

ICP-AM-MISA5

High-Purity Standards

Catalogue number: ICP-AM-MISA5

Version No: 1.1

Safety Data Sheet according to OSHA HazCom Standard (2012) requirements

Chemwatch Hazard Alert Code: 3

Issue Date: 12/06/2016 Print Date: 06/03/2017 S GHS USA EN

SECTION 1 IDENTIFICATION

Product Identifier

Product name	ICP-AM-MISA5
Synonyms	Not Available
Proper shipping name	Corrosive liquid, acidic, inorganic, n.o.s.
Other means of identification	ICP-AM-MISA5

Recommended use of the chemical and restrictions on use

Name, address, and telephone number of the chemical manufacturer, importer, or other responsible party

Registered company name	High-Purity Standards
Address	PO Box 41727 SC 29423 United States
Telephone	843-767-7900
Fax	843-767-7906
Website	highpuritystandards.com
Email	Not Available

Emergency phone number

• • •	
Association / Organisation	INFOTRAC
Emergency telephone numbers	1-800-535-5053
Other emergency telephone numbers	1-352-323-3500

SECTION 2 HAZARD(S) IDENTIFICATION

Classification of the substance or mixture

Classification

Metal Corrosion Category 1, Skin Corrosion/Irritation Category 1A

Label elements

Hazard pictogram(s)



SIGNAL WORD

DANGER

Hazard statement(s)

nazaru Statement(s)		
H290	May be corrosive to metals.	
H314	Causes severe skin burns and eye damage.	

Hazard(s) not otherwise specified

Not Applicable

Issue Date: **12/06/2016**Print Date: **06/03/2017**

P260

Do not breathe dust/fume/gas/mist/vapours/spray.

Precautionary statement(s) Response

P301+P330+P331

IF SWALLOWED: Rinse mouth. Do NOT induce vomiting.

Precautionary statement(s) Storage

P405

Store locked up.

Precautionary statement(s) Disposal

P501

Dispose of contents/container in accordance with local regulations.

SECTION 3 COMPOSITION / INFORMATION ON INGREDIENTS

Substances

See section below for composition of Mixtures

Mixtures

CAS No	%[weight]	Name
7440-45-1	0.01	cerium
7429-91-6	0.01	dysprosium
7440-52-0	0.01	<u>erbium</u>
7440-53-1	0.01	europium
7440-54-2	0.01	gadolinium
7440-60-0	0.01	<u>holmium</u>
7439-91-0	0.01	<u>lanthanum</u>
7439-94-3	0.01	lutetium
7440-00-8	0.01	<u>neodymium</u>
7440-10-0	0.01	<u>praseodymium</u>
7440-19-9	0.01	<u>samarium</u>
7440-20-2	0.01	scandium
7440-27-9	0.01	terbium
7440-29-1	0.01	<u>thorium</u>
7440-30-4	0.01	<u>thulium</u>
7440-65-5	0.01	<u>yttrium</u>
7440-64-4	0.01	ytterbium
7697-37-2	2	nitric acid
7732-18-5	Balance	<u>water</u>

SECTION 4 FIRST-AID MEASURES

Description of first aid measures

If this product comes in contact with the eyes

Eye Contact

- Immediately hold eyelids apart and flush the eye continuously with running water.
- ▶ Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids.
- ▶ Continue flushing until advised to stop by the Poisons Information Centre or a doctor, or for at least 15 minutes.
- Transport to hospital or doctor without delay.
- ▶ Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.

If skin or hair contact occurs

- ▶ Immediately flush body and clothes with large amounts of water, using safety shower if available.
- ▶ Quickly remove all contaminated clothing, including footwear.
- ▶ Wash skin and hair with running water. Continue flushing with water until advised to stop by the Poisons Information Centre.
- ► Transport to hospital, or doctor.

For thermal burns:

- Decontaminate area around burn.
- ► Consider the use of cold packs and topical antibiotics.

For first-degree burns (affecting top layer of skin)

Hold burned skin under cool (not cold) running water or immerse in cool water until pain subsides.

Skin Contact

- ▶ Use compresses if running water is not available.
- ► Cover with sterile non-adhesive bandage or clean cloth.
- ▶ Do NOT apply butter or ointments; this may cause infection.
- ▶ Give over-the counter pain relievers if pain increases or swelling, redness, fever occur.

For second-degree burns (affecting top two layers of skin)

- ▶ Cool the burn by immerse in cold running water for 10-15 minutes.
- Use compresses if running water is not available.
- ▶ Do NOT apply ice as this may lower body temperature and cause further damage.
- Do NOT break blisters or apply butter or ointments; this may cause infection.
- ▶ Protect burn by cover loosely with sterile, nonstick bandage and secure in place with gauze or tape.

To prevent shock: (unless the person has a head, neck, or leg injury, or it would cause discomfort):

Chemwatch: 9-297043 Page 3 of 15 Catalogue number: ICP-AM-MISA5

Issue Date: 12/06/2016 Print Date: 06/03/2017 **ICP-AM-MISA5**

► Lay the person flat. ▶ Elevate feet about 12 inches. Elevate burn area above heart level, if possible. Cover the person with coat or blanket. Seek medical assistance. For third-degree burns Seek immediate medical or emergency assistance. In the mean time: Protect burn area cover loosely with sterile, nonstick bandage or, for large areas, a sheet or other material that will not leave lint in wound. Separate burned toes and fingers with dry, sterile dressings. ▶ Do not soak burn in water or apply ointments or butter; this may cause infection. To prevent shock see above. For an airway burn, do not place pillow under the person's head when the person is lying down. This can close the airway. Have a person with a facial burn sit up ▶ Check pulse and breathing to monitor for shock until emergency help arrives. ▶ If fumes or combustion products are inhaled remove from contaminated area. ▶ Lay patient down. Keep warm and rested. Prostheses such as false teeth, which may block airway, should be removed, where possible, prior to initiating first aid procedures. Apply artificial respiration if not breathing, preferably with a demand valve resuscitator, bag-valve mask device, or pocket mask as trained. Perform CPR if necessary. Transport to hospital, or doctor, without delay. Inhalation Inhalation of vapours or aerosols (mists, fumes) may cause lung oedema. ▶ Corrosive substances may cause lung damage (e.g. lung oedema, fluid in the lungs). As this reaction may be delayed up to 24 hours after exposure, affected individuals need complete rest (preferably in semi-recumbent posture) and must be kept under medical observation even if no symptoms are (yet) manifested. ▶ Before any such manifestation, the administration of a spray containing a dexamethasone derivative or beclomethasone derivative may be considered. This must definitely be left to a doctor or person authorised by him/her. (ICSC13719) ▶ For advice, contact a Poisons Information Centre or a doctor at once. ▶ Urgent hospital treatment is likely to be needed. ed do **NOT** induce vomiting ▶ If vomiting occurs, lean patient forward or place on left side (head-down position, if possible) to maintain open airway and prevent aspiration. Ingestion Observe the patient carefully. Never give liquid to a person showing signs of being sleepy or with reduced awareness; i.e. becoming unconscious. Give water to rinse out mouth, then provide liquid slowly and as much as casualty can comfortably drink. Transport to hospital or doctor without delay.

Most important symptoms and effects, both acute and delayed

Version No: 1.1

Indication of any immediate medical attention and special treatment needed

For acute or short term repeated exposures to strong acids:

- Airway problems may arise from laryngeal edema and inhalation exposure. Treat with 100% oxygen initially.
- Respiratory distress may require cricothyroidotomy if endotracheal intubation is contraindicated by excessive swelling
- Intravenous lines should be established immediately in all cases where there is evidence of circulatory compromise.
- For Strong acids produce a coagulation necrosis characterised by formation of a coagulum (eschar) as a result of the dessicating action of the acid on proteins in specific tissues.

- ▶ Immediate dilution (milk or water) within 30 minutes post ingestion is recommended.
- ▶ **DO NOT** attempt to neutralise the acid since exothermic reaction may extend the corrosive injury
- Be careful to avoid further vomit since re-exposure of the mucosa to the acid is harmful. Limit fluids to one or two glasses in an adult.
- Charcoal has no place in acid management.
- ▶ Some authors suggest the use of lavage within 1 hour of ingestion.

- > Skin lesions require copious saline irrigation. Treat chemical burns as thermal burns with non-adherent gauze and wrapping.
- ▶ Deep second-degree burns may benefit from topical silver sulfadiazine.

- Figure Eye injuries require retraction of the eyelids to ensure thorough irrigation of the conjuctival cul-de-sacs. Irrigation should last at least 20-30 minutes. DO NOT use neutralising agents or any other additives. Several litres of saline are required.
- Cycloplegic drops, (1% cyclopentolate for short-term use or 5% homatropine for longer term use) antibiotic drops, vasoconstrictive agents or artificial tears may be indicated dependent on the severity of the injury
- ▶ Steroid eye drops should only be administered with the approval of a consulting ophthalmologist).

[Ellenhorn and Barceloux: Medical Toxicology]

SECTION 5 FIRE-FIGHTING MEASURES

Extinguishing media

- ▶ There is no restriction on the type of extinguisher which may be used.
- Use extinguishing media suitable for surrounding area

Special hazards arising from the substrate or mixture

Fire Incompatibility

None known

Special protective equipment and precautions for fire-fighters

Fire Fighting

Fire/Explosion Hazard

- Non combustible.
- Not considered to be a significant fire risk.
- Acids may react with metals to produce hydrogen, a highly flammable and explosive gas.
- Heating may cause expansion or decomposition leading to violent rupture of containers.

Catalogue number: ICP-AM-MISA5

ICP-AM-MISA5

Issue Date: 12/06/2016 Print Date: 06/03/2017

▶ May emit corrosive, poisonous fumes. May emit acrid smoke.

SECTION 6 ACCIDENTAL RELEASE MEASURES

Personal precautions, protective equipment and emergency procedures

See section 8

Environmental precautions

See section 12

Methods and material for containment and cleaning up

metrious and material for t	ornamient and cleaning up
Minor Spills	 Drains for storage or use areas should have retention basins for pH adjustments and dilution of spills before discharge or disposal of material. Check regularly for spills and leaks. Clean up all spills immediately. Avoid breathing vapours and contact with skin and eyes. Control personal contact with the substance, by using protective equipment. Contain and absorb spill with sand, earth, inert material or vermiculite. Wipe up. Place in a suitable, labelled container for waste disposal.
Major Spills	#

Personal Protective Equipment advice is contained in Section 8 of the SDS.

SECTION 7 HANDLING AND STORAGE

Precautions for safe handling

- Avoid all personal contact, including inhalation.
- Wear protective clothing when risk of exposure occurs.
- Use in a well-ventilated area.
- WARNING: To avoid violent reaction, ALWAYS add material to water and NEVER water to material
- Avoid smoking, naked lights or ignition sources.
- Avoid contact with incompatible materials.
- When handling, DO NOT eat, drink or smoke Safe handling
 - Keep containers securely sealed when not in use.
 - Avoid physical damage to containers.
 - Always wash hands with soap and water after handling.
 - Work clothes should be laundered separately. Launder contaminated clothing before re-use.
 - Use good occupational work practice
 - Observe manufacturer's storage and handling recommendations contained within this SDS.
 - Atmosphere should be regularly checked against established exposure standards to ensure safe working conditions are maintained.

Other information

- · Store in original containers. Keep containers securely sealed
- Store in a cool, dry, well-ventilated area.
- Store away from incompatible materials and foodstuff containers.
- Protect containers against physical damage and check regularly for leaks.
- ▶ Observe manufacturer's storage and handling recommendations contained within this SDS

Conditions for safe storage, including any incompatibilities

► DO NOT use aluminium or galvanised containers

- Check regularly for spills and leaks
- Lined metal can, lined metal pail/ can.
- Plastic pail.
- ▶ Polvliner drum.
- ▶ Packing as recommended by manufacturer.
- ▶ Check all containers are clearly labelled and free from leaks.

For low viscosity materials

Drums and jerricans must be of the non-removable head type. Suitable container

▶ Where a can is to be used as an inner package, the can must have a screwed enclosure.

For materials with a viscosity of at least 2680 cSt. (23 deg. C) and solids (between 15 C deg. and 40 deg C.):

- Removable head packaging;
- Cans with friction closures and
- ▶ low pressure tubes and cartridges

may be used.

Where combination packages are used, and the inner packages are of glass, porcelain or stoneware, there must be sufficient inert cushioning material in contact with inner and outer packages unless the outer packaging is a close fitting moulded plastic box and the substances are not incompatible with the

Storage incompatibility

- ▶ Inorganic acids are generally soluble in water with the release of hydrogen ions. The resulting solutions have pH's of less than 7.0.
- Inorganic acids neutralise chemical bases (for example: amines and inorganic hydroxides) to form salts neutralisation can generate dangerously large amounts of heat in small spaces
- The dissolution of inorganic acids in water or the dilution of their concentrated solutions with additional water may generate significant heat.
- ▶ The addition of water to inorganic acids often generates sufficient heat in the small region of mixing to cause some of the water to boil explosively. The resulting "bumping" can spatter the acid.
- ▶ Inorganic acids react with active metals, including such structural metals as aluminum and iron, to release hydrogen, a flammable gas.
- Inorganic acids can initiate the polymerisation of certain classes of organic compounds.
- Inorganic acids react with cyanide compounds to release gaseous hydrogen cyanide.
- Inorganic acids generate flammable and/or toxic gases in contact with dithiocarbamates, isocyanates, mercaptans, nitriles, nitriles, sulfides, and strong reducing agents. Additional gas-generating reactions occur with sulfites, nitrites, thiosulfates (to give H2S and SO3), dithionites (SO2), and even carbonates.

Page **5** of **15** Issue Date: 12/06/2016 Print Date: 06/03/2017

Version No: 1.1

- ► Acids often catalyse (increase the rate of) chemical reactions.
- ▶ WARNING: Avoid or control reaction with peroxides. All transition metal peroxides should be considered as potentially explosive. For example transition metal complexes of alkyl hydroperoxides may decompose explosively.
- ▶ The pi-complexes formed between chromium(0), vanadium(0) and other transition metals (haloarene-metal complexes) and mono-or poly-fluorobenzene show extreme sensitivity to heat and are explosive.

 • Avoid reaction with borohydrides or cyanoborohydrides

SECTION 8 EXPOSURE CONTROLS / PERSONAL PROTECTION

Control parameters

OCCUPATIONAL EXPOSURE LIMITS (OEL)

INGREDIENT DATA

Source	Ingredient	Material name	TWA	STEL	Peak	Notes
US OSHA Permissible Exposure Levels (PELs) - Table Z1	yttrium	Yttrium	1 mg/m3	Not Available	Not Available	[*Note: The REL also applies to other yttrium compounds (as Y).]
US NIOSH Recommended Exposure Limits (RELs)	yttrium	Yttrium metal	1 mg/m3	Not Available	Not Available	Not Available
US OSHA Permissible Exposure Levels (PELs) - Table Z1	nitric acid	Nitric acid	5 mg/m3 / 2 ppm	10 mg/m3 / 4 ppm	Not Available	TLV® Basis: URT & eye irr; dental erosion
US NIOSH Recommended Exposure Limits (RELs)	nitric acid	Aqua fortis, Engravers acid, Hydrogen nitrate, Red furning nitric acid (RFNA), White furning nitric acid (WFNA)	5 mg/m3 / 2 ppm	4 ppm	Not Available	Not Available
US ACGIH Threshold Limit Values (TLV)	nitric acid	Nitric acid	2 ppm	Not Available	Not Available	Not Available

EMERGENCY LIMITS

Ingredient	Material name	TEEL-1	TEEL-2	TEEL-3
cerium	Cerium	30 mg/m3	330 mg/m3	2,000 mg/m3
dysprosium	Dysprosium	30 mg/m3	330 mg/m3	2,000 mg/m3
europium	Europium	30 mg/m3	330 mg/m3	2,000 mg/m3
gadolinium	Gadolinium	30 mg/m3	330 mg/m3	2,000 mg/m3
holmium	Holmium	12 mg/m3	130 mg/m3	790 mg/m3
lanthanum	Lanthanum	30 mg/m3	330 mg/m3	2,000 mg/m3
lutetium	Lutetium	30 mg/m3	330 mg/m3	2,000 mg/m3
neodymium	Neodymium	30 mg/m3	330 mg/m3	2,000 mg/m3
praseodymium	Praseodymium	1.2 mg/m3	13 mg/m3	79 mg/m3
samarium	Samarium	30 mg/m3	330 mg/m3	2,000 mg/m3
scandium	Scandium	30 mg/m3	330 mg/m3	2,000 mg/m3
terbium	Terbium	1.2 mg/m3	13 mg/m3	79 mg/m3
thorium	Thorium	30 mg/m3	330 mg/m3	2,000 mg/m3
thulium	Thulium	30 mg/m3	330 mg/m3	2,000 mg/m3
yttrium	Yttrium	3 mg/m3	33 mg/m3	200 mg/m3
nitric acid	Nitric acid	Not Available	Not Available	Not Available

Ingredient	Original IDLH	Revised IDLH
cerium	Not Available	Not Available
dysprosium	Not Available	Not Available
erbium	Not Available	Not Available
europium	Not Available	Not Available
gadolinium	Not Available	Not Available
holmium	Not Available	Not Available
lanthanum	Not Available	Not Available
lutetium	Not Available	Not Available
neodymium	Not Available	Not Available
praseodymium	Not Available	Not Available
samarium	Not Available	Not Available
scandium	Not Available	Not Available
terbium	Not Available	Not Available
thorium	Not Available	Not Available
thulium	Not Available	Not Available
yttrium	Not Available	Not Available
ytterbium	Not Available	Not Available

Catalogue number: ICP-AM-MISA5

Version No: 1.1

ICP-AM-MISA5

Issue Date: 12/06/2016 Print Date: 06/03/2017

nitric acid	100 ppm	25 ppm
water	Not Available	Not Available

Exposure controls

Engineering controls are used to remove a hazard or place a barrier between the worker and the hazard. Well-designed engineering controls can be highly effective in protecting workers and will typically be independent of worker interactions to provide this high level of protection.

The basic types of engineering controls are:

Process controls which involve changing the way a job activity or process is done to reduce the risk.

Enclosure and/or isolation of emission source which keeps a selected hazard "physically" away from the worker and ventilation that strategically "adds" and "removes" air in the work environment. Ventilation can remove or dilute an air contaminant if designed properly. The design of a ventilation system must match the particular process and chemical or contaminant in use.

Employers may need to use multiple types of controls to prevent employee overexposure.

Local exhaust ventilation usually required. If risk of overexposure exists, wear approved respirator. Correct fit is essential to obtain adequate protection. Supplied-air type respirator may be required in special circumstances. Correct fit is essential to ensure adequate protection.

An approved self contained breathing apparatus (SCBA) may be required in some situations.

Provide adequate ventilation in warehouse or closed storage area. Air contaminants generated in the workplace possess varying "escape" velocities which, in turn, determine the "capture velocities" of fresh circulating air required to effectively remove the contaminant.

Appropriate engineering controls

Type of Contaminant:	Air Speed:
solvent, vapours, degreasing etc., evaporating from tank (in still air).	0.25-0.5 m/s (50-100 f/min.)
aerosols, fumes from pouring operations, intermittent container filling, low speed conveyer transfers, welding, spray drift, plating acid fumes, pickling (released at low velocity into zone of active generation)	0.5-1 m/s (100-200 f/min.)
direct spray, spray painting in shallow booths, drum filling, conveyer loading, crusher dusts, gas discharge (active generation into zone of rapid air motion)	1-2.5 m/s (200-500 f/min.)
grinding, abrasive blasting, tumbling, high speed wheel generated dusts (released at high initial velocity into zone of very high rapid air motion).	2.5-10 m/s (500-2000 f/min.)

Within each range the appropriate value depends on:

Lower end of the range	Upper end of the range
1: Room air currents minimal or favourable to capture	1: Disturbing room air currents
2: Contaminants of low toxicity or of nuisance value only.	2: Contaminants of high toxicity
3: Intermittent, low production.	3: High production, heavy use
4: Large hood or large air mass in motion	4: Small hood-local control only

Simple theory shows that air velocity falls rapidly with distance away from the opening of a simple extraction pipe. Velocity generally decreases with the square of distance from the extraction point (in simple cases). Therefore the air speed at the extraction point should be adjusted, accordingly, after reference to distance from the contaminating source. The air velocity at the extraction fan, for example, should be a minimum of 1-2 m/s (200-400 f/min) for extraction of solvents generated in a tank 2 meters distant from the extraction point. Other mechanical considerations, producing performance deficits within the extraction apparatus, make it essential that theoretical air velocities are multiplied by factors of 10 or more when extraction systems are installed or used.

Personal protection











- Safety glasses with unperforated side shields may be used where continuous eye protection is desirable, as in laboratories; spectacles are not sufficient where complete eye protection is needed such as when handling bulk-quantities, where there is a danger of splashing, or if the material may be under
- Chemical goggles.whenever there is a danger of the material coming in contact with the eyes; goggles must be properly fitted.
- Full face shield (20 cm, 8 in minimum) may be required for supplementary but never for primary protection of eyes; these afford face protection.
- Alternatively a gas mask may replace splash goggles and face shields. Eve and face protection
 - Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants. A written policy document, describing the wearing of lenses or restrictions on use, should be created for each workplace or task. This should include a review of lens absorption and adsorption for the class of chemicals in use and an account of injury experience. Medical and first-aid personnel should be trained in their removal and suitable equipment should be readily available. In the event of chemical exposure, begin eye irrigation immediately and remove contact lens as soon as practicable. Lens should be removed at the first signs of eye redness or irritation - lens should be removed in a clean environment only after workers have washed hands thoroughly. [CDC NIOSH Current Intelligence Bulletin 59], [AS/NZS 1336 or national equivalent]

See Hand protection below Skin protection

Hands/feet protection

- Elbow length PVC gloves
- ▶ When handling corrosive liquids, wear trousers or overalls outside of boots, to avoid spills entering boots.

Body protection

See Other protection below

Other protection

- Overalls PVC Apron.
- - ▶ PVC protective suit may be required if exposure severe.
 - Eyewash unit
 - ▶ Ensure there is ready access to a safety shower

Thermal hazards

Not Available

Respiratory protection

Type A Filter of sufficient capacity. (AS/NZS 1716 & 1715, EN 143:2000 & 149:2001, ANSI Z88 or national equivalent)

SECTION 9 PHYSICAL AND CHEMICAL PROPERTIES

Catalogue number: ICP-AM-MISA5

ICP-AM-MISA5

Issue Date: 12/06/2016 Print Date: 06/03/2017

Information on basic physical and chemical properties

Appearance	Colourless		
Physical state	Liquid	Relative density (Water = 1)	Not Available
Odour	Not Available	Partition coefficient n-octanol / water	Not Available
Odour threshold	Not Available	Auto-ignition temperature (°C)	Not Available
pH (as supplied)	Not Available	Decomposition temperature	Not Available
Melting point / freezing point (°C)	Not Available	Viscosity (cSt)	Not Available
Initial boiling point and boiling range (°C)	Not Available	Molecular weight (g/mol)	Not Available
Flash point (°C)	Not Available	Taste	Not Available
Evaporation rate	Not Available	Explosive properties	Not Available
Flammability	Not Available	Oxidising properties	Not Available
Upper Explosive Limit (%)	Not Available	Surface Tension (dyn/cm or mN/m)	Not Available
Lower Explosive Limit (%)	Not Available	Volatile Component (%vol)	Not Available
Vapour pressure (kPa)	Not Available	Gas group	Not Available
Solubility in water (g/L)	Miscible	pH as a solution (1%)	Not Available
Vapour density (Air = 1)	Not Available	VOC g/L	Not Available

SECTION 10 STABILITY AND REACTIVITY

Reactivity	See section 7
Chemical stability	► Contact with alkaline material liberates heat
Possibility of hazardous reactions	See section 7
Conditions to avoid	See section 7
Incompatible materials	See section 7
Hazardous decomposition products	See section 5

SECTION 11 TOXICOLOGICAL INFORMATION

Information on toxicological effects

Inhaled	The material can cause respiratory irritation in some persons. The body's response to such irritation can cause further lung damage. Corrosive acids can cause irritation of the respiratory tract, with coughing, choking and mucous membrane damage. There may be dizziness, headache, nausea and weakness. The material has NOT been classified by EC Directives or other classification systems as "harmful by inhalation". This is because of the lack of corroborating animal or human evidence. Exposure to vapours of some rare earth salts can cause sensitivity to heat, itching, and increased sensitivity of smell and taste. Other effects include inflamed airways and lung, emphysema, regional narrowing of terminal airways and cell changes.
Ingestion	Ingestion of acidic corrosives may produce burns around and in the mouth, the throat and oesophagus. Immediate pain and difficulties in swallowing and speaking may also be evident. Dysprosium is a rare earth metal - heavy type (yttrium family). There have been no reports of poisoning in workers, although the metal can cause chest X-ray abnormalities due to its high density. The material has NOT been classified by EC Directives or other classification systems as "harmful by ingestion". This is because of the lack of corroborating animal or human evidence.
Skin Contact	Skin contact with acidic corrosives may result in pain and burns; these may be deep with distinct edges and may heal slowly with the formation of scar tissue. Skin contact is not thought to have harmful health effects (as classified under EC Directives); the material may still produce health damage following entry through wounds, lesions or abrasions. Open cuts, abraded or irritated skin should not be exposed to this material Entry into the blood-stream, through, for example, cuts, abrasions or lesions, may produce systemic injury with harmful effects. Examine the skin prior to the use of the material and ensure that any external damage is suitably protected.
Еуе	If applied to the eyes, this material causes severe eye damage. Direct eye contact with acid corrosives may produce pain, tears, sensitivity to light and burns. Mild burns of the epithelia generally recover rapidly and completely.
Chronic	Repeated or prolonged exposure to acids may result in the erosion of teeth, swelling and/or ulceration of mouth lining. Irritation of airways to lung, with cough, and inflammation of lung tissue often occurs. Long-term exposure to respiratory irritants may result in airways disease, involving difficulty breathing and related whole-body problems. Substance accumulation, in the human body, may occur and may cause some concern following repeated or long-term occupational exposure. Thulium is a rare earth metal - heavy type (yttrium family). There have been no reports of poisoning in workers, although the metal can cause chest X-ray abnormalities due to its high density. Yttrium is a rare earth metal - heavy type (yttrium family). There have been no reports of poisoning in workers, although the metal can cause chest X-ray abnormalities due to its high density. Dysprosium is a rare earth metal - heavy type (yttrium family). There have been no reports of poisoning in workers, although the metal can cause chest X-ray abnormalities due to its high density.

Issue Date: 12/06/2016 Print Date: 06/03/2017

TOXICITY IRRITATION ICP-AM-MISA5 Not Available Not Available TOXICITY IRRITATION cerium Oral (rat) LD50: >5000 mg/kg^[1] Not Available TOXICITY IRRITATION dysprosium Not Available Not Available TOXICITY IRRITATION erbium Not Available Not Available TOXICITY IRRITATION europium Not Available Not Available TOXICITY IRRITATION gadolinium Not Available Not Available TOXICITY IRRITATION holmium Not Available Not Available TOXICITY IRRITATION lanthanum Not Available Not Available TOXICITY IRRITATION lutetium Not Available Not Available TOXICITY IRRITATION neodymium Not Available Not Available TOXICITY IRRITATION praseodymium Not Available Not Available TOXICITY IRRITATION samarium Not Available Not Available TOXICITY IRRITATION scandium Not Available Not Available TOXICITY IRRITATION terbium Not Available Not Available TOXICITY IRRITATION thorium Not Available Not Available TOXICITY IRRITATION thulium Not Available Not Available TOXICITY IRRITATION yttrium Not Available Not Available

Catalogue number: ICP-AM-MISA5 Version No: 1.1

Issue Date: 12/06/2016 Print Date: 06/03/2017 **ICP-AM-MISA5**

	TOVICITY	IDDITATION	
ytterbium	TOXICITY	IRRITATION	
	Not Available	Not Available	
nitrio opid	TOXICITY		IRRITATION
nitric acid	Inhalation (rat) LC50: 625 ppm/1h*t ^[2]		Not Available
	TOXICITY	IRRITATION	
water	Not Available	Not Available	
Legend:	Value obtained from Europe ECHA Registered Substances - Acute toxicity:	2.* Value obtained from manufact	turer's SDS. Unless otherwise specified data
	extracted from RTECS - Register of Toxic Effect of chemical Substances		·
NEODYMIUM	Nausea and vomiting, gastrointestinal changes, sweating recorded.		
SCANDIUM	Scandium metal on its own is not considered to be toxic.		
	Thorium and its compounds are mainly alpha particle emitters although beta an	nd gamma	
	radiation is also encountered The radiological danger is considerably more serious than the chemical danger	er in view of the long time that all t	thorium compounds remain in the organs where
THORIUM	they are deposited (mainly in bones, lungs, lymphatic glands etc.) leading to lead to l		
	WARNING: This substance has been classified by the IARC as Group 1: CA	RCINOGENIC TO HUMANS.	
	For acid mists, aerosols, vapours		
	Test results suggest that eukaryotic cells are susceptible to genetic damage w	•	
NITRIC ACID	The material may produce severe irritation to the eye causing pronounced infla The material may produce respiratory tract irritation, and result in damage to t		ınction.
	The material may cause severe skin irritation after prolonged or repeated expo	sure and may produce on contac	t skin redness, swelling, the production of
	vesicles, scaling and thickening of the skin. Oral (?) LD50: 50-500 mg/kg * [Various Manufacturers]		
CERIUM & ERBIUM &			
EUROPIUM & GADOLINIUM & HOLMIUM			
& LANTHANUM &	Lanthanide poisoning causes immediate defaecation, writhing, inco-ordination	laboured breathing, and inactivit	v
NEODYMIUM & PRASEODYMIUM &		,g,	,
SAMARIUM & THULIUM &			
YTTRIUM & YTTERBIUM			
CERIUM & DYSPROSIUM & ERBIUM & EUROPIUM &			
GADOLINIUM & HOLMIUM			
& LANTHANUM & LUTETIUM & NEODYMIUM	No starting and a starting to the starting and a st		
& PRASEODYMIUM &	No significant acute toxicological data identified in literature search.		
SAMARIUM & TERBIUM & THORIUM & THULIUM &			
YTTRIUM & YTTERBIUM &			
WATER SCANDIUM & THULIUM &			
NITRIC ACID	Asthma-like symptoms may continue for months or even years after exposure to	the material ends.	
YTTRIUM & YTTERBIUM	For typical lanthanides:		
	Symptoms of toxicity from rare earth elements include writhing, inco-ordination	i, laboured breatning, and sedatio	n.
Acute Toxicity	0	Carcinogenicity	
Skin Irritation/Corrosion	✓	Reproductivity 🛇	
Serious Eye Damage/Irritation	⊗ S ТОТ - S	ingle Exposure	
Respiratory or Skin sensitisation	STOT - Rep	eated Exposure	
Mutagenicity	○ As	piration Hazard	
			able but does not fill the criteria for classification

One - Data Not Available to make classification

SECTION 12 ECOLOGICAL INFORMATION

Toxicity

IOD AM MICAE	ENDPOINT	TEST DURATION (HR)	SPECIES	VALUE	SOURCE
ICP-AM-MISA5	Not Applicable	Not Applicable	Not Applicable	Not Applicable	Not Applicable

ICP-AM-MISA5

Issue Date: 12/06/2016 Print Date: 06/03/2017

TEST DURATION (HR) **ENDPOINT** SPECIES VALUE SOURCE cerium Not Applicable Not Applicable Not Applicable Not Applicable Not Applicable ENDPOINT **TEST DURATION (HR) SPECIES** VALUE SOURCE dysprosium Not Applicable Not Applicable Not Applicable Not Applicable Not Applicable ENDPOINT **TEST DURATION (HR) SPECIES** VALUE SOURCE erbium Not Applicable Not Applicable Not Applicable Not Applicable Not Applicable **ENDPOINT TEST DURATION (HR) SPECIES** VALUE SOURCE europium Not Applicable Not Applicable Not Applicable Not Applicable Not Applicable **ENDPOINT TEST DURATION (HR) SPECIES** VALUE SOURCE gadolinium Not Applicable Not Applicable Not Applicable Not Applicable Not Applicable **TEST DURATION (HR) ENDPOINT SPECIES** VALUE SOURCE holmium Not Applicable Not Applicable Not Applicable Not Applicable Not Applicable ENDPOINT TEST DURATION (HR) **SPECIES** VALUE SOURCE lanthanum EC50 Crustacea 1352mg/L 5 ENDPOINT **TEST DURATION (HR) SPECIES** VALUE SOURCE lutetium Not Applicable Not Applicable Not Applicable Not Applicable Not Applicable ENDPOINT TEST DURATION (HR) **SPECIES** VALUE SOURCE neodymium Not Applicable Not Applicable Not Applicable Not Applicable Not Applicable **ENDPOINT TEST DURATION (HR) SPECIES** VALUE SOURCE praseodymium Not Applicable Not Applicable Not Applicable Not Applicable Not Applicable **ENDPOINT** TEST DURATION (HR) **SPECIES** VALUE SOURCE samarium Not Applicable Not Applicable Not Applicable Not Applicable Not Applicable **ENDPOINT TEST DURATION (HR) SPECIES** VALUE SOURCE scandium Not Applicable Not Applicable Not Applicable Not Applicable Not Applicable **ENDPOINT TEST DURATION (HR) SPECIES** VALUE SOURCE terbium Not Applicable Not Applicable Not Applicable Not Applicable Not Applicable **ENDPOINT TEST DURATION (HR) SPECIES** VALUE SOURCE thorium Not Applicable Not Applicable Not Applicable Not Applicable Not Applicable **ENDPOINT TEST DURATION (HR) SPECIES** VALUE SOURCE thulium Not Applicable Not Applicable Not Applicable Not Applicable Not Applicable **ENDPOINT TEST DURATION (HR) SPECIES** VALUE SOURCE yttrium Not Applicable Not Applicable Not Applicable Not Applicable Not Applicable **ENDPOINT** SOURCE **TEST DURATION (HR) SPECIES VALUE** ytterbium Not Applicable Not Applicable Not Applicable Not Applicable Not Applicable

ICP-AM-MISA5

Issue Date: **12/06/2016**Print Date: **06/03/2017**

	ENDPOINT	TEST DURATION (HR)		SPECIES	VALUE	SOURCE
nitric acid	NOEC	16		Crustacea	107mg/L	4
	ENDPOINT	TEST DURATION (HR)	SPECIES	VA	ALUE	SOURCE
water	Not Applicable	Not Applicable	Not Applic	able No	ot Applicable	Not Applicable
Legend:	(QSAR) - Aquatic Toxicity D	xicity Data 2. Europe ECHA Registered Subs ata (Estimated) 4. US EPA, Ecotox database lata 7. METI (Japan) - Bioconcentration Data	- Aquatic To	oxicity Data 5. ECET	, ,	

Ecotoxicity:

The tolerance of water organisms towards pH margin and variation is diverse. Recommended pH values for test species listed in OECD guidelines are between 6.0 and almost 9. Acute testing with fish showed 96h-LC50 at about pH 3.5

For Lanthanoids (Formerly Lanthanides: Synonym Rare Earth Metals and their Salts):

Environmental Fate: Rare earths, such as the lanthanoids, are relatively abundant in the crust of the Earth. These elements are not • rare • -scientists once thought these substances were only found in very small amounts on the Earth Most of the lanthanides occur together in nature, and they are very difficult to separate from each other. The lanthanides form alloys, (mixtures), with many other metals, and these alloys exhibit a wide range of physical properties. Lanthanoid emissions to the environment have increased as a result of the growing industrial applications of these elements; however, robust data to evaluate the environmental fate of lanthanoids are scarce.

Atmospheric Fate: These substances react with oxygen in the atmosphere to form an oxide residue which tarnishes surfaces exposed to these elements. They burn readily in air to form oxides. Terrestrial Fate: Soil - Lanthanoids can be found in most soils. These substances are expected to strongly sorb to soil and are not expected to evaporate from soil surfaces. Plants • These substances are expected to accumulate in plants, especially duckweed.

Aquatic Fate: Rare earth chlorides are very poorly soluble in water. These substances will bind to carbonated and dissolved organic matter in water. The lanthanides react slowly with cold water and more rapidly with hot water to form hydrogen gas. The lanthanum ion is expected to have high attraction to the negatively charged humic material present in most natural waters. This mechanism will also remove lanthanum from the water column.

Ecotoxicity: These elements have a high tendency to accumulate in plants and organisms. A typical oxide of this group, cerium oxide, has low toxicity to the fathead minnow, green algae, and Daphnia water fleas. Rare earth chlorides exhibit acute aquatic toxicity at concentrations exceeding 100 ppm and chronic toxicity, persisting for more than 21 days, at concentrations greater than 30 ppm. Industrial processes have little impact on altering background levels. Lanthanum 3+ is toxic to some aquatic organisms. Dissolved lanthanum is very toxic to species of Daphnia in both chronic and acute tests and may also be toxic to other species. There seems little doubt that dissolved lanthanum has at least high acute and chronic toxicity to fresh water fish and to various species of Daphnia in soft water, although water quality appears to have a very large effect on the toxicity.

Prevent, by any means available, spillage from entering drains or water courses.

DO NOT discharge into sewer or waterways.

Persistence and degradability

Ingredient	Persistence: Water/Soil	Persistence: Air
water	LOW	LOW

Bioaccumulative potential

Ingredient	Bioaccumulation
water	LOW (LogKOW = -1.38)

Mobility in soil

Ingredient	Mobility
water	LOW (KOC = 14.3)

SECTION 13 DISPOSAL CONSIDERATIONS

Waste treatment methods

► Recycle wherever possible.

▶ Consult manufacturer for recycling options or consult local or regional waste management authority for disposal if no suitable treatment or disposal facility

Product / Packaging disposal

- can be identified.

 Treat and neutralise at an approved treatment plant. Treatment should involve: Neutralisation with soda-ash or soda-lime followed by: burial in a land-fill
- Treat and neutralise at an approved treatment plant. Treatment should involve: Neutralisation with soda-ash or soda-lime followed by: burial in a land-fill specifically licensed to accept chemical and / or pharmaceutical wastes or Incineration in a licensed apparatus (after admixture with suitable combustible material).
- ► Decontaminate empty containers with 5% aqueous sodium hydroxide or soda ash, followed by water. Observe all label safeguards until containers are cleaned and destroyed.

SECTION 14 TRANSPORT INFORMATION

Labels Required



Marine Pollutant

NO

Land transport (DOT)

UN number	3264
UN proper shipping name	Corrosive liquid, acidic, inorganic, n.o.s.

Issue Date: **12/06/2016**Print Date: **06/03/2017**

ICP-AM-MISA5

Transport hazard class(es)

Class 8
Subrisk Not Applicable

Packing group II

Environmental hazard Not Applicable

Special precautions for user

Hazard Label 8
Special provisions 386, B2, IB2, T11, TP2, TP27

Air transport (ICAO-IATA / DGR)

All transport (IOAO IAIA / E	- ,		
UN number	3264		
UN proper shipping name	Corrosive liquid, acidic	, inorganic, n.o.s. *	
	ICAO/IATA Class	8	
Transport hazard class(es)	ICAO / IATA Subrisk	Not Applicable	
	ERG Code	8L	
Packing group	II		
Environmental hazard	Not Applicable		
	Special provisions		A3A803
	Cargo Only Packing I	nstructions	855
	Cargo Only Maximum	Qty / Pack	30 L
Special precautions for user	Passenger and Cargo Packing Instructions		851
	Passenger and Cargo Maximum Qty / Pack		1 L
	Passenger and Cargo	Limited Quantity Packing Instructions	Y840
	Passenger and Cargo	Limited Maximum Qty / Pack	0.5 L
			-

Sea transport (IMDG-Code / GGVSee)

UN number	3264
UN proper shipping name	CORROSIVE LIQUID, ACIDIC, INORGANIC, N.O.S.
Transport hazard class(es)	IMDG Class 8 IMDG Subrisk Not Applicable
Packing group	II .
Environmental hazard	Not Applicable
Special precautions for user	EMS Number F-A, S-B Special provisions 274 Limited Quantities 1 L

Transport in bulk according to Annex II of MARPOL and the IBC code

Source	Product name	Pollution Category	Ship Type
IMO MARPOL (Annex II) - List of Noxious Liquid Substances Carried in Bulk	Nitric acid (70% and over) Nitric acid (less than 70%)	Y; Y	2 2

SECTION 15 REGULATORY INFORMATION

Safety, health and environmental regulations / legislation specific for the substance or mixture

CERIUM(7440-45-1) IS FOUND ON THE FOLLOWING REGULATORY LISTS

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

DYSPROSIUM(7429-91-6) IS FOUND ON THE FOLLOWING REGULATORY LISTS

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

ERBIUM(7440-52-0) IS FOUND ON THE FOLLOWING REGULATORY LISTS

Chemwatch: 9-297043

Version No: 1.1

Monographs

Page 13 of 15

Catalogue number: ICP-AM-MISA5

ICP-AM-MISA5

Issue Date: 12/06/2016 Print Date: 06/03/2017

International Agency for Research on Cancer (IARC) - Agents Classified by the IARC

US - California OEHHA/ARB - Chronic Reference Exposure Levels and Target Organs (CRELs)

US - California Permissible Exposure Limits for Chemical Contaminants

US - Hawaii Air Contaminant Limits

US - Michigan Exposure Limits for Air Contaminants

US - Oregon Permissible Exposure Limits (Z-1)

US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants

US - Washington Permissible exposure limits of air contaminants

US - Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

EUROPIUM(7440-53-1) IS FOUND ON THE FOLLOWING REGULATORY LISTS

International Air Transport Association (IATA) Dangerous Goods Regulations - Prohibited List Passenger and Cargo Aircraft

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

GADOLINIUM(7440-54-2) IS FOUND ON THE FOLLOWING REGULATORY LISTS

International Air Transport Association (IATA) Dangerous Goods Regulations - Prohibited List Passenger and Cargo Aircraft

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

HOLMIUM(7440-60-0) IS FOUND ON THE FOLLOWING REGULATORY LISTS

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

LANTHANUM(7439-91-0) IS FOUND ON THE FOLLOWING REGULATORY LISTS

International Air Transport Association (IATA) Dangerous Goods Regulations - Prohibited List Passenger and Cargo Aircraft

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

LUTETIUM(7439-94-3) IS FOUND ON THE FOLLOWING REGULATORY LISTS

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

NEODYMIUM(7440-00-8) IS FOUND ON THE FOLLOWING REGULATORY LISTS

PRASEODYMIUM(7440-10-0) IS FOUND ON THE FOLLOWING REGULATORY LISTS

International Air Transport Association (IATA) Dangerous Goods Regulations - Prohibited List

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

International Air Transport Association (IATA) Dangerous Goods Regulations - Prohibited List Passenger and Cargo Aircraft

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

SAMARIUM(7440-19-9) IS FOUND ON THE FOLLOWING REGULATORY LISTS

International Agency for Research on Cancer (IARC) - Agents Classified by the IARC Monographs

US - California OEHHA/ARB - Chronic Reference Exposure Levels and Target Organs (CRELs)

US - California Permissible Exposure Limits for Chemical Contaminants

US - Hawaii Air Contaminant Limits

Passenger and Cargo Aircraft

US - Michigan Exposure Limits for Air Contaminants

US - Oregon Permissible Exposure Limits (Z-1)

US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants

US - Washington Permissible exposure limits of air contaminants

US - Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

SCANDIUM(7440-20-2) IS FOUND ON THE FOLLOWING REGULATORY LISTS

International Agency for Research on Cancer (IARC) - Agents Classified by the IARC Monographs

US - California OEHHA/ARB - Chronic Reference Exposure Levels and Target Organs (CRELs)

US - California Permissible Exposure Limits for Chemical Contaminants

US - Hawaii Air Contaminant Limits

US - Michigan Exposure Limits for Air Contaminants

US - Oregon Permissible Exposure Limits (Z-1)

US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants

US - Washington Permissible exposure limits of air contaminants

US - Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

TERBIUM(7440-27-9) IS FOUND ON THE FOLLOWING REGULATORY LISTS

International Agency for Research on Cancer (IARC) - Agents Classified by the IARC Monographs

US - California OEHHA/ARB - Chronic Reference Exposure Levels and Target Organs (CRELs)

US - California Permissible Exposure Limits for Chemical Contaminants

US - Hawaii Air Contaminant Limits

US - Michigan Exposure Limits for Air Contaminants

- US Oregon Permissible Exposure Limits (Z-1)
- US Tennessee Occupational Exposure Limits Limits For Air Contaminants
- US Washington Permissible exposure limits of air contaminants
- US Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants
- US Toxic Substances Control Act (TSCA) Chemical Substance Inventory

THORIUM(7440-29-1) IS FOUND ON THE FOLLOWING REGULATORY LISTS

International Agency for Research on Cancer (IARC) - Agents Classified by the IARC Monographs

US - California Proposition 65 - Carcinogens

US - Pennsylvania - Hazardous Substance List

US Clean Air Act - Hazardous Air Pollutants

US Priority List for the Development of Proposition 65 Safe Harbor Levels - No Significant Risk Levels (NSRLs) for Carcinogens and Maximum Allowable Dose Levels (MADLs) for Chemicals Causing Reproductive Toxicity

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

THULIUM(7440-30-4) IS FOUND ON THE FOLLOWING REGULATORY LISTS

International Agency for Research on Cancer (IARC) - Agents Classified by the IARC Monographs

US - California OEHHA/ARB - Chronic Reference Exposure Levels and Target Organs (CRFLs)

US - California Permissible Exposure Limits for Chemical Contaminants

US - Hawaii Air Contaminant Limits

US - Michigan Exposure Limits for Air Contaminants

US - Oregon Permissible Exposure Limits (Z-1)

US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants

US - Washington Permissible exposure limits of air contaminants

US - Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

YTTRIUM(7440-65-5) IS FOUND ON THE FOLLOWING REGULATORY LISTS

Chemwatch: **9-297043** Page **14** of **15**

Catalogue number: ICP-AM-MISA5

Version No: 1.1

ICP-AM-MISA5

Issue Date: **12/06/2016**Print Date: **06/03/2017**

International Air Transport Association (IATA) Dangerous Goods Regulations - Prohibited List US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants Passenger and Cargo Aircraft US - Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants US - Alaska Limits for Air Contaminants US - Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air US - Hawaii Air Contaminant Limits Contaminants US - Washington Permissible exposure limits of air contaminants US - Idaho - Limits for Air Contaminants US - Massachusetts - Right To Know Listed Chemicals US - Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants US - Michigan Exposure Limits for Air Contaminants US ACGIH Threshold Limit Values (TLV) US - Minnesota Permissible Exposure Limits (PELs) US NIOSH Recommended Exposure Limits (RELs) US - Oregon Permissible Exposure Limits (Z-1) US OSHA Permissible Exposure Levels (PELs) - Table Z1

YTTERBIUM(7440-64-4) IS FOUND ON THE FOLLOWING REGULATORY LISTS

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

US - Pennsylvania - Hazardous Substance List

US - Rhode Island Hazardous Substance List

NITRIC ACID(7697-37-2) IS FOUND ON THE FOLLOWING REGULATORY LISTS

International Air Transport Association (IATA) Dangerous Goods Regulations - Prohibited List Passenger and Cargo Aircraft		
US - Alaska Limits for Air Contaminants		
US - California OEHHA/ARB - Acute Reference Exposure Levels and Target Organs (RELs)		
US - California Permissible Exposure Limits for Chemical Contaminants		
US - Hawaii Air Contaminant Limits		
US - Idaho - Limits for Air Contaminants		
US - Massachusetts - Right To Know Listed Chemicals		
US - Michigan Exposure Limits for Air Contaminants		
US - Minnesota Permissible Exposure Limits (PELs)		
US - Oregon Permissible Exposure Limits (Z-1)		
US - Pennsylvania - Hazardous Substance List		
US - Rhode Island Hazardous Substance List		
US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants		

- US Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants
- US Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air Contaminants

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

- US Washington Permissible exposure limits of air contaminants
- US Washington Toxic air pollutants and their ASIL, SQER and de minimis emission values
- US Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants
- US ACGIH Threshold Limit Values (TLV)
- US CWA (Clean Water Act) List of Hazardous Substances
- US EPCRA Section 313 Chemical List
- US NIOSH Recommended Exposure Limits (RELs)
- US OSHA Permissible Exposure Levels (PELs) Table Z1
- US SARA Section 302 Extremely Hazardous Substances
- US Toxic Substances Control Act (TSCA) Chemical Substance Inventory

WATER(7732-18-5) IS FOUND ON THE FOLLOWING REGULATORY LISTS

US - Pennsylvania - Hazardous Substance List

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

Federal Regulations

Superfund Amendments and Reauthorization Act of 1986 (SARA)

SECTION 311/312 HAZARD CATEGORIES

Immediate (acute) health hazard	Yes
Delayed (chronic) health hazard	No
Fire hazard	No
Pressure hazard	No
Reactivity hazard	No

US. EPA CERCLA HAZARDOUS SUBSTANCES AND REPORTABLE QUANTITIES (40 CFR 302.4)

Name	Reportable Quantity in Pounds (lb)	Reportable Quantity in kg
Nitric acid	1000	454

State Regulations

US. CALIFORNIA PROPOSITION 65

WARNING: This product contains a chemical known to the State of California to cause cancer and birth defects or other reproductive harm

US - CALIFORNIA PREPOSITION 65 - CARCINOGENS & REPRODUCTIVE TOXICITY (CRT): LISTED SUBSTANCE

Radionuclides Listed

National Inventory	Status
Australia - AICS	N (ytterbium; erbium; terbium; dysprosium; europium; yttrium; lutetium; holmium; gadolinium; thulium)
Canada - DSL	N (erbium; terbium; dysprosium; europium; holmium; gadolinium; scandium)
Canada - NDSL	N (ytterbium; cerium; neodymium; water; thorium; lanthanum; samarium; yttrium; praseodymium; lutetium; thulium; nitric acid)
China - IECSC	N (ytterbium; erbium; terbium; cerium; thorium; lanthanum; europium; praseodymium; lutetium; holmium; gadolinium; thulium)
Europe - EINEC / ELINCS / NLP	Y
Japan - ENCS	N (ytterbium; erbium; terbium; cerium; dysprosium; neodymium; water; thorium; lanthanum; europium; yttrium; praseodymium; lutetium; holmium; gadolinium; scandium; thulium; nitric acid)
Korea - KECI	N (erbium; dysprosium; thorium; praseodymium; lutetium; holmium; thulium)
New Zealand - NZIoC	N (ytterbium; erbium; terbium; dysprosium; europium; lutetium; holmium; gadolinium; scandium; thulium)
Philippines - PICCS	N (ytterbium; erbium; terbium; dysprosium; neodymium; europium; yttrium; praseodymium; lutetium; holmium; gadolinium; scandium; thulium)
USA - TSCA	Y
Legend:	Y = All ingredients are on the inventory N = Not determined or one or more ingredients are not on the inventory and are not exempt from listing(see specific ingredients in brackets)

Chemwatch: 9-297043

Catalogue number: ICP-AM-MISA5

Version No: 1.1

Page 15 of 15

Issue Date: **12/06/2016**Print Date: **06/03/2017**

SECTION 16 OTHER INFORMATION

Other information

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references.

The SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

Definitions and abbreviations

PC-TWA: Permissible Concentration-Time Weighted Average

PC-STEL: Permissible Concentration-Short Term Exposure Limit

IARC: International Agency for Research on Cancer

ACGIH: American Conference of Governmental Industrial Hygienists

STEL: Short Term Exposure Limit

TEEL: Temporary Emergency Exposure Limit。

IDLH: Immediately Dangerous to Life or Health Concentrations

OSF: Odour Safety Factor

NOAEL: No Observed Adverse Effect Level LOAEL: Lowest Observed Adverse Effect Level

TLV: Threshold Limit Value LOD: Limit Of Detection OTV: Odour Threshold Value BCF: BioConcentration Factors BEI: Biological Exposure Index

This document is copyright.

Apart from any fair dealing for the purposes of private study, research, review or criticism, as permitted under the Copyright Act, no part may be reproduced by any process without written permission from CHEMWATCH.

TEL (+61 3) 9572 4700.