406mg/L	2
	EC50	48			Crustacea			>0.89mg/L	2
cobalt	EC50	72			Algae or other	aquatic plants		0.144mg/L	2
	BCF	1344			Fish			0.99mg/L	4
	EC50	70			Algae or other			0.02mg/L	2
	NOEC	168			Algae or other	aquatic plants		0.0018mg/L	2
	ENDPOINT	TEOT	DUBATION (UP)		SPECIES			VALUE	SOURCE
			DURATION (HR)						
	LC50	96			Fish			0.0028mg/L	2
	EC50	48			Crustacea			0.001mg/L	5
copper	EC50	72	J 1 1		0.013335mg/L				
	BCF	960			ish			200mg/L	4
	EC50	96			Crustacea			0.001mg/L	5
	NOEC	96		(Crustacea			0.0008mg/L	4
	ENDPOINT	TEST	DURATION (HR)	S	PECIES		,	/ALUE	SOURCE
	LC50	96	()	F	ish).05mg/L	2
	EC50	96			Algae or other aquatic plants			3.7mg/L	4
iron	BCF	24			Crustacea			0.0000002mg/L	4
	EC50	504			Crustacea			1.49mg/L	2
	NOEC	504			ïsh).52mg/L	2
		·							·
	ENDPOINT	TEST	DURATION (HR)		SPECIES			VALUE	SOURCE
	LC50	96			Fish		0.0079mg/L		2
	EC50	48			Crustacea	rustacea		0.029mg/L	2
lead	EC50	72			Algae or other a	aquatic plants 0.0205		0.0205mg/L	2
	BCFD	8			Fish			4.324mg/L	4
	EC50	48			Algae or other a	aquatic plants 0.0217mg/L		0.0217mg/L	2
	NOEC	672			Fish			0.00003mg/L	4
								=	
	ENDPOINT		T DURATION (HR)		SPECIES			VALUE	SOURCE
	LC50		96 Fish		541mg/L		2		
magnesium	EC50		72 Algae or other aquatic plants				2		
	EC50	72			-	aquatic plants		>20mg/L	2
	NOEC	72			Algae or other	aquatic plants		>25.5mg/L	. 2
	ENDPOINT	-	TEST DURATION (HR)		SPECIE	S	VALUE		SOURCE
manganese(II) acetate	Not Applicable		Not Applicable		Not App		Not Applica	able	Not Applicable
	ENDPOINT	TEST	DURATION (HR)	S	PECIES		1	/ALUE	SOURCE
nickel	LC50	96		F	Fish		().0000475mg/L	4
HICKEI									

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				1						1.	
	EC50	72				atic plants		.0407mg/L		2	
	BCF	1440)			atic plants		.47mg/L		4	
	EC50	720		Crustac				.0062mg/L		2	
	NOEC	72		Algae	r otner aqu	atic plants	0	.0035mg/L		2	
	ENDPOINT		TEST DURATION (HR)			SPECIES	VA	LUE	SO	URCE	
potassium	EC50		24			Crustacea	40	Omg/L	5		
							1	9-			
	ENDPOINT	TES	ST DURATION (HR)	SPECI	ES			VALUE		SOURCE	
	LC50	96		Fish				>0.0262mg/L		2	
	EC50	48		Crusta	cea			>0.1603mg/L		2	
selenium	EC50	72		Algae	or other aq	uatic plants		>0.00173mg/L		2	
	BCF	504		Crusta	cea			0.711mg/L		4	
	EC50	96		Algae	or other aq	uatic plants		0.355mg/L		2	
	NOEC	72		Algae	or other aq	uatic plants		0.000547mg/L		2	
	ENDPOINT		TEST DURATION (HR)			SPECIES	VAL	UE	so	URCE	
sodium	EC50		48			Crustacea	1640)mg/L	4		
	EC50 504				Crustacea	1020)mg/L	4			
	ENDPOINT	TE	ST DURATION (HR)	SPE	CIES			VALUE		SOURCE	
	LC50	96		Fish				0.693mg/L		2	
nmonium metavanadate	EC50	48		Crustacea				2.387mg/L		2	
illionium metavanauate	EC50	72		Algae	e or other a	quatic plants		0.9894mg/L		2	
	EC50	72		Algae	Algae or other aquatic plants			1.162mg/L 2		2	
	NOEC	72		Algae	Algae or other aquatic plants				0.0168mg/L 2		
	ENDPOINT		ST DURATION (HR)	SPEC	IES			VALUE		SOURCE	
	LC50		96		Fish			0.00272mg/L		4	
	EC50	48		Crusta	acea			0.04mg/L		5	
zinc	EC50	72		Algae	or other ac	· · ·		0.106mg/L		4	
	BCF	360		Algae				9mg/L		4	
	EC50	120		Fish		0.0003				5	
	NOEC	336	i	Algae	or other ac	quatic plants		0.00075mg/L		4	
	ENDROINT		TEST DURATION (UD)			SPECIES	1/0		20	LIDCE	
nitric acid	ENDPOINT		TEST DURATION (HR)					VALUE 107mg/L		URCE	
	NOEC		16			Crustacea	10	/mg/L	4		
	ENDPOINT		TEST DURATION (HR)		SPECIES	S	VALUE		SOUF	CE	
water	Not Applicable		Not Applicable		Not Appl					Not Applicable	
							st. ppiloc	-		,	
	ENDPOINT	TE	ST DURATION (HR)	SPE	CIES			VALUE		SOURCE	
	LC50	96		Fish	Fish			>85.9mg/L		2	
ammonium phosphate, monobasic	EC50	72		Alga	e or other a	aquatic plants		>97.1mg/L		2	
MONODASIC	EC50	72				aquatic plants		>97.1mg/L		2	
	NOEC	72				aquatic plants		3.57mg/L		2	
mmonium fluorosilicate	ENDPOINT		TEST DURATION (HR)		SPECIES	S	VALUE		SOUR	CE	
onium nuorosincate	Not Applicable		Not Applicable		Not Appl	icable	Not Applica	ble	Not Ap	plicable	
thorium oxide	ENDPOINT		TEST DURATION (HR)		SPECIES		VALUE			CE	
	Not Applicable		Not Applicable		Not Appl	icable	Not Applica	ble	Not Ap	plicable	
	ENDROINT		TEST DUDATION (UD)		SDECIE	8	VALUE		SOUR	CE	
beryllium acetate	ENDPOINT Not Applicable		TEST DURATION (HR)		SPECIES		VALUE	LI-	SOUF		
	Not Applicable		Not Applicable		Not Applicable		Not Applicable		Not Ap	plicable	

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Legend:

Extracted from 1. IUCLID Toxicity Data 2. Europe ECHA Registered Substances - Ecotoxicological Information - Aquatic Toxicity 3. EPIWIN Suite V3.12 (QSAR) - Aquatic Toxicity Data (Estimated) 4. US EPA, Ecotox database - Aquatic Toxicity Data 5. ECETOC Aquatic Hazard Assessment Data 6. NITE (Japan) - Bioconcentration Data 7. METI (Japan) - Bioconcentration Data 8. Vendor Data

Ecotoxicity:

The tolerance of water organisms towards pH margin and variation is diverse. Recommended pH values for test species listed in OECD guidelines are between 6.0 and almost 9. Acute testing with fish showed 96h-LC50 at about pH 3.5

For Fluorides: Small amounts of fluoride have beneficial effects however, excessive intake over long periods may cause dental and/or skeletal fluorosis. Fluorides are absorbed by humans following inhalation of workplace and ambient air that has been contaminated, ingestion of drinking water and foods and dermal contact. Populations living in areas with high fluoride levels in groundwater may be exposed to higher levels of fluorides in their drinking water or in beverages prepared with the water. Among these populations, outdoor labourers, people living in hot climates, and people with excessive thirst will generally have the greatest daily intake of fluorides because they consume greater amounts of water.

Atmospheric Fate: Both hydrogen fluoride and particulate fluorides will be transported in the atmosphere and deposited on land or water by wet and dry deposition. Non-volatile inorganic fluoride particulates are removed from the atmosphere via condensation or nucleation processes. Fluorides adsorbed on particulate matter in the atmosphere are generally stable and are not readily hydrolyzed, although they may be degraded by radiation if they persist in the atmosphere. Fluorine and the silicon fluorides (fluosilicates, silicofluorides) are hydrolyzed in the atmosphere to form hydrogen fluoride. Hydrogen fluoride may combine with water vapour to produce an aerosol or fog of aqueous hydrofluoric acid. Inorganic fluoride compounds, with the exception of sulfur hexafluoride, are not expected to remain in the troposphere for long periods or to migrate to the stratosphere. Estimates of the residence time of sulfur hexafluoride in the atmosphere range from 500 to several thousand years. Fluoride in aerosols can be transported over large distances by wind or as a result of atmospheric turbulence. Fluorosilicic acid and hydrofluoric acid in high aquatic concentrations such as may be found in industrial waste ponds may volatilize, releasing silicon tetrafluoride and hydrogen fluoride into the atmosphere. Soluble inorganic fluorides may also form aerosols at the air/water interface or vaporize into the atmosphere whereas undissolved species generally undergo sedimentation.

Terrestrial Fate: Soils - Atmospheric fluorides may be transported to soils and surface waters through both wet and dry deposition processes where they may form complexes and bind strongly to soil and sediment. Solubilisation of inorganic fluorides from minerals may also be enhanced by the presence of bentonite clays and humic acid. Factors that influence the mobility of inorganic fluorides in soil are pH and the formation of aluminium and calcium complexes. In more acidic soils, concentrations of inorganic fluoride were considerably higher in the deeper horizons. The low affinity of fluorides for organic material results in leaching from the more acidic surface horizon and increased retention by clay minerals and silts in the more alkaline, deeper horizons. The maximum adsorption of fluoride to soil was reported to occur at pH 5.5. In acidic soils with pH below 6, most of the fluoride is in complexes with either aluminium or iron. Fluoride in alkaline soils at pH 6.5 and above is almost completely fixed in soils as calcium fluoride, if sufficient calcium carbonate is available. Fluoride is extremely immobile in soil.

Aquatic Fate: Fresh Water: - In water, the transport and transformation of inorganic fluorides are influenced by pH, water hardness and the presence of ion-exchange materials such as clays. In natural water, fluoride forms strong complexes with aluminium in water, and fluorine chemistry in water is largely regulated by aluminium concentration and pH. Below pH 5, fluoride is almost entirely complexed with aluminium and consequently, the concentration of free F- is low. Once dissolved, inorganic fluorides remain in solution under conditions of low pH and hardness and in the presence of ion-exchange material. Sea Water - Fluoride forms stable complexes with calcium and magnesium, which are present in sea water. Calcium carbonate precipitation dominates the removal of dissolved fluoride from sea water. The residence time for fluoride in ocean sediment is calculated to be 2-3 million years.

Ecotoxicity: Fluorides have been shown to accumulate in animals that consume fluoride-containing foliage. However, accumulation is primarily in skeletal tissue and therefore, it is unlikely that fluoride will biomagnify up the food chain.

For Manganese and its Compounds:

Environmental Fate: Manganese is a naturally occurring element in the environment occurring as a result of weathering of geological material. It also occurs from its use in steel manufacture/coal mining. The most commonly occurring of 11 possible oxidation states are +2, (e.g. manganese chloride or sulfate), +4, (e.g. manganese dioxide), and +7 (e.g. potassium permanganate), although the latter is unstable in the environment.

Atmospheric Fate: Elemental/inorganic manganese compounds may exist in air as suspended particulates from industrial emissions or soil erosion. Manganese-containing particles are mainly removed from the atmosphere by gravitational settling - large particles tend to fall out faster than small particles. The half-life of airborne particles is usually on the order of days, depending on the size of the particle and atmospheric conditions. Some removal by washout mechanisms such as rain may also occur, although it is of minor significance in comparison to dry deposition. Terrestrial Fate: Manganese in soil can migrate as particulate matter to air or water and soluble manganese compounds can be leached from the soil. High soil pH reduces manganese availability while low soil pH will increase availability, even to the point of toxicity. Soils high in organic matter Φ tie up Φ manganese such that high organic matter soils can be manganese deficient. Fertilization with materials containing chlorine, nitrate, and/or sulfate, can also enhance manganese uptake, (termed the anion effect). Adsorption of soluble manganese to soil/sediments increases, (cation), and organic matter increases. In some cases, adsorption of manganese to soils may not be a readily reversible process. At low concentrations, manganese may be fixed by clays and will not be released into solution readily. Bacteria and microflora can increase the mobility of manganese.

Aquatic Fate: Most manganese salts, with the exception of phosphates, carbonates, and oxides, are soluble in water. Solubility is controlled by the precipitation of insoluble forms, (species). In most oxygenated waters, the most common form is insoluble manganese oxide. Manganese chloride is the dominant form at pH 4-7, but may oxidize at pH>8 or 9.

Ecotoxicity: While lower organisms, (plankton, aquatic plants, and some fish), can significantly bioconcentrate manganese, higher organisms, (including humans), tend to maintain manganese balance. Manganese in water may be significantly concentrated at lower levels of the food chain.

Uptake of manganese by aquatic invertebrates and fish increases with temperature and decreases with pH. Fish and crustaceans appear to be the most sensitive to acute and chronic exposures. The substance has low toxicity to trout but, is moderately toxic to Coho salmon. The substance is toxic to Daphnia water fleas and moderately toxic to freshwater algae Pseudomonas putida and Photobacterium phosphoreum bacteria.

For Vanadium Compounds

Environmental Fate: Vanadium is travels through the environment via long-range transportation in the atmosphere, water, and land by natural and man-made sources, wet and dry deposition, adsorption and complexing. From natural sources, vanadium is probably in the form of less soluble trivalent mineral particles.

Atmospheric Fate: Vanadium generally enters the atmosphere as an aerosol. Natural and man-made sources of vanadium tend to release large particles that are more likely to settle near the source. Smaller particles, such as those emitted from oil-fueled power plants, have a longer residence time in the atmosphere and are more likely to be transported farther away from the site of release.

Terrestrial Fate: Soil - Transport and partitioning of vanadium in soil is influenced by pH and reduction potential. Ferric hydroxides and solid bitumens (organic) are the main carriers of vanadium in the sedimentation process. Iron acts as a carrier for trivalent vanadium and is responsible for its diffusion through molten rocks where it becomes trapped during crystallization. Vanadium is fairly mobile in neutral or alkaline soils, but its mobility decreases in acidic soils. Under oxidizing, unsaturated conditions, some mobility is observed, but under reducing, saturated conditions, vanadium is immobile. Plants - Vanadium levels in terrestrial plants are dependent upon the amount of water-soluble vanadium available in the soil as well as pH and growing conditions. The uptake of vanadium into the above-ground parts of many plants is low, although root concentrations have shown some correlation with levels in the soil. Certain legumes have been shown to be vanadium accumulators and the root nodules of these plants may contain vanadium levels three times greater than those of the surrounding soil. Fly agaric (Amanita muscaria) mushrooms are known to actively accumulate vanadium.

Aquatic Fate: Vanadium is eventually adsorbed to hydroxides or associated with organic compounds and is deposited on the sea bed. Vanadium is transported in water by solution (13%) or suspension (87%). Upon entering the ocean, vanadium is deposited to the sea bed. Only about 0.001% of vanadium entering the oceans is estimated to persist in soluble form. Sorption and biochemical processes are thought to contribute to the extraction of vanadium from sea water. Adsorption to organic matter as well as to manganese oxide and ferric hydroxide results in the precipitation of dissolved vanadium. Biochemical processes are also of importance in the partitioning from sea water to sediment.

Ecotoxicity: Some marine organisms, in particular the sea squirts, bioconcentrate vanadium very efficiently, attaining body concentrations approximately 10,000 times greater than the ambient sea water. Upon the death of the organism, the body burden adds to the accumulation of vanadium in silt. In general, marine plants and invertebrates contain higher levels of vanadium than terrestrial plants and animals. In the terrestrial environment, bioconcentration is more commonly observed amongst the lower plant phyla than in the higher, seed-producing phyla. Vanadium appears to be present in all terrestrial animals; however tissue concentrations in vertebrates are often so low that detection is difficult. The highest levels of vanadium in terrestrial mammals are generally found in the liver and skeletal tissues. No data are available regarding biomagnification of vanadium within the food chain, but human studies suggest that it is unlikely. Bioaccumulation appears to be unlikely.

Prevent, by any means available, spillage from entering drains or water courses.

DO NOT discharge into sewer or waterways.

Persistence and degradability

Ingredient	Persistence: Water/Soil	Persistence: Air
ammonium metavanadate	HIGH	HIGH
water	LOW	LOW
ammonium phosphate, monobasic	HIGH	HIGH

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Bioaccumulative potential

Ingredient	Bioaccumulation
ammonium metavanadate	LOW (LogKOW = 2.229)
water	LOW (LogKOW = -1.38)
ammonium phosphate, monobasic	LOW (LogKOW = -0.7699)

Mobility in soil

Ingredient	Mobility
ammonium metavanadate	LOW (KOC = 35.04)
water	LOW (KOC = 14.3)
ammonium phosphate, monobasic	HIGH (KOC = 1)

SECTION 13 DISPOSAL CONSIDERATIONS

Waste treatment methods

Product / Packaging disposal

- ► Recycle wherever possible.
- ▶ Consult manufacturer for recycling options or consult local or regional waste management authority for disposal if no suitable treatment or disposal facility can be identified.
- Freat and neutralise at an approved treatment plant. Treatment should involve: Neutralisation with soda-ash or soda-lime followed by: burial in a land-fill specifically licensed to accept chemical and / or pharmaceutical wastes or Incineration in a licensed apparatus (after admixture with suitable combustible material).
- ▶ Decontaminate empty containers with 5% aqueous sodium hydroxide or soda ash, followed by water. Observe all label safeguards until containers are cleaned and destroyed.

SECTION 14 TRANSPORT INFORMATION

Labels Required



Marine Pollutant

Land transport (DOT)

UN number	3264		
UN proper shipping name	Corrosive liquid, acidic, inorganic, n.o.s. (contains nitric acid)		
Transport hazard class(es)	Class 8 Subrisk Not Applicable		
Packing group	II		
Environmental hazard	Not Applicable		
Special precautions for user	Hazard Label 8 Special provisions 386, B2, IB2, T11, TP2, TP27		

Air transport (ICAO-IATA / DGR)

UN number	3264				
UN proper shipping name	Corrosive liquid, acidic	, inorganic, n.o.s. * (c	ontains nitric acid)		
Transport hazard class(es)	ICAO/IATA Class	8			
	ICAO / IATA Subrisk	Not Applicable			
	ERG Code	8L			
Packing group	oup II				
Environmental hazard	Not Applicable	Not Applicable			
	Special provisions		A3A803		
special precautions for user	Cargo Only Packing I	nstructions	855		
,	Cargo Only Maximum	Oty / Pack	30 L	1	

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Passenger and Cargo Packing Instructions	851
Passenger and Cargo Maximum Qty / Pack	1 L
Passenger and Cargo Limited Quantity Packing Instructions	Y840
Passenger and Cargo Limited Maximum Qty / Pack	0.5 L

Sea transport (IMDG-Code / GGVSee)

UN number	3264	
UN proper shipping name	CORROSIVE LIQUID, ACIDIC, INORGANIC, N.O.S. (contains nitric acid)	
Transport hazard class(es)	IMDG Class 8 IMDG Subrisk Not Applicable	
Packing group	II .	
Environmental hazard	Not Applicable	
Special precautions for user	EMS Number F-A, S-B Special provisions 274 Limited Quantities 1 L	

Transport in bulk according to Annex II of MARPOL and the IBC code

Source	Product name	Pollution Category	Ship Type
IMO MARPOL (Annex II) - List of Noxious Liquid Substances Carried in Bulk	Nitric acid (70% and over) Nitric acid (less than 70%)	Y; Y	2 2

SECTION 15 REGULATORY INFORMATION

Safety, health and environmental regulations / legislation specific for the substance or mixture

ALUMINIUM(7429-90-5) IS FOUND ON THE FOLLOWING REGULATORY LIS	STS

US - Alaska Limits for Air Contaminants	US - Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air
US - California Permissible Exposure Limits for Chemical Contaminants	Contaminants
US - Hawaii Air Contaminant Limits	US - Washington Permissible exposure limits of air contaminants
US - Massachusetts - Right To Know Listed Chemicals	US - Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants
US - Michigan Exposure Limits for Air Contaminants	US ACGIH Threshold Limit Values (TLV)
US - Minnesota Permissible Exposure Limits (PELs)	US ACGIH Threshold Limit Values (TLV) - Carcinogens
US - Oregon Permissible Exposure Limits (Z-1)	US ATSDR Minimal Risk Levels for Hazardous Substances (MRLs)
US - Pennsylvania - Hazardous Substance List	US EPCRA Section 313 Chemical List
US - Rhode Island Hazardous Substance List	US NIOSH Recommended Exposure Limits (RELs)
US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants	US OSHA Permissible Exposure Levels (PELs) - Table Z1
US - Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants	US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

ANTIMONY(7440-36-0) IS FOUND ON THE FOLLOWING REGULATORY LISTS

US - Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air
Contaminants
US - Washington Permissible exposure limits of air contaminants
US - Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants
US ACGIH Threshold Limit Values (TLV)
US Clean Air Act - Hazardous Air Pollutants
US CWA (Clean Water Act) - Priority Pollutants
US CWA (Clean Water Act) - Toxic Pollutants
US EPCRA Section 313 Chemical List
US NIOSH Recommended Exposure Limits (RELs)
US OSHA Permissible Exposure Levels (PELs) - Table Z1
US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

\parallel ARSENIC(7440-38-2) IS FOUND ON THE FOLLOWING REGULATORY LISTS

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International Agency for Research on Cancer (IARC) - Agents Classified by the IARC
Monographs

- US Alaska Limits for Air Contaminants
- US California OEHHA/ARB Acute Reference Exposure Levels and Target Organs (RELs)
- US California OEHHA/ARB Chronic Reference Exposure Levels and Target Organs
- US California Permissible Exposure Limits for Chemical Contaminants
- US Hawaii Air Contaminant Limits
- US Idaho Limits for Air Contaminants
- US Massachusetts Right To Know Listed Chemicals
- US Minnesota Permissible Exposure Limits (PELs)
- US New Jersey Right to Know Special Health Hazard Substance List (SHHSL): Carcinogens
- US Pennsylvania Hazardous Substance List
- US Tennessee Occupational Exposure Limits Limits For Air Contaminants
- US Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants
- US Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air Contaminants

- US Washington Permissible exposure limits of air contaminants
- US Washington Toxic air pollutants and their ASIL, SQER and de minimis emission values
- US ACGIH Threshold Limit Values (TLV)
- US ACGIH Threshold Limit Values (TLV) Carcinogens
- US ATSDR Minimal Risk Levels for Hazardous Substances (MRLs)
- US Clean Air Act Hazardous Air Pollutants
- US CWA (Clean Water Act) Priority Pollutants
- US CWA (Clean Water Act) Toxic Pollutants
- US EPCRA Section 313 Chemical List
- US National Toxicology Program (NTP) 14th Report Part A Known to be Human Carcinogens
- US NIOSH Recommended Exposure Limits (RELs)
- US OSHA Permissible Exposure Levels (PELs) Table Z1
- US Toxic Substances Control Act (TSCA) Chemical Substance Inventory

CADMIUM(7440-43-9) IS FOUND ON THE FOLLOWING REGULATORY LISTS

International Agency for Research on Cancer (IARC) - Agents Classified by the IARC Monographs

- US Alaska Limits for Air Contaminants
- US California Proposition 65 Priority List for the Development of MADLs for Chemicals Causing Reproductive Toxicity
- US California OEHHA/ARB Chronic Reference Exposure Levels and Target Organs (CRELs)
- US California Permissible Exposure Limits for Chemical Contaminants
- US California Proposition 65 Carcinogens
- US California Proposition 65 Maximum Allowable Dose Levels (MADLs) for Chemicals Causing Reproductive Toxicity
- US California Proposition 65 No Significant Risk Levels (NSRLs) for Carcinogens
- US California Proposition 65 Reproductive Toxicity
- US Hawaii Air Contaminant Limits
- US Idaho Acceptable Maximum Peak Concentrations
- US Idaho Limits for Air Contaminants
- US Massachusetts Right To Know Listed Chemicals
- US Michigan Exposure Limits for Air Contaminants
- US Minnesota Permissible Exposure Limits (PELs)
- US New Jersey Right to Know Special Health Hazard Substance List (SHHSL): Carcinogens
- US Oregon Permissible Exposure Limits (Z-1)
- US Oregon Permissible Exposure Limits (Z-2)
- US Pennsylvania Hazardous Substance List
- US Rhode Island Hazardous Substance List
- US Tennessee Occupational Exposure Limits Limits For Air Contaminants

- US Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants
- US Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air Contaminants
- US Washington Permissible exposure limits of air contaminants
- US Washington Toxic air pollutants and their ASIL, SQER and de minimis emission values
- US Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants
- US Wyoming Toxic and Hazardous Substances Table Z-2 Acceptable ceiling concentration, Acceptable maximum peak above the acceptable ceiling concentration for an 8-hr shift
- US ACGIH Threshold Limit Values (TLV)
- US ACGIH Threshold Limit Values (TLV) Carcinogens
- US ATSDR Minimal Risk Levels for Hazardous Substances (MRLs)
- US Clean Air Act Hazardous Air Pollutants
- US CWA (Clean Water Act) Priority Pollutants
- US CWA (Clean Water Act) Toxic Pollutants
- US EPA Carcinogens Listing
- US EPCRA Section 313 Chemical List
- US National Toxicology Program (NTP) 14th Report Part A Known to be Human Carcinogens
- US NIOSH Recommended Exposure Limits (RELs)
- US OSHA Carcinogens Listing
- US OSHA Permissible Exposure Levels (PELs) Table Z1
- US OSHA Permissible Exposure Levels (PELs) Table Z2
- US Toxic Substances Control Act (TSCA) Chemical Substance Inventory

CALCIUM(7440-70-2) IS FOUND ON THE FOLLOWING REGULATORY LISTS

- US Massachusetts Right To Know Listed Chemicals
- US Pennsylvania Hazardous Substance List

- US Rhode Island Hazardous Substance List
- US Toxic Substances Control Act (TSCA) Chemical Substance Inventory

US - Washington Permissible exposure limits of air contaminants

US - Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air

US - Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants

CHROMIUM(7440-47-3) IS FOUND ON THE FOLLOWING REGULATORY LISTS

International Agency for Research on Cancer (IARC) - Agents Classified by the IARC Monographs

- US Alaska Limits for Air Contaminants
- US California Permissible Exposure Limits for Chemical Contaminants
- US Hawaii Air Contaminant Limits
- US Idaho Limits for Air Contaminants
- US Massachusetts Right To Know Listed Chemicals
- US Michigan Exposure Limits for Air Contaminants
- US Oregon Permissible Exposure Limits (Z-1)
- US Pennsylvania Hazardous Substance List US - Rhode Island Hazardous Substance List
- US Tennessee Occupational Exposure Limits Limits For Air Contaminants
- US ACGIH Threshold Limit Values (TLV) Carcinogens US Clean Air Act - Hazardous Air Pollutants

US ACGIH Threshold Limit Values (TLV)

- US CWA (Clean Water Act) Priority Pollutants
- US CWA (Clean Water Act) Toxic Pollutants
- US EPCRA Section 313 Chemical List

Contaminants

- US NIOSH Recommended Exposure Limits (RELs)
- US OSHA Permissible Exposure Levels (PELs) Table Z1 US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

COBALT(7440-48-4) IS FOUND ON THE FOLLOWING REGULATORY LISTS

US - Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants

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International Agency for Research on Cancer (IARC) - Agents Classified by the IARC Monographs

US - Alaska Limits for Air Contaminants

US - California Permissible Exposure Limits for Chemical Contaminants

US - California Proposition 65 - Carcinogens

US - Hawaii Air Contaminant Limits

US - Idaho - Limits for Air Contaminants

US - Massachusetts - Right To Know Listed Chemicals

US - Michigan Exposure Limits for Air Contaminants

US - Minnesota Permissible Exposure Limits (PELs)

US - New Jersey Right to Know - Special Health Hazard Substance List (SHHSL): Carcinogens

US - Oregon Permissible Exposure Limits (Z-1)

US - Pennsylvania - Hazardous Substance List

US - Rhode Island Hazardous Substance List

US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants

US - Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants

COPPER(7440-50-8) IS FOUND ON THE FOLLOWING REGULATORY LISTS

US - Alaska Limits for Air Contaminants

US - California Permissible Exposure Limits for Chemical Contaminants

US - Hawaii Air Contaminant Limits

US - Idaho - Limits for Air Contaminants

US - Massachusetts - Right To Know Listed Chemicals

US - Michigan Exposure Limits for Air Contaminants

US - Oregon Permissible Exposure Limits (Z-1)

US - Pennsylvania - Hazardous Substance List

US - Rhode Island Hazardous Substance List

US - Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air Contaminants

US - Washington Permissible exposure limits of air contaminants

US - Washington Toxic air pollutants and their ASIL, SQER and de minimis emission values

US - Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants

US ACGIH Threshold Limit Values (TLV)

US ACGIH Threshold Limit Values (TLV) - Carcinogens

US ATSDR Minimal Risk Levels for Hazardous Substances (MRLs)

US Clean Air Act - Hazardous Air Pollutants

US EPCRA Section 313 Chemical List

US National Toxicology Program (NTP) 14th Report Part B.

US NIOSH Recommended Exposure Limits (RELs)

US OSHA Permissible Exposure Levels (PELs) - Table Z1

US Priority List for the Development of Proposition 65 Safe Harbor Levels - No Significant Risk Levels (NSRLs) for Carcinogens and Maximum Allowable Dose Levels (MADLs) for

Chemicals Causing Reproductive Toxicity

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

US - California OEHHA/ARB - Acute Reference Exposure Levels and Target Organs (RELs)

US - Minnesota Permissible Exposure Limits (PELs)

US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants

US - Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants

US - Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air Contaminants

US - Washington Permissible exposure limits of air contaminants

US - Washington Toxic air pollutants and their ASIL, SQER and de minimis emission values

US - Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants

US ACGIH Threshold Limit Values (TLV)

US ATSDR Minimal Risk Levels for Hazardous Substances (MRLs)

US CWA (Clean Water Act) - Priority Pollutants

US CWA (Clean Water Act) - Toxic Pollutants

US EPA Carcinogens Listing

US EPCRA Section 313 Chemical List

US NIOSH Recommended Exposure Limits (RELs)

US OSHA Permissible Exposure Levels (PELs) - Table Z1

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

IRON(7439-89-6) IS FOUND ON THE FOLLOWING REGULATORY LISTS

International Agency for Research on Cancer (IARC) - Agents Classified by the IARC Monographs

US - California OEHHA/ARB - Chronic Reference Exposure Levels and Target Organs (CRELs)

US - California Permissible Exposure Limits for Chemical Contaminants

US - Hawaii Air Contaminant Limits

US - Michigan Exposure Limits for Air Contaminants

US - Oregon Permissible Exposure Limits (Z-1)

US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants

US - Washington Permissible exposure limits of air contaminants

US - Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

LEAD(7439-92-1) IS FOUND ON THE FOLLOWING REGULATORY LISTS

International Agency for Research on Cancer (IARC) - Agents Classified by the IARC

Monographs US - Alaska Limits for Air Contaminants

US - California - Proposition 65 - Priority List for the Development of MADLs for Chemicals Causing Reproductive Toxicity

US - California Permissible Exposure Limits for Chemical Contaminants

US - California Proposition 65 - Carcinogens

US - California Proposition 65 - Maximum Allowable Dose Levels (MADLs) for Chemicals Causing Reproductive Toxicity

US - California Proposition 65 - No Significant Risk Levels (NSRLs) for Carcinogens

US - California Proposition 65 - Reproductive Toxicity

US - Hawaii Air Contaminant Limits

US - Idaho - Acceptable Maximum Peak Concentrations

US - Idaho - Limits for Air Contaminants

US - Massachusetts - Right To Know Listed Chemicals

US - Minnesota Permissible Exposure Limits (PELs)

US - New Jersey Right to Know - Special Health Hazard Substance List (SHHSL): Carcinogens

US - Pennsylvania - Hazardous Substance List

US - Rhode Island Hazardous Substance List

US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants

US - Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants

US - Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air Contaminants

US - Washington Permissible exposure limits of air contaminants

US - Washington Toxic air pollutants and their ASIL. SQER and de minimis emission values

US ACGIH Threshold Limit Values (TLV)

US ACGIH Threshold Limit Values (TLV) - Carcinogens

US Clean Air Act - Hazardous Air Pollutants

US CWA (Clean Water Act) - Priority Pollutants

US CWA (Clean Water Act) - Toxic Pollutants

US EPA Carcinogens Listing US EPCRA Section 313 Chemical List

US National Toxicology Program (NTP) 14th Report Part B. US NIOSH Recommended Exposure Limits (RELs)

US OSHA Permissible Exposure Levels (PELs) - Table Z1

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

MAGNESIUM(7439-95-4) IS FOUND ON THE FOLLOWING REGULATORY LISTS

International Agency for Research on Cancer (IARC) - Agents Classified by the IARC

Monographs US - California OEHHA/ARB - Chronic Reference Exposure Levels and Target Organs (CRELs)

US - California Permissible Exposure Limits for Chemical Contaminants

US - Hawaii Air Contaminant Limits

US - Massachusetts - Right To Know Listed Chemicals

US - Michigan Exposure Limits for Air Contaminants

US - Oregon Permissible Exposure Limits (Z-1)

US - Pennsylvania - Hazardous Substance List

US - Rhode Island Hazardous Substance List

US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants

US - Washington Permissible exposure limits of air contaminants

US - Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

MANGANESE(II) ACETATE(638-38-0) IS FOUND ON THE FOLLOWING REGULATORY LISTS

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US - Alaska Limits for Air Contaminants US - Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants US - California OEHHA/ARB - Chronic Reference Exposure Levels and Target Organs US - Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air Contaminants US - California Permissible Exposure Limits for Chemical Contaminants US - Washington Permissible exposure limits of air contaminants US - Hawaii Air Contaminant Limits US - Washington Toxic air pollutants and their ASIL, SQER and de minimis emission values US - Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants US - Idaho - Limits for Air Contaminants US - Michigan Exposure Limits for Air Contaminants US Clean Air Act - Hazardous Air Pollutants US - Minnesota Permissible Exposure Limits (PELs) US EPCRA Section 313 Chemical List US - Oregon Permissible Exposure Limits (Z-1) US OSHA Permissible Exposure Levels (PELs) - Table Z1 US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

NICKEL(7440-02-0) IS FOUND ON THE FOLLOWING REGULATORY LISTS

US - Alaska Limits for Air Contaminants US - Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air US - California OEHHA/ARB - Acute Reference Exposure Levels and Target Organs (RELs) US - Washington Permissible exposure limits of air contaminants US - California OEHHA/ARB - Chronic Reference Exposure Levels and Target Organs (CRELs) US - Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants US - California Permissible Exposure Limits for Chemical Contaminants US ACGIH Threshold Limit Values (TLV) US - California Proposition 65 - Carcinogens US ACGIH Threshold Limit Values (TLV) - Carcinogens US - Hawaii Air Contaminant Limits US ATSDR Minimal Risk Levels for Hazardous Substances (MRLs) US - Idaho - Limits for Air Contaminants US Clean Air Act - Hazardous Air Pollutants US - Massachusetts - Right To Know Listed Chemicals US CWA (Clean Water Act) - Priority Pollutants US - Michigan Exposure Limits for Air Contaminants US CWA (Clean Water Act) - Toxic Pollutants

US - Minnesota Permissible Exposure Limits (PELs) US EPCRA Section 313 Chemical List US - New Jersey Right to Know - Special Health Hazard Substance List (SHHSL): US National Toxicology Program (NTP) 14th Report Part B. Carcinogens US NIOSH Recommended Exposure Limits (RELs) US - Oregon Permissible Exposure Limits (Z-1) US OSHA Permissible Exposure Levels (PELs) - Table Z1

US - Pennsylvania - Hazardous Substance List US Priority List for the Development of Proposition 65 Safe Harbor Levels - No Significant Risk US - Rhode Island Hazardous Substance List Levels (NSRLs) for Carcinogens and Maximum Allowable Dose Levels (MADLs) for US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants Chemicals Causing Reproductive Toxicity

US - Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

POTASSIUM(7440-09-7) IS FOUND ON THE FOLLOWING REGULATORY LISTS

International Air Transport Association (IATA) Dangerous Goods Regulations - Prohibited List Passenger and Cargo Aircraft

US - Massachusetts - Right To Know Listed Chemicals

US - Pennsylvania - Hazardous Substance List

US - Rhode Island Hazardous Substance List

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

SELENIUM(7782-49-2) IS FOUND ON THE FOLLOWING REGULATORY LISTS

International Agency for Research on Cancer (IARC) - Agents Classified by the IARC US - Washington Permissible exposure limits of air contaminants Monographs US - Washington Toxic air pollutants and their ASIL, SQER and de minimis emission values

US - Alaska Limits for Air Contaminants US ACGIH Threshold Limit Values (TLV) US - California OEHHA/ARB - Acute Reference Exposure Levels and Target Organs (RELs) US ATSDR Minimal Risk Levels for Hazardous Substances (MRLs)

US - California OEHHA/ARB - Chronic Reference Exposure Levels and Target Organs US Clean Air Act - Hazardous Air Pollutants

(CRELs) US CWA (Clean Water Act) - Priority Pollutants US - Hawaii Air Contaminant Limits US CWA (Clean Water Act) - Toxic Pollutants

US - Idaho - Limits for Air Contaminants US EPA Carcinogens Listing

US - Massachusetts - Right To Know Listed Chemicals US EPCRA Section 313 Chemical List US - Minnesota Permissible Exposure Limits (PELs)

US NIOSH Recommended Exposure Limits (RELs) US - Pennsylvania - Hazardous Substance List US OSHA Permissible Exposure Levels (PELs) - Table Z1

US - Rhode Island Hazardous Substance List US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants US - Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants

US - Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air

SODIUM(7440-23-5) IS FOUND ON THE FOLLOWING REGULATORY LISTS

International Air Transport Association (IATA) Dangerous Goods Regulations - Prohibited List Passenger and Cargo Aircraft

US - Massachusetts - Right To Know Listed Chemicals

US - Pennsylvania - Hazardous Substance List

US - Rhode Island Hazardous Substance List

US CWA (Clean Water Act) - List of Hazardous Substances

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

AMMONIUM METAVANADATE(7803-55-6) IS FOUND ON THE FOLLOWING REGULATORY LISTS

US - California OEHHA/ARB - Acute Reference Exposure Levels and Target Organs (RELs)

US - Massachusetts - Right To Know Listed Chemicals

US - Pennsylvania - Hazardous Substance List

US EPCRA Section 313 Chemical List

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

ZINC(7440-66-6) IS FOUND ON THE FOLLOWING REGULATORY LISTS

International Agency for Research on Cancer (IARC) - Agents Classified by the IARC Monographs

US - California OEHHA/ARB - Chronic Reference Exposure Levels and Target Organs (CRELs)

US - California Permissible Exposure Limits for Chemical Contaminants

US - Hawaii Air Contaminant Limits

US - Massachusetts - Right To Know Listed Chemicals

US - Michigan Exposure Limits for Air Contaminants

US - Oregon Permissible Exposure Limits (Z-1) US - Pennsylvania - Hazardous Substance List

US - Rhode Island Hazardous Substance List

US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants

US - Washington Permissible exposure limits of air contaminants

US - Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants

US ATSDR Minimal Risk Levels for Hazardous Substances (MRLs)

US CWA (Clean Water Act) - Priority Pollutants

US CWA (Clean Water Act) - Toxic Pollutants

US EPA Carcinogens Listing

US EPCRA Section 313 Chemical List

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

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NITRIC ACID(7697-37-2) IS FOUND ON THE FOLLOWING REGULATORY LISTS

International Air Transport Association (IATA) Dangerous Goods Regulations - Prohibited List Passenger and Cargo Aircraft

- US Alaska Limits for Air Contaminants
- US California OEHHA/ARB Acute Reference Exposure Levels and Target Organs (RELs)
- US California Permissible Exposure Limits for Chemical Contaminants
- US Hawaii Air Contaminant Limits
- US Idaho Limits for Air Contaminants
- US Massachusetts Right To Know Listed Chemicals
- US Michigan Exposure Limits for Air Contaminants
- US Minnesota Permissible Exposure Limits (PELs)
- US Oregon Permissible Exposure Limits (Z-1)
- US Pennsylvania Hazardous Substance List
- US Rhode Island Hazardous Substance List
- US Tennessee Occupational Exposure Limits Limits For Air Contaminants

- US Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants
- US Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air Contaminants
- US Washington Permissible exposure limits of air contaminants
- US Washington Toxic air pollutants and their ASIL, SQER and de minimis emission values
- US Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants
- US ACGIH Threshold Limit Values (TLV)
- US CWA (Clean Water Act) List of Hazardous Substances
- US EPCRA Section 313 Chemical List
- US NIOSH Recommended Exposure Limits (RELs)
- US OSHA Permissible Exposure Levels (PELs) Table Z1
- US SARA Section 302 Extremely Hazardous Substances US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

WATER(7732-18-5) IS FOUND ON THE FOLLOWING REGULATORY LISTS

US - Pennsylvania - Hazardous Substance List

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

AMMONIUM PHOSPHATE, MONOBASIC(7722-76-1) IS FOUND ON THE FOLLOWING REGULATORY LISTS

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

AMMONIUM FLUOROSILICATE(16919-19-0) IS FOUND ON THE FOLLOWING REGULATORY LISTS

International Agency for Research on Cancer (IARC) - Agents Classified by the IARC Monographs

- US California OEHHA/ARB Chronic Reference Exposure Levels and Target Organs
- US Hawaii Air Contaminant Limits
- US Idaho Limits for Air Contaminants
- US Massachusetts Right To Know Listed Chemicals
- US Oregon Permissible Exposure Limits (Z-1)
- US Oregon Permissible Exposure Limits (Z-2)
- US Pennsylvania Hazardous Substance List

THORIUM OXIDE(1314-20-1) IS FOUND ON THE FOLLOWING REGULATORY LISTS

- US California Proposition 65 Carcinogens
- US Massachusetts Right To Know Listed Chemicals
- Carcinogens

- US Vermont Permissible Exposure Limits Table 7-1-A Transitional Limits for Air Contaminants
- US Washington Permissible exposure limits of air contaminants
- US Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants
- US Wyoming Toxic and Hazardous Substances Table Z-2 Acceptable ceiling concentration,
- Acceptable maximum peak above the acceptable ceiling concentration for an 8-hr shift US CWA (Clean Water Act) - List of Hazardous Substances
- US OSHA Permissible Exposure Levels (PELs) Table Z1
- US OSHA Permissible Exposure Levels (PELs) Table Z2
- US Toxic Substances Control Act (TSCA) Chemical Substance Inventory

- US New Jersey Right to Know Special Health Hazard Substance List (SHHSL):
- US Pennsylvania Hazardous Substance List

- US EPCRA Section 313 Chemical List
- US National Toxicology Program (NTP) 14th Report Part A Known to be Human Carcinogens US Priority List for the Development of Proposition 65 Safe Harbor Levels - No Significant Risk
- Levels (NSRLs) for Carcinogens and Maximum Allowable Dose Levels (MADLs) for Chemicals Causing Reproductive Toxicity
- US Toxic Substances Control Act (TSCA) Chemical Substance Inventory

BERYLLIUM ACETATE(543-81-7) IS FOUND ON THE FOLLOWING REGULATORY LISTS

International Agency for Research on Cancer (IARC) - Agents Classified by the IARC Monographs

- US Alaska Limits for Air Contaminants
- US California OEHHA/ARB Chronic Reference Exposure Levels and Target Organs (CRELs)
- US California Permissible Exposure Limits for Chemical Contaminants
- US Idaho Acceptable Maximum Peak Concentrations
- US Idaho Limits for Air Contaminants
- US Michigan Exposure Limits for Air Contaminants
- US Minnesota Permissible Exposure Limits (PELs)
- US Oregon Permissible Exposure Limits (Z-1) US - Oregon Permissible Exposure Limits (Z-2)
- US Tennessee Occupational Exposure Limits Limits For Air Contaminants
- US Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants
- US Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air Contaminants

- US Washington Permissible exposure limits of air contaminants
- US Washington Toxic air pollutants and their ASIL, SQER and de minimis emission values
- US Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants
- US Wyoming Toxic and Hazardous Substances Table Z-2 Acceptable ceiling concentration, Acceptable maximum peak above the acceptable ceiling concentration for an 8-hr shift
- US ACGIH Threshold Limit Values (TLV)
- US ACGIH Threshold Limit Values (TLV) Carcinogens
- US Clean Air Act Hazardous Air Pollutants
- US CWA (Clean Water Act) Priority Pollutants
- US CWA (Clean Water Act) Toxic Pollutants
- US EPA Carcinogens Listing
- US EPCRA Section 313 Chemical List
- US National Toxicology Program (NTP) 14th Report Part A Known to be Human Carcinogens
- US OSHA Permissible Exposure Levels (PELs) Table Z1
- US OSHA Permissible Exposure Levels (PELs) Table Z2

Federal Regulations

Superfund Amendments and Reauthorization Act of 1986 (SARA)

SECTION 311/312 HAZADD CATEGODIES

SECTION 31 I/312 HAZARD CATEGORIES		
	Immediate (acute) health hazard	Yes
	Delayed (chronic) health hazard	No
	Fire hazard	No
	Pressure hazard	No
	Poortivity hazard	No

US. EPA CERCLA HAZARDOUS SUBSTANCES AND REPORTABLE QUANTITIES (40 CFR 302.4)

	,	
Name	Reportable Quantity in Pounds (lb)	Reportable Quantity in kg
Antimony	5000	2270
Arsenic	1	0.454

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10 Cadmium 4.54 Chromium 5000 2270 5000 2270 Copper Lead 10 4.54 Nickel 100 45.4 100 Selenium 45.4 10 Sodium Ammonium vanadate 1000 454 Zinc 1000 454 1000 454 Nitric acid Ammonium silicofluoride 1000 454

State Regulations

US. CALIFORNIA PROPOSITION 65

WARNING: This product contains a chemical known to the State of California to cause cancer and birth defects or other reproductive harm

US - CALIFORNIA PREPOSITION 65 - CARCINOGENS & REPRODUCTIVE TOXICITY (CRT): LISTED SUBSTANCE

Cadmium and cadmium compounds: Cadmium, Cobalt metal powder, Lead and lead compounds: Lead, Nickel (Metallic), Thorium dioxide Listed

National Inventory	Status
Australia - AICS	N (beryllium acetate)
Canada - DSL	N (beryllium acetate)
Canada - NDSL	N (sodium; lead; calcium; zinc; potassium; ammonium metavanadate; magnesium; copper; thorium oxide; water; ammonium phosphate, monobasic; antimony; selenium; ammonium fluorosilicate; aluminium; arsenic; cobalt; nickel; iron; chromium; beryllium acetate; cadmium; manganese(II) acetate; nitric acid)
China - IECSC	N (thorium oxide; beryllium acetate)
Europe - EINEC / ELINCS / NLP	Y
Japan - ENCS	N (sodium; calcium; zinc; potassium; magnesium; copper; thorium oxide; water; ammonium phosphate, monobasic; antimony; selenium; ammonium fluorosilicate; aluminium; arsenic; cobalt; nickel; iron; chromium; beryllium acetate; cadmium; manganese(II) acetate; nitric acid)
Korea - KECI	N (beryllium acetate)
New Zealand - NZIoC	N (thorium oxide; beryllium acetate)
Philippines - PICCS	N (thorium oxide; beryllium acetate; manganese(II) acetate)
USA - TSCA	N (beryllium acetate)
Legend:	Y = All ingredients are on the inventory N = Not determined or one or more ingredients are not on the inventory and are not exempt from listing(see specific ingredients in brackets)

SECTION 16 OTHER INFORMATION

Other information

Ingredients with multiple cas numbers

Name	CAS No
aluminium	7429-90-5, 91728-14-2
calcium	7440-70-2, 8047-59-4
copper	7440-50-8, 133353-46-5, 133353-47-6, 195161-80-9, 65555-90-0, 72514-83-1
ammonium fluorosilicate	16919-19-0, 1309-32-6

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references.

The SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

Definitions and abbreviations

 ${\sf PC-TWA}.\ {\sf Permissible}\ {\sf Concentration-Time}\ {\sf Weighted}\ {\sf Average}$

PC-STEL: Permissible Concentration-Short Term Exposure Limit

IARC: International Agency for Research on Cancer

ACGIH: American Conference of Governmental Industrial Hygienists

STEL: Short Term Exposure Limit

TEEL: Temporary Emergency Exposure Limit。

IDLH: Immediately Dangerous to Life or Health Concentrations

OSF: Odour Safety Factor

TLV: Threshold Limit Value

NOAEL :No Observed Adverse Effect Level LOAEL: Lowest Observed Adverse Effect Level

LOD: Limit Of Detection OTV: Odour Threshold Value BCF: BioConcentration Factors BEI: Biological Exposure Index

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