

ICP-MS-ICS Solution AB

High-Purity Standards

Catalogue number: ICP-MS-ICS Solution AB

Version No: 2.2 Safety Data Sheet according to OSHA HazCom Standard (2012) requirements

SECTION 1 IDENTIFICATION

Product Identifier

Product name	ICP-MS-ICS Solution AB
Synonyms	ICP-MS-ICS Solution AB
Proper shipping name	Corrosive liquid, acidic, inorganic, n.o.s. (contains nitric acid and hydrofluoric acid)
Other means of identification	ICP-MS-ICS Solution AB

Recommended use of the chemical and restrictions on use

Relevant identified uses Use according to manufacturer's directions.

Name, address, and telephone number of the chemical manufacturer, importer, or other responsible party

Registered company name	High-Purity Standards
Address	PO Box 41727 SC 29423 United States
Telephone	843-767-7900
Fax	843-767-7906
Website	highpuritystandards.com
Email	Not Available

Emergency phone number

• • •	
Association / Organisation	INFOTRAC
Emergency telephone numbers	1-800-535-5053
Other emergency telephone numbers	1-352-323-3500

SECTION 2 HAZARD(S) IDENTIFICATION

Classification of the substance or mixture Classification Acute Toxicity (Oral) Category 4, Acute Toxicity (Dermal) Category 4, Metal Corrosion Category 1, Skin Corrosion/Irritation Category 1A				
abel elements				
Hazard pictogram(s)				

Hazard statement(s)

H302	Harmful if swallowed.
H312	Harmful in contact with skin.
H290	May be corrosive to metals.
H314	Causes severe skin burns and eye damage.

SIGNAL WORD

DANGER

Chemwatch Hazard Alert Code: 3

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Not Applicable					
Precautionary statement(s) Prevention				
P260	Do not breathe dust/fume/gas/mist/vapours/	'spray.			
Precautionary statement(s P301+P330+P331	IF SWALLOWED: Rinse mouth. Do NOT in	duce vomiting	j.		
Precautionary statement(s) Storage				
P405	Store locked up.				
Precautionary statement(s) Disposal				
P501	Dispose of contents/container in accordance	e with local re	egulations.		

SECTION 3 COMPOSITION / INFORMATION ON INGREDIENTS

Substances

See section below for composition of Mixtures

Mixtures

CAS No	%[weight]	Name
7429-90-5	0.05	aluminium
7440-70-2	0.05	<u>calcium</u>
7439-89-6	0.05	iron
13446-18-9	0.05 (as Mg)	magnesium nitrate
7439-98-7	0.001	molybdenum
7722-76-1	0.05 (as P)	ammonium phosphate, monobasic
7440-09-7	0.05	potassium
7440-23-5	0.05	sodium
7664-93-9	0.05 (as S)	sulfuric acid
7440-32-6	0.001	titanium
631-61-8	0.1 (as C)	ammonium acetate
12125-02-9	0.36 (as Cl)	ammonium chloride
7697-37-2	2	nitric acid
7664-39-3	0-0.49	hydrofluoric acid
7732-18-5	balance	water
7440-38-2	0.00001	arsenic
7440-22-4	0.00001	silver
7440-43-9	0.000005	cadmium
7440-48-4	0.00002	cobalt
7440-47-3	0.00001	chromium
638-38-0	0.00001 (as Mn)	manganese(II) acetate
7440-02-0	0.00002	nickel
7803-55-6	0.00002 (as V)	ammonium metavanadate
7440-66-6	0.00001	zinc
7782-49-2	0.00001	selenium
7440-50-8	0.00001	copper

SECTION 4 FIRST-AID MEASURES

Description of first aid measures

Eye Contact	 If this product comes in contact with the eyes: Immediately hold eyelids apart and flush the eye continuously with running water. Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids. Continue flushing until advised to stop by the Poisons Information Centre or a doctor, or for at least 15 minutes. Transport to hospital or doctor without delay. Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.
Skin Contact	 For thermal burns: Decontaminate area around burn. Consider the use of cold packs and topical antibiotics. For first-degree burns (affecting top layer of skin) Hold burned skin under cool (not cold) running water or immerse in cool water until pain subsides. Use compresses if running water is not available. Cover with sterile non-adhesive bandage or clean cloth.

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	 Do NOT apply butter or ointments; this may cause infection. Give over-the counter pain relievers if pain increases or swelling, redness, fever occur. For second-degree burns (affecting top two layers of skin) Cool the burn by immerse in cold running water for 10-15 minutes. Use compresses if running water is not available. Do NOT apply ice as this may lower body temperature and cause further damage. Do NOT apply ice as this may lower body temperature and cause further damage. Do NOT apply ice as this may lower body temperature and cause further damage. Protect burn by cover loosely with sterile, nonstick bandage and secure in place with gauze or tape. To prevent shock: (unless the person has a head, neck, or leg injury, or it would cause discomfort): I ay the person flat. Elevate feet about 12 inches. Elevate feet about 12 inches. Seek medical assistance. For third-degree burns Seek immediate medical or emergency assistance. In the mean time: Protect burn area cover loosely with sterile, nonstick bandage or, for large areas, a sheet or other material that will not leave lint in wound. Separate burned toes and fingers with dry, sterile dressings. Do not soak burn in water or apply ointments or butter, this may cause infection. To prevent shock see above. For an ainway burn, do not place pillow under the person's head when the person is lying down. This can close the ainway. Have and breathing to monitor for shock until emergency help arrives. If there is evidence of severe skin irritation or skin burns: Avoid further contract. Immediately remove contaminated dothing, including footwear. Fush skin under running water for 15 minutes. Avoid further contract. Immediately remove contaminated dothing, including footwear. Fush skin under running water for 15 minu
Inhalation	 Transport to hospital, or doctor, urgently. If fumes or combustion products are inhaled remove from contaminated area. Lay patient down. Keep warm and rested. Prostheses such as false teeth, which may block airway, should be removed, where possible, prior to initiating first aid procedures. Apply artificial respiration if not breathing, preferably with a demand valve resuscitator, bag-valve mask device, or pocket mask as trained. Perform CPR if necessary. Transport to hospital, or doctor, without delay. Inhalation of vapours or aerosols (mists, fumes) may cause lung oedema. Corrosive substances may cause lung damage (e.g. lung oedema, fluid in the lungs). As this reaction may be delayed up to 24 hours after exposure, affected individuals need complete rest (preferably in semi-recumbent posture) and must be kept under medical observation even if no symptoms are (yet) manifested. Before any such manifestation, the administration of a spray containing a dexamethasone derivative or beclomethasone derivative may be considered. This must definitely be left to a doctor or person authorised by him/her. (ICSC13719) For massive exposures: If dusts, vapours, aerosols, fumes or combustion products are inhaled, remove from contaminated area. Lay patient down. Keep warm and rested. Prostheses such as false teeth, which may block airway, should be removed, where possible, prior to initiating first aid procedures. Apply artificial respiration if not breathing, preferably with a demand valve resuscitator, bag-valve mask device, or pocket mask as trained. Perform CPR if necessary. If victim is conscious, give six calcium gluconate or calcium carbonate tablets in water by mouth.
Ingestion	 Transport to hospital, or doctor, urgently. For advice, contact a Poisons Information Centre or a doctor at once. Urgent hospital treatment is likely to be needed. If swallowed do NOT induce voriting. If vomiting occurs, lean patient forward or place on left side (head-down position, if possible) to maintain open airway and prevent aspiration. Observe the patient carefully. Never give liquid to a person showing signs of being sleepy or with reduced awareness; i.e. becoming unconscious. Give water to rinse out mouth, then provide liquid slowly and as much as casualty can comfortably drink. Transport to hospital or doctor without delay.

Most important symptoms and effects, both acute and delayed

See Section 11

Indication of any immediate medical attention and special treatment needed

Following acute or short term repeated exposure to hydrofluoric acid:

- Subcutaneous injections of Calcium Gluconate may be necessary around the burnt area. Continued application of Calcium Gluconate Gel or subcutaneous Calcium Gluconate should then continue for 3-4 days at a frequency of 4-6 times per day. If a "burning" sensation recurs, apply more frequently.
- Systemic effects of extensive hydrofluoric acid burns include renal damage, hypocalcaemia and consequent cardiac arrhythmias. Monitor haematological, respiratory, renal, cardiac and electrolyte status at least daily. Tests should include FBE, blood gases, chest X-ray, creatinine and electrolytes, urine output, Ca ions, Mg ions and phosphate ions. Continuous ECG monitoring may be required.
- Where serum calcium is low, or clinical, or ECG signs of hypocalcaemia develop, infusions of calcium gluconate, or if less serious, oral Sandocal, should be given. Hydrocortisone 500 mg in a four to six hourly infusion may help.
- Antibiotics should not be given as a routine, but only when indicated.
- Eye contact pain may be excruciating and 2-3 drops of 0.05% pentocaine hydrochloride may be instilled, followed by further irrigation

BIOLOGICAL EXPOSURE INDEX - BEI

These represent the determinants observed in specimens collected from a healthy worker exposed at the Exposure Standard (ES or TLV):

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Version No. 2.2						
Determinant	Index	Sampling Time	Comments			
1. Methaemoglobin in blood	1.5% of haemogle	bbin During or end of shift	B, NS, SQ			
B: Background levels occur in spe	B: Background levels occur in specimens collected from subjects NOT exposed.					
NS: Non-specific determinant; Al	so seen after exposure to other materials					
SQ: Semi-quantitative determinat	nt - Interpretation may be ambiguous; should be use	ed as a screening test or confirmatory test.				
For acute or short term repeated	exposures to fluorides:					
	stro-intestinal tract may be retarded by calcium salts					
	may be absorbed through the respiratory tract with ad 30 mins. post-exposure; 50% appears in the urine					
 For acute poisoning (endotration) 	cheal intubation if inadequate tidal volume), monitor	r breathing and evaluate/monitor blood pressure and pulse freque				
		prolongation or T-wave changes. Maintain monitor. Treat shock v	vigorously with isotonic saline (in 5% glucose) to			
restore blood volume and enhWhere evidence of hypocalca		conate (10 ml of a 10% solution) is injected to avoid tachycardia.				
	BIOLO	GICAL EXPOSURE INDEX - BEI				
These represent the determinants	s observed in specimens collected from a healthy we	orker exposed at the Exposure Standard (ES or TLV):				
Determinant	Index	Sampling Time	Comments			
Fluorides in urine	3 mg/gm creatinine	Prior to shift	B, NS			
	10mg/gm creatinine	End of shift	B, NS			
B: Background levels occur in spe	ecimens collected from subjects NOT exposed					
NS: Non-specific determinant; als	so observed after exposure to other exposures.					
SECTION 5 FIRE-FIGHTIN						
SECTION S FIRE-FIGHTIN	NG MEASURES					
Extinguishing media						
 There is no restriction on the Use extinguishing media suit 	type of extinguisher which may be used.					
 Ose exunguishing media sun 	able for surrounding area.					
Special hazards arising fro	om the substrate or mixture					
Fire Incompatibility	None known.					
Special protective equipm	ent and precautions for fire-fighters					
Fire Fighting						

SECTION 6 ACCIDENTAL RELEASE MEASURES

Personal precautions, protective equipment and emergency procedures

See section 8

Environmental precautions

See section 12

Methods and material for containment and cleaning up

Minor Spills	 Drains for storage or use areas should have retention basins for pH adjustments and dilution of spills before discharge or disposal of material. Check regularly for spills and leaks. Clean up all spills immediately. Avoid breathing vapours and contact with skin and eyes. Control personal contact with the substance, by using protective equipment. Contain and absorb spill with sand, earth, inert material or vermiculite. Wipe up. Place in a suitable, labelled container for waste disposal.
Major Spills	 Clear area of personnel and move upwind. Alert Fire Brigade and tell them location and nature of hazard. Wear breathing apparatus plus protective gloves. Prevent, by any means available, spillage from entering drains or water course. Stop leak if safe to do so. Contain spill with sand, earth or vermiculite. Collect recoverable product into labelled containers for recycling. Neutralise/decontaminate residue (see Section 13 for specific agent). Collect solid residues and seal in labelled drums for disposal. Wash area and prevent runoff into drains. After clean up operations, decontaminate and launder all protective clothing and equipment before storing and re-using.

Personal Protective Equipment advice is contained in Section 8 of the SDS.

SECTION 7 HANDLING AND STORAGE

Precautions for safe handling

Safe handling	 Avoid all personal contact, including inhalation. Wear protective clothing when risk of exposure occurs. Use in a well-ventilated area. WARNING: To avoid violent reaction, ALWAYS add material to water and NEVER water to material. Avoid smoking, naked lights or ignition sources. Avoid contact with incompatible materials. When handling, DO NOT eat, drink or smoke. Keep containers securely sealed when not in use. Avoid physical damage to containers. Always wash hands with scap and water after handling. Work clothes should be laundered separately. Launder contaminated clothing before re-use. Use good occupational work practice. Observe manufacturer's storage and handling recommendations contained within this SDS. Atmosphere should be requilarly checked against established exposure standards to ensure safe working conditions are maintained.
Other information	 Store in original containers. Keep containers securely sealed. Store in a cool, dry, well-ventilated area. Store away from incompatible materials and foodstuff containers. Protect containers against physical damage and check regularly for leaks. Observe manufacturer's storage and handling recommendations contained within this SDS.

Conditions for safe storage, including any incompatibilities

Suitable container	 DO NOT use aluminium or galvanised containers Lined metal can, lined metal pail/ can. Plastic pail. Polyliner drum. Packing as recommended by manufacturer. Check all containers are clearly labelled and free from leaks. For low viscosity materials Drums and jerricans must be of the non-removable head type. Where a can is to be used as an inner package, the can must have a screwed enclosure. For materials with a viscosity of at least 2680 cSt. (23 deg. C) and solids (between 15 C deg. and 40 deg C.): Removable head packaging; Cans with friction closures and low pressure tubes and cartridges may be used. - Where combination packages are used, and the inner packages are of glass, porcelain or stoneware, there must be sufficient inert cushioning material in contact with inner and outer packages unless the outer packaging is a close fitting moulded plastic box and the substances are not incompatible with the plastic. Material is corrosive to most metals, glass and other siliceous materials.
Storage incompatibility	For aluminas (aluminium oxide): Incompatible with hot chlorinated rubber. In the presence of chlorine titllovide may react violently and igniteMay initiate explosive polymerisation of olefin oxides including ethylene oxideProduces exothermic reaction above 200 C with halocarbons and an exothermic reaction at ambient temperatures with halocarbons in the presence of other metalsProduces exothermic reaction with oxygen diffuorideMay form explosive mixtures with oxygen diffuorideProduces exothermic reaction with oxygen diffuorideProms explosive mixtures with oxygen diffuorideProms explosive mixtures with oxygen diffuorideProduces exothermic acids and there and producing a salt Inorganic caids an expensive mixture with water with the release of hydrogen ions. The resulting solutions have pH's of less than 7.0 Inorganic caids an equation water with water with there leases of hydrogen ions. The resulting solutions have pH's of less than 7.0 Inorganic acids an equation acids in a malt paces The disolution of inorganic acids in water or the dilution of their concentrated solutions with additional water may generate significant heat The addition of water to inorganic acids in water or the dilution of their concentrated solutions with additional water may generate significant heat The addition of water to inorganic acids in water or the dilution of their concentrated solutions with additional water may generate solutions with additional water may generate significant heat The addition of water to inorganic acids in water or the dilution of their concentrated solutions with additional water may generate significant heat The addition of water to inorganic acids in a water or the dilution dista as a luminum and iron, to release hydrogen, a flammable gas Inorganic acids react with cyanic compounds to release geaseus

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	 Acetic acid: vapours forms explosive mixtures with air reacts violently with bases such as carbor tert-butoxide reacts (sometimes violently), with strong ammonia, ammonium nitrate, bromine pe isocyanates, oleum, perchloric acid, perm attacks cast iron, stainless steel and othe attacks many forms of rubber, plastics ar Hydrogen fluoride: reacts violently with strong oxidisers, ace chlorosulfonic acid, cyanogen fluoride, el burst into flame), nitrogen trifluoride, N-plit tetrafluorosilicate(2-), beta-propiolactone reacts (possibly violently) with aliphatic ar epichlorohydrin, isocyanates, metal acet vinjedene fluoride attacks glass and siliceous materials, com 	zinc producing hydrogen gas which may form an explosive mixtu (above 39 C.) anates and hydroxides (giving off large quantities of heat), oxidiser acids, aliphatic amines, alkanolamines, alkylene oxides, epichloro entafluoride, chlorosulfonic acid, chromic acid, chromium trioxide, e nanganates, phosphorus isocyanate, phosphorus trichloride, sodiu er metals, forming flammable hydrogen gas di coatings atic anhydride, alkalis, 2-aminoethanol, arsenic trioxide (with gene thylenediamine, ethyleneimine, fluorine (fluorine gas reacts vigoro nenylazopiperidine, oleum, oxygen difluoride, phosphorus pentoxi i, propylene oxide, sodium, sodium tetrafluorosilicate, sulfuric acid nines, alcohols, alkanolamines, alkylene oxides, aromatic amines, ylides, metal silicides, methanesulfonic acid, nitrogen compounds ncrete, ceramics, metals (flammable hydrogen gas may be produ exception of lead, platinum, polyethylene, wax.	rs, organic amines, acetaldehyde, potassium hydrin, acetic anhydride, 2-aminoethanol, ethylenediamine, ethyleneimine, hydrogen peroxide, um peroxide, xylene ration of heat), bismuthic acid, calcium oxide, iusly with a 50% hydrofluoric acid solution and may de, potassium permanganate, potassium d, vinyl acetate amides, ammonia, ammonium hydroxide, , organic anhydrides, oxides, silicon compounds,

SECTION 8 EXPOSURE CONTROLS / PERSONAL PROTECTION

Control parameters

OCCUPATIONAL EXPOSURE LIMITS (OEL)

INGREDIENT DATA

Source	Ingredient	Material name	TWA	STEL	Peak	Notes
US OSHA Permissible Exposure Levels (PELs) - Table Z1	aluminium	Aluminum, metal	15 mg/m3	Not Available	Not Available	Total dust; (as Al)
US OSHA Permissible Exposure Levels (PELs) - Table Z1	aluminium	Aluminum, metal- Respirable fraction	5 mg/m3	Not Available	Not Available	(as Al)
US NIOSH Recommended Exposure Limits (RELs)	aluminium	Aluminium, Aluminum metal, Aluminum powder, Elemental aluminum	10 (total), 5 (resp) mg/m3	Not Available	Not Available	Not Available
US NIOSH Recommended Exposure Limits (RELs)	molybdenum	Molybdenum metal	0.5 mg/m3	Not Available	Not Available	See Appendix D
US ACGIH Threshold Limit Values (TLV)	molybdenum	Molybdenum, as Mo	Not Available	Not Available	Not Available	TLV® Basis: LRT irr
US OSHA Permissible Exposure Levels (PELs) - Table Z1	sulfuric acid	Sulfuric acid	1 mg/m3	Not Available	Not Available	TLV® Basis: Pulm func
US NIOSH Recommended Exposure Limits (RELs)	sulfuric acid	Battery acid, Hydrogen sulfate, Oil of vitriol, Sulfuric acid (aqueous)	1 mg/m3	Not Available	Not Available	Not Available
US ACGIH Threshold Limit Values (TLV)	sulfuric acid	Sulfuric acid	0.2 mg/m3	Not Available	Not Available	Not Available
US NIOSH Recommended Exposure Limits (RELs)	ammonium chloride	Ammonium chloride, Ammonium muriate fume, Sal ammoniac fume	10 mg/m3	20 mg/m3	Not Available	TLV® Basis: Eye & URT irr
US ACGIH Threshold Limit Values (TLV)	ammonium chloride	Ammonium chloride, fume	10 mg/m3	20 mg/m3	Not Available	Not Available
US OSHA Permissible Exposure Levels (PELs) - Table Z1	nitric acid	Nitric acid	5 mg/m3 / 2 ppm	10 mg/m3 / 4 ppm	Not Available	TLV® Basis: URT & eye irr; dental erosion
US NIOSH Recommended Exposure Limits (RELs)	nitric acid	Aqua fortis, Engravers acid, Hydrogen nitrate, Red furning nitric acid (RFNA), White furning nitric acid (WFNA)	5 mg/m3 / 2 ppm	4 ppm	Not Available	Not Available
US ACGIH Threshold Limit Values (TLV)	nitric acid	Nitric acid	2 ppm	Not Available	Not Available	Not Available
US OSHA Permissible Exposure Levels (PELs) - Table Z1	hydrofluoric acid	Hydrogen fluoride	2.5 mg/m3 / 3 ppm	Not Available	5 mg/m3 / 6 ppm	See Table Z-2;(as F)
US OSHA Permissible Exposure Levels (PELs) - Table Z2	hydrofluoric acid	Hydrogen fluoride	3 ppm	Not Available	2 ppm	(Z37.28–1969)
US NIOSH Recommended Exposure Limits (RELs)	hydrofluoric acid	Anhydrous hydrogen fluoride; Aqueous hydrogen fluoride (i.e., Hydrofluoric acid); HF-A	0.5 ppm	Not Available	Not Available	[15-minute]
US ACGIH Threshold Limit Values (TLV)	hydrofluoric acid	Hydrogen fluoride, as F	Not Available	Not Available	Not Available	TLV® Basis: URT, LRT, skin, & eye irr; fluorosis; BEI
US NIOSH Recommended Exposure Limits (RELs)	arsenic	Arsenic metal: Arsenia	Not Available	Not Available	0.002 mg/m3	Ca See Appendix A

US NIOSH Recommended Exposure Limits (RELs)	silver	Silver metal: Argentum	0.01 mg/m3	Not Available	Not Available	Not Available
US OSHA Permissible Exposure Levels (PELs) - Table Z1	cadmium	Cadmium	0.005 mg/m3	Not Available	Not Available	see 1910.1027;(as Cd)
US NIOSH Recommended Exposure Limits (RELs)	cadmium	Cadmium metal: Cadmium	0.01 mg/m3	Not Available	Not Available	Ca See Appendix A [*Note: The REL applies to all Cadmium compounds (as Cd).]
US ACGIH Threshold Limit Values (TLV)	cadmium	Cadmium	Not Available	Not Available	Not Available	TLV® Basis: Kidney dam; BEI
US OSHA Permissible Exposure Levels (PELs) - Table Z1	cobalt	Cobalt metal, dust, and fume	0.1 mg/m3	Not Available	Not Available	(as Co)
US NIOSH Recommended Exposure Limits (RELs)	cobalt	Cobalt metal dust, Cobalt metal fume	0.05 mg/m3	Not Available	Not Available	TLV® Basis: Pneumonitis
US ACGIH Threshold Limit Values (TLV)	cobalt	Hard metals containing Cobalt and Tungsten carbide, as Co	0.005 mg/m3	Not Available	Not Available	Not Available
US NIOSH Recommended Exposure Limits (RELs)	chromium	Chrome, Chromium	0.5 mg/m3	Not Available	Not Available	Not Available
US NIOSH Recommended Exposure Limits (RELs)	nickel	Nickel metal: Elemental nickel, Nickel catalyst	0.015 mg/m3	Not Available	Not Available	Ca See Appendix A [*Note: The REL does not apply to Nickel carbonyl.]
US ACGIH Threshold Limit Values (TLV)	nickel	Nickel and inorganic compounds including Nickel subsulfide, as Ni - Elemental	1.5 mg/m3	Not Available	Not Available	TLV® Basis: Dermatitis; pneumoconiosis
US NIOSH Recommended Exposure Limits (RELs)	selenium	Elemental selenium, Selenium alloy	0.2 mg/m3	Not Available	Not Available	[*Note: The REL also applies to other selenium compounds (as Se) except Selenium hexafluoride.]
US NIOSH Recommended Exposure Limits (RELs)	copper	Copper metal dusts, Copper metal fumes	1 mg/m3	Not Available	Not Available	[*Note: The REL also applies to other copper compounds (as Cu) except Copper fume.]
US ACGIH Threshold Limit Values (TLV)	copper	Copper - Fume, as Cu	0.2 mg/m3	Not Available	Not Available	TLV® Basis: Irr; GI; metal fume fever; BEI
US ACGIH Threshold Limit Values (TLV)	copper	Copper - Dusts and mists, as Cu	1 mg/m3	Not Available	Not Available	TLV® Basis: Irr; GI; metal fume fever; BEI

EMERGENCY LIMITS

Ingredient	Material name		TEEL-1	TEEL-2	TEEL-3
iron	Iron	3.2 mg/m3	35 mg/m3	150 mg/m3	
magnesium nitrate	Magnesium(II) nitrate (1:2), hexahydrate		16 mg/m3	180 mg/m3	1,100 mg/m3
magnesium nitrate	Magnesium nitrate; (Magnesium(II) nitrate (1:2))		30 mg/m3	330 mg/m3	2,000 mg/m3
molybdenum	Molybdenum		30 mg/m3	330 mg/m3	2,000 mg/m3
ammonium phosphate, monobasic	Ammonium dihydrogen phosphate; (Monoammonium phosphate)		17 mg/m3	190 mg/m3	1,100 mg/m3
potassium	Potassium		2.3 mg/m3	25 mg/m3	150 mg/m3
sodium	Sodium		13 mg/m3	140 mg/m3	870 mg/m3
sulfuric acid	Sulfuric acid		Not Available	Not Available	Not Available
titanium	Titanium		30 mg/m3	330 mg/m3	2,000 mg/m3
ammonium acetate	Ammonium acetate		3.8 mg/m3	42 mg/m3	250 mg/m3
ammonium chloride	Ammonium chloride		20 mg/m3	110 mg/m3	330 mg/m3
nitric acid	Nitric acid		Not Available	Not Available	Not Available
hydrofluoric acid	Hydrogen fluoride; (Hydrofluoric acid)		Not Available	Not Available	Not Available
silver	Silver		0.3 mg/m3	170 mg/m3	990 mg/m3
cadmium	Cadmium		Not Available	Not Available	Not Available
cobalt	Cobalt		0.18 mg/m3	2 mg/m3	20 mg/m3
chromium	Chromium		1.5 mg/m3	17 mg/m3	99 mg/m3
manganese(II) acetate	Acetic acid, manganese(II) salt (2:1)		9.4 mg/m3	16 mg/m3	96 mg/m3
nickel	Nickel		4.5 mg/m3	50 mg/m3	99 mg/m3
ammonium metavanadate	Ammonium vanadate; (Ammonium vanadium oxide; Ammonium metavanadate)		0.01 mg/m3	0.11 mg/m3	80 mg/m3
zinc	Zinc		6 mg/m3	21 mg/m3	120 mg/m3
selenium	Selenium		0.6 mg/m3	6.6 mg/m3	40 mg/m3
copper	Copper		3 mg/m3	33 mg/m3	200 mg/m3
Ingredient	Original IDLH	Revis	ed IDLH		
aluminium	Not Available	Not Av	vailable		
calcium	Not Available	Not Available			
iron	Not Available	Not Av	ailable		
magnesium nitrate	Not Available	Not Av	Not Available		

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molybdenum	N.E. / N.E.	5,000 mg/m3
ammonium phosphate, monobasic	Not Available	Not Available
potassium	Not Available	Not Available
sodium	Not Available	Not Available
sulfuric acid	80 mg/m3	15 mg/m3
titanium	Not Available	Not Available
ammonium acetate	Not Available	Not Available
ammonium chloride	Not Available	Not Available
nitric acid	100 ppm	25 ppm
hydrofluoric acid	30 ppm	30 [Unch] ppm
water	Not Available	Not Available
arsenic	100 mg/m3	5 mg/m3
silver	N.E. / N.E.	10 mg/m3
cadmium	50 mg/m3 / 9 mg/m3	9 mg/m3 / 9 [Unch] mg/m3
cobalt	20 mg/m3	20 [Unch] mg/m3
chromium	N.E. / N.E.	250 mg/m3
manganese(II) acetate	N.E. / N.E.	500 mg/m3
nickel	N.E. / N.E.	10 mg/m3
ammonium metavanadate	Not Available	Not Available
zinc	Not Available	Not Available
selenium	Unknown mg/m3 / Unknown ppm	1 mg/m3
copper	N.E. / N.E.	100 mg/m3

Exposure controls

	Engineering controls are used to remove a hazard or place a barrier between the worker and the ha effective in protecting workers and will typically be independent of worker interactions to provide this has a second sec		ontrols can be highly
	The basic types of engineering controls are: Process controls which involve changing the way a job activity or process is done to reduce the risk.		
	Enclosure and/or isolation of emission source which keeps a selected hazard "physically" away from	the worker and ventilation that stra	tegically "adds" and
	"removes" air in the work environment. Ventilation can remove or dilute an air contaminant if designed	d properly. The design of a ventilation	on system must match
	the particular process and chemical or contaminant in use. Employers may need to use multiple types of controls to prevent employee overexposure.		
	Local exhaust ventilation usually required. If risk of overexposure exists, wear approved respirator. C Supplied-air type respirator may be required in special circumstances. Correct fit is essential to ensi An approved self contained breathing apparatus (SCBA) may be required in some situations. Provide adequate ventilation in warehouse or closed storage area. Air contaminants generated in the turn, determine the "capture velocities" of fresh circulating air required to effectively remove the cont	ure adequate protection. e workplace possess varying "esca	
	Type of Contaminant:		Air Speed:
	solvent, vapours, degreasing etc., evaporating from tank (in still air).		
Appropriate engineering	aerosols, fumes from pouring operations, intermittent container filling, low speed conveyer transfers, welding, spray drift, plating acid fumes, pickling (released at low velocity into zone of active generation)		
controls	direct spray, spray painting in shallow booths, drum filling, conveyer loading, crusher dusts, gas discharge (active generation into zone of rapid air motion)		
	grinding, abrasive blasting, tumbling, high speed wheel generated dusts (released at high initial velocity into zone of very high rapid air motion).		
	Within each range the appropriate value depends on:		
	Lower end of the range	Upper end of the range	
	1: Room air currents minimal or favourable to capture	1: Disturbing room air currents	
	2: Contaminants of low toxicity or of nuisance value only.	2: Contaminants of high toxicity	
	3: Intermittent, low production.	3: High production, heavy use	
	4: Large hood or large air mass in motion	4: Small hood-local control only	
	Simple theory shows that air velocity falls rapidly with distance away from the opening of a simple ext of distance from the extraction point (in simple cases). Therefore the air speed at the extraction point distance from the contaminating source. The air velocity at the extraction fan, for example, should be solvents generated in a tank 2 meters distant from the extraction point. Other mechanical considerati apparatus, make it essential that theoretical air velocities are multiplied by factors of 10 or more when	should be adjusted, accordingly, a a minimum of 1-2 m/s (200-400 f/n ons, producing performance defici	fter reference to nin) for extraction of is within the extraction

Personal protection



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Eye and face protection	 where complete eye protection is needed pressure. Chemical goggles.whenever there is a da Full face shield (20 cm, 8 in minimum) ma Alternatively a gas mask may replace split Contact lenses may pose a special hazarn lenses or restrictions on use, should be cr chemicals in use and an account of injury readily available. In the event of chemical 	d; soft contact lenses may absorb and concentrate irritants. A written reated for each workplace or task. This should include a review of len experience. Medical and first-aid personnel should be trained in their exposure, begin eye irrigation immediately and remove contact lens a n - lens should be removed in a clean environment only after workers	plashing, or if the material may be under be properly fitted. yes; these afford face protection. policy document, describing the wearing of is absorption and adsorption for the class of removal and suitable equipment should be is soon as practicable. Lens should be removed
Skin protection	See Hand protection below		
Hands/feet protection	 Elbow length PVC gloves When handling corrosive liquids, wear tro 	ousers or overalls outside of boots, to avoid spills entering boots.	
Body protection	See Other protection below		
Other protection	 Overalls. PVC Apron. PVC protective suit may be required if exp Eyewash unit. Ensure there is ready access to a safety s 		
Thermal hazards	Not Available		

Respiratory protection

Type A Filter of sufficient capacity. (AS/NZS 1716 & 1715, EN 143:2000 & 149:2001, ANSI Z88 or national equivalent)

SECTION 9 PHYSICAL AND CHEMICAL PROPERTIES

Information on basic physical and chemical properties

Appearance	Yellow		
Physical state	Liquid	Relative density (Water = 1)	Not Available
Odour	Not Available	Partition coefficient n-octanol / water	Not Available
Odour threshold	Not Available	Auto-ignition temperature (°C)	Not Available
pH (as supplied)	Not Available	Decomposition temperature	Not Available
Melting point / freezing point (°C)	Not Available	Viscosity (cSt)	Not Available
Initial boiling point and boiling range (°C)	Not Available	Molecular weight (g/mol)	Not Available
Flash point (°C)	Not Available	Taste	Not Available
Evaporation rate	Not Available	Explosive properties	Not Available
Flammability	Not Available	Oxidising properties	Not Available
Upper Explosive Limit (%)	Not Available	Surface Tension (dyn/cm or mN/m)	Not Available
Lower Explosive Limit (%)	Not Available	Volatile Component (%vol)	Not Available
Vapour pressure (kPa)	Not Available	Gas group	Not Available
Solubility in water (g/L)	Miscible	pH as a solution (1%)	Not Available
Vapour density (Air = 1)	Not Available	VOC g/L	Not Available

SECTION 10 STABILITY AND REACTIVITY

Reactivity	See section 7
Chemical stability	Contact with alkaline material liberates heat
Possibility of hazardous reactions	See section 7
Conditions to avoid	See section 7
Incompatible materials	See section 7
Hazardous decomposition products	See section 5

SECTION 11 TOXICOLOGICAL INFORMATION

Inhaled

Information on toxicological effects

Inhalation of vapours or aerosols (mists, fumes), generated by the material during the course of normal handling, may be harmful. The material can cause respiratory irritation in some persons. The body's response to such irritation can cause further lung damage. Corrosive acids can cause irritation of the respiratory tract, with coughing, choking and mucous membrane damage. There may be dizziness, headache, nausea and weakness.

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Benchel and about nouzbe are spacer in animale separed in mobiletar. My shadara. Mobiletarum have many oxfue bondue in takin marked in the set of the set								
Adde elicited in fundie relation in the instant. In the card instant, any and that decrements any			by inhalation. Molybdenum fume	may produce bronchial irritation and moderate				
As an invation of byforgs function (byforgs causes access multiple) of or explore the function of the space f		Acute effects of fluoride inhalation include irritation of nose and throat, coughing and chest discomfort. A single acute over-exposure may even cause nose						
Accesse Accesse <t< th=""><th></th><th>Acute inhalation of hydrogen fluoride (hydrofluoric acid) vapours causes severe i water in the lungs, and may cause death. The above irritation occurs even with f irritating odour, that can be detected at concentrations of about 0.04 parts per n severe inflammation and water buildup in the lungs (which may occur with 1 hou intolerable, but a vapour concentration of 30 parts per million is considered as in It is estimated that the lowest lethal concentration for a 5-minute human exposur either skin contact or inhalation may lead to low levels of calcium and magnesiur</th><th colspan="6">bleed. Acute inhalation of hydrogen fluoride (hydrofluoric acid) vapours causes severe irritation of the eye, nose and throat, delayed fever, bluing of the extremities and water in the lungs, and may cause death. The above irritation occurs even with fairly low concentrations of hydrogen fluoride. Hydrogen fluoride has a strong irritating odour, that can be detected at concentrations of about 0.04 parts per million. Higher levels cause corrosion of the throat, nose and lungs, leading to severe inflammation and water buildup in the lungs (which may occur with 1 hour of exposure). A vapour concentration of 10 parts per million is regarded as intrediately dangerous to life and health. It is estimated that the lowest lethal concentration for a 5-minute human exposure to hydrogen fluoride is in the range of 50 to 250 parts per million. Exposure by</th></t<>		Acute inhalation of hydrogen fluoride (hydrofluoric acid) vapours causes severe i water in the lungs, and may cause death. The above irritation occurs even with f irritating odour, that can be detected at concentrations of about 0.04 parts per n severe inflammation and water buildup in the lungs (which may occur with 1 hou intolerable, but a vapour concentration of 30 parts per million is considered as in It is estimated that the lowest lethal concentration for a 5-minute human exposur either skin contact or inhalation may lead to low levels of calcium and magnesiur	bleed. Acute inhalation of hydrogen fluoride (hydrofluoric acid) vapours causes severe irritation of the eye, nose and throat, delayed fever, bluing of the extremities and water in the lungs, and may cause death. The above irritation occurs even with fairly low concentrations of hydrogen fluoride. Hydrogen fluoride has a strong irritating odour, that can be detected at concentrations of about 0.04 parts per million. Higher levels cause corrosion of the throat, nose and lungs, leading to severe inflammation and water buildup in the lungs (which may occur with 1 hour of exposure). A vapour concentration of 10 parts per million is regarded as intrediately dangerous to life and health. It is estimated that the lowest lethal concentration for a 5-minute human exposure to hydrogen fluoride is in the range of 50 to 250 parts per million. Exposure by					
Though considered non-harmful, sight intraction may read from contact because of the deraies nature of the alternation. The time y cases of the deraies nature of the alternation and the material intervation. See in the set of the deraies nature of the alternation of the deraies of the deraies nature of the alternation of the deraies of the der	Ingestion	damage to the health of the individual. Ingestion of acidic corrosives may produce burns around and in the mouth, the t speaking may also be evident. Molybdenum, an essential trace element, can in large doses hamper growth an other symptoms include greying of hair, shrinking of the testicles, reduced fertili mucous membranes. Fluoride causes severe loss of calcium in the blood, with symptoms appearing	damage to the health of the individual. Ingestion of acidic corrosives may produce burns around and in the mouth, the throat and oesophagus. Immediate pain and difficulties in swallowing and speaking may also be evident. Molybdenum, an essential trace element, can in large doses hamper growth and cause loss of appetite, listlessness and diarrhoea. Anaemia also occurs, and other symptoms include greying of hair, shrinking of the testicles, reduced fertility and milk production, shortness of breath, incoordination and irritation of the mucous membranes. Fluoride causes severe loss of calcium in the blood, with symptoms appearing several hours later including painful and rigid muscle contractions of the limbs.					
Deck depictorial with add considers may produce pain, team, sexelikity to light and burns. Mid burns of the epithelia generally recover repidy and completely. Animal testing showed that a 20% solution of hydrothocics add flydogen hourids) in water cause definitioned by time definition definitioned by time definition definitioned by time definition definitioned by time definition definitioned by time definition definitioned by time definition definitioned definition. The sense definition definition definition definition definition definition definition definition definition definition. The sense definition definition definition definition definition definition definition definition definition definition. The definition definition. The definition definitedefinition definition definition definition definition definition	Skin Contact	Skin contact with the material may be harmful; systemic effects may result following absorption. Though considered non-harmful, slight irritation may result from contact because of the abrasive nature of the aluminium oxide particles. Thus it may cause itching and skin reaction and inflammation. Skin contact with acidic corrosives may result in pain and burns; these may be deep with distinct edges and may heal slowly with the formation of scar tissue. Contact of the skin with liquid hydrofluoric acid (hydrogen fluoride) may cause severe burns, erythema, and swelling, vesiculation, and serious crusting. With more serious burns, ulceration, blue-gray discoloration, and necrosis may occur. Solutions of hydrofluoric acid, as dilute as 2%, may cause severe skin burns. Fluorides are easily absorbed through the skin and cause death of soft tissue and erode bone. Healing is delayed and death of tissue may continue to spread beneath skin. Open cuts, abraded or irritated skin should not be exposed to this material						
Substance accumulation, in the human body, may occur and may cause some oncien following repeated or longetome cocupational exposure. Animal testing shorts to administum outdes may cause low disease and cance, depending on the size of the particle. The smaller the size, the greater the tendencies of causing harm. Chronici Pip hevels of motybodium can cause over-activity of the flywoid gread. Not Motybodium can cause over-activity of the flywoid gread. Chronici Repeated or problem can cause accurate over-activity of the flywoid gread. Interview of the particle. The smaller the size, and even of the costion of testic of the particle. Chronici Repeated or problem can cause accurate the swelling and/or ulceration of mouth lining. Initiation of always to lung, with cough, and rifemmation di lung size often occurs. Initiation of always to lung, with cough, with cough, and may cause the second or provide cause beer changes with and earlier the science of testic on and consortion of the bone and underlying tissue. Ingestion causes servere pains and burne in the mouth and threat and blood calcum levels are diagerously reduced. ICP-MS-ICS Solution AB IOXICITY IRRITATION aluminium IOXICITY IRRITATION Interview IOXICITY IRRITATION Interview IOXICITY IRRITATION Chronic IOXICITY IRRITATION Interview IOXICITY IRRITATION Interview Interview Interview <th>Eye</th> <th colspan="5">Direct eye contact with acid corrosives may produce pain, tears, sensitivity to light and burns. Mild burns of the epithelia generally recover rapidly and completely. Animal testing showed that a 20% solution of hydrofluoric acid (hydrogen fluoride) in water caused immediate damage in the form of total clouding of the lens and ischaemia of the conjunctiva. Swelling of the stroma of the cornea occurred within 1 hour, followed by tissue death (necrosis) of structures of the front of the</th>	Eye	Direct eye contact with acid corrosives may produce pain, tears, sensitivity to light and burns. Mild burns of the epithelia generally recover rapidly and completely. Animal testing showed that a 20% solution of hydrofluoric acid (hydrogen fluoride) in water caused immediate damage in the form of total clouding of the lens and ischaemia of the conjunctiva. Swelling of the stroma of the cornea occurred within 1 hour, followed by tissue death (necrosis) of structures of the front of the						
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ICP-MS-ICS Solution AB Not Available Not Available Not Available aluminium TOXICITY IRRITATION Oral (rat) LD50: >2000 mg/kg ^[1] Not Available TOXICITY IRRITATION Dermal (rabbit) LD50: >2500 mg/kg ^[1] Not Available Oral (rat) LD50: >2000 mg/kg ^[1] Not Available Oral (rat) LD50: S000 mg/kg ^[2] Not Available TOXICITY IRRITATION Oral (rat) LD50: S000 mg/kg ^[2] Not Available Oral (rat) LD50: S000 mg/kg ^[2] Eye (rabbit): 500 mg/24h - mild Oral (rat) LD50: S2000 mg/kg ^[1] IRRITATION Modyber IRRITATION Magnesium nitrate Eye (rabbit): 500 mg/24h - mild Modyber Skin (rabbit): S00 mg/24h - mild Modyber IRRITATION								
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molybdenum dermal (rat) LD50: >2000 mg/kg ^[1] Not Available			okin (rabbit). 500 mg/24n - Mil					
		ΤΟΧΙΟΙΤΥ		IRRITATION				
Oral (rat) LD50: >2000 mg/kg ^[1]	molybdenum	dermal (rat) LD50: >2000 mg/kg ^[1]		Not Available				
		Oral (rat) LD50: >2000 mg/kg ^[1]						

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Legend:	1. Value obtained from Europe ECHA Registered Substance: extracted from RTECS - Register of Toxic Effect of chemical		rom manufacturer's SDS. Unless otherwise specified data				
		Substances					
CALCIUM	The solid may react violently on contact with wet skin tissue, i.e. eyes, mouth, causing chemical and thermal burns. The acute effects include burns, ulceration, or tissue death, severe eye damage (corneal burns or opacification), and probable blindness. Inhalation of dust or furmes (especially from a fire involving calcium) will cause shortness of breath, nausea, headache, nose and respiratory tract irritation and in extreme, pneumonitis						
MAGNESIUM NITRATE		The material may be irritating to the eye, with prolonged contact causing inflammation. Magnesium nitrate heaxahydrate is a methaemoglobin-forming agent which if inhaled or ingested in high enough concentrations may cause fatigue, headache, dizziness. (Source: LL.O. Encyclopaedia)					
SULFURIC ACID	WARNING: For inhalation exposure <u>ONLY</u> : This substance I Occupational exposures to strong inorganic acid mists of sul	-	Broup 1: CARCINOGENIC TO HUMANS				
AMMONIUM ACETATE	Altered sleep time, muscle contraction, coma, dyspnae, hypog						
NITRIC ACID	For acid mists, aerosols, vapours Test results suggest that eukaryotic cells are susceptible to ge The material may cause severe skin irritation after prolonged of vesicles, scaling and thickening of the skin. Oral (?) LD50: 50-500 mg/kg * [Various Manufacturers]						
HYDROFLUORIC ACID	(liver and kidney damage) [Manufacturer] for hydrogen fluorid	e (as vapour)					
ARSENIC	Arsenic compounds are classified by the European Union as WARNING: This substance has been classified by the IARC Tumorigenic - Carcinogenic by RTECS criteria.						
COBALT	Allergic reactions involving the respiratory tract are usually du Attention should be paid to atopic diathesis, characterised by in Exogenous allergic alveolitis is induced essentially by allerger involved.	increased susceptibility to nasal inflat	mmation, asthma and eczema.				
CHROMIUM	On skin and inhalation exposure, chromium and its compound Tenth Annual Report on Carcinogens: Substance known to be [<i>National Toxicology Program: U.S. Dep.</i> Gastrointestinal tumours, lymphoma, musculoskeletal tumours	e Carcinogenic					
MANGANESE(II) ACETATE	Laboratory tests have shown mutagenic effects: Positive B. re	с.					
NICKEL	Tenth Annual Report on Carcinogens: Substance anticipated [<i>National Toxicology Program: U.S. Dep.</i> Oral (rat) TDLo: 500 mg/kg/5D-I Inhalation (rat) TCLo: 0.1 m	-					
COPPER	for copper and its compounds (typically copper chloride): Acute toxicity: There are no reliable acute oral toxicity result WARNING: Inhalation of high concentrations of copper fume like respiratory tract irritation with fever.		cute industrial disease of short duration. tiredness, influenza				
ALUMINIUM & CALCIUM & MOLYBDENUM & AMMONIUM PHOSPHATE, MONOBASIC & POTASSIUM & SODIUM & TITANIUM & HYDROFLUORIC ACID & WATER & CHROMIUM	No significant acute toxicological data identified in literature s	search.					
CALCIUM & AMMONIUM PHOSPHATE, MONOBASIC & POTASSIUM & SODIUM & SULFURIC ACID & AMMONIUM ACETATE & NITRIC ACID & HYDROFLUORIC ACID & AMMONIUM METAVANADATE	Asthma-like symptoms may continue for months or even years	after exposure to the material ends.					
MAGNESIUM NITRATE & ZINC	The material may cause skin irritation after prolonged or repeat scaling and thickening of the skin.	The material may cause skin irritation after prolonged or repeated exposure and may produce on contact skin redness, swelling, the production of vesicles, scaling and thickening of the skin.					
AMMONIUM CHLORIDE & NITRIC ACID & HYDROFLUORIC ACID	The material may produce severe irritation to the eye causing	pronounced inflammation.					
NITRIC ACID & HYDROFLUORIC ACID	The material may produce respiratory tract irritation, and resu	It in damage to the lung including re	duced lung function.				
COBALT & NICKEL	The following information refers to contact allergens as a grou	up and may not be specific to this pro	oduct.				
COBALT & NICKEL	WARNING: This substance has been classified by the IARC	as Group 2B. Possibly Carcinogeni	c to Humans.				
CHROMIUM & SELENIUM	The substance is classified by IARC as Group 3: NOT classifiable as to its carcinogenicity to humans.		o to maniferio.				
Acute Toxicity	~	Carcinogenicity	\otimes				
Skin Irritation/Corrosion	* *	Reproductivity	0				
Serious Eye							
Damage/Irritation Respiratory or Skin	0	STOT - Single Exposure	0				
sensitisation	\otimes	STOT - Repeated Exposure	\otimes				

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Mutagenicity	Aspiration Hazard	
	 Data available 	e but does not fill the criteria for classification e to make classification ilable to make classification
SECTION 12 ECOLOGICAL INFORMATION		
Toxicity		

ICP-MS-ICS Solution AB	ENDPOINT		TEST DURATION (HR)		SPECIE	S	VALUE		SOU	RCE
	Not Applicable		Not Applicable		Not App	licable	Not Ap	Not Applicable		pplicable
	ENDPOINT	TFS	T DURATION (HR)	SPECIE	s			VALUE		SOURCE
	LC50	96		Fish	-			0.078-0.108mg	1/L	2
	EC50	48		Crustace	ea			0.7364mg/L	<i>y</i> -	2
aluminium	EC50	96				atic plants		0.0054mg/L		2
	BCF	360		-	-	atic plants		9mg/L		4
	EC50	120		Fish				0.000051mg/L		5
	NOEC	72		Algae or	other aqu	atic plants		>=0.004mg/L		2
						1				
	ENDPOINT		TEST DURATION (HR)			SPECIES		VALUE	SC	DURCE
calcium	EC50		24			Crustacea		6934mg/L	5	
	NOEC		48			Crustacea		33.3mg/L	2	
	ENDPOINT	TES	T DURATION (HR)	SPECI	=9			VALUE		SOURCE
	LC50	96			_0			0.05mg/L		2
	EC50	96			or other ag	uatic plants		3.7mg/L		4
iron	BCF	24			cea			0.0000002mg	J/L	4
	EC50	504			cea			4.49mg/L	,	2
	NOEC	504		Fish				0.52mg/L		2
	ENDPOINT	TE	ST DURATION (HR)	SPE	CIES			VALUE		SOURCE
magnesium nitrate	LC50	96		Fish	۱		1378mg/L			2
	EC50	72		Alga	ae or other	aquatic plants		>100mg/L		2
	NOEC	72		Alga	ae or other	aquatic plants		100mg/	/L	2
	ENDPOINT	TF	ST DURATION (HR)	SPE	CIES			VALUE		SOURCE
	LC50	96	c. Solution (int)	Fish				609.1m		2
	EC50	72				aquatic plants		289.2m		2
molybdenum	BCF	336	3	-		aquatic plants		64mg/L	-	4
	EC50	336		-		aquatic plants		64mg/L		4
	NOEC	672		-	stacea			0.67mg		2
	NOEC	672	2	Crus	stacea			0.67mg	/L	2
	ENDROINT			0.05						COUDOE

	ENDPOINT	TEST DURATION (HR)	SPECIES	VALUE	SOURCE
	LC50	96	Fish	>85.9mg/L	2
ammonium phosphate, monobasic	EC50	72	Algae or other aquatic plants	>97.1mg/L	2
	EC50	72	Algae or other aquatic plants	>97.1mg/L	2
	NOEC	72	Algae or other aquatic plants	3.57mg/L	2

netopolium	ENDPOINT	TEST DURATION (HR)	SPECIES	VALUE	SOURCE
potassium	EC50	24	Crustacea	400mg/L	5
	ENDPOINT	TEST DURATION (HR)	SPECIES	VALUE	SOURCE
sodium	EC50	48	Crustacea	1640mg/L	4
	EC50	504	Crustacea	1020mg/L	4

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	ENDPOINT	TE	ST DURATION (HR)	\$	SPECIES			VALUE	SOURC
	LC50	96		1	Fish			=8mg/L	1
sulfuric acid	EC50	48		(Crustacea			=42.5mg/L	1
	EC50	240		1	Algae or other	aquatic plants		2.5000mg/L	4
	NOEC	720	0	ł	Fish			0.13mg/L	2
	ENDPOINT	TE	ST DURATION (HR)		SPECIES			VALUE	SOURC
titanium	EC50	4.5				r aquatic plants		>100mg/L	2
utanium	NOEC	4.5			Crustacea	aqualic plants		1mg/L	2
			1			1			
	ENDPOINT		TEST DURATION (HR)			SPECIES	VALU	IE	SOURCE
ammonium acetate	EC50		48			Crustacea	>919	mg/L	2
	EC50		24			Crustacea	>919	mg/L	2
	ENDPOINT	TE	ST DURATION (HR)	5	SPECIES			VALUE	SOURC
	LC50	96	. ,	F	Fish			0.08mg/L	4
	EC50	48			Crustacea			0.261mg/L	4
ammonium chloride	EC50	72				aquatic plants		166.5mg/L	4
	ECO	168			Crustacea	1		=0.025mg/L	1
	NOEC	720			Fish			0.006mg/L	4
		120						0.000mg/L	•
	ENDPOINT		TEST DURATION (HR)			SPECIES	VA	UE	SOURCE
nitric acid	NOEC		16			Crustacea		'mg/L	4
	ENDPOINT		TEST DURATION (HR)			SPECIES	VALU	E	SOURCE
	LC50		96			Fish	51mg/	L	2
hydrofluoric acid	EC50		48			Crustacea	=270n	ng/L	1
	EC50		96			Crustacea 26-48mg		ng/L	2
	NOEC		504			Fish	4mg/L		2
	ENDPOINT		TEST DURATION (HR)		SPECI	=9	VALUE		SOURCE
water	Not Applicable		Not Applicable		Not App		Not Applical	ble	Not Applicable
	ENDPOINT	TE	ST DURATION (HR)		SPECIES			VALUE	SOURC
arsenic	LC50	96			Fish			9.9mg/L	4
disenic	EC50	336	3		Algae or othe	r aquatic plants		0.63mg/L	4
	NOEC	336	3		Algae or othe	r aquatic plants		<0.75mg/L	4
	ENDPOINT	TES	T DURATION (HR)	SPE	CIES		VAL	UE	SOUR
	LC50	96		Fish			0.00	148mg/L	2
	EC50	48		Crus	stacea			024mg/L	4
silver	EC50	96			e or other aqu	uatic plants		1628837mg/L	4
	BCF	336			mg/L	4			
	EC50	48	18 Crustacea		0.00	024mg/L	4		
	NOEC	480	480 Crustacea 0.00		031mg/L	2			
	ENDPOINT		T DURATION (HR)		ECIES			LUE	SOUR
		96		Fisl				01mg/L	4
	LC50			0	stacea		0.0	033mg/L	5
	EC50	48							
cadmium	EC50 EC50	72		Alg	ae or other aq	uatic plants		18mg/L	2
cadmium	EC50			Alg: Fisl	ae or other aq	uatic plants	500	18mg/L)mg/L 0065mg/L	2 4 5

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	ENDPOINT	TEST DURATION (HR)	SPECIES	VALUE	SOURCE
	LC50	96	Fish	1.406mg/	L 2
	EC50	48	Crustacea	>0.89mg	
cobalt	EC50	72	Algae or other aquatic plants	0.144mg/	
	BCF	1344	Fish	0.99mg/L	
	EC50	70	Algae or other aquatic plants	0.02mg/L	
	NOEC	168	Algae or other aquatic plants	0.0018mg	
	ENDPOINT	TEST DURATION (HR)	SPECIES	VALUE	SOURCE
	LC50	96	Fish	13.9mg/L	4
	EC50	48	Crustacea	0.0225mg/	
chromium	EC50	72			
chronnum		1440	Algae or other aquatic plants	0.104mg/L	
	BCF		Algae or other aquatic plants	0.0495mg/	
	EC50	48	Crustacea	0.0245mg/	
	NOEC	672	Fish	0.00019mg	y/L 4
	ENDROINT		0050150	VALUE	COURCE
manganese(II) acetate	ENDPOINT	TEST DURATION (HR)	SPECIES	VALUE	SOURCE
	Not Applicable	Not Applicable	Not Applicable	Not Applicable	Not Applicable
	ENDPOINT		SPECIES	VALUE	SOURCE
		TEST DURATION (HR)	SPECIES		SOURCE
	LC50	96	Fish	0.0000475mg	
	EC50	48	Crustacea	0.013mg/L	5
nickel	EC50	72	Algae or other aquatic plants	0.0407mg/L	2
	BCF	1440	Algae or other aquatic plants	0.47mg/L	4
	EC50	720	Crustacea	0.0062mg/L	2
	NOEC	72	Algae or other aquatic plants	0.0035mg/L	2
					1
	ENDPOINT	TEST DURATION (HR)	SPECIES	VALUE	SOURCE
	LC50	96	Fish	0.693mg/	L 2
ammonium metavanadate	EC50	48 Crustacea		2.387mg/	L 2
	EC50	72	Algae or other aquatic plants	0.9894mg	y/L 2
	EC50	72	Algae or other aquatic plants	1.162mg/	L 2
	NOEC	72	Algae or other aquatic plants	0.0168m	J/L 2
	ENDPOINT	TEST DURATION (HR)	SPECIES	VALUE	SOURCE
	LC50	96	Fish	0.00272mg	-
	EC50	48	Crustacea	0.04mg/L	5
zinc	EC50	72	Algae or other aquatic plants	0.106mg/L	4
	BCF	360	Algae or other aquatic plants	9mg/L	4
	EC50	120	Fish	0.00033mg	/L 5
	NOEC	336	Algae or other aquatic plants	0.00075mg	1/L 4
	ENDPOINT	TEST DURATION (HR)	SPECIES	VALUE	SOURCE
	LC50	96	Fish	>0.0262mg/	
	EC50	48	Crustacea	>0.1603mg/	2
selenium	EC50	72	Algae or other aquatic plants	>0.00173mg	/L 2
	BCF	504	Crustacea	0.711mg/L	4
	EC50	96	Algae or other aquatic plants	0.355mg/L	2
	NOEC	72	Algae or other aquatic plants	0.000547mg	/L 2
				1	1
	ENDPOINT	TEST DURATION (HR)	SPECIES	VALUE	SOURCE
				0.0000//	L A
	LC50	96	Fish	0.0028mg/L	
copper	LC50 EC50	96 48	Fish Crustacea	0.0028mg/L	5
copper					5

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EC50 96	Crustacea	0.001mg/L	5
NOEC 96	Crustacea	0.0008mg/L	4

Legend:

Extracted from 1. IUCLID Toxicity Data 2. Europe ECHA Registered Substances - Ecotoxicological Information - Aquatic Toxicity 3. EPIWIN Suite V3.12 (QSAR) - Aquatic Toxicity Data (Estimated) 4. US EPA, Ecotox database - Aquatic Toxicity Data 5. ECETOC Aquatic Hazard Assessment Data 6. NITE (Japan) - Bioconcentration Data 7. METI (Japan) - Bioconcentration Data 8. Vendor Data

For Molybdenum:

Environmental Fate: Molybdenum is an essential micronutrient in plants and animals. It is commonly used in the manufacture of steel alloys. Based on the high concentration of molybdenum in all analyzed waste types, the exposure of the environment to molybdenum is regarded as significant. The limited amount of data regarding its toxicity makes it impossible to evaluate the potential for adverse environmental and health effects from molybdenum exposure. Molybdenum is generally found in two oxidation states in nature, Mo(IV) and Mo(VI). In oxidizing environments, Mo(VI) dominates and it is commonly present as molybdenum contains seven isotopes. Molybdenum oxidizes at elevated temperatures.

Atmospheric Fate: Molybdenum can be deposited via dry/wet deposition; however, atmospheric exposure has been identified as a minor source to terrestrial and aquatic habitats.

Terrestrial Fate: Molybdenum is a naturally occurring substance in soil. Soil molybdenum is a potentially toxic element, but no cases have been reported of molybdenum toxicity to animals from consumption of forage grown on sludge-amended soils. Microbes are expected to transform the substance.

Aquatic Fate: Molybdenum disulfide is sparingly soluble in water but oxidizes to more soluble molybdates, which are stable in water. At pH 3-5, molybdate frequently shifts to hydrogen molybdate. Low pH molybdenum is usually adsorbed to sediment composed of clay, or other minerals that are prone to weathering. Molybdenum in the water is expected to be taken up by aquatic organisms. Concentrations of the substance in sediments are by site-specific factors like flow rate, and other factors, (e.g. organic content, pH)

Ecotoxicology: Molybdenum cause adverse effects in ruminant animals. Livestock have been injured by forage grown on soils containing the element. The substance stoxicological properties in mammals are governed, to a large extent, by its interaction with copper and sulfur; residues of molybdenum alone are not sufficient to diagnose poisoning by the substance. Domestic ruminants, especially cattle, are especially sensitive to molybdenum poisoning, when copper and inorganic sulfate are deficient. The resistance of small laboratory animals, and wildlife, is at least 10X that of cattle. Mule deer are not adversely affected by the substance. The substance may have a negative impact on reproduction in domestic birds and there is inadequate data on its effects on waterfowl and most mammals.

Ecotoxicity:

The tolerance of water organisms towards pH margin and variation is diverse. Recommended pH values for test species listed in OECD guidelines are between 6.0 and almost 9. Acute testing with fish showed 96h-LC50 at about pH 3.5

For Fluorides: Small amounts of fluoride have beneficial effects however; excessive intake over long periods may cause dental and/or skeletal fluorosis. Fluorides are absorbed by humans following inhalation of workplace and ambient air that has been contaminated, ingestion of drinking water and foods and dermal contact. Populations living in areas with high fluoride levels in groundwater may be exposed to higher levels of fluorides in their drinking water or in beverages prepared with the water. Among these populations, outdoor labourers, people living in hot climates, and people with excessive thirst will generally have the greatest daily intake of fluorides because they consume greater amounts of water.

Atmospheric Fate: Both hydrogen fluoride and particulate fluorides will be transported in the atmosphere and deposited on land or water by wet and dry deposition. Non-volatile inorganic fluoride particulates are removed from the atmosphere via condensation or nucleation processes. Fluorides adsorbed on particulate matter in the atmosphere are generally stable and are not readily hydrolyzed, although they may be degraded by radiation if they persist in the atmosphere. Fluorine and the silicon fluorides (fluosilicates, silicofluorides) are hydrolyzed in the atmosphere to form hydrogen fluoride. Hydrogen fluoride may combine with water vapour to produce an aerosol or fog of aqueous hydrofluoric acid. Inorganic fluoride compounds, with the exception of sulfur hexafluoride, are not expected to remain in the troposphere for long periods or to migrate to the stratosphere. Estimates of the residence time of sulfur hexafluoride in the atmosphere range from 500 to several thousand years. Fluorise in aerosols can be transported over large distances by wind or as a result of atmospheric turbulence. Fluorosilicic acid and hydrofluoric acid in high aquatic concentrations such as may be found in industrial waste ponds may volatilize, releasing silicon tetrafluoride and hydrogen fluoride into the atmosphere. Soluble inorganic fluorides may also form aerosols at the air/water interface or vaporize into the atmosphere whereas undissolved species generally undergo sedimentation.

Terrestrial Fate: Soils - Atmospheric fluorides may be transported to soils and surface waters through both wet and dry deposition processes where they may form complexes and bind strongly to soil and sediment. Solubilisation of inorganic fluorides from minerals may also be enhanced by the presence of bentonite clays and humic acid. Factors that influence the mobility of inorganic fluorides in soil are pH and the formation of aluminium and calcium complexes. In more acidic soils, concentrations of inorganic fluoride were considerably higher in the deeper horizons. The low affinity of fluorides for organic material results in leaching from the more acidic surface horizon and increased retention by clay minerals and silts in the more alkaline, deeper horizons. The maximum adsorption of fluoride to soil was reported to occur at pH 5.5. In acidic soils with pH below 6, most of the fluoride is in complexes with either aluminium or iron. Fluoride in alkaline soils at pH 6.5 and above is almost completely fixed in soils as calcium fluoride, if sufficient calcium carbonate is available. Fluoride is extremely immobile in soil.

Aquatic Fate: Fresh Water: - In water, the transport and transformation of inorganic fluorides are influenced by pH, water hardness and the presence of ion-exchange materials such as clays. In natural water, fluoride forms strong complexes with aluminium in water, and fluorine chemistry in water is largely regulated by aluminium concentration and pH. Below pH 5, fluoride is almost entirely complexed with aluminium and consequently, the concentration of free F- is low. Once dissolved, inorganic fluorides remain in solution under conditions of low pH and hardness and in the presence of ion-exchange material. Sea Water - Fluoride forms stable complexes with calcium and magnesium, which are present in sea water. Calcium carbonate precipitation dominates the removal of dissolved fluoride from sea water. The residence time for fluoride in ocean sediment is calculated to be 2-3 million years.

Ecotoxicity: Fluorides have been shown to accumulate in animals that consume fluoride-containing foliage. However, accumulation is primarily in skeletal tissue and therefore, it is unlikely that fluoride will biomagnify up the food chain.

Prevent, by any means available, spillage from entering drains or water courses. DO NOT discharge into sewer or waterways.

Persistence and degradability

Ingredient	Persistence: Water/Soil	Persistence: Air
ammonium phosphate, monobasic	HIGH	HIGH
water	LOW	LOW
ammonium metavanadate	HIGH	HIGH

Bioaccumulative potential

Ingredient	Bioaccumulation
ammonium phosphate, monobasic	LOW (LogKOW = -0.7699)
water	LOW (LogKOW = -1.38)
ammonium metavanadate	LOW (LogKOW = 2.229)

Mobility in soil

Ingredient	Mobility
ammonium phosphate, monobasic	HIGH (KOC = 1)
water	LOW (KOC = 14.3)
ammonium metavanadate	LOW (KOC = 35.04)

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SECTION 13 DISPOSAL CONSIDERATIONS

Waste treatment methods

	 Containers may still present a chemical hazard/ danger when empty. Return to supplier for reuse/ recycling if possible.
	Otherwise:
	If container can not be cleaned sufficiently well to ensure that residuals do not remain or if the container cannot be used to store the same product, then
	puncture containers, to prevent re-use, and bury at an authorised landfill.
	Where possible retain label warnings and SDS and observe all notices pertaining to the product.
Product / Packaging	Recycle wherever possible.
disposal	 Consult manufacturer for recycling options or consult local or regional waste management authority for disposal if no suitable treatment or disposal facility can be identified.
	Treat and neutralise at an approved treatment plant. Treatment should involve: Neutralisation with soda-ash or soda-lime followed by: burial in a land-fill specifically licensed to accept chemical and / or pharmaceutical wastes or Incineration in a licensed apparatus (after admixture with suitable combustible material).
	 Decontaminate empty containers with 5% aqueous sodium hydroxide or soda ash, followed by water. Observe all label safeguards until containers are cleaned and destroyed.

SECTION 14 TRANSPORT INFORMATION

Labels Required



Marine Pollutant

Land transport (DOT)

3264		
Corrosive liquid, acidic, inorganic, n.o.s. (contains nitric acid and hydrofluoric acid)		
Class8SubriskNot Applicable		
II Contraction of the second se		
Not Applicable		
Hazard Label8Special provisions386, B2, IB2, T11, TP2, TP27		

Air transport (ICAO-IATA / DGR)

Sea transport (IMDG-Code / GGVSee)

UN number	3264
UN proper shipping name	CORROSIVE LIQUID, ACIDIC, INORGANIC, N.O.S. (contains nitric acid and hydrofluoric acid)

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Transport hazard class(es)	IMDG Class8IMDG SubriskNot Applicable
Packing group	Ш
Environmental hazard	Not Applicable
Special precautions for user	EMS NumberF-A, S-BSpecial provisions274Limited Quantities1 L

Transport in bulk according to Annex II of MARPOL and the IBC code

Source	Product name	Pollution Category	Ship Type
IMO MARPOL (Annex II) - List of Noxious Liquid Substances Carried in Bulk	Nitric acid (70% and over) Nitric acid (less than 70%)	Y; Y	2 2

SECTION 15 REGULATORY INFORMATION

Safety, health and environmental regulations / legislation specific for the substance or mixture

ALUMINIUM(7429-90-5) IS FOUND ON THE FOLLOWING REGULATORY LISTS

US - Alaska Limits for Air Contaminants	US - Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air
US - California Permissible Exposure Limits for Chemical Contaminants	Contaminants
US - Hawaii Air Contaminant Limits	US - Washington Permissible exposure limits of air contaminants
US - Massachusetts - Right To Know Listed Chemicals	US - Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants
US - Michigan Exposure Limits for Air Contaminants	US ACGIH Threshold Limit Values (TLV)
US - Minnesota Permissible Exposure Limits (PELs)	US ACGIH Threshold Limit Values (TLV) - Carcinogens
US - Oregon Permissible Exposure Limits (Z-1)	US ATSDR Minimal Risk Levels for Hazardous Substances (MRLs)
US - Pennsylvania - Hazardous Substance List	US EPCRA Section 313 Chemical List
US - Rhode Island Hazardous Substance List	US NIOSH Recommended Exposure Limits (RELs)
US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants	US OSHA Permissible Exposure Levels (PELs) - Table Z1
US - Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants	US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory
CALCIUM(7440-70-2) IS FOUND ON THE FOLLOWING REGULATORY LISTS	
US - Massachusetts - Right To Know Listed Chemicals	US - Rhode Island Hazardous Substance List
US - Pennsylvania - Hazardous Substance List	US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory
IRON(7439-89-6) IS FOUND ON THE FOLLOWING REGULATORY LISTS	
International Agency for Research on Cancer (IARC) - Agents Classified by the IARC	US - Oregon Permissible Exposure Limits (Z-1)
Monographs	US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants
US - California OEHHA/ARB - Chronic Reference Exposure Levels and Target Organs	US - Washington Permissible exposure limits of air contaminants
(CRELs)	US - Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants
US - California Permissible Exposure Limits for Chemical Contaminants	US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory
US - Hawaii Air Contaminant Limits	
US - Michigan Exposure Limits for Air Contaminants	
MAGNESIUM NITRATE(13446-18-9) IS FOUND ON THE FOLLOWING REGULATORY LIST	S
US - Massachusetts - Right To Know Listed Chemicals	US EPCRA Section 313 Chemical List
US - Pennsylvania - Hazardous Substance List	US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory
US - Rhode Island Hazardous Substance List	
MOLYBDENUM(7439-98-7) IS FOUND ON THE FOLLOWING REGULATORY LISTS	
US - Alaska Limits for Air Contaminants	US - Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants
US - Hawaii Air Contaminant Limits	US - Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air
US - Idaho - Limits for Air Contaminants	Contaminants
US - Massachusetts - Right To Know Listed Chemicals	US - Washington Permissible exposure limits of air contaminants
US - Minnesota Permissible Exposure Limits (PELs)	US ACGIH Threshold Limit Values (TLV)
US - Pennsylvania - Hazardous Substance List	US ACGIH Threshold Limit Values (TLV) - Carcinogens
US - Rhode Island Hazardous Substance List	US NIOSH Recommended Exposure Limits (RELs)
US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants	US OSHA Permissible Exposure Levels (PELs) - Table Z1
	US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory
AMMONIUM PHOSPHATE, MONOBASIC(7722-76-1) IS FOUND ON THE FOLLOWING REG	GULATORY LISTS
US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory	
POTASSIUM(7440-09-7) IS FOUND ON THE FOLLOWING REGULATORY LISTS	

POTASSIUM(7440-09-7) IS FOUND ON THE FOLLOWING REGULATORY LISTS	
International Air Transport Association (IATA) Dangerous Goods Regulations - Prohibited List	US - Rhode Island Hazardous Substance List
Passenger and Cargo Aircraft	US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory
US - Massachusetts - Right To Know Listed Chemicals	

US - Pennsylvania - Hazardous Substance List

SODIUM(7440-23-5) IS FOUND ON THE FOLLOWING REGULATORY LISTS

hemwatch: 9-405773	Page 18 of 23 Issue Date: 06/02/2
atalogue number: ICP-MS-ICS Solution AB	ICP-MS-ICS Solution AB Print Date: 06/02/2
rsion No: 2.2	
International Air Transport Association (IATA) Dangerous Goods Regulations - Prohibited List	US - Rhode Island Hazardous Substance List
Passenger and Cargo Aircraft	US CWA (Clean Water Act) - List of Hazardous Substances
US - Massachusetts - Right To Know Listed Chemicals	US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory
US - Pennsylvania - Hazardous Substance List	
SULFURIC ACID(7664-93-9) IS FOUND ON THE FOLLOWING REGULATORY LISTS	
International Agency for Research on Cancer (IARC) - Agents Classified by the IARC	US - Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminant
Monographs	US - Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air
International Air Transport Association (IATA) Dangerous Goods Regulations - Prohibited List	Contaminants
Passenger and Cargo Aircraft	US - Washington Permissible exposure limits of air contaminants
US - Alaska Limits for Air Contaminants	US - Washington Toxic air pollutants and their ASIL, SQER and de minimis emission values
US - California OEHHA/ARB - Acute Reference Exposure Levels and Target Organs (RELs)	US - Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants
US - California OEHHA/ARB - Chronic Reference Exposure Levels and Target Organs (CRELs)	US ACGIH Threshold Limit Values (TLV)
US - California Permissible Exposure Limits for Chemical Contaminants	US ACGIH Threshold Limit Values (TLV) - Carcinogens
US - Camorna Permissible Exposure Limits for Chemical Contaminants	US CWA (Clean Water Act) - List of Hazardous Substances
US - Idaho - Limits for Air Contaminants	US Drug Enforcement Administration (DEA) List I and II Regulated Chemicals
US - Massachusetts - Right To Know Listed Chemicals	US EPCRA Section 313 Chemical List
5	US National Toxicology Program (NTP) 14th Report Part A Known to be Human Carcinoger
US - Michigan Exposure Limits for Air Contaminants	US NIOSH Recommended Exposure Limits (RELs)
US - Minnesota Permissible Exposure Limits (PELs)	US OSHA Permissible Exposure Levels (PELs) - Table Z1
US - New Jersey Right to Know - Special Health Hazard Substance List (SHHSL): Carcinogens	US SARA Section 302 Extremely Hazardous Substances
US - Oregon Permissible Exposure Limits (Z-1)	US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory
US - Rhode Island Hazardous Substance List	
US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants	
TITANIUM(7440-32-6) IS FOUND ON THE FOLLOWING REGULATORY LISTS	
International Agency for Research on Cancer (IARC) - Agents Classified by the IARC	US - Oregon Permissible Exposure Limits (Z-1)
Monographs	US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants
International Air Transport Association (IATA) Dangerous Goods Regulations - Prohibited List	US - Washington Permissible exposure limits of air contaminants
Passenger and Cargo Aircraft	US - Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants
US - California OEHHA/ARB - Chronic Reference Exposure Levels and Target Organs (CRELs)	US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory
US - California Permissible Exposure Limits for Chemical Contaminants	
US - Hawaii Air Contaminant Limits	
US - Michigan Exposure Limits for Air Contaminants	
AMMONIUM ACETATE(631-61-8) IS FOUND ON THE FOLLOWING REGULATORY LISTS	
US - Massachusetts - Right To Know Listed Chemicals	US EPA Carcinogens Listing
US - Pennsylvania - Hazardous Substance List	US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory
US CWA (Clean Water Act) - List of Hazardous Substances	
AMMONIUM CHLORIDE(12125-02-9) IS FOUND ON THE FOLLOWING REGULATORY LIS	STS
US - Alaska Limits for Air Contaminants	US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants
US - California Permissible Exposure Limits for Chemical Contaminants	US - Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminar
US - Hawaii Air Contaminant Limits	US - Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air
US - Massachusetts - Right To Know Listed Chemicals	Contaminants
US - Michigan Exposure Limits for Air Contaminants	US - Washington Permissible exposure limits of air contaminants
US - Minnesota Permissible Exposure Limits (PELs)	US ACGIH Threshold Limit Values (TLV)
US - Oregon Permissible Exposure Limits (Z-1)	US CWA (Clean Water Act) - List of Hazardous Substances
US - Pennsylvania - Hazardous Substance List	US NIOSH Recommended Exposure Limits (RELs)
US - Rhode Island Hazardous Substance List	US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory
NITRIC ACID(7697-37-2) IS FOUND ON THE FOLLOWING REGULATORY LISTS	
International Air Transport Association (IATA) Dangerous Goods Regulations - Prohibited List	US - Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminar
Passenger and Cargo Aircraft	US - Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air
US - Alaska Limits for Air Contaminants	Contaminants
US - California OEHHA/ARB - Acute Reference Exposure Levels and Target Organs (RELs)	US - Washington Permissible exposure limits of air contaminants
US - California Permissible Exposure Limits for Chemical Contaminants	US - Washington Toxic air pollutants and their ASIL, SQER and de minimis emission values
US - Hawaii Air Contaminant Limits	US - Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants
US - Idaho - Limits for Air Contaminants	US ACGIH Threshold Limit Values (TLV)
US - Massachusetts - Right To Know Listed Chemicals	US CWA (Clean Water Act) - List of Hazardous Substances
US - Michigan Exposure Limits for Air Contaminants	US EPCRA Section 313 Chemical List
US - Minnesota Permissible Exposure Limits (PELs)	US NIOSH Recommended Exposure Limits (RELs)
US - Oregon Permissible Exposure Limits (Z-1)	US OSHA Permissible Exposure Levels (PELs) - Table Z1
US - Pennsylvania - Hazardous Substance List	US SARA Section 302 Extremely Hazardous Substances
US - Rhode Island Hazardous Substance List	US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory
US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants	

US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants

HYDROFLUORIC ACID(7664-39-3) IS FOUND ON THE FOLLOWING REGULATORY LISTS

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International Agency for Research on Cancer (IARC) - Agents Classified by the IARC Monographs	US - Vermont Permissible Exposure Limits Table Z-1-A	Transitional Limits for Air
US - Alaska Limits for Air Contaminants	US - Washington Permissible exposure limits of air conta	aminants
US - California OEHHA/ARB - Acute Reference Exposure Levels and Target Organs (RELs)	US - Washington Toxic air pollutants and their ASIL, SQE	ER and de minimis emission values
US - California OEHHA/ARB - Chronic Reference Exposure Levels and Target Organs	US - Wyoming Toxic and Hazardous Substances Table Z	21 Limits for Air Contaminants
(CRELs)	US - Wyoming Toxic and Hazardous Substances Table Z	Z-2 Acceptable ceiling concentration,
US - California Permissible Exposure Limits for Chemical Contaminants	Acceptable maximum peak above the acceptable ceiling	concentration for an 8-hr shift
US - Hawaii Air Contaminant Limits	US ACGIH Threshold Limit Values (TLV)	
US - Idaho - Acceptable Maximum Peak Concentrations	US ACGIH Threshold Limit Values (TLV) - Carcinogens	
US - Idaho - Limits for Air Contaminants	US ATSDR Minimal Risk Levels for Hazardous Substan	nces (MRLs)
US - Massachusetts - Right To Know Listed Chemicals	US Clean Air Act - Hazardous Air Pollutants	
US - Michigan Exposure Limits for Air Contaminants	US CWA (Clean Water Act) - List of Hazardous Substan	ices
US - Minnesota Permissible Exposure Limits (PELs)	US EPCRA Section 313 Chemical List	
US - Oregon Permissible Exposure Limits (Z-1)	US NIOSH Recommended Exposure Limits (RELs)	
US - Oregon Permissible Exposure Limits (Z-2)	US OSHA Permissible Exposure Levels (PELs) - Table 2	
US - Pennsylvania - Hazardous Substance List	US OSHA Permissible Exposure Levels (PELs) - Table 2	
US - Rhode Island Hazardous Substance List	US SARA Section 302 Extremely Hazardous Substances	
US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants	US Toxic Substances Control Act (TSCA) - Chemical Su	bstance Inventory
US - Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants		
WATER(7732-18-5) IS FOUND ON THE FOLLOWING REGULATORY LISTS		
US - Pennsylvania - Hazardous Substance List	US Toxic Substances Control Act (TSCA) - Chemical Su	bstance Inventory
ARSENIC(7440-38-2) IS FOUND ON THE FOLLOWING REGULATORY LISTS		
	LIC Weshington Dermissible synapture limits of sir contr	aminonto
International Agency for Research on Cancer (IARC) - Agents Classified by the IARC Monographs	US - Washington Permissible exposure limits of air conta	
US - Alaska Limits for Air Contaminants	US - Washington Toxic air pollutants and their ASIL, SQE	ER and de minimis emission values
US - California OEHHA/ARB - Acute Reference Exposure Levels and Target Organs (RELs)	US ACGIH Threshold Limit Values (TLV)	
US - California OEHHA/ARB - Chronic Reference Exposure Levels and Target Organs	US ACGIH Threshold Limit Values (TLV) - Carcinogens	
(CRELs)	US ATSDR Minimal Risk Levels for Hazardous Substan	ices (MRLS)
US - California Permissible Exposure Limits for Chemical Contaminants	US Clean Air Act - Hazardous Air Pollutants	
US - Hawaii Air Contaminant Limits	US CWA (Clean Water Act) - Priority Pollutants	
US - Idaho - Limits for Air Contaminants	US CWA (Clean Water Act) - Toxic Pollutants US EPCRA Section 313 Chemical List	
US - Massachusetts - Right To Know Listed Chemicals	US National Toxicology Program (NTP) 14th Report Part	t A Known to be Human Carologgans
US - Minnesota Permissible Exposure Limits (PELs)	US NIOSH Recommended Exposure Limits (RELs)	r A Known to be Human Carcinogens
US - New Jersey Right to Know - Special Health Hazard Substance List (SHHSL):	US OSHA Permissible Exposure Levels (PELs) - Table 2	71
Carcinogens	US Toxic Substances Control Act (TSCA) - Chemical Su	
US - Pennsylvania - Hazardous Substance List	00 Toxic Substances Control Act (TOCA) - Chemical Su	issuance inventory
US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants		
US - Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants		
US - Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air		
Contaminants		
SILVER(7440-22-4) IS FOUND ON THE FOLLOWING REGULATORY LISTS		
US - Alaska Limits for Air Contaminants	US - Vermont Permissible Exposure Limits Table Z-1-A	Transitional Limits for Air
US - California Permissible Exposure Limits for Chemical Contaminants	Contaminants	
US - Hawaii Air Contaminant Limits	US - Washington Permissible exposure limits of air conta	
US - Idaho - Limits for Air Contaminants	US - Wyoming Toxic and Hazardous Substances Table Z	21 Limits for Air Contaminants
US - Massachusetts - Right To Know Listed Chemicals	US ACGIH Threshold Limit Values (TLV)	
US - Michigan Exposure Limits for Air Contaminants	US CWA (Clean Water Act) - Priority Pollutants	
US - Minnesota Permissible Exposure Limits (PELs)	US CWA (Clean Water Act) - Toxic Pollutants	
US - Oregon Permissible Exposure Limits (Z-1)	US EPA Carcinogens Listing	

- US Pennsylvania Hazardous Substance List
- US Rhode Island Hazardous Substance List
- US Tennessee Occupational Exposure Limits Limits For Air Contaminants
- US Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants

CADMIUM(7440-43-9) IS FOUND ON THE FOLLOWING REGULATORY LISTS

- US EPCRA Section 313 Chemical List
- US NIOSH Recommended Exposure Limits (RELs)
- US OSHA Permissible Exposure Levels (PELs) Table Z1
- US Toxic Substances Control Act (TSCA) Chemical Substance Inventory

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Monographs

US - Alaska Limits for Air Contaminants

US - California - Proposition 65 - Priority List for the Development of MADLs for Chemicals Causing Reproductive Toxicity

- US California OEHHA/ARB Chronic Reference Exposure Levels and Target Organs (CRELs)
- US California Permissible Exposure Limits for Chemical Contaminants
- US California Proposition 65 Carcinogens
- US California Proposition 65 Maximum Allowable Dose Levels (MADLs) for Chemicals Causing Reproductive Toxicity
- US California Proposition 65 No Significant Risk Levels (NSRLs) for Carcinogens
- US California Proposition 65 Reproductive Toxicity
- US Hawaii Air Contaminant Limits
- US Idaho Acceptable Maximum Peak Concentrations
- US Idaho Limits for Air Contaminants
- US Massachusetts Right To Know Listed Chemicals
- US Michigan Exposure Limits for Air Contaminants
- US Minnesota Permissible Exposure Limits (PELs)
- US New Jersey Right to Know Special Health Hazard Substance List (SHHSL): Carcinogens
- US Oregon Permissible Exposure Limits (Z-1)
- US Oregon Permissible Exposure Limits (Z-2)
- US Pennsylvania Hazardous Substance List
- US Rhode Island Hazardous Substance List
- US Tennessee Occupational Exposure Limits Limits For Air Contaminants

COBALT(7440-48-4) IS FOUND ON THE FOLLOWING REGULATORY LISTS

- International Agency for Research on Cancer (IARC) Agents Classified by the IARC Monographs
- US Alaska Limits for Air Contaminants
- US California Permissible Exposure Limits for Chemical Contaminants
- US California Proposition 65 Carcinogens
- US Hawaii Air Contaminant Limits
- US Idaho Limits for Air Contaminants
- US Massachusetts Right To Know Listed Chemicals
- US Michigan Exposure Limits for Air Contaminants
- US Minnesota Permissible Exposure Limits (PELs)
- US New Jersey Right to Know Special Health Hazard Substance List (SHHSL):
- Carcinogens
- US Oregon Permissible Exposure Limits (Z-1)
- US Pennsylvania Hazardous Substance List
- US Rhode Island Hazardous Substance List
- US Tennessee Occupational Exposure Limits Limits For Air Contaminants
- US Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants

CHROMIUM(7440-47-3) IS FOUND ON THE FOLLOWING REGULATORY LISTS

International Agency for Research on Cancer (IARC) - Agents Classified by the IARC Monographs

- US Alaska Limits for Air Contaminants
- US California Permissible Exposure Limits for Chemical Contaminants
- US Hawaii Air Contaminant Limits
- US Idaho Limits for Air Contaminants
- US Massachusetts Right To Know Listed Chemicals
- US Michigan Exposure Limits for Air Contaminants
- US Oregon Permissible Exposure Limits (Z-1)
- US Pennsylvania Hazardous Substance List
- US Rhode Island Hazardous Substance List
- US Tennessee Occupational Exposure Limits Limits For Air Contaminants
- US Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants

MANGANESE(II) ACETATE(638-38-0) IS FOUND ON THE FOLLOWING REGULATORY LISTS

- US Alaska Limits for Air Contaminants
- US California OEHHA/ARB Chronic Reference Exposure Levels and Target Organs (CRELs)
- US California Permissible Exposure Limits for Chemical Contaminants
- US Hawaii Air Contaminant Limits
- US Idaho Limits for Air Contaminants
- US Michigan Exposure Limits for Air Contaminants
- US Minnesota Permissible Exposure Limits (PELs)
- US Oregon Permissible Exposure Limits (Z-1)
- US Tennessee Occupational Exposure Limits Limits For Air Contaminants

NICKEL(7440-02-0) IS FOUND ON THE FOLLOWING REGULATORY LISTS

- US Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants US - Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air Contaminants US - Washington Permissible exposure limits of air contaminants US - Washington Toxic air pollutants and their ASIL, SQER and de minimis emission values US - Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants US - Wyoming Toxic and Hazardous Substances Table Z-2 Acceptable ceiling concentration, Acceptable maximum peak above the acceptable ceiling concentration for an 8-hr shift US ACGIH Threshold Limit Values (TLV) US ACGIH Threshold Limit Values (TLV) - Carcinogens US ATSDR Minimal Risk Levels for Hazardous Substances (MRLs) US Clean Air Act - Hazardous Air Pollutants US CWA (Clean Water Act) - Priority Pollutants US CWA (Clean Water Act) - Toxic Pollutants US EPA Carcinogens Listing US EPCRA Section 313 Chemical List US National Toxicology Program (NTP) 14th Report Part A Known to be Human Carcinogens US NIOSH Recommended Exposure Limits (RELs) US OSHA Carcinogens Listing US OSHA Permissible Exposure Levels (PELs) - Table Z1 US OSHA Permissible Exposure Levels (PELs) - Table Z2
- US Toxic Substances Control Act (TSCA) Chemical Substance Inventory

US - Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air

- Contaminants US - Washington Permissible exposure limits of air contaminants US - Washington Toxic air pollutants and their ASIL, SQER and de minimis emission values US - Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants US ACGIH Threshold Limit Values (TLV) US ACGIH Threshold Limit Values (TLV) - Carcinogens US ATSDR Minimal Risk Levels for Hazardous Substances (MRLs) US Clean Air Act - Hazardous Air Pollutants US EPCRA Section 313 Chemical List US National Toxicology Program (NTP) 14th Report Part B. US NIOSH Recommended Exposure Limits (RELs) US OSHA Permissible Exposure Levels (PELs) - Table Z1 US Priority List for the Development of Proposition 65 Safe Harbor Levels - No Significant Risk Levels (NSRLs) for Carcinogens and Maximum Allowable Dose Levels (MADLs) for Chemicals Causing Reproductive Toxicity US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory
- US Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air Contaminants US - Washington Permissible exposure limits of air contaminants US - Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants US ACGIH Threshold Limit Values (TLV) US ACGIH Threshold Limit Values (TLV) - Carcinogens US Clean Air Act - Hazardous Air Pollutants US CWA (Clean Water Act) - Priority Pollutants US CWA (Clean Water Act) - Priority Pollutants US EPCRA Section 313 Chemical List US NIOSH Recommended Exposure Limits (RELs) US OSHA Permissible Exposure Levels (PELs) - Table Z1 US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory
- US Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants

 US Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air Contaminants

 US Washington Permissible exposure limits of air contaminants

 US Washington Toxic air pollutants and their ASIL, SQER and de minimis emission values

 US Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants

 US Clean Air Act Hazardous Air Pollutants

 US EPCRA Section 313 Chemical List

 US OSHA Permissible Exposure Levels (PELs) Table Z1
 - US TO THE ETHISSIDE EXPOSURE LEVELS (FELS) TADIE 21
 - US Toxic Substances Control Act (TSCA) Chemical Substance Inventory

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US - Alaska Limits for Air Contaminants	US - Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air
US - California OEHHA/ARB - Acute Reference Exposure Levels and Target Organs (RELs)	Contaminants
US - California OEHHA/ARB - Chronic Reference Exposure Levels and Target Organs	US - Washington Permissible exposure limits of air contaminants
(CRELs)	US - Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants
US - California Permissible Exposure Limits for Chemical Contaminants	US ACGIH Threshold Limit Values (TLV)
US - California Proposition 65 - Carcinogens	US ACGIH Threshold Limit Values (TLV) - Carcinogens
US - Hawaii Air Contaminant Limits	US ATSDR Minimal Risk Levels for Hazardous Substances (MRLs)
US - Idaho - Limits for Air Contaminants	US Clean Air Act - Hazardous Air Pollutants
US - Massachusetts - Right To Know Listed Chemicals	US CWA (Clean Water Act) - Priority Pollutants
US - Michigan Exposure Limits for Air Contaminants	US CWA (Clean Water Act) - Toxic Pollutants
US - Minnesota Permissible Exposure Limits (PELs)	US EPCRA Section 313 Chemical List
US - New Jersey Right to Know - Special Health Hazard Substance List (SHHSL):	US National Toxicology Program (NTP) 14th Report Part B.
Carcinogens	US NIOSH Recommended Exposure Limits (RELs)
US - Oregon Permissible Exposure Limits (Z-1)	US OSHA Permissible Exposure Levels (PELs) - Table Z1
US - Pennsylvania - Hazardous Substance List	US Priority List for the Development of Proposition 65 Safe Harbor Levels - No Significant Ris
US - Rhode Island Hazardous Substance List	Levels (NSRLs) for Carcinogens and Maximum Allowable Dose Levels (MADLs) for
US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants	Chemicals Causing Reproductive Toxicity
US - Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants	US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory
AMMONIUM METAVANADATE(7803-55-6) IS FOUND ON THE FOLLOWING REGULAT	DRY LISTS
US - California OEHHA/ARB - Acute Reference Exposure Levels and Target Organs (RELs)	US EPCRA Section 313 Chemical List
US - Massachusetts - Right To Know Listed Chemicals	US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory
US - Pennsylvania - Hazardous Substance List	00 Toxic Substances Control Act (TOCA) - Chemical Substance Inventory
ZINC(7440-66-6) IS FOUND ON THE FOLLOWING REGULATORY LISTS	
International Agency for Research on Cancer (IARC) - Agents Classified by the IARC	US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants
Monographs	US - Washington Permissible exposure limits of air contaminants
US - California OEHHA/ARB - Chronic Reference Exposure Levels and Target Organs	US - Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants
(CRELs)	US ATSDR Minimal Risk Levels for Hazardous Substances (MRLs)
US - California Permissible Exposure Limits for Chemical Contaminants	US CWA (Clean Water Act) - Priority Pollutants
US - Hawaii Air Contaminant Limits	US CWA (Clean Water Act) - Toxic Pollutants
US - Massachusetts - Right To Know Listed Chemicals	US EPA Carcinogens Listing
US - Michigan Exposure Limits for Air Contaminants	US EPCRA Section 313 Chemical List
US - Oregon Permissible Exposure Limits (Z-1)	US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory
US - Pennsylvania - Hazardous Substance List	
US - Rhode Island Hazardous Substance List	
SELENIUM(7782-49-2) IS FOUND ON THE FOLLOWING REGULATORY LISTS	
International Agency for Research on Cancer (IARC) - Agents Classified by the IARC	US - Washington Permissible exposure limits of air contaminants
Monographs	US - Washington Toxic air pollutants and their ASIL, SQER and de minimis emission values
US - Alaska Limits for Air Contaminants	US ACGIH Threshold Limit Values (TLV)
US - California OEHHA/ARB - Acute Reference Exposure Levels and Target Organs (RELs)	US ATSDR Minimal Risk Levels for Hazardous Substances (MRLs)
US - California OEHHA/ARB - Chronic Reference Exposure Levels and Target Organs	US Clean Air Act - Hazardous Air Pollutants
(CRELs)	US CWA (Clean Water Act) - Priority Pollutants
US - Hawaii Air Contaminant Limits	US CWA (Clean Water Act) - Toxic Pollutants
US - Idaho - Limits for Air Contaminants	US EPA Carcinogens Listing
US - Massachusetts - Right To Know Listed Chemicals	US EPCRA Section 313 Chemical List
US - Minnesota Permissible Exposure Limits (PELs)	
US - Pennsylvania - Hazardous Substance List	US NIOSH Recommended Exposure Limits (RELs)
US - Rhode Island Hazardous Substance List	US OSHA Permissible Exposure Levels (PELs) - Table Z1
US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants	US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory
US - Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants	
US - Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air	
Contaminants	
COPPER(7440-50-8) IS FOUND ON THE FOLLOWING REGULATORY LISTS	
US - Alaska Limits for Air Contaminants	US - Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air
US - California OEHHA/ARB - Acute Reference Exposure Levels and Target Organs (RELs)	Contaminants
US - California Permissible Exposure Limits for Chemical Contaminants	US - Washington Permissible exposure limits of air contaminants
US - Hawaii Air Contaminant Limits	US - Washington Toxic air pollutants and their ASIL, SQER and de minimis emission values
US - Idaho - Limits for Air Contaminants	US - Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants
US - Massachusetts - Right To Know Listed Chemicals	US ACGIH Threshold Limit Values (TLV)
LIC Michigan Expansion Limits for Air Contaminanta	US ATSDR Minimal Risk Levels for Hazardous Substances (MRLs)
US - Michigan Exposure Limits for Air Contaminants	US CWA (Clean Water Act) - Priority Pollutants
	US CWA (Clean Water Act) - Toxic Pollutants
US - Minnesota Permissible Exposure Limits (PELs)	
US - Minnesota Permissible Exposure Limits (PELs) US - Oregon Permissible Exposure Limits (Z-1)	US EPA Carcinogens Listing
US - Oregon Permissible Exposure Limits (Z-1) US - Pennsylvania - Hazardous Substance List	
US - Minnesota Permissible Exposure Limits (PELs) US - Oregon Permissible Exposure Limits (Z-1) US - Pennsylvania - Hazardous Substance List US - Rhode Island Hazardous Substance List	US EPA Carcinogens Listing
US - Minnesota Permissible Exposure Limits (PELs) US - Oregon Permissible Exposure Limits (Z-1) US - Pennsylvania - Hazardous Substance List	US EPA Carcinogens Listing US EPCRA Section 313 Chemical List US NIOSH Recommended Exposure Limits (RELs)

Federal Regulations

Superfund Amendments and Reauthorization Act of 1986 (SARA)

SECTION 311/312 HAZARD CATEGORIES

 Immediate (acute) health hazard
 Yes

 Delayed (chronic) health hazard
 No

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Fire hazard No Pressure hazard No Reactivity hazard No

Name	Reportable Quantity in Pounds (lb)	Reportable Quantity in kg
Sodium	10	4.54
Sulfuric acid	1000	454
Ammonium acetate	5000	2270
Ammonium chloride	5000	2270
Nitric acid	1000	454
Hydrofluoric acid	100	45.4
Arsenic	1	0.454
Silver	1000	454
Cadmium	10	4.54
Chromium	5000	2270
Nickel	100	45.4
Ammonium vanadate	1000	454
Zinc	1000	454
Selenium	100	45.4
Copper	5000	2270

State Regulations

US. CALIFORNIA PROPOSITION 65

WARNING: This product contains a chemical known to the State of California to cause cancer and birth defects or other reproductive harm

US - CALIFORNIA PREPOSITION 65 - CARCINOGENS & REPRODUCTIVE TOXICITY (CRT): LISTED SUBSTANCE

Cadmium and cadmium compounds: Cadmium, Cobalt metal powder, Nickel (Metallic) Listed

National Inventory	Status
Australia - AICS	Y
Canada - DSL	Y
Canada - NDSL	N (sodium; calcium; magnesium nitrate; ammonium chloride; zinc; potassium; ammonium metavanadate; titanium; copper; ammonium phosphate, monobasic; water; selenium; aluminium; molybdenum; arsenic; cobalt; nickel; sulfuric acid; iron; ammonium acetate; chromium; hydrofluoric acid; silver; cadmium; manganese(II) acetate; nitric acid)
China - IECSC	Y
Europe - EINEC / ELINCS / NLP	Y
Japan - ENCS	N (sodium; calcium; magnesium nitrate; zinc; potassium; titanium; copper; ammonium phosphate, monobasic; water; selenium; aluminium; molybdenum; arsenic; cobalt; nickel; iron; ammonium acetate; chromium; hydrofluoric acid; silver; cadmium; manganese(II) acetate; nitric acid)
Korea - KECI	Y
New Zealand - NZIoC	Y
Philippines - PICCS	N (manganese(II) acetate)
USA - TSCA	Y
Legend:	Y = All ingredients are on the inventory $N = Not$ determined or one or more ingredients are not on the inventory and are not exempt from listing(see specific ingredients in brackets)

SECTION 16 OTHER INFORMATION

Other information

Ingredients with multiple cas numbers

Name	CAS No
aluminium	7429-90-5, 91728-14-2
calcium	7440-70-2, 8047-59-4
magnesium nitrate	13446-18-9, 10377-60-3, 10213-15-7
ammonium chloride	12125-02-9, 152128-19-3
hydrofluoric acid	7664-39-3, 790596-14-4
copper	7440-50-8, 133353-46-5, 133353-47-6, 195161-80-9, 65555-90-0, 72514-83-1

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references.

The SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

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end of SDS

 $\mathsf{PC-TWA}:$ Permissible Concentration-Time Weighted Average PC-STEL: Permissible Concentration-Short Term Exposure Limit IARC: International Agency for Research on Cancer ACGIH: American Conference of Governmental Industrial Hygienists STEL: Short Term Exposure Limit TEEL: Temporary Emergency Exposure Limit。 IDLH: Immediately Dangerous to Life or Health Concentrations OSF: Odour Safety Factor NOAEL :No Observed Adverse Effect Level LOAEL: Lowest Observed Adverse Effect Level TLV: Threshold Limit Value LOD: Limit Of Detection OTV: Odour Threshold Value BCF: BioConcentration Factors

BEI: Biological Exposure Index

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